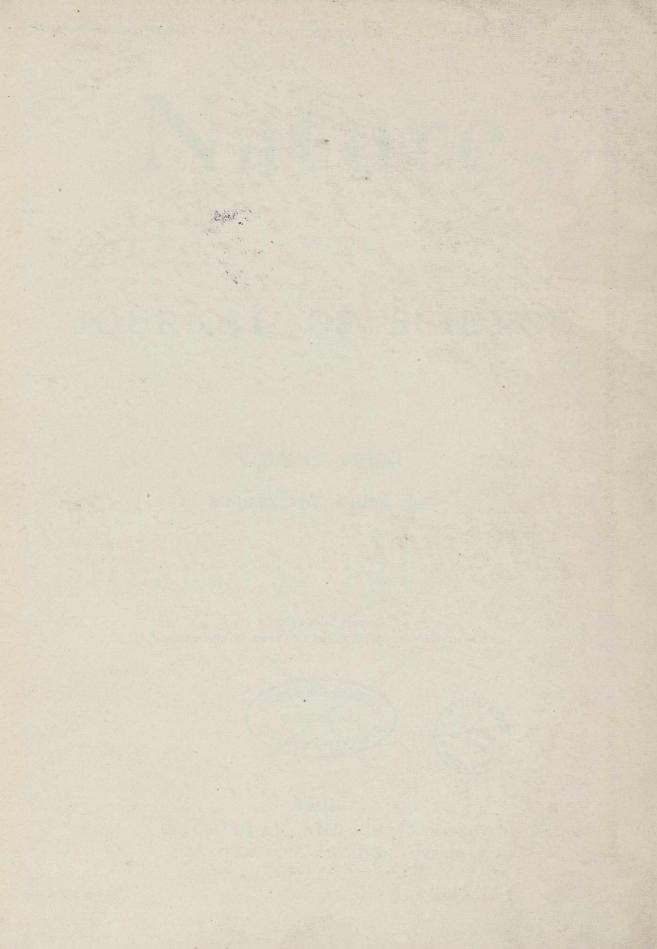


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"To the solid ground
Of Nature trusts the mind that builds for aye."—WORDSWORTH.





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INDEX

NAME INDEX

Abdouloff (I.), [D. Ivanoff and], Velocity of Disengagement of Hydrocarbons produced by the action of Indene on Fatty Organomagnesium Derivatives, 483

Abe (N.), Age and Growth of Limpets, 30

Abel (E.), and H. Schmid, Flow Kinetics: Model of Photolysis, 251

Abruzzi (Duke of the), [death], 427, [obituary article], 717 Achalme (P.), Chemical Nature of the Neutron, 559 Achard (C.), and A. Boutaric, Suspensions starting with

Proteins separated from Serum, 178

Ackermann (A. S. E.), Illuminated Fountains, 396 Ackermann (Dr. W.), Detection of Traces of Carbon Monoxide in Air, 441

Acworth (Capt. B.), Economic Significance of Coal, 478 Adams (L. È.), Moles Storing Worms, 466 Addington (J. N.), Gaseous Tubes for Lighting, 235

Adelsberger [Scheibe and], Oscillating Quartz Crystal as an Accurate Clock, 281

Agello (Warrant Officer F.), New Air Speed Record, 580 Agostinelli (C.), Concurrent Directions in a Variety V_n ,

Aitken (Dr. A. C.), awarded the Makdougall-Brisbane prize of the Royal Society of Edinburgh, 359

Aiyar (P. R. Chidambara), Earth's Influence on Sunspots,

Alcock (Lieut.-Col. A. W.), [death], 462; [obituary article], 273

Alessandri (C.), Apparent Velocity of the Surface Propagation of Earthquakes in relation to the Hypocentral Depth, 923

Alexander (Dr. F.), The Medical Value of Psychoanalysis (Review), 532

Alexander (Prof. T.), [death], 266 Alfvén (H.), Origin of Cosmic Radiation, 619

Algar (J.), Isabella M. McCarthy and Eveline M. Dick. Synthesis of Diflavones, 666

Allan (D.), Candles and Candlemaking, 269

Allcock (H. J.), and J. R. Jones, The Nonogram: the Theory and Practical Construction of Computation Charts (Review), 785

Allen (Dr. E. J.), Prof. Johannes Schmidt, 424

Allen (W. E. D.), A History of the Georgian People: from the Beginning down to the Russian Conquest in the Nineteenth Century (Review), 308

Allibone (T. E.), F. S. Edwards, and D. B. McKenzie, A New Impulse Generator for Three Million Volts, 129 Allix (A.), Progressive Darkening of the Lyons Atmo-

sphere, 179 Allmand (Prof. A. J.), Photochemical Reaction of Hydrogen and Chlorine, 656; Simple Molecules and Elementary Processes, 173

Aloisi (P.), An Andalusite Twin, 851

Alpatov (Dr.), Egg Production in Vestigial- and Long-Winged Flies, 176

Alter [Baxter and], Lead from Cyrtolite, 881

Amati (A.), [G. Mezzadroli and], Action of certain Alkaloids on the Development of Aspergillus niger, 339

Ambler (H. R.), and T. C. Sutton, Detection of traces of Carbon Monoxide in Air, 766

Amiel (J.), Slow Combustion of Benzene, 774

Anderson (J.), appointed professor of surgery in St. Andrews University, 848

Anderson (R. M.), Methods of Collecting and Preserving Vertebrate Animals, 906

Anderson, Wilson Photographs of Cosmic Ray Phenomena,

Andrewes (late Sir Frederick W.), and Mrs. Ethel M. Christie, The Hæmolytic Streptococci: their Grouping by Agglutination, 593

Anniss (L. G.), Upper Devonian Rocks of the Chudleigh Area, South Devon, 482

Anstey (Dr. Vera), appointed Cassel lecturer in commerce at the London School of Economics, 480

Antoniadi (Dr. E. M.). The Minor Details of the Planet Mars, 802

Antoniani (C.), the Phytosterol Group (2), 483; and F. Zanelli, Cholesterol of the Human Brain, 71

Appleton (Prof. E. V.), Fine Structure of the Ionosphere, 872; Thermionic Vacuum Tubes and their Applications (Review), 9; and R. Naismith, Kennelly-Heaviside Layer, 808; Weekly Measurements of Upper Atmospheric Ionisation, 522

Appleyard (R.), Charles Parsons: his Life and Work

(Review), 891

Armstrong (Dr. E. F.), Enzymes: a Discovery and its Consequences, 535; Alcohol through the Ages, 810; Chemistry and our Idiosyncrasies (*Review*), 314; and

others, Utilisation of Coal, 479

Armstrong (Prof. H. E.), Agriculture and Milk Supply, 605; An Indian Sage (Review), 672; Caution in Christening, 330; Chemistry and the Art of Living (Sir Jesse Boot Foundation lecture), 795; elected president of the Lancaster Frankland Chemical Society, 197; Huxley and Scientific Education. 684; In Chase of Truth of Alcohol, 53; the Work of Sir Edward Frankland, 196

Arakatsu (B.), Anomalous Absorption of γ-Rays, 696 Arend [Delporte and], A Remarkable Short-Lived Nova,

Arnaud (F. W. F.), elected president of the Society of Public Analysts, 359

Arrow (G. J.), and others. Edited by W. P. Pycraft, The Standard Natural History: from Ameeba to Man (Review), 152

Ashworth (Prof. J. H.), Lieut.-Col. J. Stephenson, 193 Aston (Dr. F. W.), Physical Atomic Weights, 172 Atkins (Dr. W. R. G.), [Dr. H. H. Poole and], Photocells

of the Dry Rectifier Type for the Measurement of Daylight, 850; Reversal of Current in Rectifier Photo-Cells, 547; Reversal of the Current from a Cuprous Oxide Photo-Cell in Red Light, 133

Atkinson (Ll. B.), elected an honorary member of the Institution of Electrical Engineers, 128

Atmanathan (S.), Katabatic Winds, 696

Atta (Van), [Van de Graaf, Compton and], Electrostatic Production of High Voltages, 475

Audubert (R.), Differentiation of the Electronic Effects and the Photoelectric Effects in Photovoltaic Elements, 482

Auger (P.), and G. M. Herzen, Emission of Neutrons by Aluminium under the Action of the α-Particles, 523 Aung (Maung Htin), Burmese Spirit-World, 767

Austen (Major E. E.), and A. W. McKenny Hughes, Clothes Moths and House Moths, 55

Awbery (J. H.), and Dr. E. Griffiths, Heats of Combustion of Carbon Monoxide in Oxygen and of Nitrous Oxide in Carbon Monoxide at Constant Pressure, 446

Aynsley (E. E.), Dr. T. G. Pearson and Dr. P. L. Robinson, New Experimental Evidence in the Sulphur-Hydrogen Reaction, 471

Babington (Dr. William), Centenary of the Death of, the Work of, 613

Bacharach (A. L.), Research on Vitamins (Review), 857; Vitamin C and Ascorbic Acid, 364

Backlund (Dr. N. O.), Dewaxing and Acid Refining Mineral Oils, 31

Badcock (W. C.), and Dr. E. J. Holmyard, Electricity and Magnetism for Beginners (Review), 347

Badouin (M.), An Intentional Fabrication of an Engraving of a Horse's Hoof on Granite, by means of a Stone Tool, 375

Bailey (Alice A.), Ultra-Violet Light and Fungi, 404; [G. B. Ramsey and], Tomato Late-Blight Rot, 440

Bailey (C. R.), and A. B. D. Cassie, Form and Vibrational Frequencies of the Nitrogen Dioxide Molecule, 910; Infra-Red Absorption Spectrum of Nitrogen Dioxide, 239; Infra-Red Region of the Spectrum (8), 286

Bailey (Prof. E. B.), Help from America in Reading Scottish Tectonics, 522; and Dr. J. Weir, Submarine Faulting in Kimmeridgian Times, 244

Bailey (Dr. K. C.), [T. N. Richardson and], Supersaturation of Liquids with Gases, 762

Bailey (V. A.), Quantitative Theory of Interaction between Different Species of Animals, 524

Bain (A. D. N.), Younger Intrusive Rocks of the Kudaru Hills, Nigeria, 214

Bainbridge-Bell (L.), [R. A. Watson Watt and], Recording Wireless Echoes at the Transmitting Station, 657

Baker (E. S. Stuart), The Nidification of Birds of the Indian Empire, Vol. 1 (*Review*), 384 Baker (Prof. H. B.), Photochemical Reaction of Hydrogen

and Chlorine, 27

Baker (Dr. J. F.), appointed professor of civil engineering in Bristol University, 848

Balbi (C. M. R.), Talking Pictures and Acoustics (Review), 188

Baldock (A. L.), [W. E. Cohen, A. G. Charles and], Chemical Composition of the Woods of the Ironbark Group of Eucalypts, 333

Baldwin (I. L.), [E. B. Fred, Elizabeth McCoy and], Root Nodule Bacteria and Leguminous Plants, 99

Ball (Sir Charles Arthur Kinahan), appointed Regius professor of surgery in Dublin University, 772 Ball (Dr. W. C.), [death], 866

Ballinger (Sir John), [death], 50

Balls (Dr. W. L.). Capacitance Hygroscopy, 329 Bamberger (Prof. E.), [death], 195

Bancroft (W. D.), and J. E. Rutzler, Jr., Agglomeration Theory of Sleep, 739
Banerjee (S.), [K. S. Krishnan and], Orientations of Mole-

cules in the p-Benzoquinone Crystal, 653

Banerji (Prof. A. C.), Momentum and Energy in the Special Theory of Relativity, 517 Banerji (Prof. S. K.), and K. N. Sohoni, Hydraulic

Seismographs, 547

Banks (C. A.), awarded the gold medal of Consolidated Gold Fields of South Africa, Ltd., 432

Banks (W. H.), [Dr. C. W. Davies and], Dissociation of Acetic Acid in Water, 328

Bannerman (W. B.), [death], 866 Bannister (F. A.), with chemical analyses by M. H. Hey, Identity of Mottramite and Psittacinite with Cupriferous Descloizite (Cuprodescloizite), 250

Barannikov (M. A.), Gypsy Art in Russia, 271

Barber (Dr. C. A.), [death], 298; [obituary article], 389 Barbey (Dr. A.), Les Insects Forestiers du Parc National Suisse, 234

Barbier (D.), [A. de La Baume Pluvinel and], Total Eclipse of the Sun on August 31, 1932, 922

Bardach (M.), [J. Basset, Mme. E. Wollman, M. A. Macheboeuf and], Biological Effects of Pressures, 774

van Baren (Prof. J.), [death], 231; [obituary article], 390 Baring (Wing-Comdr. the Hon. Maurice), elected a member of the Athenæum Club, 510

Barker (A. H.), Electricity, Gas and other Fuels as

Heating Agents, 32 Barker (Dr. S. G.), Fibres from the Coat of a Blackface

Lamb, 799 Barkla (Prof. C. G.), Properties of X-Radiation, 166 Barlot (J.), A New Method of Analysis of Bituminous

Schists, 774

Barnes (Col. F. P.), Mechanical Transport in India, 578 Baroni (A.), Diselenomesoxanilides and Oxyselenanilides, Lithium Alloys (1), 71; Protoselenosulphochloride, 483

Barrell [Rolt and], Sources of Light for Interferometer Work, 244; [J. E. Sears and], A New Apparatus for Determining the Relationship between Wave-Lengths of Light and the Fundamental Standards of Length, 192

Barringer (D. M.), Meteor Crater, Arizona, 579.

Barritt (J.), Methionine in Wool, 689 Barrow (W.), elected pro-chancellor of Birmingham

University, 336

Bartlett (Prof. F. C.), Remembering: a Study in Experimental and Social Psychology (Review), 309

Bartlett (R. W.), Co-operation in Marketing Dairy Products. 2 Parts (Review), 188

Barton (A. C.), Burial of a Bari Rain-maker, 29

Bary (P.), Causes of the Thixotropy of Certain Salts, 338

Basedow (Dr. H.), [obituary article], 865

Basset (J.), and M. A. Machebœuf, Biological Effects of Ultra-Pressures, 251; Mme. E. Wollman, M. A. Macheboeuf and M. Bardach, Biological Effects of Ultra-Pressures, 774
Bateman (Prof. H.), Variation Problems for a Symmetrical

Region, 472

Bates (Dr. L. F.), and B. J. Lloyd Evans, A Compact Electromagnet for General Purposes, 558

Bates (G. H.), Eradication of the Stinging Nettle, 208 Bather (Dr. F. A.), A Peculiar Visual Experience, 62; awarded the Mary Clark Thompson medal of the National Academy of Sciences, Washington, 304; re-elected president of the Palæontographical Society,

Baubiac (J.), [E. Crausse and], Modification of the Chronophotographic Method and some Applications, 482; Transitory Regimes, 559 Baudiš (Prof. J.), [obituary article], 830

Bauerman (Hilary), centenary of the birth of, 355

Baur (Dr. E.), elected a foreign member of the Linnean Society of London, 759

Bavink (B.), translated, with additional notes and bibliography, by H. S. Hatfield, The Anatomy of Modern Science: an Introduction to the Scientific Philosophy of To-day (Review), 707

Baxter and Alter, Lead from Cyrtolite, 881 Bayard-Duclaux (Mme. F.), Methods for putting in Circuit Specimens of Rock for Measuring their Electrical

Resistance, 887 Baylis (Dr. H. A.), Worms Parasitic in Cetacea, 551

Beach (late A. C. G.), Talbot's Bands, 702

Beadles (C. F.), [death], 86; [obituary article], 122

Beal (Dr.), Aluminium in Foodstuffs, 432 Bean (Prof. R. B.), The Races of Man: Differentiation

and Dispersal of Man (Review), 712

Beasley (H. G.), Hawaiian Feather Cape, 171 Beattie (Prof. J. M.), Prof. R. Donaldson, 122 Beatty (Dr. R. T.), Hearing in Man and Animals (Review),

Beauverie (J.), and Mlle. S. Monchal, Life of Green

Plants in a Confined Atmosphere, 36

Bechterev (Prof. V. N.), translated by Emma and W. Murphy, General Principles of Human Reflexology: an Introduction to the Objective Study of Personality (Review), 675

Beck (Conrad), elected president of the Royal Micro-

scopical Society, 128

Beckett (H. E.), 'Raw' Weather, 132

Beckley (V. A.), and F. McNaughtan, Distribution of Nitrates in the Soil and Root Development in Coffee,

Beckstrom (R. C.), Petroleum Production in Russia, 541 Beddington (R.), elected president of the Freshwater Biological Association, 796

Bedel (C.), Magnetic Susceptibility of Ferro-Silicons Rich

in Silicon, 375

Beebe (Dr. W.), Deep Sea Dives in the Bathysphere, etc., 776

Bell (A. H.), The Exponential and Hyperbolic Functions and their Applications (Review), 319 Bell (Prof. E. T.), The Queen of the Sciences (Review),

319

Bell (Dr. G. D. H.), appointed University demonstrator in agricultural biology in Cambridge University, 920 Bell (G. E.), [Dr. G. W. C. Kaye and], Measurement of X-Ray Tube Current and Voltage, 552

Bell (Mary), and S. E. Green, Radiometer Action and the

Pressure of Radiation, 374

Belling (Dr. J.), [death], 390; [obituary article], 575 Benedict (W. S.), [Dr. L. Harris, G. W. King and], Form and Vibrational Frequencies of the NO₂ Molecule, 621 Bennett (Dr. I.), The Practical Treatment of Diabetes

(Review), 9

Berger (P.), [G. Tiercy and], Aerological Soundings and the Wind Gradient in Switzerland, 143 Bergmann (Prof. M.), Amino-Acids, Proteins and Proteo-

lytic Enzymes, 662, 698 Berkeley (C.), Oxidase of the Crystalline Style, 94

Berliner (Dr. Arnold), Seventieth birthday of, 127 Bernacchi (Commdr. L. C.), A Very Gallant Gentleman (Review), 782

Bernal (J. D.), and D. Crowfoot, Crystal Structure of Vitamin $\vec{B_1}$ and of Adenine Hydrochloride, 911 Bernard (P. N.), and J. Guillerm, Transmissible Lysis

of the Cholera Vibrion, 887

Bernatzik (Dr. H. A.), mit einem beitrag von B. Struck, Äthiopen des Westens: Forschungsreisen in Portugiesisch-Guinea, Band 1 und 2 (Review), 599

Bertholf (Prof. Ll. M.), Eyes and Colour Change, 331 Berthon (R.), Heats of Moistening Silica Gel with Various Liquids, 106

Bertrand (G.), and L. Silberstein, Importance of Sulphates as Manure, 215

Besicovitch (A. S.), Almost Periodic Functions (Review), 384

Betz [Volmar and], Emetic Derivatives of Lactic Acid, 410

Beutel and Kutzelnigg, Fluorescence of Zinc Oxide, 917 Bhattacharya (A. K.), [Prof. N. R. Dhar, B. L. Mukerji and], Kinetics of the Iodine-Oxalate Reaction, 840

Bickham (S. H.), bequest by, to Cambridge University, 884

Bierry (H.), B. Gouzon and Mlle. C. Magnan, Application of the Iodometric Method to the Estimation of Sugar in the Blood, 667

van Biesbroeck (Prof.), Comets, 245

Biot (M.), Critical torsional Oscillations of a Rotating Accelerated Shaft, 631

Biquard (P.), An Optical Method for the Measurement of the Absorption of Ultra-Sound Waves by Liquids, 375

Birch (T. W.), and Dr. W. J. Dann, Estimation and Distribution of Ascorbic Acid (Vitamin C) and Glutathione in Animal Tissues, 469; Dr. L. J. Harris and S. N. Ray, Hexuronic (Ascorbic) Acid as the Antiscorbutic Factor, and its Chemical Determination, 273

Bjerknes (Prof. V. F. K.), elected a foreign member of the Royal Society, 798

Black (F. A.), The Calendar and its Reform (Review), 859

Blackett (P. M. S.), appointed professor of physics at Birkbeck College, 480; Cosmic Radiation (Symons Memorial Lecture), 429; [Dr. J. Chadwick, G. Occhialini and], New Evidence for the Positive Electron, 473; and G. Occhialini, Photography of the Tracks of Penetrating Radiation, 286; Wilson Photographs of Cosmic Ray Phenomena, 589

Blacktin (Dr. S. C.), Interaction between Soot Films and

Oil. 873

Blackwelder (Prof. E.), Age of Meteor Crater, 404

Blaringhem (L.), A Case of Atavism in a Strain of the Opium Poppy, 178 Blatchford (A. H.), Diffraction of X-Rays by Liquid

Sulphur, 813

Blau (Marietta), and Hertha Wambacher, Photographic Detection of Protons Liberated by Neutrons (2), 287 Bledisloe (Lord), Economic Value of Agricultural Science,

197

Bliss (E. W.), Tabulation of World Weather (5), 35

Boardman (E. M.), [E. T. Burton and], Effects of Solar Eclipse on Audio Frequency Atmospherics, 81 Bloch (A. M.), [D. M. Newitt and], Slow Combustion of

Ethane at High Pressures, 214

Boas (Prof. F.), Aims of Anthropological Research, 196 Bobrovnikoff (N. T.), Halley's Comet in its Apparition of 1909–1911, 282

Bocking (T. G.), Magnetic Data and Mine Surveying, 768 Boggio (T.), Curvature of Lines of Varieties, 71

Bohr (Prof. N.), Light and Life, 421; 457

Bolam (Dr. T. R.), The Donnan Equilibria: and their Application to Chemical, Physiological and Technical Processes (Review), 859

Boman (late Dr. E.), Archæological Researches in the

Argentine, 63

Bond (Dr. W. N.), Numerical Examples in Physics

(Review), 320
Bone (Prof. W. A.), awarded the medal of the Society of Chemical Industry; the work of, 793; Photographic Analysis of Explosion Flames, 300; Researches on Gaseous Combustion (Bakerian Lecture). 494; and others, Gaseous Combustion at High Pressures, 245

Bonner (J.), [K. V. Thimann and], Growth Hormone of Plants (2), 631

Bonot (A.), Modification of the Seric Proteins Isolated by the Acetone Method and Myxoprotein, 106 Boratynski (K.), and A. Nowakowski, Modifications of

Phosphoric Anhydride, 595

Bordier (H.), Production of Merget's Phenomenon by d'Arsonvalisation with Short Waves, 375

Boring (Dr. Alice M.), C. C. Liu and S. C. Chou, Handbook of North China Amphibia and Reptiles, 304

Borradaile (L. A.), and F. A. Potts. With chapters by Prof. L. E. S. Eastham and J. T. Saunders, The Invertebrata: a Manual for the use of Students (Review), 76

van den Bos (W. H.), Measures of Double Stars at the Union Observatory, Johannesburg, 476

Bossolasco (M.), Magnetic Measurements at Mogadiscio, 560

Bossuet (R.), Examination of Minerals for the Alkali Metals, 923; Photographic Sensibility of the Lines of the Alkaline Metals in the Oxyacetylene Flame, 482

Bostock (Dr. J.), The Neural Energy Constant: a Study of the Bases of Consciousness (Review), 80

Boswell (Prof. P. G. H.), [Dr. L. S. B. Leakey, Prof. H. Reck, A. T. Hopwood, Dr. J. D. Solomon and], The Oldoway Human Skeleton, 397

Bottoni (O.), Relationships between the Chemical Composition of the Soil and that of its Colloidal Contents, 560

Boulton (Prof. W. S.) the title of emeritus professor conferred upon, 336 Bouquet (Rev. A. C.), A Study of the Ordinary Argu-

ments for the Existence and Nature of God; Religious Experience; its Nature, Types and Validity; Phases of the Christian Church: a Short View of its History (Review), 677

Bourdon (B.), and others, A History of Psychology in Autobiography. Vol. 2 (Review), 321

Bourne (G.), Vitamin C in the Adrenal Gland, 874 Bourne (Prof. G. C.), [death], 390; [obituary article], 496 Boutaric (A.), [C. Achard and], Suspensions prepared starting with Proteins separated from Serum, 178; and Mlle. Madeleine Roy, Influence of Radio-Active Radiations on the Flocculation of Colloids, 738

Boutry (G. A.), [P. Fleury and], Exact Measurement of

Photographic Densities, 738
Bowden (Dr. F. P.), S. D. D. Morris and Dr. C. P. Snow, Absorption Spectrum of Vitamin A at Low Temperatures, 582

Bowden (Dr. P.), and T. Moore, Absorption Spectrum of the Vitamin E Fraction of Wheat-Germ Oil, 512

Bower (S. Morris), Summer Thunderstorms, 473 Bower (W. R.), and Prof. R. Satterly, Practical Physics. Third edition (Review), 491

Bowles (E. O.), [W. B. Marshall and] New Fossil Fresh Water Molluses from Ecuador, 589

Bowser (W. E.), and J. D. Newton, Effect of Weed-killers on the Soil, 880

Boycott (Prof. A. E.), Local Lists of Animals, 94 Boyle (C.), and J. J. Ryan, Grass Silage, 410

Boys (Dr. C. V.), A Destructive Lightning Flash, 765; Progressive Lightning: a New Stereoscope, 492 Bracewell (S.), [A. Brammall and], Garnet in the Dart-

moor Granite, 250

Brackett (F. S.), and E. S. Johnstone, Functions of Radiation in the Physiology of Plants, 331 Braddick (Dr. H. J. J.), and Prof. R. W. Ditchburn,

Absorption of Light in Cæsium Vapour, 132 Bradfield (A. E.), A. R. Penfold and J. L. Simonsen,

Essential Oil from the Wood of Eremophila Mitchelli, 667

Bradley (A. J.), and A. H. Jay, Quartz as a Standard for Accurate Lattice-Spacing Measurements, 813; and Phyllis Jones, X-Ray Investigation of the Copper-Aluminium Alloys, 589

Bradley (H.), Physics in the Boot and Shoe Industry, 756 Bragg (Sir William), Old Trades and New Knowledge (Review), 860; presentation of a Portrait of, to the Royal Institution, 647; The Crystals of the Living Body, 125; The World of Sound (Review), 860 Bragg (Prof. W. L.), Structure of Alloys, 749 Brailsford (Dr. J. F.), appointed radiological demonstrator

in anatomy in Birmingham University, 248 Brammall (A.), and S. Bracewell, Garnet in the Dart-

moor Granite, 250

Brandt (L.), Yangtse-Kiang Flood of 1931, 280

Brata (Luang), [Dr. C. F. Powell and], Positive Ion Emission from Oxide Catalysts, 168 Brazier (C. E.), and L. Génaux, The Earthquake of March

2, 1933, 596

Bricout (P.), A Magnetic Apparatus for the Determina-tion of Thicknesses, 595

Bridgman (Prof. P. W.), High Pressure Research, 280; The Physics of High Pressure (Review), 259 Brierley (Prof. W. B.), elected president of the Association

of Economic Biologists, 304

Briers (C. T.), Fauna of Hot Springs in North America,

Briggs (Prof. H.), [Prof. P. F. Kendall and], Formation of Rock Joints and the Cleat of Coal, 922

Briggs (Dr. L. J.), appointed director of the U. S. Bureau of Standards, 580

Brightman (R.), The Cost of a New World Order (Review),

Brillouin (M.), Linear Partial Differential Equations in the Plane, 410

Bristowe (W. S.), Insect-Eating in Siam, 234

British Drug Houses, Ltd., Guide to the B.P., 1932, 128 Broadhurst (E.), and J. D. Campbell, Geology and Petrology of the Mount Leinster District, North-East Victoria, 703 Brock (G. C.), A Method of Preparing a Filter for the

3130 Mercury Line, 410

Brocklehurst (Prof. R. J.), Neuro Humoral Mechanisms, 65

Brockway (L. O.), and L. Pauling, Determination of the Structures of the Hexafluorides of Sulphur, Selenium and Tellurium by the Electron Diffraction Method, 739

van den Broek (Prof. J. A.), Elastic Energy Theory (Review), 347

de Broglie (Prof. L.), Théorie de la quantification dans la nouvelle mécanique (Review), 639

Broniewski (W.), and S. Jaslan, Influence of Oxygen on the Properties of Copper, 338

Brooks (Dr. C. E. P.), Climate: a Handbook for Business Men, Students and Travellers. Second edition (Review), 9; and Theresa M. Hunt, Variations of Wind Direction in the British Isles since 1341, 814

Brooks (Harriet), (Mrs. Frank Pitcher), [obituary article], 865

Brooks (Matilda Moldenhauer), Penetration of m-Bromophenol Indophenol and of Guaiacol Indophenol into Valonia Ventricosa, 631

Brotherton (C. R.), Gift to Leeds University, 665 Brown (F. J.), Life-History of the Fowl Tapeworm, Davainea proglottina, 276

Brown (Prof. J. Macmillan), Education and Economic

Recovery, 538
Brown (R.), Nitrogen Fixation in the Genus Lolium, 169

Brown (Prof. W. Langdon), elected a member of the Athenæum Club, 510

Browne (W. M.), Photographing the Moon's Shadow on the Earth, 697

Browne (W. R.), Possible Correlation of certain Pre-Cambrian Granites of Australia and some Deductions Therefrom, 667

Bruhat (G.), and A. Guinier, Improvement of the Photoelectric Polarimeter, 630

Brunt (D.), Adiabatic Lapse-Rate for Dry and Saturated Air, 814

Bryce (J.), Economic Possibilities of Rice Grass, 368 Bryce (Prof. T. H.), [N. H. W. Maclaren and], Early Stages in the Development of Cavia, 922 Brydon (H. B.), Canals on Mars, 518

Buch (K.), Boric Acid in Sea Water and its Effect on the Carbon Dioxide Equilibrium, 688

Buck (Dr. P. H.), (Te Rangi Hiroa), Ethnology of Manihiki and Rakahanga, Cook Islands, 588

Buckley and Snyder, Spreading of Liquids on Solid Surfaces, 407

Bugnard (L.), P. Gley and A. Langevin, Recording and Measurement of the Blood Pressure, 375

Bullen (K. E.), [Dr. H. Jeffreys and], Corrections to the Times of the P Wave in Earthquakes, 97

Bunker (J. W. M.), [R. S. Harris and], Bacterial Detoxification, 244

Burch (F. P.), [obituary article], 390

Burden (Mrs. R. G.), Gift for Research into Mental Problems and Disorders, 159

Buret (R.), [C. Dufraisse and], The Dissociable Organic Oxides, 70

Burge (W. S.), Power Station Efficiency, 518

Burger (Dr. H. C.), [Dr. L. S. Ornstein, Dr. W. J. H. Moll and], Objektive Spektralphotometrie (Review), 824 Burgers (Dr. W. G.), Recrystallisation Power and Shear

Hardening in Aluminium Single Crystals, 326 Burgevin (H.), Nitrogenous Manuring of Legumes, 660

Burkitt (Prof. A. N. St. G. H.), elected president of the Linnean Society of N.S.W., 759

Burkitt (M. C.), Racial Distribution in Palæolithic Europe, 540; [E. J. Wayland and], Stone Age Culture in Uganda, 730

Burnell (Maisie), [H. E. Dadswell and], Identification of the Coloured Woods of the Genous Eucalyptus, 280

Burrows (H.), Prof. E. C. Dodds, and N. M. Kennaway, Some Effects observed in Mice under Continued Treatment with Œstrin, 801

Burt (C. E.), and May Danheim Burt, Reptiles and Amphi-

bians of the Pacific Islands, 767

Burt (Prof. C. L.), appointed Heath Clark lecturer of London University for 1933, 336

Burt (May Danheim), [C. E. Burt and], Reptiles and

Amphibians of the Pacific Islands, 767 Burton (E. T.), and E. M. Boardman, Effects of Solar Eclipse on Audio Frequency Atmospherics, 81

Burton (M.), Locomotion in Sponges, 356

Bushe-Fox (Dr. J. P.), Roman Richborough, 539 Butler (Dr. J. A. V.), and Miss L. C. Connell, Rôle of the Solvent in Electrolytic Dissociation, 800

Cadman (Sir John), awarded the gold medal of the Institution of Mining and Metallurgy, 432; awarded the Melchett medal of the Institute of Fuel, 581

Cairns [Common and], Colonisation of a Disused Millpond at Hillsborough, Co. Down, 702 Cairns (H.), and A. E. Muskett, Phytophthora megasperma

causing Pink Rot of the Potato, 277 Calder (J. W.), [E. W. Hullett and], Fluorescence in

Rye-grasses, 474 Calder (Mary G.), the Kidston Collection of Fossil Plant Slides (2), 922

Calderwood (Dr. W. L.), Salmon Hatching and Salmon Migrations: being the Buckland Lectures for 1930

(Review), 42

Calman (Dr. W. T.), A Dodecapodous Pycnogonid, 242 and Isabella Gordon, Dodecolopoda mawsoni, a New Species representing a New Genus of Pycnogonida, 774

Calvet (J.), [C. Matignon and], Ageing of the Aluminium-Beryllium Alloys after Tempering, 886

Cambier (R.), and L. Leroux, Estimation of Organic Nitrogen in the Presence of Nitrates by Kjeldahl's Method, 179

Campbell (Prof. A. N.), Constitution of Binary Alloys at Room Temperature, 438

Campbell (J. D.), [E. Broadhurst and], Geology and Petrology of the Mount Leinster District, North-East Victoria, 703 Campbell (J. S.), Nuclear Moments of the Gallium Iso-

topes 69 and 71, 204

Campbell (Sir Malcolm), Motor-car Speed Record, 303 Campbell (Dr. N. R.), Convention and Fact, 237; Measurement of Visual Sensations, 850

Campin (R. H.), Underground Lighting in Mines, 465 Candler (W. E.), awarded a Rayleigh prize of Cambridge University, 408

Cannata (C.), An Extension of Woo's Formula, 339 Cantacuzène (J.), and A. Tchekirian, Presence of Vanadium in Certain Tunicates, 35

Capatos (L.), [N. Perakis and], Constant Paramagnetism of Metallic Rhenium, 559

di Capua (A.), [A. Quilico and], Aspergillin, the Pigment of the Spores of Aspergillus niger (1), 851; (2), 924

Carey (Gladys), and Lilian Fraser, Embryology and Seedling Development of Aegiceras majus, Gaertn., 287

Carey (Commdr. W. M.), [death], 646 Carline (G. R.), [obituary article], 86

Carlini (Mlle.), [E. Carrière and], Decomposition of Thiosulphuric Acid in Dilute Solution at the Boiling Point, 286

Carlow (C. A.), elected president of the Mining Institute

of Scotland, 650 Carmichael (J.), 'Bull-Dog' Calf in African Cattle, 878 Carpanese (T.), Mineral Deposits of Monte Rosso di

Verra, 775; (2), 924 Carpenter (Dr. G. D. Hale), appointed Hope professor of zoology in Oxford University; the work of, 159

Carpenter (Sir Harold), Metals in the Service of Human Life and Industry, 733 Carr (Dr. F. H.), and W. Jewell, Characteristics of Highly

Active Vitamin A, 92

Carré (P.), and D. Libermann, Mechanism of the Reaction of Phosphorus Pentachloride on Neutral Alkyl Sulphites, 143; Reaction of Phosphorus Pentachloride on the Neutral Aryl Sulphites, 667 Carrière (E.), and Mlle. Carlini, Decomposition of Thio-

sulphuric Acid in Dilute Solution at the Boiling

Point, 286

Carroll (J.), The Potato Eelworm (Heterodera schachtii) in the Irish Free State, 850

Carroll (Prof. J. A.), Zones of Apparent Inhibition of Sunspots on the Solar Disc, 548

Carr-Saunders (Prof. A. M.), and P. A. Wilson, The Professions (Review), 863

Carter (C. S.), [obituary article], 829 Carty (Dr. J. J.), [obituary article], 85; awarded the John J. Carty medal and award of the U.S. National Academy of Sciences, 754

Cassie (A. B. D.), Structure of Triatomic Molecules, 438; [C. R. Bailey and], Form and Vibrational Frequencies of the Nitrogen Dioxide Molecule, 910; Infra-Red Absorption Spectrum of Nitrogen Dioxide, 239; Infra-Red Region of the Spectrum (8), 286

Castelfranchi (Prof. G.), translated by Dr. W. S. Stiles and Dr. J. W. T. Walsh, Recent Advances in Atomic

Physics, 2 vols. (Review), 319

Castellani (Sir Aldo), Climate and Acclimatization: some Notes and Observations (Review), 712

Casterat (N.), Discovery of Cave Paintings in the Pyrenees, 19

Castle (W. E.), and C. E. Keeler, Blood Group Inheritance in the Rabbit; Tests for Linkage between the Blood-Group Genes and Other Known Genes of the Rabbit,

Cavinato (A.), Thermal Expansion in Crystals and Haüy's Law, 71

Cawood (Dr. W.), and H. S. Patterson, A New Temperature Recorder, 332

Cayeux (L.), Constitution of the Devonian Phosphates of Tennessee, 666; Modes of Existence of Glauconite in Limestone, 143; Rôle of the Trilobites in the Genesis of the Palæozoic Deposits of Calcium Phosphate, 850

Chadwick (Dr. J.), The Neutron (Bakerian Lecture), 794; P. M. S. Blackett and G. Occhialini, New Evidence

for the Positive Electron, 473

Chamberlain (Prof. C. J.), Methods in Plant Histology. Fifth edition (Review), 44 Champetier (G.), Action of Solutions of Orthophosphoric

Acid on Ordinary Cellulose, 703 Champion (H. G.), Twisted Trees, 133

Chapin (J. P.), Bird Migration in Tropical Africa, 331 Chapman (F.), Origin of Tektites, 876; Palæozoic Fossils from Victoria, 703

Chapman (F. S.), and others, Northern Lights: the Official Account of the British Arctic Air Route Expedition, 1930–1931 (Review), 317

Chapman (F. W.), Cathode Ray Photography of Random Electrical Transients, 620

Chapman (P. F.), [Dr. M. Nierenstein and], the Authorship of the "Ordinall of Alchimy", 520

Chapman (Prof. S.), Atoms, Molecules and the Atmosphere, 271; elected president of the Royal Meteorological Society, 200

Chapman (V. J.), elected Frank Smart student in botany in Cambridge University, 408

Chappell (E.), Career of Samuel Pepys, 533

Charles (A. G.), [W. E. Cohen, A. L. Baldock and], Chemical Composition of the Woods of the Ironbark group of Eucalypts, 333

Charrier (G.), and E. Ghigi, Action of Alkylmagnesium Iodides on (1:9)-Benzanthrone-(10), 107; Action of Ammonia on Acenaphthenequinone, 107

Chartrou (J.-J.), Pétroles naturels et artificiels (Review), 455

Chatelet (M.), A Pyridine-Iodine Molecular Association,

Chatley (Prof. H.), Number 60 in Time Measurements, 914 Chaudron (G.), [A. Girard and], Dissociation of Cubic Ferric Oxide, 447; [J. Herenguel and], Sublimation of Magnesium in a Vacuum and Casting in an Atmosphere of Argon, 179

Cheesman (E. E.), Hybrid Bananas, 552

Cherubino (S.), Classification of Hyper-elliptic Surfaces from the Real Point of View, 339

Cherbuliez (E.), and F. Meyer, New Researches on Casein, 215

Chevenard (P.), awarded the medal of the Sociéte d'Encouragement pour l'Industrie Nationale; the work

Child (J. M.), Elements of Coordinate Geometry (Review), 603

China (W. E.), Cattle Blood-Sucking Habit of a Bug, 806 Chodat (F.), and M. Junquera, Reduction of Methylene Blue by an Endomyces at the Expense of its Endo-

cellular Hydrogen Donators, 815 Chodat (Prof. R.), presented with the Linnean Medal of the Linnean Society of London; the work of, 793

Choucroun (Mlle.), A Correct Arrangement for Electrophoresis, 630

Chowdhury (K. A.), Identification of Indian Sleeper

Woods, 30

Christie (Mrs. Ethel M.), [late Sir Frederick W. Andrewes and], The Hæmolytic Streptococci: their Grouping by Agglutination, 593

Chudoba (Dr. K.), Die Feldspäte und ihre praktische

Bestimmung ((Review), 384

Churcher (B. G.), and A. J. King, Scales of Loudness,

Churchward (J. G.), Geographical Distribution of Tilletia Species on Wheat in Australia in 1931, 631

de la Cierva (J.), awarded the Elliott Cresson medal of the Franklin Institute, 871

Cilento (Dr. R. W.), The Medical Sciences in Relation to

Depopulation, 519 Cimino (M.), The Einsteinian Correction of Time in a Planetary Motion, 107

Clark (Miss A. R.), [Mrs. E. W. Sexton and], Further Mutations in the Amphipod Gammarus chevreuxi, Sexton, 201

Clark (Dr. E. L.), [Dr. E. R. Clark and], Disposal of Debris by Blood-vessels, 880

Clark (Dr. E. R.), and Dr. E. L. Clark, Disposal of Debris by Blood-vessels, 880

Clark (F. H.), Hydrocephalus, a Hereditary Character in the House Mouse, 288

Clark (Dr. G. A.), appointed professor of physiology in Sheffield University, 736

Clarke (Dr. Ernest), [obituary article], 85

Claude (G.), Manufacture of Commercial Oxygen, 70

Claxton, (T. F.), Climate of Hong-Kong, 104 Clay (Prof. J.), Earth-Magnetic Effect and the Corpuscular Nature of (Cosmic) Ultra-Radiation, 136

Clay (Dr. R. S.), and T. H. Court, The History of the Microscope: Compiled from Original Instruments and Documents, up to the Introduction of the Achromatic Microscope (Review), 219

Clayton (Dr. G. C.), the Institute of Chemistry, 357 Clement (M. A.), Mexican and Eucadorian Copper and

Bronze Axes, 279 Clements (D. I.), N. H. Howes, and G. P. Wells, Is Plasticine Edible? 330

Clerk (Sir Dugald), Bequests for Science, 235

Close (Col. Sir Charles), The Map of England: or About England with an Ordnance Map (Review), 260

Clowes (A. J.), Influence of the Pacific on the Circulation in the South-West Atlantic Ocean, 189

Coales (J. F.), Errors Experienced on a Ship's Directionfinder, 846

Coates (Dr. W. H.), Special Posts in Industrial Management, 268

Cockeroft (Dr. J. D.), and Dr. E. T. S. Walton, Disintegration of Light Elements by Fast Protons, 23

Cockerell (Prof. T. D. A.), Sacred Sandstone of the Mayas, 656; Snails and Changes in Sea-level, 277

Cockerham (G.), Nitrogen Content of Normal and Leafroll Potatoes, 375

Cohen (W. E.), A. L. Baldock, and A. G. Charles, Chemical Composition of the Woods of the Ironbark group of Eucalypts, 333

Colani (Mlle. M.), Artificial Ridges of the Valves of Lamellibranchs (Northern Annam), 523

Colebrook (F. M.), High Selectivity Tone Corrected Receiving Circuits, 442

Collenette (C. L.), Effect of the Preying of Birds on Butterflies, 200

Collip (Prof. J. B.), Dr. H. Selye, and Prof. D. L. Thomson, Gonad-stimulating Hormones in Hypophysectomised Animals, 56

Combe (A. D.), and others, Geology of South-West Ankole and Adjacent Territories, 172; and W. C. Simmons, The Volcanic Area of Bufumbira. Part 1: The Geology of the Volcanic Area of Bufumbira, South-West Uganda; with Notes on the Petrology and Economic Geology (Review), 821

Comber (Prof. N. H.), An Introduction to the Scientific Study of the Soil. Second edition (Review), 316

Common and Cairns, Colonisation of a Disused Millpond

at Hillsborough, Co. Down, 702 Compton (Prof. A. H.), Cosmic Rays, 807; Latitude Variation of Cosmic Rays, 769; Nature of Cosmic Rays, 713

Compton (K. T.), Accommodation Coefficient of Gaseous Ions at Cathodes, 631

Compton (P.), The Genius of Louis Pasteur (Review), 40 Compton [Van de Graaf, Van Atta and], Electrostatic Production of High Voltages, 475

Connell (Miss L. C.), [Dr. J. A. V. Butler and], Rôle of the Solvent in Electrolytic Dissociation, 800

Conant (Prof. J. B.), elected president of Harvard University, 700; and Emma M. Dietz, Structural Formulæ of the Chlorophylls, 131

Conti (Prince Ginori), 'Red Coal', 679 Conzemius (Dr. E.), Shamans and Spiritism on the Mosquito Coast, Honduras, 135

Cook (Prof. C. W.), [death], 538

Cook (Major F. C.), appointed chairman of the Road Research Board, 686

Cook (Dr. J. W.), and Prof. E. C. Dodds, Sex Hormones and Cancer-producing Compounds, 205; Prof. E. C. Dodds, and C. L. Hewett, A Synthetic Estrusexciting Compound, 56

Cooper (G. A.), [Dr. C. Schuchert and], Brachiopod genera of the Sub-Orders Orthoidea and Pentameroidea, 244

Cooper (J. W.), and T. C. Denston, A Textbook of Pharmacognosy (Review), 419

Copland (Prof. D. B.), appointed Alfred Marshall lecturer for 1933–34 in Cambridge University, 884

Corbet (Dr. A. S.), The Bacterial Growth Curve and the History of Species, 61

Corkill (Dorothy E.), [R. J. Lythgoe and], Measurement of Visual Acuity, 98

Cornford (Prof. F. M.), Before and After Socrates (Review),

Cornish (Dr. Vaughan), Lessons in Visualisation from the Royal Academy, 644; Flow of Water between Moving Boundaries, 731

Correns (Prof. C. E.), [death], 353; [obituary article], 537 Costa (C. de A. Martins), appointed director of the Brazilian Meteorological Service, 581

Costantin (J.), Objections to the Mycorrhizal Theory, 447; Variations in the Virulence of Degenerescence in the Potato, 850

Coste (J. H.), Interaction between Soot Films and Oil, 691 Cotton (A.), Circular Magnetic Dichroism and Magnetic Rotatory Dispersion, 70; Construction of Thermoelectric Elements, 559; and M. Schérer, Magnetic Rotatory Dispersion of a Coloured Diamagnetic Compound, Thiobenzophenone, 215

Cotton (F. T.), [R. W. Fenning and], A Bomb Calorimeter Determination of the Heats of Formation of Nitrous Oxide and Carbon Dioxide, 446

Coulson (A.), Diabase Rocks at the You Yango near Geelong, 71; The Older Volcanic and Tertiary Marine Series at Curlewis near Geelong, 71

Cournot (J.), and Mlle. Louise Halm, Measurement of the Degree of Polishing in view of the Determination of the Amount of Corrosion of Rustless Steels, 738

Court (T. H.), [Dr. R. S. Clay and], The History of the Microscope: Compiled from Original Instruments and Documents, up to the Introduction of the Achromatic Microscope (Review), 219

Courtauld (Dr. S.), elected a member of the Athenæum

Club, 200

Courtauld (W. J.), gift to Cambridge University, 444

Coustal (R.), Connexion between the Two General Methods for the Preparation of Phosphorescent Zinc Sulphide,

Coventry (J. E. C.), [S. M. Naudé and], Intensity of Cosmic Radiation in South Africa, 411

Cowan (S. L.), Action of Quarternary Ammonium Salts on Nerve, 658

Coward (T. A.), [death], 158; [obituary article], 297

Cox (E. G.), and Dr. E. L. Hirst, Constitution of Vitamin $C_1, 402$

Cox (L. R.), Fossiliferous Siliceous Boulders from the Anglo-Egyptian Sudan, 70

Craigie (J. H.), Union of Pycniospores and Haploid Hyphæ in *Puccinia Helianthi* Schw., 25
Crampton (C.), [C. E. Horton and], A Radio Compass

developed in H.M. Signal School, 846 Crausse (E.), and J. Baubiac, Modification of the Chrono-

photographic Method and Some Applications, 482; Transitory Regimes, 559 Crawford (B. H.), [W. S. Stiles and], The Luminous Efficiency of Rays entering the Eye Pupil at Different Points, 250

Crawford (M. E. F.), E. O. V. Perry, and Dr. S. S. Zilva, Vitamin Content of Australian, New Zealand and English Butters, 770

Crawford (M. F.), and A. M. Crooker, Nuclear Moment of

Arsenic, 655

Crawhall (T. C.), Iron-working in the Bahr el Ghazal, 474 Crew (Prof. F. A. E.), Non-disjunction in the Fowl, 446 Crocco (G. A.), Stability of 'Instrumental' Flight, 71

Crommelin (Dr. A. C. D.), Halley's Comet in 1909-11, 282 Crooker (A. M.), [M. F. Crawford and], Nuclear Moment of Arsenic, 655

Crosby (Elizabeth C.), [G. K. Huber and], A Phylogenetic Consideration of the Optic Tectum, 739

Crossland (Dr. C.), Distribution of the Polychæte Worm, Syllis ramosa, McIntosh, 242

Crosthwait (Col. H. L.), Mount Everest, 10 Crow (Dr. A. D.), and W. E. Grimshaw, Rate of Burning of Colloidal Propellants, 60. The Combustion Problem of Internal Ballistics, 805

Crowden (G. P.), Muscular Work, Fatigue and Recovery (Review), 321

Crowfoot (D.), [J. D. Bernal and], Crystal Structure of Vitamin B₁, and of Adenine Hydrochloride, 911 Crowther (J. G.), Osiris and the Atom (*Review*), 320; The

ABC of Chemistry (Review), 895

Crump (L. M.), Eugenics and Marriage Laws, 540

Culey (Alma G.), Mineralogy of the Narrabeen Series of New South Wales, 523

Cummer (Dr. C. L.), A Manual of Clinical Laboratory
 Methods. Third edition (Review), 116
 Cunningham (Dr. Brysson), Canadian Water Power

Development in 1932, 788 Cunningham (J. T.), and D. M. Reid, Pelvic Filaments of

Lepidosiren, 913

Curie (Mme. Irène), and F. Joliot, Conditions of Emission of Neutrons by the Action of α-Particles on the Light Elements, 447

Curie (M.), and S. Takvorian, Radioactivity of a Neodymium-Samarium Fractionation, 702

Curry (J.), and Dr. G. Herzberg, Extension of the Visible Absorption System of NO2 to Longer Wave-lengths,

Curtis (A. L.), British Coals, 580 Curtis (Prof. W. E.), Arc Spectrum of Iodine, 398 Cushing (Prof. H. W.), elected a foreign member of the

Royal Society, 798 Cuthbertson (J. W.), Fatigue Limit of Medium Carbon Steel, 140; Fatigue-resisting Properties of Aluminium Alloys at High Temperatures, 660

Dadswell (H. E.), and Maisie Burnell, Identification of the Coloured Woods of the Genus Eucalyptus, 280 D'Agostino (O.), [N. Parravano and], Velocity of Dis-

solution of Industrial Aluminas in Fused Cryolite, 107 Dakin (Prof. W. J.), Unusual Occurrence of Pelagic Organisms, 239

Dale (Miss A. B.), The Form and Properties of Crystals: An Introduction to the Study of Minerals and the use of the Petrological Microscope (Review), 44

Dale (Sir Henry), Ultramicroscopic Organisms and the Troubles which they cause, 370

Daly (Prof. I. de Burgh), appointed professor of physiology in Edinburgh University, 700

Dampier (Sir William Cecil), appointed a Development Commissioner, 581

Dana (Prof. E. S.), A Treatise on Mineralogy: with an extended treatise on Crystallography and Physical Mineralogy. Fourth edition, revised and enlarged by Prof. W. E. Ford (Review), 318

Dangeard (P. A.), A Centrosomic Apparatus in Species of

the Genus Lonicera, 886

Dann (Dr. W. J.), Hexuronic (Ascorbic) Acid as the Antiscorbutic Factor, and its Chemical Determination, 274; Methylnornarcotine, Glycuronic Acid, and Vitamin C, 24; [T. W. Birch and], Estimation and Distribution of Ascorbic Acid (Vitamin C) and Glutathione in Animal Tissues, 469

Darby (Dr. H. H.), Insects and Micro-climates, 839

Darling (Dr. F. F.), [Dr. W. Orr and], Sterility in Domestic-ated Animals, 200
 Darlington (Dr. C. D.), Chromosomes and Plant-breeding

(Review), 5; Recent Advances in Cytology (Review), 5 Darzens (G.), and M. Meyer, New General Method for the Synthesis of Aldehydes, 483

Das (Prof. B. K.), [M. Rahimullah and], Making Whole

Mounts of Vertebrate Skeletons, 171
Dasgupta (Dr. Surendranath), A History of Indian Vol. 1. Second impression. Philosophy. (Review), 855

David (P.), Radiation of the Radio-transmitter of the Eiffel Tower, 179

Davidson (Dr. A. M.), and Dr. P. H. Gregory, Development of Fuseaux, Aleuriospores, and Spirals on Detached Hairs infected by Ringworm Fungi, 836 Davidson (Dr. J.), Effect of Rainfall-Evaporation Ratio

on Insects inhabiting the Soil Surface, 837 Davie (J. H.), Fixation of Mitochondria, 59

Davies (Dr. C. W.), and W. H. Banks, Dissociation of Acetic Acid in Water, 328

Davies (D. A. Bryn), The Ordovician Rocks of the Trefriw District, North Wales, 595

Davies (G. M.), A French-English Vocabulary in Geology and Physical Geography (Review), 604

Davies (Col. L. M.), The Genera Dictyoconoides, Lockhartia, and Rotalia, 279
Davis (Prof. W. M.), Glacial Epochs of the Santa Monica

Mountains, California, 288

Davison (Dr. C.), The Recent Japanese Earthquake, 351 Davison (F. T.), elected president of the American Museum of Natural History, 236

Davy (C. H.), and C. H. Sparks, High Pressure Boilers, 90 Dawe (M. T.), appointed director of agriculture and forests, Palestine, 543

Dawson (Prof. H. M.), Ionisation of Sulphuric Acid, 375 Deaglio (R.), Photoelectric effect in the Mono-crystals of Cuprite, 887

Deane (C.), New Species of Corylophidæ (Coleoptera), 287 Dearborn (Dr. N.), Fur-bearing Animals in Michigan, 127 Debierne (A.), New Radioactive Substances, 630

Debye (Prof. P.), elected a foreign member of the Royal Society, 798; Relations between Stereo-chemistry and Physics (Faraday lecture), 498

Dee (P. I.), appointed a university demonstrator in the Dept. of Physics of Cambridge University, 811

Defant (Dr. A.), awarded the Alexander Agassiz medal of the U.S. National Academy of Sciences, 754

Dei (C.), Sensitivity and Approximation of Measurements of the Amplification Coefficients of Triodes, 923

Delporte and Arend, A Remarkable Short-lived Nova,

Demontvignier (M.), [A. Labarthe and], Measuring and Recording Rapidly Variable Pressures, 179 Denbigh (K. G.), and Prof. R. Whytlaw-Gray, Higher

Homologues of Sulphur Hexafluoride, 763

Densmore (Miss). Yuman Music, 98

Denston (T. C.), [J. W. Cooper and], A Textbook of Pharmacognosy (Review), 419
Denton (Prof. F. M.), Dimensions of Fundamental Units,

Desmarez [Muller and], Differential Microscopic Characters of the Bone of Adult Cynocephalus and of Human Bone, 739

Detwiler (S. R.), Segmentation of Spinal Nerves in Salamander Embryos, 739

Deulofeu (Dr. V.), A Rule for the Rotatory Direction of the Acetylated Aldonic Nitriles, 548

Deutschbein (O.), [Prof. R. Tomaschek and], Fluorescence

of Pure Salts of the Rare Earths, 473

Devaux (J.), [C. Maurain and], Electrical Conductivity and Atmospheric Condensation Nuclei during Voyage to Greenland, 35; Total Calorific Radiation in Greenland, 106

De Vito (G.), [E. Parisi and], Distribution of Nitrates and 'Organisation' of Nitrogen in the Leaves of Green

Plants, 107

Dewey and others, Falling Water-Level in the Chalk

under London, 882

Dhar (Prof. N. R.), A. K. Bhattacharya, and B. L. Mukerji, Kinetics of the Iodine-Oxalate Reaction, 840; and Atma Ram, Presence of Formaldehyde in Dew, 800

Dick (Eveline M.), [J. Algar, Isabella M. McCarthy and], Synthesis of Diflavones, 666

Dick (Dr. George F.), and Dr. Gladys H. Dick, awarded the Cameron prize of Edinburgh University, 177

Dick (Dr. Gladys H.), [Dr. George F. Dick and], awarded the Cameron prize of Edinburgh University, 177

Dick (J.), appointed assistant lecturer in mechanical engineering in Sheffield University, 736

Dickens (Dr. F.), Iodoacetic Acid, Glutathione and Tissue Glyoxalase, 130

Dickinson (Dr.), Motor Car Lights on the Road, 20 Dickinson (Dr. S.), resignation of, from the University College of South Wales and Monmouthshire, 444

Dietz (Emma M.), [J. B. Conant and], Structural Formulæ

of the Chlorophylls, 131

Dinsdale (A.), First Principles of Television (Review), 456 Ditchburn (Prof. R. W.), Frequency Distribution of Resonance Radiation; Separating Isotopes by the use of Resonance Radiation, 106; Transmission of Resonance Radiation through a Gas, 738; [Dr. H. J. J. Braddick and], Absorption of Light in Casium Vapour, 132

Divoux (J.), [A. Donzelot and], Compensating Plate Currents, 106

Dixon (Prof. H. H.), Bast Sap, 367; Strepsinema Stage in Reduction, 437

Dixon (Dr. M.), appointed university lecturer in Biochemistry in Cambridge University, 665

Dixon (S.), Relation of Food to Disease, 333 Dixon (late Prof. W. E.), Memorials to, 832

Dobbie (Dr. J. C.), Effects of Tidal Displacements of the Vertical on Latitude Variation, 769

Dobinski (S.), Dielectric Constant of Liquid Phosphorus,

Dobson (Dr. G. M. B.), 'Raw' Weather, 28

Dobzhansky (T.), [A. H. Sturtevant and], Genetics of Certain Chromosome Anomalies in Drosophila melanogaster, 627

Dodds (Prof. E. C.), [H. Burrows, N. M. Kennaway and], Some Effects Observed in Mice under Continued Treatment with Estrin, 801; [Dr. J. W. Cook and], Sex Hormones and Cancer-producing Compounds, 205; [Dr. J. W. Cook, C. L. Hewett and], A Synthetic Estrus-exciting Compound, 56

Dodé (M.), [C. Matignon, H. Moureu and], Causes of the Simultaneous Production of 1-butene and 2-butene,

Dodwell (Prof. H. H.), elected a member of the Athenæum

Club, 510

Donaldson (Prof. R.), [death], 86; [obituary article], 122 Donatien (A.), [E. Sergent, L. Parrot, F. Lestoquard and], Experimental Suppression of Sexual Reproduction in Theileria dispar, 143

Donnan (Prof. F. G.), elected an honorary member of the Bunsen Society, 871

Dono (Tsurumatsu), The Copper Age in Ancient China, 243; 917

Dony (O.), Inverse Sublimation, 731

Donzelot (P.), and J. Divoux, Compensating Plate Currents, 106

Douglas (J. A.), Geology of the Marcapata Valley in Eastern Peru, 70

Drilhon (Mme. Andrée), Glucose and Shedding the Shell of Crustaceans, 483

Druce (Dr. G. Claridge), The Comital Flora of the British Isles (Flora Comitalis Britannicæ: Fl. Com. Brit.) (Review), 309

Druce (Dr. J. G. F.), Kronman's Rhenium (Review), 224 Dryden (L.), Xenohelix in the Maryland Miocene, 775

Drysdale (Dr. C. V.), appointed to the Safety in Mines Research Board, 91

Dubois (G.), Stratigraphic Subdivision of the Schistogreywacke Complex of the Vosges, 339

Dubois (P.), Action of Hydrogen Peroxide on Permanganate, 923

Dubouloz (P.), Yield of Fluorescence of Sodium Salicylate, 850

Duckert (R.), [H. Paillard and], Catalytic Oxidation of Acenaphthene, 815

Duclaux (Prof. M. J.), Structure of Cellulose, 553

Dufay (J.), and Ssu-Pin Liau, Absorption of Light in Interstellar Space, 923

Dufour (R.), Initial Electrolytic Overvoltage in the Disengagement of Hydrogen on Mercury, 482

Dufraisse (C.), and R. Buret, The Dissociable Organic Oxides, 70; and J. A. Monier, Jr., Dissociable Organic Oxides, 887

Dufton (A. F.), Inheritance of Intelligence in Man, 763 Dunn (Dr. J. T.), elected president of the Society of Chemical Industry, 432; Public Analysis and the Food Industry (Review), 822

Dunsheath (P.), Ionisation in Cable Dielectrics, 404 Dupin (P.), and M. Teissié-Solier, Vortices Produced by Obstacles revolving round an Axis parallel to the General Direction of Flow, 179

Durell (C. V.), Advanced Algebra. Vol. 1 (Review), 319 Durham (Bishop of), Ethical Conditions of Scientific Method (Fison lecture), 902

Durst (C. S.), Breakdown of Steep Wind Gradients in Inversions, 35; Intrusion of Air into Anticyclones, 737; Thermal Balance of a Water Drop or Ice Particle Suspended in the Atmosphere, 35; Variations in the Structure of Wind over Different Surfaces, 814

Duthie (E. S.), [Prof. J. B. Gatenby and], Cultures of Snail's Tissue, 474

Dwyer (F. P.), and D. P. Mellor, Occurrence of β-Cristobalite in Australian Opals, 524

Dyer (Dr. B.), and Dr. C. A. Mitchell, The Society of Public Analysts and other Analytical Chemists (Review), 822

Dyson (Sir Frank), Complimentary luncheon to, 391

Earl (J. C.), and N. F. Hall, Chemical Changes Involved in the Formation of Aminoazo-compounds (2), 667

Ebner (R.), [F. Werner and], Results of a Zoological Expedition to Morocco (1930): (5) Scorpions, 287 Eckersley (T. L.), Polarisation of Echoes from the Ken-

nelly-Heaviside Layer, 512 Eddington (Sir Arthur), Method of Least Squares, 374;

The Expanding Universe (Review), 637

Edgell (J. W.), appointed Davy bacteriologist in Bristol University, 848

Edisbury (J. R.), [Dr. R. A. Morton and], Absorption Spectrum of the Unsaponifiable Matter from Wheat-Germ Oil, 618

Edmond (F.), appointed to the Safety in Mines Research Board, 91

Edwards (F. S.), [T. E. Allibone, D. B. McKenzie and], A New Impulse Generator for Three Million Volts, Edwards (George), the work of, T. E. James, 124

Edwards (Capt. L. N.), Evolution of Early American Bridges, 429

Edwards (K. H. R.), Constitution and Temperament in

Man, 832

Edwards (T. I.), Temperature and Seed Germination, 404 van Eckelen (M.), and A. Emmerie, A Carotene Derivative giving with Antimony Trichloride an Absorption Band at 610-630 mu, 275

Egal (A.), A new Method of Realisation of Thermoelectric Phenomena, 410

Egerton (A.), and M. Milford, Fusion of Carbon, 169; and F. Ll. Smith, Hydrocarbon Combustion in an Engine, 725

Eggert (Prof. J.), translated by Dr. S. J. Gregg, Physical Chemistry (Review), 419; und Dr. R. Schmidt, Einführung in die Tonphotographie: photographische Grundlagen der Lichtton-Aufzeichnung (Review), 781 Eibenschütz (G.), Nature of Electrodynamic Forces, 923

Eichholz (Dr. A.), [death], 195

Einaudi (R.), Magnetic Behaviour of the Oxygen Molecule,

Einstein (Prof. A.), appointed professor of mathematical physics at the Collège de France, 581; The Method of Theoretical Physics (Herbert Spencer lecture), 867 Elford (W. J.), Principles of Ultra-filtration as applied

to Biological Problems, 178

Elkin (Dr. A. P.), The Cultural and Racial Clash in Australia, 520

Elkin (Prof. W. L.), [death], 866

Elliott (Dr. K. A. C.), elected a fellow of Selwyn College, Cambridge, 920

Ellis and Mott, Internal Conversion of γ-Rays, 517

Ellsworth (H. V.), Canadian Minerals of the Rare Elements, 660

Ellsworth (J.), The Double System R Z Cassiopeiæ with Eclipses, 850

Elmhirst (R.), Quantitative Studies between the Tide Marks, 767

Elmore (F. E.), bequest to Cambridge University, 700 Elsasser (Dr. W.), A Possible Property of the Positive

Electron, 764
Elvey (Dr. C. T.), The Gegenschein or Counterglow, 590
Emeléus (Dr. K. G.), appointed professor of experimental
physics in Belfast University, 628

Emeléus (Dr. H. J.), awarded the Harrison memorial

prize, 91 Emmerie (A.), An Inhibitor of the Antimony Trichloride Test for Vitamin A in Cod Liver Oil, 364; [M. van Eckelen and], A Carotene Derivative giving with Antimony Trichloride an Absorption Band at 610-630 mµ, 275

Emmerson (T.), [Prof. R. Whiddington and], Scattering

of Electrons in Helium, 814

Emmons (Prof. W. H.), Batholiths and Ore Deposits, 730

Emschwiller (G.), Action of Gaseous Hydrogen Iodide on some Iodine Derivatives of Hydrocarbons, 738 Enright (J. J.), H. E. Friesell, and M. O. Trescher, Cause

and Nature of Dental Caries, 918

Enriques (F.), Certain Invariant Series of Groups of Points

on an Algebraic Surface, 559
Enriques (Prof. P.), [death], 22; [obituary article], 265
Eriksson [Svedberg and], Hæmocyanin of Octopus vulgaris, 137

Esclangon (E.), Remarkable Epoch of Solar Activity, 559 Evans (B. J. Lloyd), [Dr. L. F. Bates and], A Compact Electromagnet for General Purposes, 558

Evans (Dr. C.), [R. C. T. Evans and], A Complex Solar Halo, 613

Evans (I. H. N.), Excavations at Kuala Selinsing, Perak, 135

Evans (R. C. T.), and Dr. C. Evans, A Complex Solar Halo, 613

Evans (Dr. R. H.), appointed lecturer in civil engineering

in Leeds University, 772 Eyring (Dr. H.), awarded the American Association prize, 304; Zero Point Energy and the Separation of Isotopes, 739

Fabris (A.), Absorptive Power of Soil for Pyrophosphoric Acid, 775

Fairbrother (F.), and E. Nightingale, General Science. Part 1 (Review), 420

Falkenhagen (Prof. H.), Elektrolyte (Review), 382 Famiani (V.), Capacity for Food Consumption after Fasting; Reconstructive Food Value of the Embryos of various Cereal and Leguminous Seeds, 339; Nutritive Value of Wheat Embryos, 107 Fantham (Prof. H. B.), Biology's Message for Civilisa-

tion, 197

Farmer (E.), The Causes of Accidents (*Review*), 895 Farmer (Dr. E. H.), and R. A. E. Galley, Catalytic

Hydrogenation of Olefinic Compounds, 60 Farran (G. P.), Unusual Occurrence of Pelagic Organisms, 240

Faure (Prof. J. C.), The Phases of Locusts in South Africa, 423

Favarger (P.), [H. Paillard and], Chlorination of Acenaphthene, 815

Fawcus (Lieut.-Gen.), Health of the British Army during

1931, 509 Fawcett (Prof. C. B.), elected president of the Institute of British Geographers, 51

Fawsitt (Prof. C. E.), Viscosity Measurements of Liquids by the Oscillating Disc Method, 97

de Fazi (R.), Chemical Constitution of Cholesterol, and a New Isomeride of Cholesterol, 924

Feather (Dr. N.), appointed a university demonstrator in the department of physics of Cambridge University, 811

Fedden (A. H. R.), awarded a silver medal of the Royal Aeronautical Society, 510

Fedotov (D. M.), Carboniferous Molluses of the Donetz Basin, 404

Fell (Dr. H. B.), and Dr. A. Robison, Glycogen in Cartilage, 62

Fenning (R. W.), and F. T. Cotton, A Bomb Calorimeter Determination of the Heats of Formation of Nitrous Oxide and Carbon Dioxide, 446

Ferguson (Dr. A.), A Vanished World (Review), 75; Edmond Halley, 153; Old Wine and New Bottles (Review), 417; Surface Tension and its Measurement, 66

Fermor (Dr. L. L.), The Place of Geology in the Life of a Nation, 227

Ferrari (A.), and G. Trampetti, Behaviour of Mixtures of Zinc Oxide and Anhydrous Zinc Chloride at High Temperature, 339

Fewkes (Dr. V. J.), American Archæologists in Yugoslavia, 393; Late Neolithic Fortress, Homolka,

Bohemia, 207

Finch (Prof. G. I.), and A. G. Quarrell, Crystal Structure and Orientation in Thin Films, 877; Determination of Crystal Lattice Constants by Electron Diffraction, 842; Structure of Magnesium, Zinc and Aluminium Films, 482; and R. W. Sutton, Control of Ignition-Coil Discharge Characteristics, 374

Findlay (Prof. A.), A Rebuke to Scientists, 1

Firth (Prof. R.), Anthropological Teaching and Research in Australia, 540; Indices of a Stable Population, 519 Fisher (Eileen E.), The 'Sooty Moulds' of Some Australian Plants, 107

Fisher (Miss N. I.), and Miss F. M. Harmer, Photographic

Sensitisers for the Infra-Red, 475 Fisher (Dr. R. A.), Number of Mendelian Factors in Quantitative Inheritance, 400; Statistical Methods for Research Workers. Fourth edition (Review), 383; Statistical Tables (Review), 893

Fisk (Prof. H. W.), [obituary article], 229 Fitzpatrick (H. M.), The Trees of Ireland, 35

Filon (Prof. L. N. G.), elected vice-chancellor of London

University, 177
Flajolet (P.), Transparency of the Atmosphere in the Lyons Region, 339

Fleming (Sir Ambrose), Television, 539 Fletcher (C. J. M.), [C. N. Hinshelwood and], Kinetics of the Decomposition of Molecules of Intermediate Complexity, 24

Fletcher (T. B.), Indian Microlepidoptera, 474

Fletcher (Sir Walter Morley), [death], 866

Fleure (Prof. H. J.), Archæology and Folk-Tradition, 431 Fleury (P.), and G. A. Boutry, Exact Measurement of Photographic Densities, 738

Fleuss (H. A.), [obituary article], 298 Foëx (E.), [R. Maire, G. Malençon and], Etiology of 'Bayoud', a Disease of the Date Palm, 922

Fonda, Particle Size and X-Ray Spectroscopy, 332 Forbes (A. C.), Some Legendary and Historical References to Irish Woods, and their Significance, 246

Forbes (Dr. H. O.), [obituary article], 460 Forbes (Dr. J. G.), Diphtheria, Past and Present: its Ætiology, Distribution, Transmission and Preven-

tion (*Review*), 150 Forbin (V.), Le Pétrole de Mésopotamie et son Pipe-

Line, 864

Ford (J. A.), Eskimo Culture Sequence, 243

Forestier (H.), Influence of the Magnetic Field on the Electrolysis of Nickel Salts, 70

Forster Cooper (C.), awarded a grant from the Balfour

Fund of Cambridge University, 736 Fortune (Dr. R. F.), Omaha Secret Societies, 331

Fosse (R.), P. de Graeve and P. E. Thomas, A New Plant Principle: Uric Acid, 179; Altantoic Acid in the Higher Plants, 702

Foster (A. W.), Some Measurements of the Thermoelectric Powers of Nickel and Nickel-Chromium Alloys in the Neighbourhood of their Curie Points, 814

Fotheringham (Dr. J. K.), Divisions of Time in Ancient Mesopotamia, 299; Eclipse of Hi and Ho, 881

Fournier (G.), and M. Guillot, Absorption of the β-Rays by Matter, 179; Relation between the Absorption of β-Rays by Organic Compounds and the Molecular Structure of the Latter, 447

Fowler (Sir Henry), elected president of the Institute of

Metals, 467

Fowler (H. W.), Fishes from the Tonga Archipelago, 63 Fox (Dr. Cyril), elected president of the Prehistoric Society of East Anglia, 520

Fox (Dr. C. S.), Bauxite and Aluminous Laterite. Second

edition (Review), 347

Fox (Prof. H. Munro), Reversible Stoppage of the Blood Circulation in Sabellids, 26; and M. L. Johnson, Control of Respiratory Movements in Crustacea, 514

Foxon (G. E. H.), Meaning of Neoteny and Pædogenesis, 93; Pelvic Fins of the Lepidosiren, 732; 913

Fraenkel (E.), and J. Zellner, Comparative Plant Chemistry (24), 888

Francis (W. L.), Output of Electrical Energy by Frog-Skin, 805

François (F.), Precipitation of Antimony Iodide and its Hydrolysis, 923

Franke (A.), A. Kroupa and S. Hadzidimitriu, A Synthesis

of α-alkyladipic Acids, 287 Fraser (A. H. H.), and Dr. D. Robertson, Nutritional

Condition of Sheep and Susceptibility to Stomach Worm, 94 Fraser (Lilian), [Gladys Carey and], Embryology and

Seedling Development of Aegiceras majus, Gaertn., 287 Fred (E. B.), I. L. Baldwin and Elizabeth McCoy, Root

Nodule Bacteria and Luguminous Plants, 99 Freeman (Dr. J. R.), Earthquake Damage and Earthquake Insurance (Review), 381; [obituary article], 266

Frenkel (Prof. J.), Wave Mechanics: Elementary Theory (Review), 860

Freudenberg (K.), [W. Kuhn und], Drehung der Polarisa-

tionsbene des Lichtes (*Review*), 677 von Freyberg (Prof. B.), Die Geologische Erforschung Thüringens in älterer Zeit: ein Beitrag zur Geschichte der Geologie bis zum Jahre 1843 (Review),

Fridenson (A.), [A. Girard, G. Sandulesco, J. J. Rutgers and], A New Crystallised Sex Hormone, 71

Friedheim (E.), Biological Signification of Melanogenesis,

Friend (Dr. J. Newton), A Text-Book of Physical Chemistry. Vol. 1: General Properties of Elements and Compounds (Review), 417

Friesell (H. E.), [J. J. Enright, M. O. Trescher and], Cause and Nature of Dental Caries, 918

Fritsch (K.), Flower-Visiting Insects in Styria, 1913, 852 Furtauer (R.), [J. Kisser and], Influence of certain Chemical Agents on the Carbon Dioxide Output of Germinating Seeds of Pisum saturim and Triticum

Gaarder (T.), and R. Spärck, Norwegian Oyster Pools, 806 Gabiano (P.), [R. de Mallemann and], Magnetic Rotatory Power of Chlorine and of Hydrochloric Acid Gas, 286

Gaden (H.), Proverbes et maximes Peuls et Toucouleurs traduits, expliqués et annotés (Review), 347

Gaede (Prof. W.), awarded the Duddell medal of the Physical Society; the work of, 195; presented with the Duddell medal of the Physical Society, 901

Gairdner (Alice E.), Sporangia containing Spermatozoids in Ferns, 621

Gale (Prof. R. C.), War and Post-War Explosives (Review), 452

Gallais (F.), Silver Iodomercurate, 215

Galley (R. A. E.), [Dr. E. H. Farmer and], Catalytic Hydrogenation of Olefinic Compounds, 60

Gamow (Dr. G.), Fundamental State of Nuclear a-Particles, 618; Mechanism of γ-Excitation by

integration, 57; Nuclear Energy Levels, 433
Garard (Prof. I. D.), An Introduction to Organic Chemistry (Review), 420

Gardiner (A. A. M.), Hermaphrodite Frog, 330

Gardiner (Prof. J. Stanley), The John Murray Expedition, 640

Garfias (V. R.), Oil Reserves and Production, 868

Garner (Prof. W. E.), Decomposition and Detonation of Solids, 65; and C. H. Moon, Acceleration of the Decomposition of Crystals of Barium Azine by the Emission from Radium Emanation, 513; and H. R. Hailes, Thermal Decomposition and Detonation of

Mercury Fulminate, 286 Garrod (Sir Archibald E.), The Inborn Factors in Disease:

an Essay (Review), 314

Garrod (Miss Dorothy), Further Remains of Palestine Man, 19

Garssen (J. E.), Magnetic Susceptibility of some Mixtures of Substances of Large Electric Moment, 523

Gatenby (Prof. J. B.), and E. S. Duthie, Culture of Snail's Tissue, 474

Gates (Prof. R. R.), Evolution and Philosophy (Review), 380; Inheritance of Flower-Size, 136; Phylogeny in the Genus Oenothera, 589; Prof. C. Correns, 537 Gattermann (L.), Revised by H. Wieland, translated by

Dr. W. McCartney, Laboratory Methods of Organic Chemistry (Review), 78

Gautheret (R.), [A. Guilliermond and], Microchemical Characters of the Oxyflavonic Compounds, 410

Gauzit (J.), Estimation of Atmospheric Ozone by Visual

Photometry, 36
Gay (L.), and J. Soulié, A Boiling Point Apparatus for the Determination of the Dew Points and Boiling Points of Mixtures of Volatile Liquids, 630

Gayford (Squadron-Leader O. R.), and Flight-Lieut. G. E. Nicholetts, New Record for Non-Stop Flight, 232; awarded the British Silver Medal of the Royal Aeronautical Society, 685;

Geddes (late Sir. Patrick), tributes to, 466

Genard (Jean), Molecular Fluorescence of Antimony, 132 Génaux (L.), [C. E. Brazier and], The Earthquake of March 2, 1933, 596

Gerlach (Dr. W.), and Dr. E. Schweitzer, Foundations and Methods of Chemical Analysis by the Emission Spectrum (translation) (Review), 420

Ghermanesco (M.), Orthogonal Polynomials with two Variables, 107

Ghigi (E.), [G. Charrier and], Action of Alkylmagnesium Iodides on (1:9)-Benzanthrone-(10), 107; of Ammonia on Acenaphthenequinone, 107

Ghiron (D.), [G. R. Levi and], Magnesium Chlorite and Double Chlorites of Copper with Magnesium, Barium, and Thallium, 775

Ghosh (Prof. P. N.), and A. K. Sen Gupta, Ultra-Violet Bands of Oxide of Phosphorus, 841

Gibson (Prof. C. S.), Organic Compounds of Gold, 130 Gifford (A. C.), Origin of the Solar System, 518

Gifford (Emma), Primes and Factors (Review), 785 Gifford (E. W.), Yuman Ethnology, 915

Gigante (R.), A Non-Parasitic Alteration of the Olive, 851 Gilding (H. P.), conferment upon, of the title of reader in experimental physiology, 177, 372

Gill (D.), Aerial Survey in Relation to Economic Geology,

160

Gion (L.), Photolysis of Aqueous Solution of Ammonia, 410

Girard (A.), and G. Chaudron, Dissociation of Cubic Ferric Oxide, 447; [G. Sandulesco, A. Fridenson and], A New Crystallised Sex Hormone, 71

Gire (G.), Thermal Decomposition of the Magnesium

Silicides, 923 Giroud (P.). [C. Nicolle, J. Laigret and], Passage of the Virus of Exanthematic Fevers by the Digestive Canal in the Rat, 375

Glasspoole (J.), Rainfall over the British Isles, 1820 to 1929, 737

Glazebrook (Sir Richard), awarded the gold medal of the Royal Aeronautical Society, 359 Gleason (G. H.), and A. C. Loonam, Sewage Purification

by a New Process, 697

Gley (P.), [L. Bugnard, A. Langevin and], Recording and Measurement of the Blood Pressure, 375

Gobert (E. G.), and H. Vaufrey, Iberomarusian, 367 Godwin (H. and M. E.), Age of Maglemose in Britain, 551 Godwin (M. E.), [H. Godwin and], Age of Maglemose in Britain, 551

Gold (Col. E.), Climate of Hong-Kong, 236

Goldschmidt (Prof. S.), Stereochemie (*Review*), 783 Goldschmidt (Prof. V.), [death], 718; [obituary article], 791

Golikere (R. K.), Through Wonderlands of the Universe (Review), 823

Goodall (W. M.), [J. P. Schafer and], Characteristics of the Ionosphere, 804

Goodrich (Prof. E. S.), Nephridia of Amphioxus, 439 Gorbach (G.), and A. Schonbeck, Influence of Hydrocyanic Acid on Bacterial Proteases, 287

Gordon (Isabella), [Dr. W. T. Calman and], Dodecolopoda mawsoni, a new species representing a new genus

of Pyenogonida, 774 Gorman (M. J.), Two Forms of Sampler used in Estimating the Number of Plants per Acre in Botanical Analyses of Grasslands, 410

Gortner (Prof. R. A.), Machine Age's Starvation Predicted, 393

Gott (Sir Benjamin), [death], 298; [obituary article], 426 Goude (H.), Improvement of Colour of Apples, 331

Gouzon (B.), [H. Bierry, Mlle. C. Magnan and], Application of the Iodometric Method to the Estimation of Sugar in the Blood, 667

Govaert (F.), Estimation of the Halogens in Organic Substances by the Sodammonium Method, 179

de Graeve (P.), [R. Fosse, P. E. Thomas and], Altantoic Acid in the Higher Plants, 702; A New Plant Principle: Uric Acid, 179

de Graaf (Van), Compton and Van Atta, Electrostatic Production of High Voltages, 475

Graham (Mrs.), Gift to Leeds University, 772

Grainger (J.), and Rachel M. Heafford, Some Effects of the ordinary Tobacco Mosaic upon the Developmental Anatomy of the Host Plant, 814 Grandidier (G.), et G. Petit, Zoologie der Madagascar

(Review), 748

Granet (Prof. M.), translated by Dr. E. D. Edwards, Festivals and Songs of Ancient China (Review), 346 Granier (J.), Conducting Properties of India-rubber Heavily Loaded with Lamp-Black, 738

Grant (G. H.), [C. N. Hinshelwood and], Upper Pressure Limit in the Explosive Chain Reaction between Hydrogen and Oxygen, 361

Grant (R.), Occurrence of Ovulation Without 'Heat' in the Ewe, 802

Gravely (Dr. F. H.), and T. N. Ramachandran, Hindu Metal Images, 279

Gray (Dr. J.), Directional Control of Fish Movements, 774; Importance of Zoological Advice to the State, 88; Muscular Movements of Fishes, 825

Gray (J. D. A.), appointed senior pathological officer in the department of preventive medicine of Bristol University, 848

Gray (Prof. J. G.), Self-Erecting Gyrostats, 250 Gray (J. L.), and Pearl Moshinsky, Genetic Psychology (1), 922

 Gray (Dr. R. W.), Peterhead Sealers and Whalers, 904
 Gredy (Mlle. B.), The Acetylene Linkage, 774; [R. Lespieau and], Study of Some α-Ethylene Oxides, 447 Green (S. E.), [Mary Bell and], Radiometer Action and

the Pressure of Radiation, 374

Green (W. J.), The Hydroxyl Group and Soap Film Structure, 873

Greene (F. A.), appointed a fellow of King's College, London, 177

Greenwood (Dr. T.), The Development of Indian Thought (Review), 855

Gregory (Dr. P. H.), [Dr. A. M. Davidson and], Development of Fuseaux, Aleuriospores, and Spirals on Detached Hairs by Ringworm Fungi, 836 Greig (J. W.), H. E. Merwin and E. S. Shepherd, Volatile

Transport of Silica, 768

Grenet (G.), The Geothermic Gradient in Limogne, 143 Griffith (M.), appointed lands director of the Welsh Plant Breeding Station of the University College of Wales, 720

Griffiths (Dr. E.), [J. H. Awbery and], Heats of Combustion of Carbon Monoxide in Oxygen and of Nitrous Oxide in Carbon Monoxide at Constant Pressure, 446

Grimpe (G.), und E. Wagler. Herausgegeben von G. Grimpe. Die Tierwelt der Nord-und Ostsee. Lief.

21, Teil 1 d₂ und Teil 2g (*Review*), 116

Grimshaw (W. E.), [Dr. A. D. Crow and], Rate of Burning of Colloidal Propellants, 60; The Combustion Problem of Internal Ballistics, 805

Grosset (H.), Plant Geography of Ulijanowsk, 696 Grosrey (A.), [G. Tiercy and], Width of a Photographic

Stellar Spectrum for Stars of the Type A5, 888 Grout (Prof. F. F.), Petrography and Petrology: a Textbook (Review), 317

Groves (J.), [death], 462

Grumez (Mme. M.), [Mme. Ramart-Lucas and], Colour and Structure of Oximes and Semicarbazones, 70

Guareschi (C.), Internal Ear of Amphibia, 659

Guérin (P.), Hydrocyanic Acid in *Glyceria aquatica*, 106 Guerrieri (E.), Periodicity in the Progressive Course of the Rainfall at Capodimote during 1833-1931, 107 Gudger (Dr. E. W.), The Whale Shark in the Waters around Ceylon, 165

Guichard, Atomic Weight of Iodine, 738 Guild (J.), Reversal of Current in Rectifier Photo-Cells,

Guillaume (Dr. Ch.-Ed.), Invar, 658

Guillerm (J.), [P. N. Bernard and], Transmissible Lysis of the Cholera Vibrion, 887

Guilliermond (A.), and R. Gautheret, Microchemical Characters of the Oxyflavonic Compounds, 410

Guillot (M.), [G. Fournier and], Absorption of the B-Rays by Matter, 179; Relation between the Absorption of β-Rays by Organic Compounds and the Molecular Structure of the Latter, 447 Guinier (A.), [G. Bruhat and], Improvement of the

Photoelectric Polarimeter, 630 Gulland (Dr. J. M.), and T. F. Macrae, Action of Proteo-

lytic Enzymes on the Oxytocic Principle of the Pituitary Gland, 470

Gulick (A.), Evolutionist and Missionary, John Thomas Gulick: Portrayed through Documents and Discussions (Review), 532

Gunn (Dr. R.), Origin of the Planetary System, 405 Günther (P. L.), [Prof. F. A. Paneth and], Chemical Detection of Artificial Transmutation of Elements, Gunther (Dr. R. T.), The Astrolabes of the World. 2 vols. (Review), 819

Gupta (A. K. Sen), [Prof. P. N. Ghosh and], Ultra-Violet Bands of Oxide of Phosphorus, 841

Gurney (Dr. R. W.), appointed a research fellow in Bristol

University, 772 Gurwitsch (Dr. A.), Mitogenetic Radiation of Nerve, 912; Unter Mitwirkung von Lydia Gurwitsch, Die mitogenetische Strahlung, zugleich zweiter Band der "Probleme der Zellteilung" (Review), 79

Gustafson (T.), and B. Kullenberg, Inertia Currents in

the Baltic, 586

Gutzeit (G.), and R. Monnier, Utilisation of some Azo Derivatives of Oxyquinoline as Reagents in Qualitative Analysis, 144 Guyatt (Dr. B. L.), [Prof. H. D. Kay and], Experimental

Rickets as a Phosphorus Deficiency Disease, 468

Gwynn (A. M.), appointed entomologist in the department of agriculture, Nigeria, 871 Gyllenberg (W.), Proper Motions in the Lund Zone of

the Astronomische Gesellschaft Catalogue, 245 Gysin (M.), Petrographical Researches in the Haut-Katanga (1), 411; (2), 815; (3), 887

de Haas (Prof. W. J.), E. C. Wiersma and Prof. H. A. Kramers, Temperature below 0.27° K., 719

Haberlandt (H.), Luminescence Investigations with Fluorites (2), 851

Hada (Y.), [T. Tamura and], Growth of Corals, 172 Haddon (E. B.), Tribes of the Southern Sudan (Review),

Hadfield (Prof. G.), appointed professor of pathology in Bristol University, 848

Hadfield, Bt. (Sir Robert), elected an honorary member of the Academy of Sciences of the U.S.S.R., 272; luncheon to, 355

Haenny (C.), Magnetic Double Refraction of some Cerium Salts in Aqueous and Non-Aqueous Solutions,

887 Hägg (Prof. G.), Vacant Positions in the Iron Lattice of Pyrrhotite, 167

Haig-Brown (R. L.), Silver: the Life-Story of an Atlantic

Salmon (*Review*), 42 Hailes (H. R.), [Prof. W. E. Garner and], Thermal Decomposition and Detonation of Mercury Fulminate, 286 Haitinger (M.), [E. Haschek and], A Simple Method of

determining Colour, 411

Haldane (Prof. J. B. S.), appointed professor of genetics at University College, London, 336; The Causes of Evolution (*Review*), 709; The Inequality of Man: and other Essays (*Review*), 529

Hall (E. R.), Mammals of Vancouver Island, 879

Hall (N. F.), [J. C. Earl and], Chemical Changes Involved in the Formation of Aminoazo-Compounds (2), 667

Hall (P.), appointed a university lecturer in mathematics in Cambridge University, 848

Hallett (Sir Frederic G.), [death], 195

Halley (Edmond), Correspondence and Papers of, arranged and edited by E. F. MacPike, 153
Hallimond (A. F.), and E. F. Herroun, Magnetic Pro-

perties of Certain Igneous Rocks, 338 Halm (Mlle. Louise), [J. Cournot and], Measurement of the Degree of Polishing in view of the Determination of the Amount of Corrosion of Rustless Steels, 738 (H.), Ribaucour's Transformation

Hamburger Spherical Representation (3), 339

Hamer (Miss F. M.), [Miss N. I. Fisher and], Photographic Sensitisers for the Infra-Red, 475 Hamilton (Sir Ian), installed as Lord Rector of Edinburgh

University, 372

Hanley (F.), [R. Sayce and], Kiln-dried Poultry Manure, 198

Harden (Prof. A.), Alcoholic Fermentation (Bedson

Lecture), 756
Harding (Col. E. W.), The Flyfisher and the Trout's
Point of View: New Light on Flyfishing Theory and Practice (Review), 42

Harding (J. W.), Semi-Conductors in a Magnetic Field, 731

Hardy (A. C.), Use of Oil and Coal in Ships, 479

Hardy (Sir William), Food Storage at Low Temperature, 459

Harker (Dr. A.), Metamorphism: a Study of the Transformations of Rock-Masses (Review), 310

Harkins (Prof. W. D.), The Neutron and Neuton, the New Element of Atomic Number Zero, 23

Harmer (Sir Sidney), re-elected president of the Ray Society, 466

Harradon (H. D.), Prof. H. W. Fisk, 229 Harris (Dr. L.), W. S. Benedict, and G. W. King, Form and Vibrational Frequencies of the NO₂ Molecule, 621

Harris (Dr. L. J.), [T. W. Birch, S. N. Ray and], Hexuronic (Ascorbic) Acid as the Antiscorbutic Factor, and its Chemical Determination, 273

Harris (R. S.), and J. W. M. Bunker, Bacterial Detoxification, 244

Harris (R. V.), The Strawberry 'Yellow-edge' Disease, 730 Harris (R. W.), Control of the Tsetse Fly, 463

Harrison (C. A.), Broad Highway of Soviet Education, 920

Harrisson (T. H.), and P. A. D. Hollom, Spread of the Great Crested Grebe in Britain, 135

Hartley (Sir Harold), Priestley's Service to Chemistry, 555

Hartmann (Dr. C. G.), Sexual Cycle in the Rhesus Monkey, 626

Hartog (Sir Philip), Priestley as a Scientific Man and Theorist, 555

Hartridge (Prof. H.), A Method of Extending the Frequency Range of the Cathode Ray Tube, 95; De-

tection of Traces of Carbon Monoxide in Air, 654 Haschek (E.), and M. Haitinger, A Simple Method of Determining Colour, 411

Haskell (Prof. E. E.), [death], 353

Haslewood (G. A. D.), appointed a research student at the Research Institute of the Cancer Hospital, 509 Hatfield (Dr. W. H.), awarded the Bessemer gold medal

of the Iron and Steel Institute, 236

Hatton (J. L. S.), [death], 86; [obituary article], 230 Haughton (Dr. S. H.), Geology of South Africa, 624

Hautot (A.), Structure of the K Line of Boron, 595; [M. Morand and], Fine Structure of the Carbon Line $K\alpha$, 143

Haworth (Prof. W. N.), [Prof. A. Szent-Györgyi and], Hexuronic Acid (Ascorbic Acid) as the Antiscorbutic Factor, 24

Hawes (Dr. W. B.), awarded the Junior Moulton medal and prize of the Institution of Chemical Engineers, 510

Hawkes (Hilda K.), [L. Hawkes and], Sandfell Laccolith and 'Dome of Elevation', 214

Hawkes (L.), and Hilda K. Hawkes, Sandfell Laccolith and 'Dome of Elevation', 214

Hawkins (Prof. H. L.), elected president of the 1934 Congress of the South-Eastern Union of Scientific

Societies, 920 Hawksworth (D.), Ancestor Spirits among the Nuba, 403 Hawley (H.), Phytosteryl Acetate Test for examining Butter Fats, 814

Hayford (Miss Phyllis), Galactic Rotation, 173

Headlam-Morley (K.), appointed secretary of the Iron and Steel Institute, 272

Heafford (Rachel M.), [J. Grainger and], Some Effects of the Ordinary Tobacco Mosaic upon the Develop-

mental Anatomy of the Host Plant, 814 Heard (R. D.), H. W. Kinnersley, J. R. O'Brien, Prof. R. A. Peters, and V. Reader, Vitamin B_4 and Adenine, 617

Hedenius (Astrid), [Prof. The Svedberg and], Molecular Weights of the Blood Pigments of the Invertebrates, 325

Hedin (Sir Sven), translated by E. G. Nash, Jehol: City of Emperors (Review), 184

Hedges (Dr. E. S.), Liesegang Rings, 169; Liesegang Rings: and other Periodic Structures (Review), 316 Hée (Mme. A.), Earthquakes of Northern Africa, 807

Heilbron (Prof. I. M.), appointed professor of organic chemistry in Manchester University, 772; J. C. E. Simpson, Hydroxyl Group in Ergosterol and Cholesterol, 438

Hemming (H.), Aerial Survey in Relation to Economic Geology, 160

Henderson (Prof. G. G.), presidential address to the Chemical Society, 498 Herdman (D. W.), [C. Squire and], The Museums of

Malta, Cyprus and Gibraltar, 430

Herdman (H. F. P.), Soundings in the Scotia Sea, 440 Herenguel (J.), and G. Chaudron, Sublimation of Magnesium in a Vacuum and Casting in an Atmosphere of

Argon, 179 Hermitte (Dr. L. C. D.), appointed demonstrator in pathology in Sheffield University, 884

Hernegger (F.), Detection of Uranium in Spring Waters

and Deposits, 704 Herrick (C. J.), Functions of the Olfactory Parts of the

Cerebral Cortex, 739

Herring (E.), Methods of Warming and Ventilating the Masonic Peace Memorial Building, 541

Herrman (L.), and Prof. L. Hogben, Intellectual Resemblance of Twins, 446 Herroun (E. F.), [A. F. Hallimond and], Magnetic Pro-

perties of Certain Igneous Rocks, 338

Hershey (Prof. J. W.), Components of the Atmosphere and Synthetic Gases in Relation to Animal Life, 238 Herzberg (Dr. G.), [J. Curry and], Extension of the Visible

Absorption System of NO2 to longer Wave-Lengths,

Herzberger (Dr. M.), Strahlenoptik (Review), 747

Herzen (G. M.), [P. Auger and], Emission of Neutrons by Aluminium under the action of the α-Particles, 523 von Hevesy (Prof. G.), Chemical Analysis by X-Rays and

its Applications (Review), 39; Properties of the Atom (Review), 4; and Dr. M. Pahl, Range of Radiation from Samarium, 434

Hewett (C. L.), [Dr. J. W. Cook, Prof. E. C. Dodds and], A Synthetic Oestrus-exciting Compound, 56

Hey (M. H.), A Possible Source of Error in the Determination of Symmetry from Optical Extinction-Angles, 630; Studies on the Zeolites (5), 630

Heyl (P. R.), The Philosophy of a Scientific Man (Review), 491

Heyroth (F. F.), and J. R. Loofbourow, Irradiation of

Nucleic Acids and Uracil, 92 Heyward (Jean), Two Species of Ophiocytium Nageli in Victoria, O. terrestris n.s. and O. arbuscula Raben-

horst, 107

Hibben (Dr. J. G.), [death], 866 Hickinbottom (Dr. W. J.), Alkylanilines with Tertiary Alkyl Groups, 762

Hicks (Prof. C. Stanton), Scientific Centralisation in the British Empire, 397

Hicks Prof. G. Dawes), Berkeley (Review), 322

Hickson (Prof. S. J.), Gorgonacea from the Great Barrier Reef, 879

Hilger, Ltd. (Adam), Spectrograph for Study of Fibrous

Substances, 844 (Prof. A. V.), Living Machinery (Review), 860; Physical Nature of the Nerve Impulse, 233; 497; 501 Hill (Dr. B.), Mortality from Whooping-Cough, 269

Hill (Sir Leonard), 'Raw' Weather, 28; 241

Hilton (Prof. H.), Plane Algebraic Curves. Second edition

(Review), 383 Himus (G. W.), Fuel Testing: Laboratory Methods in

Fuel Technology (*Review*), 259 Hinds (G. H.), [F. R. W. Hunt and], Rate of Burning of Colloidal Propellants, 206

Hingston (Major R. W. G.), A Naturalist in the Guiana Forest (Review), 78

Hinshelwood (C. N.), and C. J. M. Fletcher, Kinetics of the Decomposition of Molecules of Intermediate Complexity, 24; and G. H. Grant, Upper Pressure Limit in the Explosive Chain Reaction between Hydrogen and Oxygen, 361; M. Hughes and A. C. Rolfe, Reaction between Hydrogen and Oxygen, 625; Combination of Hydrogen and Oxygen in a Silver Vessel, 142

Hiriyanna (M.), Outlines of Indian Philosophy (Review), 855

Hiroa (Te Rangi), (Dr. P. H. Buck), Ethnology of Manihiki and Rakahanga, Cook Islands, 588

Hirst (Dr. E. L.), [E. G. Cox and], Constitution of Vitamin C, 402; E. G. V. Percival and F. Smith, Constitution of Ascorbic Acid, 617

Hitchen (C. S.), Skiddaw Granite and its Residual Products, 482

Hoagland (H.), Impulses from Sensory Nerves of Catfish, 631

Hoar (T. P.), awarded the Gordon Wigan prize for chemistry of Cambridge University, 284

Hobson (Bernard), bequests for Geological Research, 500 Hobson (Prof. E. W.), [death], 576

Hocart (R.), Orientation of Arsenolite and of Senarmontite by Mica, 851

Hodge (W. V. D.), appointed a university lecturer in mathematics in Cambridge University, 848 Hodges (Prof. C.), The Background of International

Relations: Our World Horizons, National and International (Review), 45

Hodgson (Dr. W. C.), Forecasting the Herring Fishery off East Anglia, 98

van Hoepen (Dr. E. C. N.). A New South African Culture?

Hoffmann (J.), Alkali- and Barium-Silver Chlorides, 704 Hogben (Prof. L.), [L. Herrman and], Intellectual Resemblance of Twins, 446

Hogbin (Dr. I.), Ceremonial Exchange in Polynesia, 439 Holden (Lord), re review of Bond's "Numerical Examples in Physics", 467

Holland (Sir Thomas), elected president of the Geological Society of London, 432

Holland (T.), and others, edited by Mary Adams, Science in the Changing World (Review), 674

Hollis (H. P.), A Numerical Coincidence, 550

Hollom (P. A. D.), [T. H. Harrisson and], Spread of the Great Crested Grebe in Britain, 135

Holm (Dr. H. T.), [death], 195

Holmes (Dr. E. G.), appointed a university lecturer in biochemistry in Cambridge University, 665

Holmes (Dr. W. H.), [death], 718 Holmes (W. M.), The Mornington Earth Tremor of Sept. 3, 1932, 71

Holmyard (Dr. E. J.), Pliny's Chemical Knowledge (Review), 305; Thomas Norton and the "Ordinall of Alchimy", 520; [W. C. Badcock and], Electricity and Magnetism for Beginners (Review), 347

Holzl (F.), and W. Stockmair, Complex Anion of Buff's Compound and of Bunsen's Salt, 251

Honda (Prof. Kôtarô), A New Alloy, 'Stainless-Invar', 587

Hope (Lord Charles), Further Light on the Schneider Mediumship, 549 Hope-Jones, Time Measurement: Old and New, 67

Hopkin and Williams, Ltd., Organic Reagents for Metals, 396

Hopkins (B. S.), and L. L. Quill, Use of Non-Aqueous Solvents in the Study of the Rare Earth Group,

Hopkins (Sir Frederick Gowland), Installation of, as president of the British Association, 50

Hopwood (A. T.), [Dr. L. S. B. Leakey, Prof. H. Reck, Prof. P. G. H. Boswell, Dr. J. D. Solomon and], The Oldoway Human Skeleton, 397 Hornblower (G. D.), Dragons, 806

Horton (C. E.), and Crampton, A Radio Compass Developed in H.M. Signal School, 846

Horton (W.), Gift to Liverpool University, 557

Houstoun (Dr. R. A.), Vision and Colour Vision (Review), 532

Howard (B. A.), The Proper Study of Mankind (Review), 676

Howell (A. B.), Jumping Rodents, 623

Howes (N. H.), [D. I. Clements, G. P. Wells and], Is Plasticine Edible ?, 330

Huber (G. K.), and Elizabeth C. Crosby, A Phylogenetic Consideration of the Optic Tectum, 739

Hudson (H.), [Prof. T. M. Lowry and], Optical Rotatory Power (4), 374

Hudspeth (Major H. M.), appointed to the Safety in Mines Research Board, 91

Huffer (Dr. C. M.), [Prof. J. Stebbins and], Absorption of

Light in the Galaxy, 769 Hughes (A. H.), and Prof. E. K. Rideal, Rate of Oxidation of Monolayers of Unsaturated Fatty Acids, 446 Hughes (A. W. McKenny), [Major E. E. Austen and],

Clothes Moths and House Moths, 55

Hughes (J. V.), The Spurious Ring Exhibited by Fluorescent Screens, 558

Hughes (M.), [Hinshelwood, Rolfe and], Reaction between Hydrogen and Oxygen, 625

Hull (Miss Eleanor), Heathen Baptism in Early Britain, 403

Hullett (E. W.), and J. W. Calder, Fluorescence in Ryegrasses, 474

Hulme, Internal Conversion of γ-Rays, 99

Hulthén (Prof. E.), and R. Rydberg, Predissociation and Pressure Effects in the Band Spectrum of Aluminium Hydride, 470

Humason (Dr.), Spectral Types of Faint Stars, 209 Humberstone (T. Ll.), The New Buildings for the University of London, 903

Humphreys (F. E.), and H. Phillips, Examination of Leather for the Presence of Extractable Chromium Compounds, 813

Hunsaker (Commdr. J. C.), awarded the Daniel Guggenheim medal, 871

Hunt (F. R. W.), and G. H. Hinds, Rate of Burning of Colloidal Propellants, 206

Hunt (Dr. J. Middlemass), bequest to Liverpool University, 51

Hunt (Theresa M.), [Dr. C. E. P. Brooks and], Variations of Wind Direction in the British Isles since 1341, 814

Hunter (Dr. J. de Graaff), Time Determination, 515 Hunter (Miss M. M.), awarded the Wyse studentship of Cambridge University, 884

Hurst (Dr. C. C.), Inheritance of Intelligence in Man, 764; The Mechanism of Creative Evolution (Review), 780

Huxley (Dr. L.), [death], 718 Huxley (L. G. H.), [G. H. Munro and], Atmospherics in

Australia, 592

Iimori (S.), Periodicity in the Solarisation of Calcite, 619 Illing (Prof. V. C.), Migration of Oil and Natural Gas, 475 Imamura (A.), T. Kodaira and H. Imamura, Earthquake Series at Nagusa, Japan, 136

Imamura (H.), [A. Imamura, T. Kodaira and], Earth-

quake Series at Nagusa, Japan, 136 Innes (Dr. R. T. A.), [death], 390; [obituary article], 461 Ionescu (T. V.), and C. Mihul, Ionised Gases in the Magnetic Field, 106; High-Frequency Discharge in

Gases, 887 Iriki (Dr. S.), Chromosomes of Amphibians and Fishes, 659 Isikawa (Dr. Tomoyosi), Physiological Standards and Occupational Characteristics of Bodily Functions of the Japanese, 588

Ivanoff (D.), and I. Abdouloff, Velocity of Disengagement of Hydrocarbons produced by the Action of Indene

on Fatty Organomagnesium Derivatives, 483 Ivimey-Cook (Dr. W. R.), appointed assistant lecturer and demonstrator in botany in University College, Cardiff, 736

Bands in the Thermoluminescence Iwase (Ei-ichi), Spectrum of Fluorite from Obira, Japan, 909

Jabloński (Dr. A.), Efficiency of Anti-Stokes Fluorescence

in Dyes, 839 Jacek (W.), Velocity of the Solution of Marble in Acids (3), 887

Jackson (A.), Egyptian Neolithic Barley, 652
 Jackson (D. A.), Structure of the Lines of the Arc
 Spectrum of Silver, 691

Jackson (Sir Herbert), Photographic Graticules, 766; Prof. J. Millar Thomson, 610

Jackson (Sir John), [death], 496 Jackson (L. C.), Principal Magnetic Susceptibilities of Some Paramagnetic Crystals at Low Temperatures,

Jacob (S. M.), Census of Nigeria, 516

Jacqué (L.), Alteration of Steels by Hydrogen, 36 Jacquet (P.), Adsorption of Colloids by Metallic Surfaces, etc., 702; Strains in Electrolytic Copper Deposited in the Presence of Colloids, 70

Jaffray (J.), High-Frequency Currents produced by

High-Tension Magnetos, 559

James (Dr. E.), appointed junior assistant bacteriologist and demonstrator in Sheffield University, 884

James (T. E.), Tercentenary of Samuel Pepys, 228; The Old Ashmolean, Oxford, 716; The Work of George Edwards, F.R.S., 124

Jaslan (S.), [W. Broniewski and], Influence of Oxygen on the Properties of Copper, 338
 Jausseran (C.), Evolution of the Latent Image, 410

Jay (A. H.), [A. J. Bradley and], Quartz as a Standard for Accurate Lattice-Spacing Measurements, 813

Jeffery (G. H.), [Dr. A. I. Vogel and], Limiting Nobilities of some Monovalent Ions and the Dissociation Constant of Acetic Acid at 25°, 27

Jeffreys (Dr. H.), Evolution of Hydrodynamics (Review), 313; and K. E. Bullen, Corrections to the Times of the P-Wave in Earthquakes, 97 Jeffreys (W. Rees), Transport Problems of the Empire,

126

Jeffs (Prof. R. E.), [death], 646

Jenkins (Dr. R. L.), Inheritance of Acquired Characters,

Jenkins (Rhys), A Cornish Engineer, Arthur Woolf, 124 Jenkins (S. H.), [Dr. A. G. Norman and], Lignin Content of Cellulose Products, 729

Jenks (A. E.), Minnesota Pleistocene Homo, 739

Jensen (H. L.), The Actinomycetales (4), 631 Jespersen (Dr. P.), Food of the Herring in Icelandic Waters, 884

Jevons (Dr. W.), Band-Spectra of Diatomic Molecules, 102

Jewell (W.), [Dr. F. H. Carr and], Characteristics of Highly Active Vitamin A, 92

Jezewski (M.), Application of the Resonance Method to the Determination of the Dielectric Constants of Aqueous Electrolytes, 448

Job (P.), Constitution of Hydrochloric Acid Solutions of Cobalt Chloride, 338

Johnson (Dr. C. H.), and A. Mead, Line Absorption of Chromic Salts in Relation to Co-ordination, 399

Johnson (M. L.), [Prof. H. Munro Fox and], Control of Respiratory Movements in Crustacea, 514

Johnson (Dr. R. C.), Band Spectra of Diatomic Molecules, 102

Johnstone (E. S.), [F. S. Brackett and], Functions of Radiation in the Physiology of Plants, 331 Johnstone (Prof. J.), [death], 22; [obituary article], 157

Joliot (F.), [Mme. Irène Curie and], Conditions of Emission of Neutrons by the Action of α-Particles on the Light Elements, 447

Jones (Brindley), [W. Singleton and], Effects of the Addition of Tellurium to Lead, 696

Jones (E. Gwynne), Hyperfine Structure of Perturbed Series, 813

Jones (Grinnell), and S. K. Talley, Viscosity of Aqueous

Solutions, 475 Jones (Dr. H. Spencer), elected an honorary fellow of

Jesus College, Cambridge, 213
Jones (J. R.), [H. J. Allcock and], The Nomogram: the Theory and Practical Construction of Computation Charts (Review), 785

Jones (J. Stinton), appointed consultant engineer for heating, etc., in connexion with the new buildings of London University, 408

Jones (Phyllis), [A. J. Bradley and], X-Ray Investigation of the Copper-Aluminium Alloys, 589

Jones (Sir Robert), [death], 86

Jones (Tudor), The Neuro-Muscular Junction and Curare,

Jones (W. J.), Recent Developments on Electric Lighting, 302

Joos (Prof. G.), Lehrbuch der theoretischen Physik (Review), 221 Jouast (R.), [N. Stoyko and], Apparent Velocity of Short

Radio-Electric Waves, 887

Jouquet (Prof. E.), The Work of the late Prof. A. C. E. Rateau, 650

Juliard (A.), Retarding Action of Glass on Landolt's Reaction, 338

Junquera (M.), Combined Influence of pH and Glucose on the Permeability of Yeast to Methylene Blue, 411; [F. Chodat and], Reduction of Methylene Blue by an Endomyces at the Expense of its Endocellular Hydrogen Donators, 815

Kahane (E.), [L. Lematte and], Silica in the Organism and the Siliceous Particles of the Blood, 523

Kahn (R. F.), appointed university lecturer in economics and politics in Cambridge University, 848

Kailan (A.), and O. Stuber, Velocity of Catalysed Hydrogenations, 251

Kamienski (Prof.), Photoconductive Effect in Carborundum Crystals, 475

Kanda (Dr. S.), Comets of A.D. 868 and 1366, 31

Kanga (Prof. D. D.), Is Man Ethically Fit for the Bounties of Science? 233

Karl (A.), Analysis of Primary Radioactive Minerals, 143; Preparation of the Metallic Tungstates, 923

Karrer (Prof. P.), H. Salomon and K. Schöpp, Constitution of Dehydro-Ascorbic Acid, 800

Katino (S.), [K. Oguma and], Chromosome Numbers in

Vertebrates, 403 Kautter (Dr. C. T.), Origin of the Chile Saltpetre, 556 Kay (Prof. H. D.), and Dr. B. L. Guyatt, Experimental Rickets as a Phosphorus Deficiency Disease, 468

Kaye (Dr. G. W. C.), and G. E. Bell, Measurement of X-Ray Tube Current and Voltage, 552; and G. C. Sherratt, Velocity of Sound in Gases in Tubes, 338 Kayser (Prof. H.), und Prof. H. Konen, Handbuch der

Spectroscopie. Band 8, Lief. 1 (Review), 824 Keeler (C. E.), [W. E. Castle and], Blood Group Inheritance in the Rabbit; Tests for Linkage between the Blood-Group Genes and other known Genes of the

Rabbit, 775 Keeley (T. C.), [Prof. F. A. Lindemann and], Helium Liquefaction Plant at the Clarendon Laboratory, Oxford, 191

Keesom (Prof.), Low Temperature Research, 768

Keggin (J. F.), Structure of the Molecule of 12-Phosphotungstic Acid, 908 Kellogg (E. S.), The California Ground Squirrel Control

Program, 721

Kemp (G. S.), [death], 50 Kemp (Dr. S.), [J. M. Wordie and], Ice in the Weddell Sea, 916

Kemula (W.), and S. Mrazek, Measurements of the Absorption of Ultra-Violet Rays by Methane, Ethane and Normal Butane in the Gaseous condition, 106

Kendall (Prof. J.), W. W. Smith and T. Tait, Calcium Isotope with Mass 41 and the Radioactive Halfperiod of Potassium, 688

Kendall (Prof. P. F.), and Prof. H. Briggs, Formation of Rock Joints and the Cleat of Coal, 922

Kennaway (N. M.), [H. Burrows, Prof. E. C. Dodds and], Some Effects Observed in Mice under continued Treatment with Œstrin, 801 Kennedy (J. M.), and Miss D. M. Noakes, Cost of Elec-

tricity Supply and Distribution, 358 Kennedy and Thorndike, Relativity of Time, 136

Kennelly (Prof. A. E.), Conference of the Symbols, Units and Nomenclature, etc., 775; elected president of the International Scientific Radio Union, 55

Kerr (T.), The Pituitary in Lepidosiren and its Development, 523

Kershaw (J. W.), resignation of the lectureship in mechanical engineering in Sheffield University, 884 Keynes (J. M.), The Means to Prosperity (Review), 451

Khan (M. A. R.), Disintegrating Action of Roots of Trees, 844

Khouvine (Mme. Y.), and G. Nitzberg, Identification and Biochemical Oxidation of α-Glucoheptulite, 339

Kikoin (Dr. I.), and Dr. M. Noskov, A New Type of Photoelectric Effect in Cuprous Oxide in a Magnetic Field, 725

King (A. J.), [B. G. Churcher and], Scales of Loudness, 760

King (G. W.), [Dr. L. Harris, W. S. Benedict and], Form and Vibrational Frequencies of the NO2 Molecule, 621 King (J. J. F.-X.), [obituary], 298

King (L. A. L.), and Miss Agnes A. Meikle, A Fly Pest of Timothy Grass, 837

King (R. O.), Oxidation and the Lubricating Properties of Oil, 476

King (W. W.), Downtonian and Dittonian of Great Britain and North-West Europe, 738

Kinnersley (H. W.), [R. D. Heard, J. R. O'Brien, Prof. R. A. Peters, V. Reader and], Vitamin B₄ and Adenine, 617

Kinoshita (M.), and K. Uchiyama, Size of Fog Droplets,

Kirkbride and Norrish, Photochemical Decomposition of

Diazomethane, 404
Kirsch (G.), and R. Trattner, Atomic Disintegration with

Emission of Neutrons, 852

Kisser (J.), and R. Furtauer, Influence of certain Chemical Agents on the Carbon Dioxide Output of Germinating Seeds of Pisum sativum and Triticum vulgare, 251 and J. Schubert, Influence of Treatment of Seeds with Chemical Stimulants on the Cell Growth of the Rootlets, 251

Klatzow (L.), [Dr. S. P. McCallum and], Conductivity of Mixtures of Gases, 841

Kleeman (Prof. R. D.), The Atomic and Molecular Forces of Chemical and Physical Interaction in Liquids and Gases, and their Effects (Review), 80

Klemenz (K.), Archimedean Bodies in Spaces of Several Dimensions, 703

Kling (A.), and A. Soulier, Accidental Ignition of Vapours of Petrols by Electric Sparks, 106 Klingenheim (A.), Vai Script, 915

Klima (J.), Chemistry of the Lichens (2), 888 Knight (W. F. J.), Symbolism of the Maze, 98 Knox (E. G. V.), elected a member of the Athenæum Club, 200

Knox-Shaw (Dr. H.), The Extra-Galactic Nebulæ, 247 Knudsen (Prof. V. O.), Architectural Acoustics (Review),

Koblmuller (L. O.), and R. Vierthaler, Transferring Single-Cell Cultures to Solid Nutrient Media, 287

Kodaira (T.), [I. Imamura, H. Imamura and], Earthquake Series at Nagusa, Japan, 136

Koeing (Friedrich), Centenary of the death of, 51 Koller (C.), and Thelma Townson, Spermatogenesis in

Drosophila obscura, Fallen (1), 447; and G. Pfeiffer, Constitution of Pinastric Acid; Galabratic Acid, 411 Koltzoff (Prof. N. K.), and V. N. Schröder, Artificial

Control of Sex in the Progeny of Mammals, 329 Konen (Prof. H.), [Prof. H. Kayser und], Handbuch der

Spectroscopie. Band 8, Lief. 1 (Review), 824 Kopper (H.), and A. Pongratz, Raman Effect (24),

251 Koshy (Prof. T. K.), Structure and Division of Somatic

Chromosomes in Allium, 362 Kraft (K.), [Dr. F. Micheel and], Constitution of Vitamin

C, 274 Kramer (Mlle. A.), The Heterosides of Philyrea latifolia,

631

Kramers (Prof. H. A.), [Prof. W. J. de Haas, E. C. Wiersma and], Temperature below 0.27° K., 719

Kremann (R.). and L. Lammermayer, Electrolysis of Aluminium Alloys containing Iron as a Model of the Electrolytic Purification of Molten Aluminium from Iron, 251

Krishna (S.), and S. Ramaswami, Calorific Values of some Indian Woods, 281

Krishnan (K. S.), elected secretary of the Indian Association for the Cultivation of Science; appointed Mahendralal Sircar research professor in physics of the Indian Association, 907; and S. Banerjee, Orientations of Molecules in the p-Benzoquinone Crystal, 653; and S. M. Mitra, Negative Polarisation in Fluorescence, 204

Kronman (Dr. E. S.), Rhenium (Opyt monografii elementa

No. 75), (Review), 224

Kropp (B.), The Crustacean Chromatophore Activator and

the Gonads of the Rat, 631

Kuhn (Dr. W.), Fundamental Laws of Optical Rotatory Power, 771; und K. Freudenberg, Drehung der Polarisationsebene des Lichtes (Review), 677

Kukarkin (Dr.), Nova of March 20, 590 Kullenberg (B.), [T. Gustafson and], Inertia Currents in the Baltic, 586; [Dr. H. Pettersson and], Boundary Tides in the Kattegat, 586 Kunhardt (Lieut.-Col. J. C. G.), [death], 231

Kürti (G.), Magneto-Rotation in Coloured Glass and Rock

Salt, 411

Kutzelnigg [Beutel and], Fluorescence of Zinc Oxide, 917 Kuznetzov (B. A.), Rodents of the Semipalatin District of Kazakstan, 516

L. (W. W.), Ideas of 'Time' and 'Events', 727 Labarthe (A.), and M. Demontvignier, Measuring and Recording Rapidly Variable Pressures, 179

Laboccetta (L.), Effective Integration of Discontinuous Functions (2), 71

Lacey (S.), Development of the Use of Gas, 478

Lagatu (H.), and L. Maume, Comparative Composition, in the Vine, of Homologous Leaves, etc., 815

Lagotala (H.), Borings in the Marly Limestones of Renéville (French Congo); The Stratigraphic Scale of the Niari Limestones, 215; Continental Formation Subjacent to the Limestones of Niari, French Congo, 851; Geology of the Comba Region, 411; Stratigraphical Study of the Mindouli-Mines Region, 144

Laigret (J.), [C. Nicolle, P. Giroud and], Passage of the Virus of Exanthematic Fevers by the Digestive Canal

in the Rat, 375

Lainé (P.), Magnetic Double Refraction of Liquid Oxygen, 850; Magnetic Properties of Liquid Ozone, 702

Laing (Dr. B. M.), David Hume (Review), 321

Laing (Dr. E. V.), Tree Roots, 176-Lake (P.), A Photogrammetric Survey in the Pamir (Review), 744

Lalande (A.), Freezing Points of Binary Mixtures of Ethyl Alcohol and Ethyl Ether, 447

Lallemand (C.), The World Crisis and the Gold Standard, 70

Lamb (Sir Horace), Hydrodynamics. Sixth edition (Review), 313

Lamb (J. H.), Electrification of an Ice Factory, 833

Lammermayer (L.), [R. Kremann and], Electrolysis of Aluminium Alloys containing Iron as a Model of the Electrolytic Purification of Molten Aluminium from Iron, 251

Lancaster-Jones (E.), awarded the B.O.I.M.A. prize of

the Institute of Physics, 467

Lang (K. C.), [L. H. Martin and], Thermal Conductivity of Water, 813

Langevin (A.), [L. Bugnard, P. Gley and], Recording and Measurement of the Blood Pressure, 375

Lapage (G.), Cultivation of Parasitic Nematodes, 583 Lapp (C.), Rotatory Power of Quinine Salts in Aqueous

Solution, 703 Larke (Sir William), The Iron and Steel Industry, 335

(Sir Joseph), The Astronomical Radiative Larmor Stability, 805

de Laszlo (Dr. H.), Electron Diffraction by Vapours, 803 Latimer [Libby and], Radioactivity of some Rare Earth Elements, 368

Latreille (Pierre André), centenary of the death of, 159

de Laubenfels (M.), Californian Sponges, 588

Laure (Y.), Combustion Pressures in Closed Vessels of

Air-Benzene Mixtures, 447 Leak (H.), and T. Priday, Migration from and to Great

Britain, 125

Leakey (Dr. L. S. B.), Fossiliferous Deposits in the Homa-Kendu Area, Kenya Colony, 886; Prof. H. Reck, Prof. P. G. H. Boswell, A. T. Hopwood and Dr. J. D. Solomon, The Oldoway Human Skeleton, 397 Lecomte (J.), Infra-Red Absorption Spectra of some

Halogen Derivatives of Methane, 738

Lecoq (R.), Rôle of the B Vitamins in the Utilisation of the Glycides by the Organism of the Pigeon, 630

Lee (C. E.), Hancock's Steam Omnibus, 647

Lee (Dr. K.), Science and the Textile Industry, 162 Lees (A.), Wave Equations and the Conservation of Energy, 402 Leeson (H. S.), and K. Mellanby, Insects and Micro-

Climates, 363 Leeuw (Dr. J. J. van der), Inadequacy of Economic

Sanctions, 867 Le Fèvre (Dr. R. J. W.), Volumes of Alkyl Groups and their Orienting Powers, 655

Legendre (Adrien Marie), Centenary of the death of, 18

Leicester (P.), Geology of Rangoon, 916

Leigh (Col. L.), Railway Electrification Experience, 649 Lematte (L.), and E. Kahane, Silica in the Organism and the Siliceous Particles of the Blood, 523

Leon (A. L.), bequest to London University, 408

Leonard (Prof. A. G.), [death], 231 Leprince-Ringuet (L.), [J. J. Trillat and], Molecular Phenomena at the Surface of Separation of Oil and Water, 850

Leroux (L.), [R. Cambier and], Estimation of Organic Nitrogen in the Presence of Nitrates by Kjeldahl's Method, 179

Le Roux (P.), Pleochroism of Iceland Spar in the Infra-Red Spectrum, 447 Lespieau (R.), and Mlle. B. Gredy, Study of some α -

Ethylene Oxides, 447

Lestoquard (F.), [E. Sergent, A. Donatien, L. Parrot and], Experimental Suppression of Sexual Reproduction in Theileria dispar, 143

Levaillant (R.), Action of Acid Chlorides on Orthoformic

Esters, 36

Lever (R. J. A. W.), Early History of the British Solomon Islands, 587

Leverhulme (Viscount), elected president of the Institution of Chemical Engineers, 510

Levi (G. R.), and D. Ghiron, Magnesium Chlorite and Double Chlorites of Copper with Magnesium, Barium, and Thallium, 775 Levi-Civita (Prof. T.), Caratteristiche dei sistemi differ-

enziali e propagazione ondosa. Lezioni raccolte dal Dr. G. Lampariello (Review), 45

Levy (Prof. H.), The Irresponsibility of Science, 162

Lewis (G. N.), Hydrogen Isotope, 590 Lewis (Sir Thomas), appointed a member of the Medical Research Council, 581

Lewis (W. V.), appointed university demonstrator in geography in Cambridge University, 628

Liau (Ssu-Pin), [J. Dufay and], Absorption of Light in Interstellar Space, 923

Libby and Latimer, Radioactivity of Some Rare Earth Elements, 368

Libermann (D.), P. Carré and], Mechanism of the Reaction of Phosphorus Pentachloride on Neutral Alkyl Sulphites, 143; Reaction of Phosphorus Pentachloride on the Neutral Aryl Sulphites, 667

Lien-Teh (Dr. Wu), [Dr. K. Chimin Wong and], History of Chinese Medicine: being a Chronicle of Medical Happenings in China from Ancient Times to the Present Period (Review), 527

Lighthall (W. D.), Plan of Hochelaga, 64 Lindemann (Prof. F. A.), and T. C. Keeley, Helium Liquefaction Plant at the Clarendon Laboratory, Oxford, 191

Linehan (P. A.), and Prof. S. P. Mercer, Fluorescence in Rye-grasses, 474; Fluorescence of Lolium Seedlings in Ultra-Violet Light, 202 .

Ling-Chao (Chien), [Ny Tsi-Ze and], Influence of Pressure on the Photographic Sensibility to various Monochromatic radiations, 286

Lingood (F. L.), [S. G. Paine, Freda Schimmer, T. C. Thrupp and], Relationship of Micro-Organisms to the

Decay of Stone, 178 Link (F.), The Partial Eclipse of the Moon on Sept. 14, 1932, 179

Linton (R.), Social Organisation and Marriage in Madagascar, 843

Lipschütz (A.), Transplantation of Ovaries preserved outside the Organism, 143

Lipscomb (A. G.), Cellulose Acetate: its Manufacture and Applications (Review), 454

Littauer (S. B.), [M. Morse and], Fields in the Calculus of Variations, 631

Little (J. E.), gift to Cambridge University, 700 Lock (G.), Cannizzaro's Reaction (2), 560

Lockyer (Dr. W. J. S.), Spectrum of γ Cassiopeiæ, 134 Lods (Prof. A.), translated by Prof. H. S. Hooke, Israel:

from the Beginnings to the Middle of the Eighth Century (Review), 315

Loeb (E. M.), Patwin and Maidu Cult Origins, 659 Loewy (Maurice), centenary of the birth of, 497

Lombard (A.), The Virgulian and the Stratigraphy of the Portlandian of the Col du Marchairuz Region,

Longfield (T. E.), Subsidence of London, 558 Longley (Dr. W. H.), Taxonomy and Evolution, 863 Lönnqvist (C.), Frequency of the Nova Phenomenon, 64 Loofbourow (J. R.), [F. F. Heyroth and], Irradiation of Nucleic Acids and Uracil, 92

Loomis (Dr. A. L.), [Dr. H. T. Stetson and], Tidal Shifts in the Earth's Crust, 137

Loonam (A. C.), [G. H. Gleason and], Sewage Purification by a New Process, 697

Loose (L.), and Dr. W. H. Pearsall, Synthesis of Protein

by Green Plants, 362 Loughnane (J. B.), Insect Transmission of Virus A of Potatoes, 838

Lovat (Lord), [death], 266; [obituary article], 352 Lowe (W. P.), The Trail that is always New (Review), 636

Lowndes (A. G.), Sexual Reproduction in Copepods, 240

Lowry (H. V.), [Dr. J. Prescott and], Elementary Trigonometry (Review), 420

Lowry (Prof. T. M.), Modern Aspects of Chemistry in Space (Review), 563; and H. Hudson, Optical Rotatory Power (4), 374

Lu [A. Travers and], Separation of Phosphoric, Arsenic and Vanadic Acids from Aluminium, 595; Volumetric Determination of Lead, 523

Lucas (A.), Forensic Chemistry and Scientific Criminal Investigation. Second edition (Review), 115 Lucas (R.), and M. Schwob, Anomalous Dispersion in

Magnetic and Electric Double Refraction, 630 Ludford (R. J.), Differences in the Growth of Transplant-

able Tumours in Plasma and Serum Culture Media,

Ludlam [Ritchie and], A Gaseous Oxidation, 64 Lugard (Lord), elected president of the Institut Colonial International, 683

Lukacs (L.), and J. Zellner, Chemistry of the Higher Fungi (22), 888

Lundmark (Dr. K.), Mass of Eros, 31

Luthi (R.), [Prof. J. Weigle and], Debye's Dispersion of Nitrobenzene, 327

Luyten (W. J.), Faint Stars with Common Proper Motion,

Lydall (F.), Electrification of Railways in Britain, 19 Lynch (Col. A.), The Case Against Einstein (Review), 260; 622

Lyons (Sir Henry), elected president of the Institute of Physics, 797

Lyot (B.), Direct Observation of Solar Prominences at Meudon, 70; 332

Lythgoe (R. J.), and Dorothy E. Corkill, Measurement of Visual Acuity, 98

MacAdam (Dr. W.), elected professor of clinical medicine in Leeds University, 772

McAlpine and Smyth, Polarity of Hydrocarbon Vapours, 517

MacBride (Prof. E. W.), appointed chairman of the Advisory Committee on Fishery Research of the Development Commissioners, 871

McCallien (W. J.), Preservation of Fossil Bones, 694 McCallum (Dr. S. P.), and L. Klatzow, Conductivity of

Mixtures of Gases, 841 McCarthy (Isabella M.), [J. Algar, Eveline M. Dick and], Synthesis of Diflavones, 666

McCoy (Elizabeth), [E. B. Fred, I. L. Baldwin and], Root Nodule Bacteria and Leguminous Plants, 99

McClintock (Barbara), A Correlation of Ring-shaped Chromosomes with Variegation in Zea mays, 631 Maccoll [Taylor and], Motion of a Cone at High Speeds in Air, 552

McConnell (C. H.), Nerve-Net of Hydra, 207 MacCulloch (Rev. Canon J. A.), Medieval Faith and Fable (Review), 80

McCurdy (N. R.), [R. A. Watson and], Cyclone Season in the South Indian Ocean, 517; Pilot Balloon Observations at Mauritius, 626

Macdougall (A. J.), appointed assistant lecturer in metallurgy in Sheffield University, 736

Mace (C. A.), appointed reader in psychology at Bedford College, London, 772

Macelwane (Rev. J. B.), and Rev. F. W. Sohon, Introduction to Theoretical Seismology. Part 2: Seismometry, by Rev. F. W. Sohon (Review), 824

MacGregor (Dr. M. E.), [death], 86; [obituary article], 123

Machebeuf (M. A.), [J. Basset and], Biological Effects of Ultra-Pressures, 251; [Mme. E. Woolman, M. Bardach and], Biological Effects of Ultra-Pressures,

Machek (G.), Action of Gaseous Cyanogen on Phenols

and Naphthols (2 and 3), 851 McKenzie (D. B.), [T. E. Allibone, F. S. Edwards and], A New Impulse Generator for Three Million Volts, 129 Mackenzie (J. E.), and H. W. Melville, Measurement of the Diffusion Coefficients of Bromine-Hydrogen and

Bromine-Carbon Dioxide, 250

Mackereth (J.), [death], 718 Mackie (H. B.), Principles of Pharmacy (*Review*), 895 Maclaren (N. H. W.), and Prof. T. H. Bryce, Early Stages in the Development of Cavia, 922

McLaughlin (D. B.), Suggested Mechanism of Class Be Stars, 739

McLennan (Prof. J. C.), and R. Turnbull, Ultra-Violet Absorption Bands of Xenon, 214 Maclagan (D. S.), An Ecological Study of the Lucerne

'Flea', 556

McNaughtan (F.), [V. A. Beckley and], Distribution of Nitrates in the Soil and Root Development in Coffee, 878

McPetrie (J. S.), Production of Electronic Oscillations with a Two-Electrode Valve, 691

Macrae (A.), Vocational Guidance and the Health of the Industrial Worker, 243

Macrae (T. F.), [Dr. J. M. Gulland and], Action of Proteolytic Enzymes on the Oxytocic Principle of the Pituitary Gland, 470 McVittie (Dr. G. C.), Milne's Theory of the Expansion of

the Universe, 533

McWhirter (W.), [obituary article], 427 Maddocks (W. R.), appointed lecturer in metallurgy in Sheffield University, 736 Magnan (A. and C.), Hot Wire Apparatus for the Study

of the Air Movements produced by the Flapping Wing of a Bird or Insect, 923

Magnan (Mlle. C.), [H. Bierry, B. Gouzon and], Application of the Iodometric Method to the Estimation of Sugar in the Blood, 667

Mahanti (P. C.), Band Spectra of Barium Oxide (BaO), 402

Mains (E. B.), Host Specialisation of Erysiphe graminis Tritici, 739

Maire (R.), E. Foëx and G. Malençon, Etiology of 'Bayoud', a Disease of the Date Palm, 922 Maulik (S. A.), Early Stages of Hispine Beetles, 171 Maurain (C.), The Earthquake of March 2, 1933, 596

Maiyabe (Prof. N.), [Prof. N. Nasu and], Landslide at Toge, Japan, 99

Major (Prof. R. H.), Classical Descriptions of Disease: with Biographical sketches of the Authors (Review), 895

Majorana (Q.), New Photoelectric Phenomenon Exhibited by Metallic Sheets, 559; Action of a Periodic Light Beam on Metallic Sheets, 447; A New Photoelectric Experiment, 71; A New Photoelectric Experiment, 107
Malençon (G.), [R. Maire, E. Foëx and], Etiology of
'Bayoud', a Disease of the Date Palm, 922

Malins (Majorie E.), [Prof. J. H. Priestley, Lorna I.

Scott and], Cambial Activity, 375

Malisoff (W. M.), Meet the Sciences (Review), 639 de Mallemann (R.), and P. Gabiano, Magnetic rotatary Power of Chlorine and of Hydrochloric Acid Gas, 286 Mallock (R. M.), An Electrical Calculating Machine, 880

Mankin (Winifred R.), Application of Optical Spectroscopy to Analysis of Tumour Tissue, 668 Mann, An Optically Active Inorganic Salt, 808

Mann (L. McL.), Preservation of Fossil Bones, 366 Marconi (Marchese), Radio Communications by Very Short Electric Waves, 292

Marinesco (N.), Preparation of Colloids by Ultra-Sonic Dispersion, 410

Mark (Prof. H.), Structure of Cellulose, 591 Markham (S. F.), [Sir Henry Miers and], The Museums and Art Galleries of British Africa, 430; Report on the Museums of Canada; Directory of Museums and Art Galleries in Canada, Newfoundland, Bermuda, British West Indies, British Guiana and the Falkland Islands, 84

Marinesco (N.), and J. J. Trillat, Action of Ultra-Sounds on Photographic Plates, 667

Marmite Food Extract Co., Ltd, The Medicinal and Dietetic Value of Marmite, 616

Marett (Dr. R. R.), Faith, Hope and Charity in Primitive

Religion (Review), 9

Marsh (M. C.), Transmission of Heat through Fabrics, 558 Marshall (A.), Chemical Examination of Explosives (Review), 186; Explosives: their History, Manufacture, Properties and Tests. Second edition. Vol 3 (Review), 452

Marshall (C. F. D.), Hancock's Steam Omnibus, 647 Marshall (Sir John), awarded the gold medal of the Royal

Asiatic Society of Bombay, 683

Marshall (J. F.), and J. Staley, A New British Record of Orthopodomyia pulchripalpis, Rondani (Diptera, Culicidæ), 435

Marshall (W. B.), and E. O. Bowles, New Fossil Fresh Water Molluses from Ecuador, 589

Martin (H.), Present Uses and Future Development of Spray Spreaders, 768

Martin (Dr. A. R.), Rôle of the Solvent in Electrolytic Dissociation, 584

Martin (L. H.), and K. C. Lang, Thermal Conductivity of Water, 813

Martindale (Dr. W. H.), The Extra Pharmacopæia of Martindale and Westcott. Twentieth edition. In 2 vols. Vol. 1: (Review), 6

Marvin (Prof. F. S.), A Symposium on Science (Review), 674

Mason (T. N.), and F. V. Tideswell, Gob Fires, 521 Massee (A. M.), Some Injurious and Beneficial Mites on Top and Soft Fruits, 136

Massey and Mohr, Electron Scattering by Atoms, 368

Mathias (E.), Study of Fulminating Material, 482 Mathieu (J. P.), Compounds of Tartaric Acid and Chromium, 106

Matignon (C.), H. Moureu and M. Dodé, Causes of the Simultaneous Production of 1-butene and 2-butene, 738; and J. Calvet, Ageing of the Aluminium-Beryllium Alloys after Tempering, 886; and M. Séon, Action of Steam on Hexane, 286; Action of Steam on Heavy Petroleum Oils and on certain Cyclic Hydrocarbons, 523; Action of Steam on Methane, 215

Matuyama (E.), Band Spectra which Appear near Visible Triplet Lines of Mercury, 58

Maulik (S.), Antennal Secretion in Insecta, 516

Maume (L.), [H. Lagatu and], Comparative Composition, in the Vine, of Homologous Leaves, etc., 815 Maunder (Mrs. A. S. D.), The Sothic Cycle, 332

Maurain (C.), and J. Devaux, Electrical Conductivity and Atmospheric Condensation Nuclei during a Voyage to Greenland, 35; Total Calorific Radiation in Green-

Mawson (Dr. C. A.), appointed for biochemical investigations at the Research Institute of the Cancer Hos-

pital, 500

Mawson (Sir Douglas), Geology and Glaciation of some Islands of the Southern Ocean, etc., 847

Maxwell (Sir George), Case for a Colonial Educational Conference, 812

Maxwell, Bt. (Sir Herbert), The Borrowed Days, 515 Maxwell (E. A.), awarded a Smith's prize of Cambridge University, 408

Mayr (K.), Definite Integrals with Bessel's Functions, 703 Mead (A.), [Dr. C. H. Johnson and], Line Absorption of Chromic Salts in Relation to Co-ordination, 399 Mead (Dr. Margaret), The Changing Culture of an Indian

Tribe (Review), 711

Megaw (E. C. S.), Ultra-Short-Wave Radio Research, 269 Mégroz (R. L.), Ronald Ross: Discoverer and Creator (Review), 40

Meikle (Miss Agnes A.), [L. A. L. King and], A Fly Pest of Timothy Grass, 837

Meksyn (Dr. D.), Neutrons, 366

Melchett (Lord), acceptance of presidency of the British Science Guild, 798
Meldrum (Dr. A. N.), Priestley as a Practical Chemist,

801; the Scientific Work of Priestley, 555 Meldrum (late Dr. N. U.), and Dr. F. J. W. Roughton,

Carbonic Anhydrase and the State of Carbon Dioxide in Blood, 874 Mellanby (Prof. E.), resignation of the chair of pharmaco-

logy in Sheffield University, 884

Mellanby (K.), [H. S. Leeson and], Insects and Micro-Climates, 363 Climates,

Melland (F. H.), Witchcraft in Africa, 195

Mellor (D. P.), [F. P. Dwyer and], Occurrence of β-Cristobalite in Australian Opals, 524

Mellor (Dr. J. W.), A Comprehensive Treatise on Inorganic and Theoretical Chemistry. Vol. 12 (Review), 638
Melton (Prof. F. A.), and W. Schriever, Meteor Craters,
100; Probable Meteorite Scars in Carolina, 624

Melville (H. W.), and H. L. Roxburgh, Upper Limit in Explosive Chain Reactions, 690; [J. E. Mackenzie and], Measurement of the Diffusion Coefficients of Bromine-Hydrogen and Bromine-Carbon Dioxide,

Mémery (H.), L'Influence Solaire et les Progrès de la Météorologie, 591; Remarkable Epoch of Solar Activity, 559

Menzel (D. H.), a Simple Derivation of the Dissociation Formula, 739

Mercier (F.), [L. J. Mercier and], New Method of Preparing Marrubiine, 143

Mercer [Lineham and], Fluorescence in Rye-grasses, 474 Mercer (Prof. S. P.), [P. A. Lineham and], Fluoresence of Lolium Seedlings in Ultra-Violet Light, 202

Merck (Dr. W.), [obituary article], 298

Mercier [L. J.), and F. Mercier, New Method of Preparing Marrubiine, 143

Mercier (R.), Paramagnetism of the Ion of Dissolved Cobalt, 338

Merrill (Dr. E. D.), elected a foreign member of the Linnean Society of London, 759

Merrill (Dr. P. W.), Radial Velocities of Variable Stars, 917

van der Merwe (D. S.), Vanished Races in South Africa,

Merwin (H. E.), [J. W. Greig, E. S. Shepherd and], Volatile Transport of Silica, 768

Messer (M.), An Agricultural Atlas of England and Wales. Second edition (Review), 420

Metropolitan-Vickers Electrical Co., Ltd., Research Work of the, 126; 649

Metz (K.), A Fauna from the Hochwipfel Strata of the Carinthian Alps; the Nassfeld Strata from Schulterkofel Westwards, 287

Meyer (A.), and M. Tuot, Dehydration of some Tertiary Alcohols by Anhydrous Copper Sulphate, 851

Meyer (F.), [E. Cherbuliez and], New Researches on Casein, 215

Meyer (M.), [G. Darzens and], New General Method for

the Synthesis of Aldehydes, 483
Mezincesco (P.), [E. F. Terroine, Mlle. Simone Valla and],
Utilisation of Sulphur and Nitrogen from Cystine

at the Level of Endogenous Protein Metabolism, 483

Mezzadroli (G.), and A. Amati, Action of certain Alkaloids on the Development of Aspergillus niger, 339 Micheel (Dr. F.), and K. Kraft, Constitution of Vitamin C,

Michels (Dr. W. C.), [Dr. W. R. Smythe and], Advanced Electrical Measurements (Review), 322

Middleton (A.), [W. E. Williams and], Fine Structure of

the Resonance Ag I Lines, 692 Middleton (A. D.), The Grey Squirrel (Review), 45 Miers (Sir Henry), and S. F. Markham, The Museums and Art Galleries of British Africa, 430; Report on the Museums of Canada; Directory of Museums and Art Galleries in Canada, Newfoundland, Bermuda, British West Indies, British Guiana, and the Falk-

land Islands, 84
Mihul (C.), [T. V. Ionescu and], Ionised Gases in the

Magnetic Field, 106
Mihul (Mme. Irène), [T. V. Ionescu and], High-Frequency Discharge in Gases, 887

Mikkelsen (Capt. E.), Spread of Eskimo in Greenland, 367 Miles (Dr. G. H.), [H. J. Welch and], Industrial Psychology in Practice (Review), 567; The Problem of Incentives in Industry (Review), 321

Mill (Dr. H. R.), An Approach to Geography, 68 Milford (M.), [A. Egerton and], Fusion of Carbon, 169 Miller (Dr. D.), Chilean Insect Parasites for New Zealand,

Miller (Dr. J. L.), and J. E. L. Robinson, awarded the Institute of Physics prize, 467

Millis (C. T.), Education for Trades and Industries: a Historical Survey (Review), 44

Mitchell (A. Crichton), Diurnal Incidence of Disturbance in the Terrestrial Magnetic Field, 446

Mitchell (Dr. A. D.), and Dr. A. M. Ward, Modern Methods

in Quantitative Chemical Analysis (Review), 531

Mitchell (Dr. C. A.), [Dr. B. Dyer and], The Society of Public Analysts and other Analytical Chemists (Review), 822

Mitchell (J. H.), Pit-Head Baths, 207

Mitolo (M.), Avitaminosis and Intoxication (1), 411; (2), 284

Mitra (Kalipada), Mother Goddess in India, 207

Mitra (Prof. S. K.), and H. Rakshit, Recording Wireless Echoes at the Transmitting Station, 657

Mitra (S. M.), [K. S. Krishnan and], Negative Polarisa-

tion in Fluorescence, 204 Mittasch (Dr. A.), und Dr. E. Theis, Von Davy und Döbereiner bis Deacon: ein halbes Jahrhundert Grenzflächenkatalyse (Review), 150

Modi (Sir Jivanji Jamshedji), [obituary article], 645 Moelwyn-Hughes (E. A.), [C. N. Hinshelwood, A. C. Rolfe and], Combination of Hydrogen and Oxygen in a Silver Vessel, 142 Mohr (C. B. O.), and F. H. Nicoll, Inelastic Electron

Scattering in Gases; Large Angle Scattering of Electrons in Gases, 143; [Massey and], Electron Scattering by Atoms, 368

Mohr (Dr. O. L.), Woolly Hair in a Nordic Pedigree, 695

Moir, Bt. (Sir Ernest), [obituary article], 901 Molliard (M.), Attenuation of the Chlorophyll shown by Parasitic Green Plants, 178

Moll (Dr. W. J. H.), [Dr. L. S. Ornstein, Dr. H. C. Burger and], Objektive Spektralphotometrie (Review), 824

Monchal (Mlle. S.), [J. Beauverie and], Life of Green Plants in a Confined Atmosphere, 36

Mondain-Monval (P.), and R. Wellard, Direct Oxidation

of Acetylene by Air, 851 Monod (J.), Axial Gradient in the Ciliated Infusoria, 339

Monier, Jr. (J. A.), [C. Dufraisse and], Dissociable Organic

Monier-Williams (G. W.), Freezing-Point of Milk, 702 Monnier (R.), [G. Gutzeit and], Utilisation of some Azo Derivatives of Oxyquinoline as Reagents in Qualitative Analysis, 144

Monod (Dr. T.), Saharan Rock-Engravings, 551

Moody (A. F.), Water-Fowl and Game-Birds in Captivity: some Notes on Habits and Management (Review), 316

Moon (C. H.), [Prof. W. E. Garner and], Acceleration of the Decomposition of Crystals of Barium Azide by the Emission from Radium Emanation, 513

Moore (Prof. E. H.), [death], 195
Moore (Prof. J. P.), Land Leeches, 63
Moore (T.), [Dr. P. Bowden and], Absorption Spectrum
of the Vitamin E Fraction of Wheat-Germ Oil, 512

Moppett (W.), Experiments in which the Allantoic Membrane of the Chick is Exposed to X-Radiation, 667

Morand (M.), and A. Hautot, Fine Structure of the Carbon Line K_a , 143

Morgan (Prof. T. H.), The Scientific Basis of Evolution (Review), 380

Morgan (William), Centenary of the death of, the work of, 612

Morita [Obata, Yosida and], Aeroplane Noises, 208 Morris (Sir Daniel), [death], 231; [obituary article], 266

Morris (S. D. D.), [Dr. F. P. Bowden, Dr. C. P. Snow and], Absorption Spectrum of Vitamin A at Low Temperatures, 582

Morrison (Prof. J. T. J.), [death], 718

Morse (M.), and S. B. Littauer, Fields in the Calculus of Variations, 631

Morton (Dr. R. A.), and J. R. Edisbury, Absorption Spectrum of the Unsaponifiable Matter from Wheat-Germ Oil, 618 Moshinsky (Pearl), [J. L. Gray and], Genetic Psychology

(1), 922

Mott [Ellis and], Internal Conversion of γ-Rays, 517 Mott [H. M. Taylor and], Internal Conversion of γ-Rays,

Mourant (A. E.), Dehydration of Thomsonite, 630 Moureu (H.), [C. Matignon, M. Dodé and], Causes of the Simultaneous Production of 1-butene and 2-butene, 738

Mrazek (S.), [W. Kemula and], Measurements of the Absorption of Ultra-Violet Rays by Methane, Ethane and Normal Butane in the Gaseous Condition, 106

Mukerji (B. L.), [Prof. N. R. Dhar, A. K. Bhattacharya and], Kinetics of the Iodine-Oxalate Reaction, 840 Müller (Dr. A.), Crystal Structure of the Normal Paraffins,

Muller and Desmarez, Differential Microscopic Characters of the Bone of Adult Cynocephalus and of Human

Bone, 739 Muller (J. A.), and Mlle. Eglantine Peytral, The Sudden Pyrogenation of Ketene, 375

Müller (W. J.), Theory of Passivity Phenomena (18), 888 Munn (L.), Prehistoric Gold-Mining in South India, 730 Munro (G. H.), and L. G. H. Huxley, Atmospherics in Australia, 592

Munro (Prof. J. W.), Infestation of Stored Products by

Insects, 82 Murison (C. A.), [Prof. G. P. Thomson and], Electron Diffraction by Films of Grease, 237; N. Stuart and], Crystalline State of Thin Spluttered Films of Platinum,

Murphy (Prof. P. A.), awarded the Boyle medal of the Royal Dublin Society, 683

Murray (Sir Hubert), and others, Depopulation in Papua, 519

Muskett (A. E.), [H. Cairns and], Phytophthora megasperma causing Pink Rot of the Potato, 277

Myers (Dr. C. S.), Business Rationalisation, its Dangers and Advantages considered from the Psychological and Social Standpoints (Review), 567; The Absurdity of any Mind-Body Relation, 579

Myrbach (Dr. O.), Wanderers Wetterbuch: Einführung in das Verständnis der Wettervorgänge (Review), 786

Nádai (Dr. A.), assisted by A. M. Wahl, Plasticity: a Mechanics of the Plastic State of Matter (Review),

Naismith (R.), [Prof. E. V. Appleton and], Kennelly-Heaviside Layer, 808; Weekly Measurements of

Upper Atmospheric Ionisation, 522

Nall (G. H.), The Life of the Sea Trout, especially in Scottish Waters: with Chapters on the Reading and Measuring of Scales (Review), 42

Nancarrow (H. A.), Determination of the Thermal Conductivities of Rocks, 702

Narlikar (Prof. V. V.), The Constant of Gravitation, G, 134

Nasu (Prof. N.), and Prof. N. Miyabe, Landslide at Toge, Japan, 99

Natta (G.), and R. Pirani, Solid Solutions by Precipitation, etc. (2), 107

Naudé (S. M.), and J. E. C. Coventry, Intensity of Cosmic

Radiation in South Africa, 411 Naumann (Prof. E.), Grundzüge der regionalen Limno-

logie (Review), 223

Navashin (M.), Origin of Spontaneous Mutations, 436 Neave (Dr. S. A.), assisted by F. J. Griffin, The History of the Entomological Society of London, 1833-1933 (Review), 678

Needham (Dr. N. J. T. M.), appointed Sir William Dunn reader in biochemistry in Cambridge University, 408 Nef (Prof. J. U.), The Rise of the British Coal Industry.

2 vols. (Review), 311 Negretti and Zambra, Humidity Tables, 589 Němec (Prof. B.), sixtieth birthday of, 355

Neville (G. H. J.), [Dr. R. G. W. Norrish and], Photosensitised Decomposition of Ozone by Chlorine,

Newbigin (Dr. Marion I.), Southern Europe: a Regional and Economic Geography of the Mediterranean Lands (Italy, Spain, Portugal, Greece, Albania and Switzerland, (Review), 317

Newitt (D. M.), and A. M. Bloch, Slow Combustion of Ethane at High Pressures, 214

Newman (Sir George), The Debt of Preventive Medicine to Harvey and the College of Physicians (Harveian Oration), 407

Newton (J. D.), [W. E. Bowser and], Effect of Weed-Killers

on the Soil, 880

Nicholetts (Flight-Lieut. G. E.), [Squadron-Leader O. R. Gayford and], New Record for Non-Stop Flight, 232; awarded the British silver medal of the Royal Aeronautical Society, 685 Nichols (H. W.), and P. C. Orr, Bakelite Impregnation

of Fossil Bones, 127

Nichols (Dr. J. E.), Origin of Curls and Twists in Wool

Fibres, 201

Nicoll (F. H.), [C. B. O. Mohr and], Inelastic Electron Scattering in Gases; Large Angle Scattering of Electrons in Gases, 143

Nicolle (Dr. C.), 'Diffusion' and Disease, 879; J. Laigret and P. Giroud, Passage of the Virus of Exanthematic Fevers by the Digestive Canal in the Rat, 375

Niebuhr (Carsten), bicentenary of the birth of, 354

Nielsen (E.), The Biology of Spiders: with especial Reference to the Danish Fauna. 2 vols. (Review), 149

Nierenstein (Dr. M.), Incunabula of Tannin Chemistry (Review), 823; and P. F. Chapman, The Authorship of the "Ordinall of Alchimy", 520

Niggli (Prof. P.), The Process of Metamorphism in Rocks

(Review), 310 Nightingale (E.), [F. Fairbrother and], General Science. Part 1 (Review), 420

Nisbet (R. H.), Fourier Analysis and Vowel Curves, 401

Nitzberg (G.), [Mme. Y. Khouvine and], Identification and Biochemical Oxidation of a-Glucoheptulite, 339 Noakes (Miss D. M.), [J. M. Kennedy and], Costs of Electricity Supply and Distribution, 358

Nolan (J. J.), and J. G. O'Keeffe, Multiply-charged Large

Ions, 106 Nolan (P. J.), Influence of Condensation Nuclei and Dust

Particles on Atmospheric Ionisation, 922 Norman (Dr. A. G.), and S. H. Jenkins, Lignin Content of

Cellulose Products, 729

Norrish [Kirkbride and], Photochemical Decomposition of Diazomethane, 404

Norrish (Dr. R. G. W.), and G. H. J. Neville, Photosensitised Decomposition of Ozone by Chlorine, 544 Norwood (Dr. C.), Use of the English Language, 197; 741

Noskov (Dr. M.), [Dr. I. Kikoin and], A New Type of Photoelectric Effect in Cuprous Oxide in a Magnetic

Field, 725

du Noüy (Dr. P. Lecomte), Improvements to the Hydrogen Electrode for the Measurement of Hydrogen Ion Concentration in Solutions, 179; Surface Tension of Colloidal Solutions, and the Action of Light on Soap Solutions, 689

Novi (Prof. I.), Influence of Sodium and Potassium Ions

in Yeast Growth, 63

Nowakowski (A.), [K. Boratynski and], Modifications of Phosphoric Anhydride, 595

Nuttall (Mrs. Zelia), [death], 830

Obata, Morita and Yosida, Aeroplane Noises, 208 Oberholser (H. C.), Birds of the Natuna Islands, 207

O'Brien (J. R.), [R. D. Heard, H. W. Kinnersley, Prof. R. A. Peters, V. Reader and], Vitamin B₄ and Adenine, 617

Occhialini (G.), [P. M. S. Blackett and], Wilson Photographs of Cosmic Ray Phenomena, 589; [Dr. J. Chadwick, P. M. S. Blackett and], New Evidence for the Positive Electron, 473; [P. M. S. Blackett and], Photography of the Tracks of Penetrating Radiation, 286

Odelstierna (Prof. E. G.), [death], 538

Offord (J. M.), elected president of the Quekett Microscopical Club, 272

Öfverholm, Electrification of the Swedish State Railways, 199

Ogden (C. K.), Bentham's Theory of Fictions (Review), 152

Oguma (K.), and S. Kakino, Chromosome Numbers in Vertebrates, 403

Ohlsson (J.), Tables for Galactic Co-ordinates, 100 Okada (Dr. Yaichiro), Sponges from the North-Western Pacific, 279

O'Keeffe (J. G.), [J. J. Nolan and], Multiply-charged Large Ions, 106

Oldenbach (Rev. F. L.), [obituary], 830

O'Malley (L. S. S.), Indian Caste Customs (Review), 638 Opik (E.), Perturbations of Comets by the Stars, 209

Orchard (O. B.), [W. H. Read and], Disinfection of Glasshouses with Sulphur, 844

Ornstein (Dr. L. S.), Dr. W. J. H. Moll und Dr. H. C. Burger, Objektive Spektralphotometrie (*Review*), 824 Orr (P. C.), [H. W. Nichols and], Bakelite Impregnation of Fossil Bones, 127

Orr (Dr. W.), and Dr. F. F. Darling, Sterility in Domesticated Animals, 200

Orton (Prof. J. H.), Some Limiting Factors in the Environment of the Common Limpet, P. vulgata, 693; Strange Spatfall of the Common Mussel on the Common Cockle, 513; Miss M. W. Parke and W. C. Smith, Breeding of Oysters (O. edulis) at Port Erin,

Osawa (S.), Abnormal Movability of the Heart, 241 Osborn (Prof. H. F.), Aristogenesis, 768; Biological Inductions from the Evolution of the Proboscidea, 775; China and American Scientific Expeditions, 87; elected honorary president of the American Museum of Natural History, 236

Osborn (T. G. B.), J. G. Wood and T. B. Paltridge, Growth and Reaction to Grazing of the Perennial Saltbush, Atriplex vesicarium, 631

Osborne (N. S.), and others, Steam Tables, 624 Osborne (Prof. W. A.), 'Raw' Weather, 515 Osgood (Dr. C. M.), Great Bear Lake Indians, 623

Pagden (H. T.), appointed entomologist of the British Solomon Islands Protectorate, 687

Pahl (Dr. M.), [Prof. G. Hevesy and], Range of Radiation

from Samarium, 434
Paillard (H.), and R. Duckert, Catalytic Oxidation of Acenaphthene, 815; and P. Favarger, Chlorination of Acenaphthene, 815

Paine (S. G.), F. L. Lingood, Freda Schimmer, and T. C. Thrupp, Relationship of Micro-Organisms to the

Decay of Stone, 178
Paltridge (T. B.), [T. G. B. Osborn, J. G. Wood and],
Growth and Reaction to Grazing of the Perennial Saltbush, Atriplex vesicarium, 631

Panay (T. N.), Integral Radiator (Black Body) of Electrically Heated Carbon, 179

Pandalai (Madhusudanan), [G. Gopala Rao and], Sodium Sulphite in Photographic Developing Solutions, 100 Paneth (Prof. F. A.), and P. L. Günther, Chemical De-tection of Artificial Transmutation of Elements, 652

Pariselle, Change of Sign of the Rotatory Power and of Mutarotation, 702

Parisi (E.), and G. De Vito, Distribution of Nitrates and 'Organisation' of Nitrogen in the Leaves of Green Plants, 107

Park (Dr. W. H.), awarded the Public Welfare medal of

the U.S. National Academy of Sciences, 754
Parke (Miss M. W.), [Prof. J. H. Orton, W. C. Smith
and] Breeding of Oysters (O. edulis) at Port Erin, 26

Parker (G. H.), Cellular Transmission of Neurohumoral Substances in Melanophore Reactions, 776

Parkinson (Dr. J.), Central African Volcanoes (Review),

Parr (late Prof. S. W.), The Analysis of Fuel, Gas, Water and Lubricants. Fourth edition (*Review*), 860
Parravano (N.), and O. D'Agostino, Velocity of Dissolution of Industrial Aluminas in Fused Cryolite, 107 Parrington (F. R.), awarded the Balfour studentship in

biology of Cambridge University, 336 Parrot (L.), [E. Sergent, A. Donatien, F. Lestoquard and], Experimental Suppression of Sexual Reproduction in Theileria dispar, 143

Parsons (C. W.), Penguin Embryos, 695

Partington (Prof. J. R.), The Scientific Work of Joseph Priestley, 348

Pascal (P.), and Mme. Réchid, Preparation of the Dimetaphosphates, 666

Pascher (Prof. A.), elected a foreign member of the Linnean Society of London, 759

Patterson (H. S.), [Dr. W. Cawood and], A New Tempera-

ture Recorder, 332 Patterson (Prof. T. S.), Priestley as a Practical Chemist,

Patton (R. T.), Ecological Studies in Victoria (1), 703 Paul (Dr. J. Harland), The Last Cruise of the Carnegie (Review), 114

Pauling (L.), [L. O. Brockway and], Determination of the Structures of the Hexafluorides of Sulphur, Selenium and Tellurium by the Electron Diffraction Method, 739

Payman (Dr. W.), Recent Developments in Coal-Mining Explosives, 846

Payne-Scott (Miss R.), Relative Intensity of Spectral Lines in Indium and Gallium, 365

Pearl (Prof. R.), Influence of Density of Population upon Egg Production in Drosophila, 176

Pearson (Dr. J.), Marine Biology in Ceylon, 54 Pearson (Dr. T. G.), Carbonyls of Lithium, Rubidium and Cæsium, 166; [E. E. Aynsley, Dr. P. L. Robinson and], New Experimental Evidence in the Sulphur-Hydrogen Reaction, 471

Pearsall (Dr. W. H.), [L. Loose and], Synthesis of Protein by Green Plants, 362; and P. Ullyott, Measurement of Light for Biological Purposes, 694

Pearson (Dr. J.), Whale Shark in the Waters around

Ceylon, 729 Peck (B. M.), Illumination of Nebulosity surrounding a Nova. 808

Peddie (Prof. W.), Phenomenal Regression to the Real

Object, 544

Peers (Sir Charles Reed), appointed a trustee of the British Museum, 164; awarded the Royal gold medal of the Royal Institute of British Architects, 236; presented with the Royal gold medal of the Royal Institute of British Architects; the work of, 497; The Ancient Monuments of England, 253 Penfold (A. R.), [A. E. Bradfield, J. L. Simonsen and],

Essential Oil from the Wood of Eremophila Mitchelli,

Penlington (H. N.), Unemployment and the Schools, 578 Penrose (H. J.), World's Record for a Two-Seater Aircraft, 160

Pepys (Samuel), and Christ's Hospital, 267; Tercentenary of; and Royal Society, 299; Tercentenary of, 354

Perakis (N.), and L. Capatos, Constant Paramagnetism of Metallic Rhenium, 559

Pérard (A.), and M. Romanowski, First Comparisons of the National Standards of Electrical Resistance, 887 Percival (Prof. E.), Trout Fishing in New Zealand, 163

Percival (E. G. V.), [Dr. E. L. Hirst, F. Smith and], Constitution of Ascorbic Acid, 617

Perier (A.), Influence, in a Homogeneous Ethnic Group, of the Variation of the Cephalic Index on that of the Alveolo-Palatinal and Superior Facial Indices, 815 Perret (A.), [R. Perrot and], The Cyanide-Cyanamide

Equilibrium, 375

Perrot (R.), [A. Perret and], The Cyanide-Cyanamide Equilibrium, 375

Perry (E. V. V.), [M. E. F. Crawford, Dr. S. S. Zilva and], Vitamin Content of Australian, New Zealand and English Butters, 770

Petermann (B.), translated by M. Fortes, The Gestalt Theory and the Problem of Configuration (Review),

Peters (Prof. R. A.), [R. D. Heard, H. W. Kinnersley, J. R. O'Brien, V. Reader and], Vitamin B₄ and Adenine, 617

Petit (G.), [G. Grandidier et], Zoologie der Madagascar (Review), 748

Petrie (Sir Flinders), Old Gaza, 868

Pettersson (Dr. H.), and B. Kullenberg, Boundary Tides in the Kattegat, 586

Peytral (Mlle. Eglantine), [J. A. Muller and], The Sudden Pyrogenation of Ketene, 375 Pfeiffer (G.), [G. Koller and], Constitution of Pinastric

Acid; Galabratic Acid, 411

Philips Industrial, Philips Sodium Lamp, 440 Phillips (H.), [F. E. Humphreys and], Examination of Leather for the Presence of Extractable Chromium Compounds, 813 Philp (J.), [Dr. F. W. Sansome and], Recent Advances in

Plant Genetics (Review), 185

Piaggio (Prof. H. T. H.), Ambiguity in Sign of Spearman's General Factor, 170; Cantor: the Mathematician of the Infinite (Review), 418

Picard (M.), and A. Stampa, A New Form of Silver Voltameter, 666

Piccard (Prof. J.), [death], 646

Pickett (Lucy W.), Crystal Structure of Diphenyl Series, 513

Pickles (Dr. S. S.), Rubber Research, 273

Picon, Preparation and Properties of Thallous Thio-

carbonate, 179; Thorium Sulphide, 70 Pierson (R. K.), Fusion of Pycniospores with Filamentous Hyphæ in the Pycnium of the White Pine Blister Rust, 728

Pietschmann (V.), Three New Fish Species (Cyprinides) from Asia Minor, 704

Pignol (Dr. R.), Rainfall in Dahomey, 208

Pilgrim (Dr. G. E.), British Museum (Natural History) Catalogue of the Pontian Carnivora of Europe in the

Department of Geology (Review), 454

Piper (Dr. S. H.), Laboratory Vacuum Technique, 65 Pippard (Prof. A. J. S.), appointed professor of civil engineering at the Imperial College—City and Guilds College, 480

Pippard (Prof. S.), Contribution of Science in the Develop-

ment of Aeronautics, 65

Pirani (R.), [G. Natta and], Solid Solutions by Precipitation, etc. (2), 107
Pirie (Dr. N. W.), appointed university demonstrator in

biochemistry in Cambridge University, 628 Pitt-Rivers (Capt. G.), Ethnogenics, 89

Plaskett (Prof. H. H.), Observational and Theoretical

Astronomy, 648
Plasmann (Dr. J.), Variable Star Delta Orionis, 661
Player (E. S.), Be Your Own Weather Prophet: a Book for the Holidays and After (Review), 712

Pluvinel (A. de La Baume), and D. Barbier, Total Eclipse

of the Sun on August 31, 1932, 922

Podhradsky (Dr. J.), Influence of Living-Space upon

Growth, 29

Pohl (Prof. R. W.), Einführung in die Mechanik und Akustik. Zweite Auflage (Review), 568; translated by Winifred M. Deans, Physical Principles of Mechanics and Acoustics (*Review*), 320

Polanyi (Dr. M.), appointed professor of physical chemistry

in Manchester University; the work of, 902
Pollard (E. C.), Entry of the Disintegrating Alpha
Particle into the Nitrogen Nucleus, etc., 375; Heights of Nuclear Potential Barriers, 97; Heights of Nuclear Potential Barriers and Nuclear Structure, 398; Law of Force between Neutron and Proton, 814; Protons Produced in the Artificial Disintegration of the Nitrogen Nucleus, 482

Pongratz (A.), [H. Kopper and], Raman Effect (24), 251 Ponte (G.), Volcanic Fumaroles, 923 Poole (Dr. H. H.), and Dr. W. R. G. Atkins, Photocells of the Dry Rectifier Type for the Measurement of Daylight, 850; Reversal of Current in Rectifier Photocells, 547; Reversal of the Current from a Cuprous Oxide Photocell in Red Light, 133

Poole (Dr. J. H. J.), Behaviour of Neon Discharge Tubes in a Flashing Capacity Circuit by means of a Cathode Ray Oscillograph, 35; Radioactivity of Samarium and the Formation of Hibernium Halos, 654

Porter (Prof. J. G.), [death], 830 Porter (Dr. T. C.), [death], 496

Posejpal (V.), Atomic Radius of Carbon in the Diamond, 410

Potts (F. A.), [L. A. Borradaile and], The Invertebrata: a Manual for the use of Students. With chapters by Prof. L. E. S. Eastham and J. T. Saunders

(Řeview), 76 Poultney (F. C.). [Prof. R. Whiddington and], Energy of the Beams in Electron Diffraction, 814

Powell (Dr. C. F.), and Luang Brata, Positive Ion Emis-

sion from Oxide Catalysts, 168

Praeger (Dr. R. Lloyd), Origin of the Irish Flora and Fauna, 279, 579; Some Noteworthy Irish Plants, 172 Prashad (Dr. B.), the late Lieut.-Col. A. W. Alcock, 906 Pratesi (P.), Action of Alkaline Hypoiodites on the Pyrrolic Aldehydes, 924; Thiocyano- and Thiopyrroles and Pyrrole Disulphides, 411

Prescott (Dr. J.), and H. V. Lowry, Elementary Trigono-

metry (Review), 420

Prévost (C.), An Iodo-silver-benzoic Complex and its Application to the Oxidation of Ethylene Compounds into α-Glycols, 774

Prévot (E.), Influence of the Daily Oscillations of the Vertical on the Results of High Precision Levelling,

Price (E. W.), Trematodes of Marine Mammals, 243

Price (H.), An Account of some Further Experiments with Rudi Schneider; a Minute-by-Minute Record of 27 Séances (Review), 489; Further Light on the Schneider Mediumship, 658

Price (J. B.), Winter Territory amongst Birds, 730

Priday (T.), [H. Leak and], Migration from and to Great Britain, 125

Priestley (Joseph), bicentenary of the birth of, 354; 498 Priestley (Prof. J. H.), Lorna I. Scott and Marjorie E. Malins, Cambial Activity, 375

Prince (Major C. E.), The Practical Applications of Light-

Sensitive Apparatus, 430

Prins (J. A.), Diffraction of Electrons in Amorphous and in Crystalline Antimony, 760

Prosad (Prof. K.), and S. Sharan, Supersonic Vibrations set up in a Zinc Bar undergoing Transverse Vibrations, 803

Proudman (Prof. J.), elected professor of oceanography in Liverpool University; the work of, 391

Pryde (J.), and R. T. Williams, A New Unsaturated Derivative of Glucuronic Acid, 57

Przibram (Prof. K.), Recrystallisation and Blue Rock-Salt (1, 2, 3), 287; Recrystallisation and Coloration

(3); Plasticity and Hardness of Crystals of Alkali Metal Halides, 560 Pugsley (H. W.), Daffodils, Narcissi and their Hybrids, 517 Pull (J. H.), The Flint Miners of Blackpatch (Review), 677

Purser (J.), appointed professor of civil engineering in Dublin University, 772

Pyne (G. T.), and J. J. Ryan, Colloidal Calcium Phosphate of Milk, 35

Quagliariello (G.), Presence in the Bile of an Enzyme which dehydrogenates Stearic Acid, 411

Quarrell (A. G.), [Prof. G. I. Finch and], Crystal Structure and Orientation in Thin Films, 877; Determination of Crystal Lattice Constants by Electron Diffraction, 842; Structure of Magnesium, Zinc and Aluminium Films, 482

Quastel (Dr. J. H.), Glutathione, Iodoacetic Acid and

Glucose Metabolism, 206

Quilico (A), and A. di Capua, Aspergillin, the Pigment of the Spores of Aspergillus niger (1), 851; (2), 924 Quill (L. L.), [B. S. Hopkins and], Use of Non-Aqueous

Solvents in the Study of the Rare Earth Group, 739 Quintin (Mlle. M.), Application of Debye's Theory to Solutions of Copper Sulphate, 630; The Electromotive Force of the Chain, 482

Radin (Dr. P.), Social Anthropology (Review), 315

Rae (Prof. R.), appointed professor of agriculture in Reading University, 772
Raghavan (Dr. M. D.), The Malabar House, 29
Rahimullah (M.), and Prof. B. K. Das, Making Whole Mounts of Vertebrate Skeletons, 171

Rakshit (H.), [Prof. S. K. Mitra and], Recording Wireless Echoes at the Transmitting Station, 657

Ram (Atma), [Prof. N. R. Dhar and], Presence of Formaldehyde in Dew, 800

Ramachandran (T. N.), [Dr. F. H. Gravely and], Hindu Metal Images, 279

Ramage (G. R.), and Prof. R. Robinson, Synthesis of Chrysene and certain Derivatives, 205

Ramage (H.), Applications of the Spectroscope to Biology,

Ramart-Lucas (Mme.), and Mme. M. Grumez, Colour and Structure of Oximes and Semi-carbazones, 70; and Mlle. Wohl, Colour and Structure of Amides, 286

Ramaswami (S.), [S. Krishna and], Calorific Values of some Indian Woods, 281

Rambaud (R.), Trans γ-oxycrotonic Acid, 483 Ramsey (C. G.), and H. R. Sleeper, Architectural Graphic Standards: for Architects, Engineers, Decorators,

Builders and Draftsmen (*Review*), 224 Ramsey (G. B.), and Alice A. Bailey, Tomato Late-Blight Rot, 440

Randoin (Mme. Lucie), et H. Simonnet, Les vitamines (Review), 258

Rankine (Prof. A. O.), elected president of the Physical Society, 510

Rao (A. S.), Spectra of Bromine: Br V, VII and IV, 170

Rao (G. Gopala), and Madhusudanan Pandalai, Sodium Sulphite in Photographic Developing Solutions, 100

Rao (Dr. K. R.), Spectra of Bromine. Br V, VII and IV,

Ratcliffe (J. A.), appointed university lecturer in the department of physics of Cambridge University, 811; and E. L. C. White, Fine-Structure of the Ionosphere, 873; Heaviside Layer, 807; [C. T. R. Wilson, E. L. C. White and], An Automatic Recording Method for Wireless Investigations of the Ionosphere, 522

Rateau (late Prof. A. C. E.), the work of, 650 Rây (Sir Prafulla Chandra), Life and Experiences of a Bengali Chemist (Review), 672; Seventieth Birthday Commemoration Volume, 866

Ray (S. N.), [T. W. Birch, Dr. L. J. Harris and], Hexuronic (Ascorbic) Acid as the Antiscorbutic Factor, and its

Chemical Determination, 273

Rayleigh (Lord), Beryllium and Helium, 724

Rayner (Dr. M. C.), Mycorrhiza in the Genus Citrus, 399 Read (Prof. J.), Growth of Space Chemistry (Review), 783; Humour and Humanism in Baeyer's Laboratory, 294

Read (J.), Elementary Textile Design and Fabric Struc-

ture (Review), 187
Read (W. H.), and O. B. Orchard, Disinfection of Glass-

houses with Sulphur, 844

Reader (V.), [R. D. Heard, H. W. Kinnersley, J. R. O'Brien, Prof. R. A. Peters and], Vitamin B_4 and Adenine, 617 Réchid (Mme.), [P. Pascal and], Preparation of the

Dimetaphosphates, 666

Reck (Prof. H.), [Dr. L. S. B. Leakey, Prof. P. G. H. Boswell, A. T. Hopwood, Dr. J. D. Solomon and], The Oldoway Human Skeleton, 397

Redman (Dr. R. O.), appointed university lecturer in the department of astrophysics of Cambridge University,

Rée (Dr. A.), [obituary article], 425

Rees (H. G.), Iron and Copper in Liver and Liver Extracts,

Rees (W. J.), appointed temporary assistant lecturer in botany in Birmingham University, 248

Regan (Dr. C. Tate), elected a foreign member of the Royal Danish Academy, 650

Regener (Prof. E.), Absorption of Penetrating Radiation, 880; Energy of Cosmic Rays, 130

Reh (Miss Emma), Archæological Exploration in Oaxaca, Mexico, 509

Reich (W. S.), Glycogen Triacetate, 106 Reid (D. M.), [J. T. Cunningham and], Pelvic Filaments of Lepidosiren, 913

Reid (F. H.), Technical Education, 883

Renshaw (G.), The Dodo and the Aphanapteryx, 728 Reynolds (Dr. Doris L.), appointed lecturer in geology

in Durham University, 811

Reynolds (Prof. S. H.), Geology of the Bristol District, 65 Ribaud (Prof. G.), Solution of the Problem of Heterochrome Photometry of Incandescent Lamps, 667; Traité de pyrométrie optique (Review), 151

Richard (J.), Use of the Cattaneo Wave Pump for Utilising

the Movements of the Sea, 179

Richards (H. W. H.), Steam, Electric and Diesel-Electric Traction, 302

Richardson (Dr. L. F.), Photography of Faint Transient Light-Spots, 401

Richardson (T. N.), and Dr. K. C. Bailey, Supersaturation of Liquids with Gases, 762

Richter (C. F.), [H. O. Wood and], Californian Earthquake

of March 10, 686 Rickard (Dr. T. A.), Man and Metals: a History of Mining in Relation to the Development of Civilisation. 2 vols. (*Review*), 743
Rideal (Prof. E. K.), [A. H. Hughes and], Rate of Oxida-

tion of Monolayers of Unsaturated Fatty Acids, 446

Rigg (Sir Edward), [death], 576

Riiser-Larsen (Capt. H.), Norwegian Antarctic Expedition,

Rinne (Prof. F.), [death], 462

Ripper (Elizabeth A.), Stromatoporoids of the Lilydale Limestone (1), 106 Risteen (Dr. A. D.), [death], 266

Ritchie (Prof. J.), Sir J. Arthur Thomson, 296; The Musk-Rat in Britain, 385

Ritchie and Ludlam, A Gaseous Oxidation, 64

Ritschel (R.), Hyperfine Structure in Aluminium, 58 Ritt (Prof. J. F.), Differential Equations from the Algebraic Standpoint (Review), 456

Ritter (Dr. E.), Die Sprache im internationalen Verkehr. Second edition; Die Weltverkehrssprache, 102

Roach (Dr. W. A.), Distribution of Molybdenum, 202 Robb (W.), Inheritance of Grain Colour in Oats, 517

Robbins (F.), Dr. R. T. A. Innes, 461 Roberts (H. A.), [obituary article], 49 Roberts (O. F. T.), Eddy Diffusion, 916

Robertson (Sir Charles Grant), Training the University Graduate, 268

Robertson (Dr. D.), [A. H. H. Fraser and], Nutritional Condition of Sheep and Susceptibility to Stomach Worm, 94

Robertson (Dr. T. Brailsford), edited by Jane W. Robertson, The Spirit of Research (Review), 111

Robin (Dr. P. A.), Animal Lore in English Literature (Review), 257

Robinson (Dr. B. W.), Electrometer Triode in the X-Ray Ionisation Spectrometer, 546

Robinson (I. V.), The Battersea Power Station, 720 Robinson (J. E. L.), [Dr. J. L. Miller and], awarded the

Institute of Physics prize, 467 Robinson (Dr. P. L.), [E. E. Aynsley, Dr. T. G. Pearson and], New Experimental Evidence in the Sulphur-Hydrogen Reaction, 471

Robinson (Prof. R.), [G. R. Ramage and], Synthesis of Chrysene and certain Derivatives, 205

Robison (Dr. R.), [Dr. H. B. Fell and], Glycogen in Cartilage, 62

Roebuck (Dr. John), the work of, 196

Rogers (Dr. A. W.), elected president of the Royal Society of South Africa, 22; 687

Rogers (M. J.), Know Your San Diego, 903 Rogers (W. S.), Root Studies, III, 695

Roget (S. R.), A Dictionary of Electrical Terms: including Telegraphy, Telephony and Wireless. edition (*Review*), 322

Rolf (Dr. B), Temperature of the Stratosphere, 280 Rolfe (A. C.), [C. N. Hinshelwood, E. A. Moelwyn-Hughes and], Combination of Hydrogen and Oxygen in a Silver Vessel, 142; Reaction between Hydrogen and

Oxygen, 625 Rolt and Barrell, Sources of Light for Interferometer

Work, 244 Romanowski (M.), [A. Pérard and], First Comparisons of the National Standards of Electrical Resistance,

887 Roscoe (Sir Henry Enfield), Centenary of the birth of, 18

Rosenhain and Stott, Energy absorbed in the Cold Working of Metals, 769

Rosenhead (Dr. L.), Instability of Liquid Surfaces, 175 Ross (Dr. T. A.), An Introduction to Analytical Psychotherapy (Review), 639

Rossi (A.), Crystalline Structure of the Compound LaAl4, 924

Rossier (P.), Diameter of some Stars of the Cluster M7, 888; Influence of the Absolute Magnitude of a Star on the Width of the Lines of Stellar Hydrogen, 483 Rossman (Dr. J.), The Law of Patents for Chemists

(Review), 115

Roth (W. E.), [obituary], 612 Rouard (P.), Reflecting Power of Metals in very Thin Plates, 36; Variations of Phase by Reflection on very Thin Metallic Films, 410 Roubaud (E.), Desert Anhydrobiosis and its Influence

on the Animal Cycle of Schistocerca peregrina, 774

Roughton (Dr. F. J. W.), [late Dr. N. U. Meldrum and], Carbonic Anhydrase and the State of Carbon Dioxide in Blood, 874

Rowan (Prof. W.), Bird Migration (3), 288

Rowland (Rev. J. P.), Recent Magnetic Disturbances, 764; Wensleydale Earthquake, 128

Roxburgh (H. L.), [H. W. Melville and], Upper Limit in Explosive Chain Reactions, 690

Roy (Mlle. Madeleine), [A. Boutaric and], Influence of

Radioactive Radiations on the Flocculation of Colloids, 738 Royer (L.), Orientation of certain Crystals by Hydrar-

gillite, 596

Ruedemann (R.), Palæozoic Planktonic Faunas of North America, 775

Rupe (Prof. H.), Adolf von Baeyer als Lehrer und Forscher, 294

Rupp (E.), Polarisation Phenomena in Electron Scattering, 209

Ruse (H. S.), Measurement of Spatial Distance in a Curved Space-Time, 250

Russ (Prof. S.), 'Raw' Weather, 131

Russell (Prof. H. N.), Composition of the Stars (Halley Lecture), 832; elected president of the American Association, 164, 304

Ruszkowski (J. S.), Development of Gyrocotyle, 243

Rutgers (J. J.), [A. Girard, G. Sandulesco, A. Fridenson and], A New Crystallised Sex Hormone, 71

Rutherford (Lord), Artificial Transmutation of the Elements (Boyle Lecture), 832; Recent Researches on the Transmutation of the Elements, 388; Scientific Problems of the Pacific, 831

Rutzler, Jr. (J. E.), [W. D. Bancroft and], Agglomeration Theory of Sleep, 739

Ryan (J. J.), [C. Boyle and], Grass Silage, 410; [G. T. Pyne and], Colloidal Calcium Phosphate of Milk,

Rydberg (R.), [Prof. E. Hulthén and], Predissociation and Pressure Effects in the Band Spectrum of Aluminium Hydride, 470

Saal (R. N. J.), and C. G. Verver, Measurement of Volatility Range of Lubricating Oils, 661

Sabatier (Dr. P.), awarded the Franklin medal of the Franklin Institute, 871

Sabatucci (N.), Toxic Action and Elimination of Nicot-

ine (1, 2), 483
Sakai (Dr. Tsune), Spider Crabs from Japan, 551
Salaman (Dr. R. N.), Protective Inoculation against a Plant Virus, 468

Salisbury (Prof. E. J.), Influence of Man on Vegetation, 919; Intensive Natural History (Review), 343; The East Anglian Flora, 336

Salisbury (F. O.), Offer to St. Andrews University of a replica of a portrait of Dr. E. Harkness, 372

Salmon (late C. E.), edited by W. H. Pearsall, Flora of

Surrey (Review), 77 Salomon (H.), [Prof. P. Karrer, K. Schöpp and], Constitution of Dehydro-Ascorbic Acid, 800

Salt (Dr. G.), elected Stringer fellow of King's College, Cambridge, 141

Salt (Lieut. J. S. A.), A Simple Method of Surveying from Air Photographs, 650

Salvatori (A.), Nature and Value of Bezssonoff's Reaction for the C Factor, 339

Sanderson (I. T.), awarded a grant from the Balfour Fund of Cambridge University, 736

Sandulesco (G.), [A. Girard, A. Fridenson, J. J. Rutgers and], A New Crystallised Sex Hormone, 71

Sandford (Dr. K. S.), Volcanic Craters in the Libyan Desert, 46

Sansome (Dr. F. W.), and J. Philp, Recent Advances in Plant Genetics (Review), 185

Santon (L.), Some Results Obtained with a Supersonic Flower, 523

Sarasin (Dr. F.), Prehistory of South-East Asia, 879

Sargent (Sir Percy), [death], 158

Sarrant (A.), France and Great Britain in Africa, 831 Sassoon (Sir Philip), Aeronautics and the Royal Air Force, 428

Satterly (Prof. J.), [W. R. Bower and], Practical Physics. Third edition (Review), 491

Savage (R. M.), Factors controlling Date of Spawning in Frogs, 587

Sayce (Prof. A. H.), [death], 195; [obituary article], 296 Sayce (R.), and F. Hanley, Kiln-dried Poultry Manure, 198

Scattergood (B.), gift to Leeds University Observatory, 665

Schaarschmidt (Prof. A.), [death], 195 Schafer (J. P.), and W. M. Goodall, Characteristics of the Ionosphere, 804

Schaffer (J.), Tissue Structure of Homologous Organs (Prepuce Glands) in Mice and Rats, 703 Schaffert (R.), Form and Vibrational Frequencies of the

Nitrogen Dioxide Molecule, 911

Scharff (Dr. R. F.), Past Wanderings in Europe of Two African Animals, 922

Scheibe and Adelsberger, Oscillating Quartz Crystal as an Accurate Clock, 281

The Circular Magnetic Dichroism and Schérer (M.), Abnormal Magnetic Rotatory Dispersion of Solutions of Cobalt Chloride, 70; [A. Cotton and], Magnetic Rotatory Dispersion of a Coloured Diamagnetic Compound, Thiobenzophenone, 215

Scheuchzer (Johann Jacob), bicentenary of the death of,

902

Schimmer (Freda), [S. G. Paine, F. L. Lingood, T. C. Thrupp and], Relationship of Micro-Organisms to the Decay of Stone, 178 Schintlmeister (J.), Ionisation of H-Rays in Different

Gases, 251

Schmid (H.), [E. Abel and], Flow Kinetics: Model of Photolysis, 251

Schmidt (Prof. J.), English edition by Dr. H. G. Rule, A Text-Book of Organic Chemistry. Second edition (Review), 116

Schmidt (Prof. Johannes), [death], 298; [obituary article], 424

Schmidt (Dr. Julius), [obituary article], 718 Schober (H.), Spectra of Rhenium (3), 411; (4), 851

Schofield (Dr. R. K.), Capacitance Hygroscopy and some

of its Applications, 96
Schokalsky (J.), The Circumnavigation of the François
Joseph Archipelago by the North, 70

Schonbeck (A.), [G. Gorbach and], Influence of Hydrocyanic Acid on Bacterial Proteases, 287

Schonland (Dr. B. F. J.), Atmospheric Electricity (Review), 785; and Viljoen, Penetrating Rays from Thunderclouds, 916

Schopfer (W.), Biometric Researches on the Spores of Mucor, 483 Schöpp (K.), [Prof. P. Karrer, H. Salomon and], Constitu-

tion of Dehydro-Ascorbic Acid, 800

Schriever (W.), [Prof. F. A. Melton and], Meteor Craters, 100; Probable Meteorite Scars in Carolina, 624

Schröder (V. N.), [Prof. N. K. Koltzoff and], Artificial Control of Sex in the Progeny of Mammals, 329

Schubert (J.), [J. Kisser and], Influence of Treatment of Seeds with Chemical Stimulants on the Cell Growth of the Rootlets, 251

Schuchert (Prof. C.), Gondwana Land Bridges, 696; and G. A. Cooper, Brachiopod Genera of the Sub-orders Orthoidea and Pentameroidea, 244

Schuster (Sir Arthur), Biographical Fragments (Review),

Schweidler (Prof. E.), Die Aufrechterhaltung der elektrischen Ladung der Erde (Review), 785

Schweitzer (A.), translated by C. T. Campion, My Life and Thought: an Autobiography (Review), 785

Schweitzer (Dr. E.), [Dr. W. Gerlach and], Foundations and Methods of Chemical Analysis by the Emission Spectrum (Translation), 420

Schweizer (Jean), Periodicity in Tree Growth in the Tropics, 624

Schwinner (R.), Geology of Eastern Styria: the Rocks and their Inter-Relationships, 560

Schwob (M.), [R. Lucas and], Anomalous Dispersion in Magnetic and Electric Double Refraction, 630

Scofield (N. B.), Legal Size Limits and their Effect upon Fisheries, 832

Scott (Dr. A.), Preservation of Metallic Objects in Museums, 906

Scott (Lorna I.), [Prof. J. H. Priestley, Marjorie E. Malins and], Cambial Activity, 375

Scott-Hill (Rear-Admiral W.), Coal for Sea Transport, 479 Scourfield (D. J.), New Species of Cyclops at Tenby, 135 Seaber (W. M.), Barium as a Normal Constituent of Brazil Nuts, 814

Searle (V. H. L.), The Electrical Age: being further Everyday Marvels of Science (Review), 491

Sears (J. E.), and H. Barrell, A New Apparatus for Determining the Relationship between Wave-Lengths of Light and the Fundamental Standards of Length, 192

Segrè (E.), Quadrupole Lines in X-Ray Spectra (2) 411 von Seidlitz (Prof. W.), Der Bau der Erde und die Bewegungen ihrer Oberfläche: eine Einführung in die Grundfragen der allgemeinen Geologie (Review),

Seligman (Brenda Z.), [Prof. C. G. Seligman and], Pagan Tribes of the Nilotic Sudan (Review), 345 Seligman (Prof. C. G.), and Brenda Z. Seligman, Pagan

Tribes of the Nilotic Sudan (Review), 345 Selwood (P. W.), Influence of Light on Paramagnetic

Susceptibility, 761 Selye (H.), [Prof. J. B. Collip, Prof. D. L. Thomson and], Gonad-stimulating Hormones in Hypophysectomised Animals, 56

Semerano (Dr. G.), Il polarografo: sua teoria e appli-

cazioni (Review), 260 Semple (Prof. Ellen Churchill), The Geography of the Mediterranean Region: its Relation to Ancient History (Review), 317

Semple (Dr. J. E.), awarded the Raymond Horton-Smith prize of Cambridge University, 772

Semple (Prof. J. G.), Composite Surfaces in Higher Space,

593

Séon (M.), [C. Matignon and], Action of Steam on Heavy Petroleum Oils and on certain Cyclic Hydrocarbons, 523; Action of Steam on Hexane, 215, 286 Sergent (E.), A. Donatien, L. Parrot and F. Lestoquard,

Experimental Suppression of Sexual Reproduction in Theileria dispar, 143 Servigne, Existence of a Polonium Acetylacetonate, 375

Seshaiya (R. V.), Style-Sac of Gastropods, 30 Sexton (Mrs. E. W.), and Miss A. R. Clark, Further Mutations in the Amphipod Gammarus chevreuxi, Sexton, 201

Seymour (A. B.), [death], 830 Sezawa (Prof. Katsutada), Viscous Damping of Vibrating Metal Bars, 803

Shapera (Dr. I.), Chastity in Bechuanaland, 516 Shapiro (H. A.), The Kitchen middens at Gordon's Bay, 411

Shapley (Dr. H.), awarded the Rumford medal of the American Academy of Arts and Sciences, 871; Galactic Dimensions, 739

Sharan (S.), [Prof. K. Prosad and], Supersonic Vibrations set up in a Zinc Bar undergoing Transverse Vibrations, 803

Sharpe (Sir Montagu), Middlesex in British, Roman and Saxon Times. Second edition (Review), 677

Sharpey-Schafer (Sir Edward), impending retirement and work of, 86

Shaw (Dr. Knox), Motion of the Spiral Nebulæ, 697

Sheehy (E. J.), Nutritive Value of Pastures, 30 Sheffield (Dr. F. M. L.), Virus Diseases and Intracellular Inclusions in Plants, 325

Shepard (Prof. F. P.), Submarine Valleys, 331 Shepherd (E. S.), [J. W. Greig, H. E. Merwin and], Volatile Transport of Silica, 768

Sheppard (T.), retirement from editorship of the Naturalist. 304

Sherratt (G. C.), [Dr. G. W. C. Kaye and], Velocity of Sound in Gases in Tubes, 338

Shield (late A. M.), proposed foundation of a studentship and a readership in Cambridge University, 811 Shire (E. S.), elected a fellow of King's College, Cambridge, Shrubsall (Dr. F. C.), and Dr. A. C. Williams, Mental Deficiency Practice: the Procedure for the Ascertainment and Disposal of the Mentally Defective

(Review), 456 Shull (A. F.), Time of Embryonic Segregation in Aphids as Determined from Intermediate Types, 775

Sibaiya (L.), [Prof. B. Venkatesachar and], Hyperfine Structure and Nuclear Moments, 844

Sidgwick (Dr. N. V.), elected a member of the Athenaum

Club, 200 Siegbahn (Prof. M.), Extreme Ultra-Violet and the Very Soft X-Ray Region (Guthrie Lecture), 901 Sigerist (Dr. H. E.), translated by Margaret Galt Boise,

Man and Medicine: an Introduction to Medical Knowledge (Review), 894

Silberstein (L.), [G. Bertrand and], Importance of Sulphates as Manure, 215

Silvestri (Prof. F.), elected a foreign member of the Linnean Society of London, 759 Simmons (W. C.), [A. D. Combe and], The Volcanic Area

of Bufumbira. Part 1: The Geology of the Volcanic Area of Bufumbira, South-West Uganda; with Notes on the Petrology and Economic Geology (Review), 821 Simonds (Col. G.), Safety of Life from Fire, 269

Simonnet (H.), [Mme. Lucie Randoin et], Les vitamines

(Review), 258
Simonsen (J. L.), [A. E. Bradfield, A. R. Penfold and],
Essential Oil from the Wood of Eremophila Mitchelli,

Simpson (Dr. G. C.), Low Auroras and Auroral Sounds, 828; The Astronomical Radiative Stability, 875

Simpson (J. B.), The Late-Glacial Re-advance Moraines of the Highland Border West of the River Tay, 522 Simpson (J. C. E.), [Prof. I. M. Heilbron and], Hydroxyl

Group in Ergosterol and Cholesterol, 438 Singleton (W.), and Brindley Jones, Effects of the Addition

of Tellurium to Lead, 696

de Sitter (Prof. W.), Kosmos (Review), 487 Skarżyński (Dr. B.), An Oestrogenic Substance from Plant Material, 766

Skene (Dr. MacGregor), Charles and Erasmus, 66 Skorko (E.), Absorption Bands of Iodine Vapour at High

Temperatures, 366 Sleeper (H. R.), [C. G. Ramsey and], Architectural Graphic Standards: for Architects, Engineers, Decorators, Builders and Draftsmen (Review), 224 Slipher (Dr. V. M.), awarded the Gold medal of the

Royal Astronomical Society, 91; awarded the Henry Draper medal of the U.S. National Academy of Sciences, 754; Spectra of the Planets (George Darwin Lecture), 734

Smith (Dr. C. J.), Intermediate Physics (Review), 532 Smith (Eng.-Capt. E.C.), Scientific Centenaries in 1933, 14 Smith (E. Lester), Systems of Four Immiscible Liquid Layers, 167 Smith (Dr. E. W.), Submarine Telephone Cables at

Carrier Frequencies, 431

Smith (Sir Frank), The Travel of Wireless Waves (Kelvin lecture), 642

Smith (F.), [Dr. E. L. Hirst, E. G. V. Percival and], Constitution of Ascorbic Acid, 617

Smith (F. Ll.), [A. Egerton and], Hydrocarbon Combustion in an Engine, 725

Smith (Dr. J. C.), Chlorination of Sodium Benzoate, 28 Smith (Dr. J. Henderson), appointed head of the department of plant pathology at Rothamsted, 303

Smith (late J. Perrin), Phylogeny of Ammonites, 368 Smith (Dr. Kenneth M.), awarded the Snell Memorial

medal of the National Institute of Agricultural Botany, 164; Plant Virus Research, 915

Smith (S.), with notes on fossils by C. J. Stubblefield, Tremadoc Shales in the Tortworth Inlier, Gloucestershire, 214

Smith (W. C.), [Prof. J. H. Orton, Miss M. W. Parke and],

Breeding of Oysters (O. edulis) at Port Erin, 26 Smith (W. W.), [Prof. J. Kendall, T. Tait and], Calcium Isotope with Mass 41 and the Radioactive Half-Period of Potassium, 688

Smith-Rose (Dr. R. L.), Electrical Properties of Soil for Alternating Currents: with particular reference to Radio-Frequencies, 142

Smoluchowski (R.), Magnetic Quenching of Tellurium

Fluorescence, 914

Smyth (Corisande) and M. Young, Facial Growth in Children, 29

Smyth [McAlpine and], Polarity of Hydrocarbon Vapours,

Smythe (F. S.), Kamet Conquered (Review), 224

Smythe (Dr. W. R.), and Dr. W. C. Michels, Advanced Electrical Measurements (Review), 322

Snape (Dr. H. L.), [death], 353

Snow (Dr. C. P.), [Dr. F. P. Bowden, S. D. D. Morris and], Absorption Spectrum of Vitamin A at Low Tempera-

Snyder [Buckley and], Spreading of Liquids on Solid Surfaces, 407

Soddy (Prof. F.), The Interpretation of the Atom (Review),

Sohon (Rev. F. W.), Seismometry (Introduction to Theoretical Seismology, Revs. J. B. Macelwane and F. W. Sohon, Part 2), (Review), 824 Sohoni (K. N.), [Prof. S. K. Banerji and], Hydraulic

Seismographs, 547

Sokolowski (J.), Stork in Western Europe, 403

Solomon (Dr. J. D.), awarded the Daniel-Pidgeon fund of the Geological Society of London, 616; [Dr. L. S. B. Leakey, Prof. H. Reck, Prof. P. G. H. Boswell, A. T. Hopwood and], The Oldoway Human Skeleton, 397

Sommerfeld (Prof. A.), presented with the James Scott prize of the Royal Society of Edinburgh, 687

Sorel (R.), Tincture of Iodine and Asepsis, 596 Sørensen (late Dr. W.), revised by Dr. C. With and K. L. Henriksen, The Arachnid Group, Opiliones-Laniatores, 843

Sorge (Dr. E.), Thickness of the Greenland Ice, 807 Soulié (J.), [L. Gay and], A Boiling Point Apparatus for the Determination of the Dew Points and Boiling

Points of Mixtures of Volatile Liquids, 630 Soulier (A.), [A. Kling and], Accidental Ignition of Vapours of Petrols by Electric Sparks, 106

Soulsby (B. H.), [death], 86; [obituary article], 230 Spärck (R.), [T. Gaarder and], Norwegian Oyster Pools,

Sparks (C. H.), [C. H. Davy and], High Pressure Boilers,

Späth (W.), Spectroscopic Detection of Small Quantities

of Elements, 30 Speiser (Dr. E. A.), Tell-Billa, 806 Spence (H. S.), Pitchblende at Great Bear Lake, Canada, 208

Spencer (C. J.), Electric Trolley Omnibuses, 869

Spencer (H. S.), Radium in Canada, 55

Spencer (Dr. L. J.), Meteoric Craters, 172; Meteoric Irons and Silica-Glass from the Meteorite Craters of Henbury (Central Australia) and Wabar (Arabia), 250; Origin of Tektites, 117; 596; 876

Spielmann (Dr. P. E.), Bitumen in Embalming, 588 Spurrell (W. R.), appointed reader in physiology at Guy's Hospital Medical School, 480

Squire (C.), and D. W. Herdman, The Museums of Malta, Cyprus and Gibraltar, 430

Stace (Dr. W. T.), The Theory of Knowledge and Existence (Review), 455

Stagg (J. M.), Report of British Polar Year Expedition,

Staley (J.), [J. F. Marshall and], A new British Record of Orthopodomyia pulchripalpis, Rondani (Diptera, Culicidæ), 435

Stalker (Prof. A. M.), [obituary article], 85

Stamp (Dr. L. D.), Annual Conference of the Geographical Association, 68; Geology in the Life of a Nation, 226; The Trinkler Expedition to Central Asia (Review), 600

Stampa (A.), [M. Picard and], A New Form of Silver Voltameter, 666

Stansfield (Prof. H.), Dimensions of Fundamental Units, 59

Starkey (J. L.), Excavations at Tell Duweir, Southern Palestine, 1932-33, 897

Starks (Prof. E. C.), [death], 462; [obituary article], 576 Statham (Col. J. C. B.), [death], 830

Stead (E. F.), The Life Histories of New Zealand Birds (Review), 603

Stead (G.), Elementary Physics. Fourth edition (Review), 748

Stebbing (Prof. E. P.), Lord Lovat, 352

Stebbing (Dr. L. S.), title of professor conferred upon, by London University, 811

Stebbins (Prof. J.), and Dr. C. M. Huffer, Absorption of Light in the Galaxy, 769

Steele (Dr. S.), A Theory of Fuel-Knock, 724 Steggall (Prof. J. E. A.), retirement from University

College, Dundee, 628

Stein (Sir Aurel), On Ancient Central-Asian Tracks: Brief Narrative of Three Expeditions in Innermost Asia and North-Western China (Review), 415

Steinburg (Prof.), Developments in Highway Engineering, 334

Stephen (A. C.), Scottish Marine Fauna, 446

Stephenson (G.), [obituary article], 50

Stephenson (Mrs. G.), gift to Cambridge University, 848 Stephenson (Lieut.-Col. J.), [obituary article], 193 Stephenson (Prof. T. A.), Lunar Periodicity in Repro-

duction, 622

Sterki (Dr. V.), [death], 353 Stetson (Prof. H. T.), Variation Effect in Latitude, Correlatable with the Moon, 437; and Dr. A. L. Loomis, Tidal Shifts in the Earth's Crust, 137

Stewardson (E. A.), Dissociation of Nitrous Oxide in the Glow Discharge, 364

Stewart (Dr. T. D.), Anomalous Eskimo Vertebræ, 125 Stiles (W. S.), and B. H. Crawford, The Luminous Efficiency of Rays entering the Eye Pupil at Different Points, 250

Stock (A.), and E. Wiborg, Electrolysis of Boron Hydrides, 212

Stockmair (W.), [F. Holzl and], Complex Anion of Buff's Compound and of Bunsen's Salt, 251

Stoll (Prof. A.), Researches in Biochemistry, 541

Stone (Prof. O.), [death], 231

Stone (Dr. W.), elected an honorary member of the British Ornithologists' Union, 723

Stonehaven (Lord), nominated as president of the Institution of Naval Architects, 432

Stoner (Dr. E. C.), Atomic Moments of Ferromagnetics, 433; Interatomic Distances and Ferromagnetism, 814

Stott (V. H.), Measurement of the Viscosity of a Molten Metal by means of an Oscillating Disc, 850; [Rosenhain and], Energy Absorbed in the Cold Working of

Metals, 769 Stoughton (Dr. R. H.), appointed professor of horticulture

in Reading University, 248 Stoy (R. H.), awarded a Smith's prize of Cambridge University, 408

Stoyko (N.), and R. Jouast, Apparent Velocity of Short Radio-Electric Waves, 887

Strachan (C.), awarded a Rayleigh prize of Cambridge University, 408

Stracke (Dr. G.), The Reinmuth Planet, 1932HA, 369

Stratton (Prof. F. J. M.), elected president of the Royal Astronomical Society, 236 Strong (Dr. W. M.), Nutritional Aspects of Depopulation

and Disease, 519 Stuart (C. M.), [obituary article], 194

Stuart (N.), [Prof. G. P. Thomson, C. A. Murison and], Crystalline State of Thin Spluttered Films of Plati-

Stuber (O.), [A. Kailan and], Velocity of Catalysed Hydrogenations, 251

Sturt (Mary), Francis Bacon: a Biography (Review), 490 Sturtevant (Prof. A. H.), Chromosome Mechanics (Review), 5; and T. Dobzhansky, Genetics of certain Chromosome Anomalies in Drosophila melanogaster, 627

Subbaraya (T. S.), [Prof. B. Venkatesachar and], Predictions of Nuclear Moments, 552

Subov (Prof. N. N.), Voyage around Franz Joseph Land, 359

Süe (P.), Determination of Niobium by Orthoxyquinoline, 738

Suga (T.), [Dr. T. Takamine, A. Yamagihara and], Anode Spot in a Neon Tube, 584

Sugden (J. W.), Restriction in Range of the Long-Billed Curlew, 439

Sullivan (J. W. N.), and others, The Outline Series.

11 vols. (Review), 8
Sushkin (late P.), Distribution of Birds in the Russian Altai and North-West Mongolia, 551

Sussmilch (C. A.), Association of the Tertiary Alkaline Rocks of N.S.W. with late Tertiary Tectonic Lines,

Sutton (Dr. L. E.), presented with the Meldola medal of the Institute of Chemistry, 357

Sutton (R. W.), [Prof. G. I. Finch and], Control of Ignition-

Coil Discharge Characteristics, 374
Sutton (T. C.), [H. R. Ambler and], Detection of Traces
of Carbon Monoxide in Air, 766

Svedberg (Prof. The), and Astrid Hedenius, Molecular Weights of the Blood Pigments of the Invertebrates, 325; and Eriksson, Hæmocyanin of Octopus vulgaris,

137 Svensson (E.), Isotope Effect in the Spectrum of Cadmium Hydride, 28

Swietoslawski (W.), and E. Wardzinski, The Ternary

Heteroazeotrope, 887 Szent-Györgyi (Prof. A.), Chemical and Biological Effects of Ultra-Sonic Radiation, 278; Identification of Vitamin C, 225; and Prof. W. N. Haworth, 'Hexuronic Acid' (Ascorbic Acid) as the Antiscorbutic Factor, 24

Tait (T.), [Prof. J. Kendall, W. W. Smith and], Calcium Isotope with Mass 41 and the Radioactive Half-Period of Potassium, 688

Takamine (Dr. T.), T. Suga and A. Yanagihara, Anode Spot in a Neon Tube, 584

Taki (I.), Abnormal Chitons, 367

Takvorian (S.), [M. Curie and], Radioactivity of a Neodymium-Samarium Fractionation, 702

Talley (S. K.), [Grinnell Jones and], Viscosity of Aqueous Solutions, 475

Talman (C. F.), Minimum Climatic Temperature, 431 Tamura (T.), and Y. Hada, Growth of Corals, 172

Taverner (E.), and others, Salmon Fishing (Review), 42 Tawney (Prof. R. H.), Land and Labour in China (Review), 712

Taylor (Prof. H. S.), presented with the Mendel medal of Villanova College, 759

Taylor (H. McC.), elected a research fellow of Clare

College, Cambridge, 736
Taylor (H. M.), and Mott (N. F.), Internal Conversion of

Y-Rays, 99

Taylor (J. E.), [Prof. R. Whiddington and], Inelastic Scattering of Electrons in Helium at Zero Angle, 814 Taylor (Dr. Monica), Sir Bertram Windle: a Memoir (Review), 307

Taylor (Sir William), [death], 195 Taylor and Maccoll, Motion of a Cone at High Speeds in Air, 552

Tckekirian (A.), [J. Cantacuzène and], Presence of Vanadium in certain Tunicates, 35

Teegan (J. A. C.), Amplification of the Ionisation pro-

duced by Radioactive Sources, 277 Teissié-Solier (M.), [P. Dupin and], Vortices produced by Obstacles revolving round an Axis Parallel to the General Direction of Flow, 179

Tellegen (B. D. H.), Interaction between Radio-Waves? 840

Terada (Torahiko), Image of the Physical World in Cinematography, 358; and T. Utigasaki, Forest Fires and Weather, 161

Terbea (E.), Vox Naturæ (Review), 491

de Terra (Dr. H.), Geologische Forschungen im westlichen K'unlun und Karakorum-Himalaya (Review), 600

Terroine (E. F.), P. Mezincesco and Mlle. Simone Valla, Utilisation of Sulphur and Nitrogen from Cystine at the Level of Endogenous Protein Metabolism, 483; and Mlle. Simone Valla, Comparative Value of various Protein Foods in Growth, 375

Thackeray (A. D.), elected to the Sheepshanks exhibition

of Cambridge University, 665

Thayer (Prof. W. S.), Osler and other Papers (Review), 40 Theis (Dr. E.), [Dr. A. Mittasch und], Von Davy und Döbereiner bis Deacon: ein halbes Jahrhundert Grenzflächenkatalyse (Review), 150 Thimann (K. V.), and J. Bonner, Growth Hormone of

Plants (2), 631 Thomas (C. W.), Patents, Trade Marks and Designs: their Commercial Aspect and Development (Review),

Thomas (Dr. H. Hamshaw), The Old Morphology and the New, 47

Thomas (Dr. John), [death], 158; [obituary article],

Thomas (P. E.), [R. Fosse, P. de Graeve and], A New Plant Principle: Uric Acid, 179; Allantoic Acid in the Higher Plants, 702

Thommen (H.), Heavy Current Circuit-Breakers, 103 Thompson (Prof. C. M.), [death], 86; [obituary article], 158; bequest to University College, Cardiff, 372 Thompson (John), the work of, 232

Thompson (J. E.), Gods of Maya and Aztec, 171

Thompson (Dr. J. J.), Logarithmetica Britannica: being a Standard Table of Logarithms to Twenty Decimal Places. Part 5: Numbers 50,000 to 60,000 (Review), 785

Thompson (Miss Laura Maud), Archæology of the Marianas Islands, 623

Thomson (Prof. D. L.), [Prof. J. B. Collop, H. Selye and], Gonad-Stimulating Hormones in Hypophysectomised Animals, 56

Thomson (Prof. G. P.), and C. A. Murison, Electron Diffraction by Films of Grease, 237; N. Stuart, and C. A. Murison, Crystalline State of Thin Spluttered Films of Platinum, 522

Thomson (Sir J. Arthur), [death], 231; [obituary article], 296; Scientific Riddles (Review), 315

Thomson (Prof. J. Millar), [death], 462; [obituary article], 610

Thomson (Dr. J. P.), Australian Water Supply, 797 Thomson (late Dr. J. Stuart), Anatomy of the Tortoise, 396

Thomson (N.), Direct Recording of Relative Intensities by Means of a Microphotometer, 558

Thon (Dr. N.), Neutron, Proton and Positron, 878 Thornton (J. W. de Witt G.), appointed lecturer in pharmacology in Bristol University, 848

Thorndike [Kennedy and], Relativity of Time, 136 Thorpe (Prof. J. F.), elected president of the Institute of Chemistry, 357; extension of appointment to the Safety in Mines Research Board, 91

Thorpe (Dr. W. H.), Tracheal and Blood Gills in Aquatic Insect Larvæ, 549

Thouless (Dr. R. H.), Phenomenal Regression to the Real

Object, 261; 544
Thrupp (T. C.), [S. G. Paine, F. L. Lingood, Freda Schimmer and], Relationship of Micro-Organisms to the Decay of Stone, 178

Thurnwald (Prof. R.), Economics in Primitive Communi-

ties (Review), 315 Ticehurst (Dr. C. B.), A History of the Birds of Suffolk

(Review), 568
Tideswell (F. V.), [T. N. Mason and], Gob Fires, 521 Tiercy (G.), Mean Absorption Coefficient in a Variable Star of the Cepheid Type, 888; Respective Phases of Ionisation Maxima and Light Maxima in a Cepheid, 483; Respective Phases of Minimum Ionisation and Light Minimum in the Cepheids, 815; Variation of Ionisation and the Spectrum Variation of some Cepheids, 215; and P. Berger, Aerological Soundings and the Wind Gradient in Switzerland, 143; and A. Grosrey, Width of a Photographic Stellar Spectrum for Stars of Type A5, 888

Todd, (Dr. C.), The Presence of a Bacteriophage for B. salmonicida in River Waters, 360

Tomaschek (Prof. R.), and O. Deutschbein, Fluorescence of Pure Salts of the Rare Earths, 473

Tompion (Thomas), Memorial to, 497

Tompkins (F. C.), Annual Meeting of the Science Masters' Association, 65

Townson (Thelma), [C. Koller and], Spermatogenesis in

Drosophila obscura, Fallen (1), 447

Trampetti (G.), [A. Ferrari and], Behaviour of Mixtures of Zinc Oxide and Anhydrous Zinc Chloride at High Temperature, 339

Trattner (R.), [G. Kirsch and], Atomic Disintegration

with Emission of Neutrons, 852

Travers (A.), and Lu, Separation of Phosphoric, Arsenic and Vanadic Acids from Aluminium, 595; Volumetric Determination of Lead, 523

Travers (Prof. M. W.), Discovery of the Rare Gases, 65 Tredre (W. H. F.), Lightning Investigation in South

Africa, 19

Trehin (R.), Influence of Temperature on the Adsorption of Aqueous Solutions of Hydrochloric Acid in the Extreme Ultra-Violet, 179

Tremblot (R.), A Variable Star with Eclipses of Short Period, 815

Trescher (M. O.), [J. J. Enright, H. E. Friesell and], Cause and Nature of Dental Caries, 918 Trevithick (Richard), Centenary of the death of, 577;

Memorial Exhibition at the Science Museum, 497 Trillat (J. J.), and L. Leprince-Ringuet, Molecular

Phenomena at the Surface of Separation of Oil and Water, 850; [N. Marinesco and], Action of Ultra-Sounds on Photographic Plates, 667

Trinkler (Dr. E.), Geographische Forschungen im westlichen Zentralasien und Karakorum-Himalaya (Re-

view), 600

Trojan (Prof. E.), Light-Producing Powers of Sponges, 728 Trombe (F.), Preparation of Metallic Neodymium Free from Iron and Silicon, 595

Troup (Prof. R. S.), Problems of British Forestry, 900 Truchet (R.), Oxidation of the True Acetylenic Hydro-

carbons by Selenious Oxide, 595 Trueman (Prof. A. E.), appointed professor of geology in

Bristol University, 848

Tschermak (E.), Petaloid Formation of the Calyx in

Phaseolus multiflorus, 560

Tsi-Ze (Ny), and Chien Ling-Chao, Influence of Pressure on the Photographic Sensibility to various Monochromatic Radiations, 286

Tsuya (Prof. H.), Lowering of the Japan Sea Floor, 368 Tubangui (M. A.), Schistosomiasis in the Philippines, 99 Tuot (M.), [A. Meyer and], Dehydration of some Tertiary Alcohols by Anhydrous Copper Sulphate, 851 Turnbull (C.), Early Days of the Turbine, 20

Turnbull (Prof. H. W.), Matrices and Continued Fractions, 922

Turnbull (R.), [Prof. J. C. McLennan and], Ultra-Violet Absorption Bands of Xenon, 214

Turner (Dr. Dorothy), concerning the review of Badcock and Holmyard's "Electricity and Magnetism for

Beginners'', 686
Turner (F. J.), Piedmontite in Quartz-Muscovite-Schist from the Shotover Valley, Western Otago, New Zealand, 630

Tutton (Dr. A. E. H.), Diamonds (Review), 255; Prof. V.

Goldschmidt, 791 Tyndall (Prof. A. M.), Gaseous Ions, 65

Ubbelohde (A. R.), Reaction Cells in Chain Reactions, 328 Uchiyama (K.), [M. Kinoshita and], Size of Fog Droplets,

Ullyott (P.), [Dr. W. H. Pearsall and], Measurement of Light for Biological Purposes, 694

Unwin (Prof. W. C.), [death], 427; [obituary article],

Urbain (P.), Relative Impermeability of the Plastic Sediments towards Rain Water, Spring Water and various Alkaline Solutions, 739

Ure (S. G. M.), presented with the Osborne Reynolds medal of the Institution of Chemical Engineers, 510

Utigasaki (T.), [T. Terada and], Forest Fires and Weather, 161

Uvarov (B. P.), Phases in South African Locusts, 423 Uwins (Capt. C. F.), awarded the British silver medal of the Royal Aeronautical Society, 685

Valla (Mlle. Simone), [E. F. Terroine and], Comparative Value of various Protein Foods in Growth, 375; [E. F. Terroine, P. Mezincesco and], Utilisation of Sulphur and Nitrogen from Cystine at the level of Endogenous Protein Metabolism, 483

Vallet (P.), Decomposition of some Complex Platinum Compounds at Progressively Increasing Tempera-

tures, 143

Vargha (L.), Triphenylmethyl Derivative of Vitamin C, 363

Vaufrey (H.), [E. G. Gobert and], 'Iberomarusian', 367 Vaufrey (R.), Pleistocene Dating of Gafsa Hills, Tunisia. 659

Vène (R.), Sadi Carnot, 303

Venkatesachar (Prof. B.), and L. Sibaiya, Hyperfine Structure and Nuclear Moments, 844; and T. S. Subbaraya, Predictions of Nuclear Moments, 552

Verhaeghe (J.), Faraday Effect of a Camphor Derivative, 666

Vernon (H. M.), Estimation of Solar Radiation in Relation to its Warming Effect on the Human Body, 737

Vernotte (P.), Natural Convection of Heat in Air when the Heating is very Small, 179 Verver (C. G.), [R. N. J. Saal and], Measurement of Vola-

tility Range of Lubricating Oils, 661

Vick (C.), New Proper Motions of Stars from Bergedorf Observatory, 625 Vierthaler (R.), [L. O. Koblmuller and], Transferring

Single-Cell Cultures to Solid Nutrient Media, 287 Vigoureux (P.), and S. Watts, Temperature Coefficient

of the Weston Standard Cell, 374 Viljoen [Schonland and], Penetrating Rays from Thunder-

clouds, 916 de Villiers (Prof. C. G. S.), The 'Tail' of the Male American

Toad, Ascaphus, 692

Virtanen (Prof. A. I.), Nitrogen-Uptake of Plants, 534 Vogel (Dr. A. I.), Substituted $\beta\gamma$ -Diphenyladipic Acids and Derivatives of Chrysene, 402; G. H. Jeffery, Limiting Mobilities of some Monovalent Ions and the Dissociation Constant of Acetic Acid at 25°, 27

Volmar and Betz, Emetic Derivatives of Lactic Acid, 410 Volterra (E.), 'Linked' Elasticity and its Mathematical Representation, 107

Vonwiller (O. U.), Diffraction Gratings Used with Grazing Incidence, 668

Vossler (K.), translated by O. Oeser, The Spirit of Language in Civilization (Review), 152

Waddington (C. H.), Heterogony and the Chemical Ground-Plan of Animal Growth, 134; Induction by Coagulated Organisers in the Chick Embryo, 275 van der Waerden (Prof. B. L.), Die gruppentheoretische

Methode in der Quantemmechanik (Review), 531 Wagler (E.), [G. Grimpe und], Herausgegeben von G. Grimpe, Die Tierwelt der Nord- und Ostsee. Lief. 21.

Teil 1.d2 und Teil 2.g (Review), 116 Wahl (A. M.), [Dr. A. Nádai and], Plasticity: a Mechanics of the Plastic State of Matter (Review), 383

Wakker, Gold-bearing Strata of the Region of St.-Yrieix (Haute-Vienne), 483 Waksman (Prof. S. A.), Principles of Soil Microbiology.

Second edition (Review), 316

Walker (Dr. F.), Differentiated Sills in Skye, 440

Walker (Sir Gilbert), Correlation of Meteorological Data, 284

Walker (Sir James), awarded the Gunning Victoria Jubilee prize of the Royal Society of Edinburgh, 359

Walker (W. C.), The "Leeds Portrait" of Joseph Priestley,

Walsh (Dr. J. W. T.), Everyday Photometry with Photoelectric Cells, 660

Walters (M. H. H.), awarded a Rayleigh prize of Cambridge University, 408

Walton (Dr. E. T. S.), [Dr. J. D. Cockcroft and], Disintegration of Light Elements by Fast Protons, 23 Wambacher (Hertha), [Marietta Blau and], Photographic

Detection of Protons Liberated by Neutrons (2), 287 Warburg (E. F.), [Sir Oscar Warburg and], Cultivated Oaks, 905

Warburg (Sir Oscar), and E. F. Warburg, Cultivated Oaks, 905

Ward (Dr. A. M.), [Dr. A. D. Mitchell and], Modern Methods in Quantitative Chemical Analysis (Review),

Ward (Prof. H. B.), elected permanent secretary of the American Association, 304

Wardzinski (E.), [W. Swietoslawski and], The Ternary Heteroazeotrope, 887

Warren (A. E.), Xanthophores in Fundulus, 287

Waterfield (Dr. R. L.), Recent Observations of Mars, 845 Waterhouse (G. A.), Australian Hesperiidæ (3), 631

Watkins (A.), Archaic Tracts Round Cambridge (Review), 491

Watson (A. G. D.), elected a fellow of King's College, Cambridge, 593

Watson (E. L. G.), The Common Earth (Review), 636 Watson (F. R. B.), Production of a Vacuum in an Air Tank by means of a Steam Jet, 369

Watson (Prof. G. N.), The Marquis and the Land Agent:

a Tale of the Eighteenth Century, 67 Watson (R. A.), and N. R. McCurdy, Cyclone Season in the South Indian Ocean, 517; Pilot Balloon Observations at Mauritius, 626

Watson (W.), Textile Design and Colour: Elementary Weaves and Figured Fabrics. Third edition (Review),

Watt (R. A. Watson), appointed superintendent of the Radio Department of the National Physical Laboratory, 720; Cathode Ray Oscillography, 66; and L. Bainbridge-Bell, Recording Wireless Echoes at the Transmitting Station, 657

Watts (Dr. J.), [death], 266 Watts (S.), [P. Vigoureux and], Temperature Coefficient of the Weston Standard Cell, 374

Wayland (E. J.), Archæological Discoveries in Uganda, 439; and M. C. Burkitt, Stone Age Culture in Uganda, 730

Wayling (H. G.), Priestley's Associations with London, 350

Webb (Beatrice), [Sidney Webb and], Methods of Social Study (Review), 748

Webb (Sidney), and Beatrice Webb, Methods of Social

Study (Review), 748 scheider (R.), Photochemical Kinetics; Photo-Wegscheider chemical Transformation of O-Nitrobenzaldehyde (II), 287

Wehmeyer (L. E.), The Genus Diaporthe, Nits, 880 Weigle (Prof. J.), Dispersion in the Hertzian Domain, 144; and R. Luthi, Debye's Dispersion of Nitrobenzene, 327

Weir (Dr. J.), [Prof. E. B. Bailey and], Submarine Faulting in Kimmeridgian Times, 244

Weiss (Prof. F. E.), elected president of the Linnean Society of London, 798

Welch (H. J.), and Dr. G. H. Miles, Industrial Psychology

in Practice (Review), 567 Welch (M. B.), Daily Shrinkage and Swelling of Wood; Longitudinal Variation of Timber during Seasoning, 667; Moisture Content of Wood, 524

Wellard (R.), [P. Mondain-Monval and], Direct Oxidation of Acetylene by Air, 851

Wells (A. L.), The Shrimp Industry of Leigh-on-Sea, 270 Wells (G. P.), [D. I. Clements, N. H. Howes and], Is Plasticine Edible?, 330

Wells (H. G.), After Democracy: Addresses and Papers on the Present World Situation (Review), 183

Wenckebach (K. F.), Mechanism of Sudden Heart-Failure in cases of Beriberi, 251

Went (Prof. F. A. F. C.), elected a foreign member of the Royal Society, 798

Werner (F.), A New Snake from the Cyclades Islands. 287; and R. Ebner, Results of a Zoological Expedition to Morocco (1930): (5) Scorpions, 287

Werner (Dr. S.), Elastic Electron Scattering in Gases, 726 Westcott (G. F.), Science Museum Handbook of Pumping Machinery, 542

Westphal (Prof. W. H.), Physik: ein Lehrbuch für Studierende an den Universitäten und technischen Hochschulen. Dritte Auflage (Review), 895

Weston (W. A. R. Dillon), Sporulation of Helminthosporium avence in Artificial Culture, 435

Wheeler (Dr. R. E. M.), Verulamium, 1932, 300 Whiddington (Prof. R.), Electron Polarisation?, 908; and T. Emmerson, Scattering of Electrons in Helium, 814; and F. C. Poultney, Energy of the Beams in Electron Diffraction, 814; and J. E. Taylor, Inelastic Scattering of Electrons in Helium at Zero Angle, 814 Whipple (Dr. F. J. W.), Air Waves from Experimental

Explosions, 138; Relations between the Combination Coefficients of Atmospheric Ions, 522; The Wetand-Dry Bulb Hygrometer, 374

Whipple (R. S.), Evolution of the Microscope (Review), 219 White (Dr. C. M.), awarded the Moulton medal of the Institution of Chemical Engineers, 510

White (E. L. C.), [J. A. Ratcliffe and], Fine-Structure of the Ionosphere, 872; Heaviside Layer, 807; [C. T. R. Wilson, J. A. Ratcliffe and], An Automatic Recording Method for Wireless Investigations of the Ionosphere, 522

White (J. W.), bequest to Bristol University, 299 White (Dr. J. H.), The History of the Phlogiston Theory (Review), 531

Whitehead (Prof. A. N.), Adventures of Ideas (Review), 746

Whitehead (S.), edited, with a preface, by E. B. Wedmore, Dielectric Phenomena. Solid Dielectrics (Review), 384 3: Breakdown of

Whitelegge (Sir Arthur), [death], 646 Whittaker (Prof. E. T.), Variation Problems for a Symmetrical Region, 472

Whytlaw-Gray (Prof. R.), [K. G. Denbigh and], Higher Homologues of Sulphur Hexafluoride, 763

Wiborg (E.), [A. Stock and], Electrolysis of Boron Hydrides, 212

Wick (G. C.), Motion of an Electron in a Crystalline Lattice, 71

Wickes (W. H. V.), [death], 231 Wieland (Prof. G. R.), Origin of Angiosperms, 360 Wiersma (E. C.), [Prof. W. J. de Haas, Prof. H. A. Kramers and], Temperature below 0·27° K., 719

Willding-Jones (C. L.), appointed assistant superintendent of agriculture, Gold Coast, 871 Willemse (Dr. W. A.), Constitution-Types in Delinquency:

Practical Applications and Bio-physiological Foundations of Kretschmer's Types (*Review*), 188 Williams (Dr. A. C.), [Dr. F. C. Shrubsall and], Mental

Deficiency Practice: the Procedure for the Ascertainment and Disposal of the Mentally Defective (Review), 456

Williams (A. F.), The Genesis of the Diamond. 2 vols. (Review), 255

Williams (A. S.), Seasonal Changes on Jupiter, 808 Williams (Prof. B.), Great Britain and the Conferences of the Institut Colonial International, 683

Williams (D. L. H.), awarded a silver medal of the Royal Aeronautical Society, 510

Williams (Dr. E. J.), Spectrum and Latitude Variation of Penetrating Radiation, 511 Williams (F. E.), Depopulation in Papua, 519

Williams (R. T.,, [J. Pryde and], A New Unsaturated Derivative of Glucuronic Acid, 57

Williams (W. E.), and A. Middleton, Fine Structure of the Resonance Ag I Lines, 692

Williamson (A. V.), elected reader in geography in Leeds University, 372

Williamson (E. B.), [death], 646

Willimott (S. G.), Solanine Poisoning, 813 Willis (Prof. Bailey), Earthquakes in the Holy Land: a Correction, 550; Isthmian Links, 696

Wilson (A. H.), appointed a university lecturer in mathematics in Cambridge University, 848; awarded the Adams prize of Cambridge University, 736 Wilson (Dr. C. B.), The Copepods of the Woods Hole

Region, Mass., 698 Wilson (C. T. R.), J. A. Ratcliffe and E. L. C. White, An Automatic Recording Method for Wireless In-

vestigations of the Ionosphere, 522 Wilson (Dr. E. D.), [Dr. V. K. Zworykin and], Photocells and their Application. Second edition (Review), 860

Wilson (G. F.), Pollination in Orchards, 659 Wilson (J. C.), High-Frequency Electric Discharge in Gases, 546

Wilson (P. A.), [Prof. A. M. Carr-Saunders and], The Professions (Review), 863

Wilson (V.), The Corallian Rocks of Yorkshire (1), 595 Winckworth (R.), The British Marine Mollusca, 334 Wingfield-Stratford (Dr. E.), They That Take the Sword

(Review), 635

Winton (Dr. A. L.), and Dr. Kate Barber Winton, The Structure and Composition of Foods. Vol. 1 (Review), 316

Winton (Dr. F. R.), recommended as reader in physiology

in Cambridge University, 736 Winton (Dr. Kate Barber), [Dr. A. L. Winton and], The Structure and Composition of Foods. Vol. 1 (Review), 316

Wodlinger (M. H.), Bibliographical Survey of Vitamins, 1650-1930: with a section on Patents (Review), 568 Wohl (Mlle.), [Mme. Ramart-Lucas and], Colour and Structure of Amides, 286

Wojtusiak (R. J.), Antagonistic Action of Ultra-Violet Rays on the Phototaxy of Daphnia magna Straus,

Wolf (Prof. Max), [obituary article], 353

Wolff (Prof. G.), Leben und Erkennen: Vorarbeiten zu einer biologischen Philosophie (Review), 529

Wollman (Mme. E.), [J. Basset, M. A. Macheboeuf, M. Bardach and], Biological Effects of Ultra-Pressures, 774

Wong (Dr. K. Chimin), and Dr. Wu Lien-Teh, History of Chinese Medicine: being a Chronicle of Medical Happenings in China from Ancient Times to the Present Period (*Review*), 527

Wood (H. O.), and C. F. Richter, Californian Earthquake of March 10, 686

Wood (J. G.), [T. G. B. Osborn, T. B. Paltridge and], Growth and Reaction to Grazing of the Perennial Saltbush, Atriplex vesicarium, 631

Wood (Prof. R. W.), Remarkable Optical Properties of the Alkali Metals, 582

Wood (W. A.), Selective Lattice Distortion in Wires under Torsion, 842

Woodward (Sir Arthur Smith), The Second Piltdown Skull, 242

Woolf (Arthur), A Cornish Engineer, Rhys Jenkins, 124

Woolley (C. L.), Ritual at Ur, 356 Woolley (Dr. R. v. d. R.), Microphotometry of the Solar Spectrum from 4040 to 4390 A., 917

Wooster (Dr. W. A.), Electrometer Triode in the X-Ray Ionisation Spectrometer, 545

Worcester (Marquis of), work of, 267

Wordie (J. M.), and Dr. S. Kemp, Ice in the Weddell Sea, 916

Worley (Prof. F. P.), Forest Fires in Relation to Soil Fertility, 787

Wright (A. R.), [obituary article], 49 Wright (E. E.), The Kerr Cell, 702

Wright (Dr. Orville), awarded the Franklin medal of the Franklin Institute, 870

Wright (Prof. W.), Sir Bertram Windle (*Review*), 307 Wynne-Jones (W. F. K.), Acid Strength and its Dependence upon the Nature of the Solvent, 374

Xavier (R. P.), retirement from the directorship of the Brazilian Meteorological Service, 581

Yanagihara (A.), [Dr. T. Takamine, T. Suga and], Anode Spot in a Neon Tube, 584

Yapp (W. B.), Teaching of Biology, 203

Yetts (W. P.), appointed professor of Chinese art and archæology at the Courtauld Institute of Art, 336 Yonge (Prof. C. M.), Crystalline Style in Gastropods, 915

Yosida [Obata, Morita and], Aeroplane Noises, 208 Younge (J. Z.), Physiology of the Iris (1), 178

Young (L.), presented with the Sir Edward Frankland medal and prize of the Institute of Chemistry, 357 Young (M.), [Corisande Smyth and], Facial Growth in Children, 29

Zagar (F.), Increase in Mass of a Planet by the Effect of Cosmic Dust (1), 775

Zeiss (Carl), New Lens of High Rapidity for Cinematography, 475

Zellner (J.), [E. Fraenkel and], Comparative Plant Chemistry (24), 888; [L. Lukacs and], Chemistry of the Higher Fungi (22), 888

Zworykin (Dr. V. K.), and Dr. E. D. Wilson, Photocells and their Application. Second edition (Review), 860

Zilva (Dr. S. S.), Indophenol Reducing Capacity of Lemon Juice and its Fractions in Relation to Vitamin C Activity, 363; [M. E. F. Crawford, E. O. V. Perry and], Vitamin Content of Australian, New Zealand and English Butters, 770

Zirkle (Dr. C.), Discovery of Sexuality in Plants, 392 Zuckerman (Dr. S.), Ancient and Modern Man, 367

TITLE INDEX

Academic: Assistance Council, Formation of an, 793; Freedom, Nationalism and, 853

Accidents, The Causes of, E. Farmer (Review), 895
Acenaphthene: Catalytic Oxidation of, H. Paillard and
R. Duckert, 815; Chlorination of, H. Paillard and P. Favarger, 815 Acetic Acid: at 25°, Dissociation Constant of, Limiting

Mobilities of some Monovalent Ions and the, Dr. A. I. Vogel and G. H. Jeffery, 27; in Water, Dissociation of, Dr. C. W. Davies and W. H. Banks, 328

Acetylated Aldonic Nitriles, A Rule for the Rotatory
Direction of the, Dr. V. Deulofeu, 548

Acetylene: Direct Oxidation of, by Air, P. Mondain-Monval and R. Wellard, 851; Linkage, Mlle. B.

Greedy, 774
Acetylenic Hydrocarbons, Oxidation of the true, by Selenious Oxide, R. Truchet, 595

Acid Strength and its Dependence upon the Nature of the Solvent, W. F. K. Wynne-Jones, 374

Acoustics, Architectural, Prof. V. O. Knudsen (Review),

Acquired Characters, Inheritance of, Dr. R. L. Jenkins, 95 Actinomycetales, The, H. L. Jensen (4), 631

Adenine: Hydrochloride, Crystal Structure of Vitamin B₁, and of, J. D. Bernal and D. Crowfoot, 911; Vitamin B₄ and, R. D. Heard, H. W. Kinnersley, J. R. O'Brien, Prof. R. A. Peters, and V. Reader, 617 Adiabatic Lapse-Rate for Dry and Saturated Air, D.

Brunt, 814

Adrenal Gland, Vitamin C in the, G. Bourne, 874 Advanced Study, Institute of: Appointments at the, 88; Work of the, 89

Egiceras majus, Gaertn., Embryology and Seedling Development of, Gladys Carey and Lilian Fraser, 287 Aerological Soundings and the Wind Gradient in Switzer-

land, G. Tiercy and P. Berger, 143 Aeronautics: and the Royal Air Force, Sir Philip Sassoon, 428; Contribution of Science in the Development

of, Prof. S. Pippard, 65 Aeroplane Noises, Obata, Morita and Yosida, 208 Africa, France and Great Britain in, A. Sarraut, 831 African Animals, Two, Past Wanderings in Europe of, Dr. R. F. Scharff, 922

Atlas of England and Wales, An, M. Agricultural: Second edition (Review), 420; Botany, Messer. National Institute of, 430; Crops, Varieties of, 594; History in Germany, 721; Research, Farmers' Guide to, 1931, 163; in Great Britain, 90; Science, Economic Value of, Lord Bledisloe, 197; Seeds,

Sowing, 248; Societies, 869 Agriculture and Milk Supply, Prof. H. E. Armstrong, 605

Ailsa Craig, Bird Life on, 903

Akron, Loss of the, 499

Air-Benzene Mixtures, The Combustion Pressures in Closed Vessels of, Y. Laure, 447

Aircraft: in relation to Petroleum Technology, 160;

Two-seater, Height Record, H. J. Penrose, 160

Air: -Map, An Experimental, 357; Movements produced by the Flapping Wing of a Bird or Insect, A Hot Wire Apparatus for the Study of the, A. Magnan and C. Magnan, 922; Speed Record, New, Warrant-Officer F. Agello, 580; Waves from Experimental Explosions, Dr. F. J. W. Whipple, 138
Alai-Pamir Expedition 1928, Wissenschaftliche Ergeb-

nisse der, Herausgegeben von Dr. H. v. Ficker und Dr. W. R. Rickmers. Teil 1: Geodätische, Topographische und Glaziologische Ergebnisse. Von Dr. Geodätischer und R. Finsterwalder. Band 1: glaziologischer Teil. Band 2: Kartenbeilagen. Teil 2: Geologische Untersuchungen im nordwestlichen Pamir-Gebiet und mittleren Transalai. Von Dr. L. Band 1: Stratigraphie (ausschliesslich

Quartar), Tektonik. Band 2: Quartäre Ablagerungen, Morphologie. Teil 3: Beiträge zur Faunistik des Pamir-Gebietes. Von Dr. W. F. Reinig. Band 1: Ökologie und Tiergeographie. Band 2: Systematischer Teil (Review), 744

Alaska: Destruction of Predatory Animals in, 695;

Excavation in, 161

Alchimy", "Ordinall of: The Authorship of the, Dr. M. Nierenstein and P. F. Chapman, 520; Thomas Norton and the, Dr. E. J. Holmyard, 520 Alcohol: In Chase of Truth of, Prof. H. E. Armstrong,

53; -Petrol as a Motor Fuel, 464; Power, 341; through the Ages, Dr. E. F. Armstrong, 810
Alcoholic Fermentation, Prof. A. Harden, 756
Aldehydes, Synthesis of, G. Darzens and M. Meyer, 483
Algebra, Advanced, C. V. Durell. Vol. 1 (Review), 319 Algebraic Surface, Certain Invariant Series of Groups of Points on an, F. Enriques, 559

α-Alkyladipic Acids, A Synthesis of, A. Franke, A. Kroupa and S. Hadzidimitriu, 287
Alkylanilines with Tertiary Alkyl Groups, Dr. W. J.

Hickinbottom, 762

Alkyl Groups, Volumes of, and their Orienting Powers, Dr. R. J. W. Le Fèvre, 655 Alkylmagnesium Iodides, Action of, on (1:9)-Ben-

zanthrone-(10), G. Charrier and Elisa Ghiga, 107

Alkali: - and Barium-Silver Chlorides, J. Hoffmann, 704; Metal Halides, Crystals of, Plasticity and Hardness of, Prof. K. Przibram, 560; Metals: Examination of Minerals for the, R. Bossuet, 923; Remarkable Optical Properties of the, Prof. R. W. Wood, 582; in the Oxyacetylene flame, Photographic Sensibility of the Lines of the, R. Bossuet, 482

Allantoic Acid in the Higher Plants, Rôle of, R. Fosse, P. de Graeve and P. E. Thomas, 702 Allantonic Membrane of the Chick Exposed to X-Radia-

tion, W. Moppett, 667

Allium, Structure and Division of Somatic Chromosomes

in, Prof. T. K. Koshy, 362 Alloys, Structure of, Prof. W. L. Bragg, 749

Alpha Particle, The Entry of the Disintegrating, into the Nitrogen Nucleus, etc., E. C. Pollard, 375 Aluminium: Alloys at High Temperatures, Fatigue-

Resisting Properties of, J. W. Cuthbertson, 660; Alloys Containing Iron as a Model of the Electrolytic Purification of Molten Aluminium from Iron, Electrolysis of, R. Kremann and L. Lammermayer, 251; -Beryllium Alloys, Ageing of the, after Tempering, C. Matignon and J. Calvet, 886; Hydride, Band Spectrum of, Predissociation and Pressure Effects in the, Prof. E. Hulthén and R. Rydberg, 470; Hyperfine Structure in, R. Ritschl, 58; in Food-Stuffs, Dr. Beal, 432; Separation of Phosphysical Research of Phosphysic phoric, Arsenic and Vanadic Acids from, A. Travers and Lu, 595; Single Crystals, Recrystallisation Power and Shear Hardening in, Dr. W. G. Burgers, 326

American: Academy of Arts and Sciences, Award of the Rumford medal to Dr. H. Shapley, 871; Association, Election of Officers, 164; Prof. H. N. Russell elected president, and Prof. H. B. Ward elected permanent secretary; award of the American permanent secretary; award of the American Association prize to Dr. H. Eyring, 304; Pronghorn increasing in numbers, 849

Amides, Colour and Structure of, Mme. Ramart-Lucas and Mlle. Wohl, 286

Amino-Acids, Proteins and Proteolytic Enzymes, Prof. M. Bergmann, 662; 698

Aminoazo-Compounds, Chemical Changes Involved in the Formation of the, J. C. Earl and N. F. Hall (2), Ammonia, Aqueous Solution of, Photolysis of, L. Gion, 410 Ammonites, Phylogeny of, late J. Perrin Smith, 368 Amor, The Minor Planet, 476 Amphibia, Internal Ear of, C. Guareschi, 659

Amphibians and Fishes, Chromosomes of, Dr. S. Iriki, 659 Amphioxus, Nephridia of, Prof. E. S. Goodrich, 439 Analysts: Public, and the Food Industry, Dr. J. T. Dunn

(Review), 822; Society of Public, and other Analytical Chemists, Dr. B. Dyer and Dr. C. A. Mitchell (Review), 822

Ancestor Spirits among the Nuba, D. Hawksworth, 403 Ancient Monuments of England, The, Sir Charles Peers,

253

Andalusite Twin, An, P. Aloisi, 851 Andaman and Nicobar Islands, Census Report, 199 Andes in Ecuador, Age of the, 589

Angiosperms, Origin of, Prof. G. R. Wieland, 360

Animal: Growth, Heterogony and the Chemical Ground-Plan of, C. H. Waddington, 134; Life, Components of the Atmosphere, and Synthetic Gases in Relation to, Prof. J. W. Hershey, 238; Lore in English Literature, Dr. P. A. Robin (*Review*), 257; Myths (*Review*), 257

Animals, Vertebrate, Collecting and Preserving, Methods

of, R. M. Anderson, 906

Ankole, South-West, and Adjacent Territories, Geology of, A. D. Combe, A. W. Groves and W. C. Simmons, 172 Annuaire Astronomique Camille Flammarion, 1933, 369 Annual Register, The, 1932, Edited by Dr. M. Epstein (Review), 824

Annuario del Observatorio Astronomico de Madrid, 1933.

405

Anode Spot in a Neon Tube, Dr. T. Takamine, T. Suga and A. Yanagihara, 584

Antarctic: Australian Sector in the, 577; British Claims

in the, 270; Geology and Glaciation, 847 Antennal Secretion in Insecta, S. Maulik, 516 Anthropological: and Ethnological Sciences,

national Congress of, 719; Fieldwork, Aeroplane and Camera in (*Review*), 599; Research, Aims of, Prof. F. Boas, 196; Teaching and Research in, Prof. R. Firth, 540

Anthropology, Early Publication in, 53 Anticyclones, Intrusion of Air into, C. S. Durst, 737

Antimony: Iodide, Precipitation of, and its Hydrolysis, F. François, 923; Molecular Fluorescence of, Jean

Genard, 132; Poisoning from Enamelled Vessels, 915 Anti-Veterinary Propaganda, 90 Aphanapteryx, The Dodo and the, G. Renshaw, 728 Aphids, Time of Embryonic Segregation in, A. F. Shull,

Apples, Colour of, Improvement of, H. Goude, 331

April Showers, 481

Aquaria, 270

Aquatic Insect Larvæ, Tracheal and Blood Gills in. Dr.

W. H. Thorpe, 549

Aqueous: Electrolytes, Determination of the Dielectric Constants of, Application of the Resonance Method to the, M. Jezewski, 448; Solutions, Viscosity of, Grinnell Jones and S. K. Talley, 475 Archaic Tracks Round Cambridge, A. Watkins (Review),

491

Archæological Exploration in Oaxaca, Mexico, Miss Emma Reh, 509

Archæology: and Folk-Tradition, Prof. H. J. Fleure, 431; Ethics and, 561

Archimedean Bodies in Spaces of Several Dimensions, K. Klemenz, 703

Architectural Graphic Standards: for Architects. and Engineers, Decorators, Builders C. G. Ramsey and H. R. Sleeper (Review), 224

Arctic Ice passes Cape Farewell, 105

Argentine, Archæological Researches in the, late Dr. E. Boman, 63

Aristogenesis, Prof. H. F. Osborn, 768

Armstrong College, Report for 1931–32 of the Standing Committee for Research, 920

Arsenic, Nuclear Moment of, M. F. Crawford and A. M. Crooker, 655

Arsenolite and Senarmontite, Orientation of, by Mica, R. Hocart, 851

Arts in Industry, The, 73

Ascaphus, The 'Tail' of the Male American Toad, Prof. C. G. S. de Villiers, 692

Ascorbic Acid: Prof. A. Szent-Györgyi and Prof. W. N. Haworth, 24; Constitution of, Dr. E. L. Hirst, E. G. V. Percival and F. Smith, 617; Vitamin C and, A. L. Bacharach, 364; Estimation and Distribution of, and Glutathione in Animal Tissues, T. W. Birch and Dr. W. J. Dann, 469

Asia: Central, Research in, 705; Innermost (Review), 415; South-East, Prehistory of, Dr. F. Sarasin, 879 Ashmolean, The Old: Oxford, T. E. James, 716; The

250th Anniversary of, 793

Asparagus Growing, Bulletin on, 722 Aspergillin: The Pigment of the Spores of Aspergillus niger, A. Quilico and A. di Capua, (1), 851, (2), 924 Aspergillus niger, Action of Certain Alkaloids on the Development of, G. Mezzadroli and A. Amati, 339

Astrolabes: (Review), 819; of the World, The, Dr. R. T.

Gunther. 2 vols. (Review), 819
Astronomical: Notes for January, 31; February, 137; March, 281; April, 476; May, 625; June, 769; Radiative Stability, The: Sir Joseph Larmor, 805; Dr. G. C. Simpson, 875

Astronomy, Observational and Theoretical, Prof. H. H.

Plaskett, 648
Athenæum Club, Election to: Dr. S. Courtauld, E. G. V. Knox and Dr. N. V. Sidgwick, 200; Wing-Commdr. the Hon. Maurice Baring, Prof. W. Langdon Brown and Prof. H. H. Dodwell, 510

Äthiopen des Westens: Forschungsreisen in Portugiesisch-Guinea, Dr. H. A. Bernatzik. Band 1 und 2

(Review), 599

Atlantic Ocean: Southern, Life in the, 34; South-West Influence of the Pacific on the Circulation in the,

A. J. Clowes, 189

Atmosphere: Components of the, and Synthetic Gases in Relation to Animal Life, Prof. J. W. Hershey, 238; in the Lyons Region, Transparency of the, P. Flagolet, 339

Atmospheric: Ionisation, Influence of Condensation Nuclei and Dust Particles on, Prof. P. J. Nolan, 922; Ions, Relations between the Combination Coefficients of, F. J. W. Whipple, 522; Ozone, Estimation of, by Visual Photometry, J. Gauzit, 36; Pollution, Investigation of, 758 Atmospherics in Australia, G. H. Munro and L. G. H.

Huxley, 592

Atom: The Interpretation of the, Prof. F. Soddy (Review), 4; Properties of the, Prof. G. Hevesy (Review),

Atomic: and Molecular Forces of Chemical and Physical Interaction in Liquids and Gases, and their Effects, Prof. R. D. Kleeman (Review), 80; Charge, Determination of, Early History of the, 569; Disintegration with Emission of Neutrons, G. Kirsch and R. Trattner, 852; Number Zero, The New Element of, the Neutron and Neuton, Prof. W. D. Harkins, 23; Physics, Recent Advances in, Prof. G. Castelfranchi. Translated by Dr. W. S. Stiles and Dr. J. W. T. Walsh. 2 vols. (Review), 319; Weights, International, 405

Atoms, Molecules and the Atmosphere, Prof. S. Chapman, 271

Atriplex vesicarium, Growth and Reaction to Grazing of the Perennial Saltbush, T. G. B. Osborn, J. G. Wood and T. B. Paltridge, 631

Audio Frequency Atmospherics, Effects of Solar Eclipse on, E. T. Burton and E. M. Boardman, 81

Auroras, Low, Dr. G. C. Simpson, 828

Australia: Anthropological Teaching and Research in, Prof. R. Firth, 540; Atmospherics in, G. H. Munro and L. G. H. Huxley, 592; The Cultural and Racial Clash in, Dr. A. P. Elkin, 520; Improvement of Beef Cattle in, 576; Possible Correlation of Certain Pre-Cambrian Granites of, W. R. Browne, 667; Prickly Pear in, Eradication of, 613

Australian: Dairy Cattle Research Council, 356; Entomology, Bibliography of, 1775-1930, 466; periidæ, G. A. Waterhouse (3), 631; Opals, β-Cristobalite in, Occurrence of, F. P. Dwyer and D. P. Mellor, 524; Timbers, Chemistry of, 333; Water Supply, Dr. J. P. Thomson, 797

Autogiro, New, 646

Avitaminosis and Intoxication, M. Mitolo (1), 411; (2),

β-Rays: Relation between the Absorption of, by Organic Compounds and the Molecular Structure of the Latter, G. Fournier and M. Guillot, 447; Absorption of the, by Matter, G. Fournier and M. Guillot, 179

B. salmonicida in River Waters, Presence of a Bacterio-

phage for, Dr. C. Todd, 360

Bacon: Francis, a Biography, Mary Sturt (Review), 490 Bacterial: Detoxification, R. S. Harris and J. W. M. Bunker, 244; Growth Curve, The, and the History of Species, Dr. A. S. Corbet, 61; Proteases, Influence of Hydrocyanic Acid on, G. Gorbach and A. Schonbeck, 287

Baeyer, Adolf, als Lehrer und Forscher, Prof. H.

Rupe, 294

Bananas, Hybrid, E. E. Cheeseman, 552

Band Spectra which appear near Visible Triplet Lines of Mercury, E. Matuyama, 58

Baltic, Inertia Currents in the, T. Gustafson and B. Kullenberg, 586

Bari Rain-Maker, Burial of a, A. C. Barton, 29

B Vitamins, Rôle of the, in the Utilisation of the Glycides by the Organism of the Pigeon, R. Lecoq, 630

Barium: Azide, Acceleration of the Decomposition of Crystals of, by the Emission from Radium Emanation, Prof. W. E. Garner and C. H. Moon, 513; Oxide (BaO), Band Spectra of, P. C. Mahanti, 402

Barley: 409; Egyptian Neolithic, A. Jackson, 652 Barnacles, Young, Change their Coats, 594

Barracuda Fisheries, 773 Bast-Sap, Prof. H. H. Dixon, 367

Batholiths and Ore Deposits, Prof. W. H. Emmons, 730 Bathysphere, Deep Sea Dives in the, Dr. W. Beebe, 776 Battersea Power Station, I. V. Robinson, 720

Bau der Erde, Der, und die Bewegungen ihrer Oberfläche: eine Einführung in der Grundfragen der allgemeinen Geologie, Prof. W. von Seidlitz (Review), 318

Bauxite and Aluminous Laterite, Dr. C. S. Fox. Second edition (Review), 347

'Bayoud', a Disease of the Date Palm, Etiology of, R. Maire, E. Foëx and G. Malençon, 922

Bechuanaland, Chastity in, Dr. I. Shapera, 516

B.D.H.: Guide to the B.P. 1932, 128; Products, 834 Belfast University, Dr. K. G. Emeléus appointed professor of experimental physics, 628

Belgrade, University of, Observatory of the, Annuaire of the, for 1933, 518

Bengali Chemist, Life and Experiences of a, Sir Prafulla Chandra Rây (Review), 672

Bentham's Theory of Fictions, C. K. Ogden (Review), 152

Benzene, Slow Combustion of, J. Amiel, 774

p-Benzoquinone Crystal, Orientations of Molecules in the, K. S. Krishnan and S. Banerjee, 653

Beriberi, Mechanism of Sudden Heart-Failure in Cases of, K. F. Wenckebach, 251 Berkeley, Prof. G. Dawes Hicks (*Review*), 322

Bērūnī, Das Vorwort zur Drogenkunde des. Eingeleitet, übersetzt und erlautert von M. Meyerhof (Review), 491

Beryllium and Helium, Lord Rayleigh, 724

Bessemer gold medal of the Iron and Steel Institute, Award of the, to Dr. W. H. Hatfield, 236

Bezssonoff's Reaction for the C Factor, Nature and Value

of, A. Salvatori, 339 Bildtelegraphie und des Fernsehens, Handbuch der: Grundlagen, Entwicklungsziele und Grenzen der elektrischen Bildfernübertragung. Bearbeitet und herausgegeben von Prof. F. Schröter (Review), 781

Binary Alloys, Constitution of, at Room Temperature, Prof. A. N. Campbell, 438

Biochemistry, Researches in, Prof. A. Stoll, 541 Biographical Fragments, Sir Arthur Schuster (*Review*),

Biology: and Philosophy (Review), 529; Teaching of, W. B. Yapp, 203

Biology's Message for Civilisation, Prof. H. B. Fantham,

Biotechnology, 597

Bird: Life on Ailsa Craig, 903; Migration, Prof. W. Rowan (3), 288; in Tropical Africa, J. P. Chapin,

Birds: Courtship of, 392; Frequency of, 885; Migration and Plumage Coloration of, 921; Movements of Winter Flocks of, 69; New Zealand, The Life Histories of, E. F. Stead (Review), 603; of Suffolk, A History of the, Dr. C. B. Ticehurst (Review), 568; of the Indian Empire, The Nidification of, E. C. Stuart Baker. Vol. 1 (Review), 384; of the Russian Altai and North-West Mongolia, Distribution of, late P. Sushkin, 551; Summer Visiting, Immigration of, 481; Wild, Protection of, 337; Winter Congregations of, 69; Winter Territory Amongst, J. B. Price, 730

Birmingham University: appointment of Dr. J. F. Brailsford as radiological demonstrator in anatomy and W. J. Rees as temporary assistant lecturer in botany, 248; W. Barrow elected pro-chancellor; conferment of title of emeritus professor on Prof. W. S. Boulton and Prof. J. T. J. Morrison, 336; Foundation of a Poynting lecture, 736

Birthday Honours, King's, 830 Birth Control Association, National, Annual Report of the, 271

Bitumen in Embalming, Dr. P. E. Spielmann, 588 Bituminous Schists, A New Method of Analysis of, J. Barlot, 774

Blackboard Coloured Diagrams: Biological Series, Rana temporaria, 3 parts (Review), 786

Blackthorn Winter, The, 445

Blood: Pigments of the Invertebrates, Molecular Weights of the, Prof. The Svedberg and Astrid Hedenius, 325; Pressure, Recording and Measurement of the, L. Bugnard, P. Gley and A. Langevin, 375; -Vessels, Disposal of Debris by, Drs. E. R. and E. L. Clark, 880

Botany in Russia, 554

Bond's Numerical Examples in Physics, concerning the review of, Lord Holden, 467

Boric Acid in Sea Water and its Effect on the Carbon Dioxide Equilibrium, K. Buch, 688

Boron: Hydrides, Electrolysis of, A. Stock and E. Wiborg, 212; Structure of the K Line of, A. Hautot,

'Borrowed Days', The: 445; Sir Herbert Maxwell, Bt., 515

Botanical Analyses of Grasslands, Two Forms of Sampler used in Estimating the number of Plants per Acre in, M. J. Gorman, 410

Brachiopod Genera of the Sub-Orders Orthoidea and Pentameroidea, Dr. C. Schuchert and G. A. Cooper,

Brazil Nuts, Barium as a Normal Constituent of, W. M. Seaber, 814

Brazilian Meteorological Service, Retirement of R. P. Xavier as director; Appointment of C. de A. Martins Costa, 581

Bridges, Early American, Evolution of, Capt. L. N. Edwards, 429

Bristol: District, Geology of the, Prof. S. H. Reynolds, 65; University: bequest to, by J. W. White, 299; Report for 1931–32, 665; Dr. R. W. Gurney appointed a research fellow, 772; Dr. J. F. Baker appointed professor of civil engineering, Prof. A. E. Trueman professor of geology, Prof. G. Hadfield professor of pathology, J. W. de Witt G. Thornton lecturer in pharmacology, J. D. A. Gray senior pathological officer, and J. W. Edgell dairy bacteriologist, 848

British: Architects, Royal Institute of: award of the Royal gold medal of the, to Sir Charles Peers, 236; 497; Army, Health of the, during 1931, Lieut .-Gen. Fawcus, 509; Association: Installation of Sir Frederick Gowland Hopkins as president, 50; Leicester meeting of the, 680; Mathematical Tables. Vol. 1: Circular and Hyperbolic Functions, Exponential Sine and Cosine Integrals, Factorial (Gamma) and Derived Functions, Integrals of Probability Integral. Prepared by the Committee for the Calculation of Mathematical Tables. Vol. 2: Emden Functions: Being Solutions of Emden's Equation, together with Certain Associated Functions (Review), 318; Coals, A. L. Curtis, 580; County Flora (Review), 309; Electrical and Allied Industries Association, Annual Report of the, 267; Forestry, Problems of, Prof. R. S. Troup, 900; Geographers, Institute of, Election of officers, 51; Industries Fair: 87; Scientific Instrument Section, 139; Marine Mollusca, The, R. Winckworth, 334; Museum: Sir Charles Reed Peers appointed a trustee of the, 164; (Natural History): Acquisitions at the, 163; 301; 464, 465; 795; Catalogue of the Pontian Carnivora of Europe in the Department of Geology, Dr. G. E. Pilgrim (Review), 454; Ornithologists' Union, Dr. W. Stone elected an honorary member of the, 723; Polar Year Expedition, Report of, J. M. Stagg, 464; Science Guild Research and Development Lecture, 124; Solomon Islands: Early History of the, R. J. A. W. Lever, 587; Protectorate, H. T. Pagden appointed entomologist to the, 687; Zoologists, Association of, Annual Meeting of the, 88

Broadcasting, Empire, 16; 160 Bromine: -Hydrogen and Bromine-Carbon Dioxide, Measurement of the Diffusion Coefficients of, J. E. Mackenzie and H. W. Melville, 250; Spectra of, Br V, VII and IV, A. S. Rao and Dr. K. R. Rao, 170

Brush-Turkeys, Nest-Building of, 233

'Buchan Cold-Spell': The First, 177; Second, 521; Third, the Ice Saints, 665; Fourth, 921

Buff's Compound, Complex Anion of, and of Bunsen's

Salt, F. Holzl and W. Stockmair, 251 Bufumbira, The Volcanic Area of, Part 1: The Geology of the Volcanic Area of Bufumbira, South-West Uganda; with notes on the Petrology and Economic Geology, A. D. Combe and W. C. Simmons (Review),

'Bull-Dog' Calf in African Cattle, J. Carmichael, 878

Bull Frogs Calling, 629

Bunsen Society, Prof. F. G. Donnan elected an honorary member of the, 871

Burmese Spirit-World, Maung Htin Aung, 767 Business Rationalisation, its Dangers and Advantages considered from the Psychological and Social Standpoints, Dr. C. S. Myers (*Review*), 567 Butterflies, Preying of Birds on, Effect of, C. L. Collenette,

Butters, Australian, New Zealand and English, Vitamin Content of, M. E. F. Crawford, E. O. V. Perry and Dr. S. S. Zilva, 770

Butyl Alcohol, Catalytic Dehydration of, by Alumina, Causes of the Simultaneous Production of 1-Butene and 2-Butene in the Course of the, C. Matignon, H. Moureu and M. Dodé, 738

Cabbages and related Green Crops, 21

Cacao Research at the Imperial College of Tropical Agriculture, 580

Cadmium Hydride, Isotope Effect in the Spectrum of, E. Svensson, 28

Cahn Hill-Improvement Scheme, Wales, M. Griffiths appointed lands director, 720

Calanus: finmarchicus, Life History of, 843; Metamorphosis of, 105

Calcite, Periodicity in the Solarisation of, S. Iimori, 619

Calcium: Isotope with Mass 41 and the Radioactive Half-Period of Potassium, Prof. J. Kendall, W. W. Smith and T. Tait, 688; Phosphate, Palæozoic Deposits of, Rôle of the Trilobites in the Genesis of the, L. Cayeux, 850

Calculating Machine, An Electrical, R. M. Mallock, 880 Calculus of Variations, Characterisation of Fields in the, M. Morse and S. B. Littauer, 631

Calendar: of Nature Topics, 17, 34, 69, 105, 141, 177, 213, 248, 284, 337, 373, 409, 445, 481, 521, 557, 594, 629, 665, 701, 737, 773, 812, 849, 885, 921; The, and its Reform, F. A. Black (Review), 859
California Ground Squirrel Control Program, E. S.

Kellogg, 721 Californian Earthquake of March 10, H. O. Wood and C. F. Richter, 686

Cambial Activity, Prof. J. H. Priestley, Lorna I. Scott and Marjorie E. Malins, 375

Cambridge University: Dr. G. Salt elected Stringer fellow at King's College, 141; Opening of the Royal Society Mond Laboratory, 210; Dr. H. Spencer Jones elected an honorary fellow of Jesus College, 213; gift to the Molteno Institute of Parasitology by the Rockefeller Foundation; gift by the Goldsmiths' Company; award of the Gordon Wigan prize for chemistry to T. P. Hoar, 284; award of the Balfour studentship in biology to F. R. Parrington, 336; Dr. N. J. T. M. Needham appointed Sir William Dunn reader in biochemistry; V. J. Chapman elected Frank Smart student in botany; E. A. Maxwell and R. H. Stoy awarded Smith's prizes; W. E. Candler, C. Strachan and M. H. H. Walters awarded Rayleigh prizes, 408; gift by W. J. Courtauld, 444; A. G. D. Watson and E. S. Shire elected fellows of King's College, 593; W. V. Lewis appointed University demonstrator in geography, and Dr. N. W. Pirie University demonstrator in biochemistry, 628; Dr. M. Dixon appointed University lecturer in biochemistry; Dr. E. G. Holmes University lecturer in place of Dr. M. Dixon; A. D. Thackeray elected Sheepshanks exhibitioner, 665; bequest by F. E. Elmore; gift by J. E. Little, 700; award of the Adams prize to A. H. Wilson; grants made from the Balfour fund to C. Forster Cooper and I. T. Sanderson; Dr. F. R. Winton recommended for the readership in physiology; H. McC. Taylor elected a research fellow of Clare College, 736; award of the Raymond Horton-Smith prize to Dr. J. E. Semple, 772; Dr. R. O. Redman appointed University lecturer in astrophysics; J. A. Ratcliffe University lecturer in physics; Dr. N. Feather and P. I. Dee University demonstrators in physics; proposed foundation of a studentship in human anatomy and a readership in pharmacology under the Shield bequest, 811; R.F. Kahn appointed University lecturer in economics and politics, A. H. Wilson, W. V. D. Hodge and P. Hall University lecturers in mathematics; gift by Mrs. G. Stephenson, 848; bequest by S. H. Bickham; Prof. D. B. Copland appointed Alfred Marshall lecturer for 1933-34; award of the Wyse studentship to Miss M. M. Hunter, 884; Dr. G. D. H. Bell appointed University demonstrator in agricultural biology; Dr. K. A. Č. Elliott elected a fellow of Selwyn College, 920

Canada: National Research Council of: Report of the, for 1931–32, 869; Work of the, 648; Newfoundland, Bermuda, British West Indies, British Guiana and the Falkland Islands, Directory of Museums and Art Galleries in, Sir Henry A. Miers and S. F. Markham, 84; Radium in, Assays of, H. S. Spencer, 55; The Museums of, Sir Henry A. Miers and S. F. Markham,

Canadian: Minerals of the Rare Elements, H. V. Ellsworth, 660; Water Power Development in 1932, Dr. Brysson Cunningham, 788

Cancer: Hospital (Free) London, Research Grant for the; Appointment of Research Workers, 500, 509; -Producing Compounds, Sex Hormones and, Dr. J. W. Cook and Prof. E. C. Dodds, 205

Candles and Candle-Making, D. Allan, 269

Cannizzaro's Reaction, G. Lock (2), 560

Cantor: Georg, Gesammelte Abhandlungen: Mathematischen und Philosophischen Inhalts mit erläuternden Anmerkungen sowie mit Ergänzungen aus dem Briefwechsel Cantor-Dedekind, Herausgegeben von E. Zermelo (Review), 418; The Mathematician of the Infinite, Prof. H. T. H. Piaggio (Review), 418

Carbon: Dioxide in the Sea, 282; Fusion of, A. Egerton and M. Milford, 169; in the Diamond, Atomic Radius of, V. Posejpal, 410; Line $K\alpha$, Fine-Structure of the, M. Morand and A. Hautot, 143; Monoxide: Heats of Combustion of, in Oxygen, and of Nitrous Oxide in Carbon Monoxide at Constant Pressure, J. H. Awbery and Dr. E. Griffiths, 446; in Air, Detection of Traces of, Prof. H. Hartridge, 654; Dr. W. Ackermann, 441; H. R. Ambler and T. C. Sutton, 766

Carbonic Anhydrase and the State of Carbon Dioxide in Blood, late Dr. N. U. Meldrum and Dr. F. J. W.

Roughton, 874

Carboniferous Molluses of the Donetz Basin, D. M. Fedotov, 404

Carbonyls of Lithium, Rubidium and Cæsium, Dr. T. G. Pearson, 166

Carborundum Crystals, Photoconductive Effect in, Prof. Kamienski, 475

Caribou in Canada, Migration of, 373

Carnegie Institution, Central American Research by the, 809

Carnegie, The Last Cruise of the, Dr. J. H. Paul (Review), 114

Carotene Derivative, A, giving with Antimony Trichloride an Absorption Band at 610-630 mu, M. van Eekelen and A. Emmerie, 275

Cartilage, Glycogen in, Dr. H. B. Fell and Dr. R. Robison, 62

Casein, New Researches on, E. Cherbuliez and F. Meyer, 215

Cassiopeiæ, y, Spectrum of, Dr. W. J. S. Lockyer, 134 Catfish, Impulses from Sensory Nerves of, H. Hoagland, 631

Cathode Ray: Oscillography, R. A. Watson Watt, 66; Oscillograph, New Applications of the, 103; Photography of Random Electrical Transients, F. W. Chapman, 620; Tube, A Method of Extending the Frequency Range of the, Prof. H. Hartridge, 95 'Catkin' Radio Valve, The, 735

Cattle: Beef, Improvement of, in Australia, 576; Blood-

Sucking Habit of a Bug, W. E. China, 806 Cave Paintings in the Pyrenees, N. Casterat, 19 Cavia, Early Stages in the Development of, N. H. W. Maclaren and Prof. T. H. Bryce, 922

Cellulose: Acetate: its Manufacture and Applications, A. G. Lipscomb (Review), 454; Products, Lignin Content of, Dr. A. G. Norman and S. H. Jenkins, 729; Structure of: Prof. H. Mark, 591; Prof. M. J. Duclaux, 553

Cenco News Chats, 200 Central-Asian Tracks: On Ancient, Brief Narrative of Three Expeditions in Innermost Asia and North-Western China, Sir Aurel Stein (Review), 415

Cephalic Index, Influence of the Variation of the, on that of the Alveolo-Palatinal and Superior Facial Indices,

A. Perier, 815 Cepheid, The Respective Phases of Ionisation Maxima and Light Maxima in a, G. Tiercy, 483

Cepheids: Respective Phases of Minimum Ionisation and Light Minimum in the, G. Tiercy, 815; Variation of Ionisation and the Spectrum Variation of Some, G. Tiercy, 215

Cerebral Cortex, Functions of the Olfactory Parts of the,

C. J. Herrick, 739

Cerebro-Spinal Meningitis, 128 Cetacea, Worms Parasitic in, Dr. H. A. Baylis, 551

Ceylon, Marine Biology in, Dr. J. Pearson, 54 Chain: Reactions, Reaction Cells in, A. R. Ubbelohde, 328; The Electromotive Force of the, Mlle. Quintin,

Charles and Erasmus, Dr. MacGregor Skene, 66

Chemical: Analysis: by the Emission Spectrum, Foundations and Methods of, Dr. W. Gerlach and Dr. E. Schweitzer (translated) (Review), 420; by X-Rays and its Applications, Prof. G. von Hevesy (Review), 39; Quantitative, Modern Methods in, Dr. A. D. Mitchell and Dr. A. M. Ward (Review), 531; Disarmament, 413; Engineers, Institution of, election of officers; award of the Osborne Reynolds medal to S. G. M. Ure, the Moulton medal to Dr. C. M. White, and the Junior Moulton medal and prize to Dr. W. B. Hawes, 510; Incendiary and Bacterial Weapons Special Committee, Report of, 413; Industry: Petroleum Technology and, 145; Society of: Dr. J. T. Dunn elected president of the, 432; Prof. W. A. Bone awarded the medal of the, 793; Physics, Journal of, No. 1, 271; Society, Present Position and Future of the, Prof. G. G. Henderson, 498; Subjects, The Elder Pliny's Chapters on, Part 2. Edited, with translation and notes, by Dr. K. C. Bailey (Review), 305

Chemistry: and our Idiosyncrasies, Dr. E. F. Armstrong (Review), 314; and the Art of Living, Prof. H. E. Armstrong (Sir Jesse Boot Foundation Lecture), 795; Applied, Reports of the Progress of, Vol. 17, 1932 (Review), 786; Federal Council for, 300; Forensic and Scientific Criminal Investigation, A. Lucas. Second edition (*Review*), 115; Inorganic and Theoretical, A Comprehensive Treatise on, Dr. J. W. Mellor. Vol. 12 (Part 1), (Review), 638; in Space, Modern Aspects of, Prof. T. M. Lowry (Review), 563; Institute of, Address by Dr. G. C. Clayton; presentation of the Meldola medal to Dr. L. E. Sutton, and the Sir Edward Frankland medal and prize to L. Young, 357; of the Higher Fungi, L. Lukacs and J. Zellner (22), 888; of the Lichens, J. Klima (2), 888; Organic: An Introduction to, Prof. I. D. Garard (Review), 420; A Text-Book of, Prof. J. Schmidt. English edition by Dr. H. G. Rule. Second edition (Review), 116; Laboratory Methods of, L. Gatter-Revised by H. Wieland. Translated by Dr. W. McCartney (*Review*), 78; Practical (*Review*), 78; Physical, Prof. J. Eggert. Translated by Dr. S. J. Gregg (Review), 419; A Text-Book of, Dr. J. Newton Friend. Vol. 1 (Review), 417; International Congress of, 53; Plant, Comparative, E. Fraenkel and J. Zellner (24), 888; Progress of, vol. 29, for 1932, Annual Reports on the (*Review*), 676; Raman Spectra and, 263; Space, Growth of, Prof. J. Read (Review), 783; Tannin, Incunabula of, Dr. M. Nierenstein (Review), 823; The A.B.C. of, J. G. Crowther (Review), 895

Chick Embryo, Induction by Coagulated Organisers in the, C. H. Waddington, 275

Chile Saltpetre, Origin of the, Dr. C. T. Kautter, 556 Chilean Insect Parasites for New Zealand, Dr. D. Miller, 283

Chimie, anorganischen, Handbuch der, Herausgegeben von Prof. R. Abegg, Dr. F. Auerbach und Prof. I. Koppel. Band 4, Abt. 3, Teil 2B, Lief. 2 (Review), 491

Chimney: Emissions, Reduction of, 234; Smoke, Measurement of, 430

China, Ancient: Copper Age in, Tsurumatsu Dono, 243; 917; Festivals and Songs of, Prof. M. Granet. Translated by Dr. E. D. Edwards (*Review*), 346; and American Scientific Expeditions, Dr. H. F. Osborn, 87; An Imperial City of (Review), 184; Land and Labour in, Prof. R. H. Tawney (Review),

Chinese: Chemical Society, Journal of the, No. 1,907; Medicine: History of, being a Chronicle of Medical Happenings in China from Ancient Times to the Present Period, Dr. K. Chimin Wong and Dr. Wu Lien-Teh (Review), 527 Chitons, Abnormal, I. Taki, 367

Chlorine: Hydrogen and, Photochemical Reaction of: Prof. H. B. Baker, 27; Prof. A. J. Allmand, 656; Magnetic Rotatory Power of, and of Hydrochloric Acid Gas, R. de Mallemann and P. Gabiano, 286 Chlorophyll, Attenuation of the, shown by Parasitic Green Plants, M. Molliard, 178

Chlorophylls, Structural Formulæ of the, Prof. J. B. Conant and Emma M. Dietz, 131

Cholera Vibrion, Transmissible Lysis of the, P. N. Bernard and J. Guillerm, 887

Cholesterol: Chemical Constitution of, and a New Isomeride of Cholesterol, R. de Fazi, 924; of the Human Brain, C. Antoniani and F. Zanelli, 71

Christening, Caution in, Prof. H. E. Armstrong, 330 Christian Church: Phases of the, a Short View of its History, Rev. A. C. Bouquet (*Review*), 677

Chromic Salts, Line Absorption of, in Relation to Coordination, Dr. C. H. Johnson and A. Mead, 399

Chromosome: Mechanics, Prof. A. H. Sturtevant (Review), 5; Numbers in Vertebrates, K. Oguma and S. Kakino, 403

Chromosomes and Plant-Breeding, Dr. C. D. Darlington (Review), 5

Chronophotographic Method, A Modification of the, and

Some Applications, E. Crausse and J. Baubiac, 482 Chrysene: Substituted βγ-Diphenyladipic Acids and Derivations of, Dr. A. I. Vogel, 402; Synthesis of, and Certain Derivatives, G. R. Ramage and Prof. R. Robinson, 205

Cinematography, New Lens of High Rapidity for, Carl Zeiss, 475

Circuit-Breakers, Heavy Current, H. Thommen, 103 Circular Magnetic Dichroism and Magnetic Rotatory Dispersion, A. Cotton, 70

Citrus, Mycorrhiza in the Genus, Dr. M. C. Rayner, 399

Civilisation and War (Review), 635
Clarendon Laboratory, Oxford, Helium Liquefaction
Plant at the, Prof. F. A. Lindemann and T. C. Keeley,

191 Climate: a Handbook for Business Men, Students and Travellers, Dr. C. E. P. Brooks. Second edition (Review), 9; and Acclimatization: Some Notes and Observations, Sir Aldo Castellani (Review), 712

Climatic Temperature, Minimum, C. F. Talman, 431 Clinical Laboratory Methods, A Manual of, Dr. C. L. Cummer. Third edition (Review), 116

Clothes Moths and House Moths, Major E. E. Austen, assisted by A. W. McKenny Hughes, 55

Cloud Photography, International, 577 Clover Mites invade Dwelling-Houses, 557

Coal: Economic Significance of, Capt. B. Acworth, 478; for Sea Transport, Rear-Admiral Scott-Hill, 479; Formation of Rock Joints and the Cleat of, Profs. P. F. Kendall and H. Briggs, 922; Industry, British: Early History of the (*Review*), 311; The Rise of the, Prof. J. U. Nef. 2 vols. (Review), 311; Low Temperature Carbonisation of, 31; Mining Explosives, Recent Developments in, Dr. W. Payman, 846; The Utilisation of: 633; A. C. Hardy, and others, Symposium on the, 478

Cobalt: Chloride: Constitution of Hydrochloric Acid Solutions of, P. Job, 338; Solutions of, Circular Magnetic Dichroism and Abnormal Magnetic Rotatory Dispersion of, M. Scherer, 70; Ion of Dissolved, Paramagnetism of the, R. Mercier, 338

Cæsium Vapour, Absorption of Light in, Dr. H. J. J.

Braddick and Prof. R. W. Ditchburn, 132 Coffee, Root Development in, Distribution of Nitrates in the Soil and, V. A. Beckley and F. McNaughtan,

Col du Marchairuz Region, The Virgulian and the Stratigraphy of the Portlandian of the, A. Lombard, 815

Colloidal: Propellants, Rate of Burning of, Dr. A. D. Crow and W. E. Grimshaw, 60; F. R. W. Hunt and G. H. Hinds, 206; Solutions, Surface Tension of, and the Action of Light on Soap Solutions, Dr. D. Lander M. W. 1989.

P. Lecomte du Noüy, 689 Colloids: Adsorption of, by Metallic Surfaces and its Influence on the Adherence of Electrolytic Deposits, P. Jacquet, 702; Flocculation of, Influence of Radioactive Radiations on the, A. Boutaric and Mlle. Madeleine Roy, 738; Preparation of, by Ultrasonic Dispersion, N. Marinesco, 410 Colonial: Educational Conference, Case for a, Sir George

Maxwell, 812; Office Appointments, 55; 871 Colonisation of a Disused Millpond at Hillsborough, Co. Down, Common and Cairns, 702

Colorado, University of, Studies, Nov., 885 Coloration Problems in Animals, 577

Colour, A Simple Method for Determining, E. Haschek and M. Haitinger, 411

Comba Region, Geology of the, H. Lagotala, 411 Comenius in England: the Visit of Jan Amos Komenský

(Comenius), the Czech Philosopher and Educationist, to London in 1641-1642; its Bearing on the Origins of the Royal Society, on the Development of the Encyclopædia, and on Plans for the Higher Education of the Indians of New England and Virginia, as described in Contemporary Documents. Selected, translated and edited, with an Introduction and Tables of Dates, by R. Fitzgibbon Young (Review), 306

Comet: Halley's, in 1909–11, Dr. A. C. D. Crommelin, 282; New, 1933a., 281; Peltier-Whipple, 173; Pons-Winnecke, Periodic, 697

Comets: 661; 845; Prof. van Biesbroeck, 245; of A.D. 868 and 1366, Dr. S. Kanda, 31; Perturba-

tions of, by the Stars, E. Opik, 209

Comital Flora of the British Isles (Flora Comitalis Britannicæ: Fl. Com. Brit.): The, being the Distribution of British (including a number of Non-Indigenous) Plants throughout the 152 Vice-Counties of Great Britain, Ireland and the Channel Islands, with the Place of Growth, Elevation, World-Distribution, Grade, Chief Synonyms and First Names by which the Plants were recorded as British. By Dr. G. Claridge Druce. With an original coloured Map showing the Botanical Vice-Counties presented by W. J. Patey (Review), 309

Commonwealth Fund fellowships, awards of, 736 Compensating Plate Currents, Method of, P. Donzelot

and J. Divoux, 106

Composite Surfaces in Higher Space, J. G. Semple, 886 Compressed Air Wind Tunnel at the National Physical Laboratory, New, 442 Cone, Motion of a, at High Speeds in Air, Taylor and

Maccoll, 552

Consolidated Gold Fields of South Africa, Ltd., awards of the, 432

Constants and Numerical Data, Annual Tables of, Vol. 8, 2 parts; Vol. 9; Tables des matières du Vol. 9 (Review), 892

Convention and Fact, Dr. N. R. Campbell, 237

Co-operative Industrial Research, 817

Copepods: of the Woods Hole Region, Mass., The, Dr. C. B. Wilson, 698; Sexual Reproduction in, A. G. Lowndes, 240

Copper: -Aluminium Alloys, X-Ray Investigation of the, A. J. Bradley and Phyllis Jones, 589; Properties of, Influence of Oxygen on the, W. Broniewski and S. Jasian, 338

Corallian Rocks of Yorkshire, V. Wilson (1), 595 Corals, Growth of, T. Tamura and Y. Hada, 172

CORRESPONDENCE

Acetic Acid: at 25°, Dissociation Constant of, Limiting Mobilities of Some Monovalent Ions and the, Dr. A. I. Vogel and G. H. Jeffery, 27; in Water, Dissociation of, Dr. C. W. Davies and W. H. Banks, 328

Acetylated Aldonic Nitriles, A Rule for the Rotatory Direction of the, Dr. V. Deulofeu, 548

Acquired Characters, Inheritance of, Dr. R. L. Jenkins,

Adenine: Hydrochloride, Crystal Structure of Vitamin B₁, and of, J. D. Bernal and D. Crowfoot, 911; Vitamin $\rm B_4$ and, R. D. Heard, H. W. Kinnersley, J. R. O'Brien, Prof. R. A. Peters and V. Reader, 617 Adrenal Gland, Vitamin C in the, G. Bourne, 874

Animal Life, Components of the Atmosphere and Synthetic Gases in Relation to, Prof. J. W. Hershey, 238

Alkali Metals, Remarkable Optical Properties of the, Prof. R. W. Wood, 582 Alkyl Groups, Volumes of, and their Orienting Powers,

Dr. R. J. W. Le Fèvre, 655

Alkylanilines with Tertiary Alkyl Groups, Dr. W. J. Hickinbottom, 762

Allium, Somatic Chromosomes in, Structure and Division of, Prof. T. K. Koshy, 362

Aluminium: Hydride, Band Spectrum of, Predissociation and Pressure Effects in the, Prof. E. Hulthén and R. Rydberg, 470; Hyperfine Structure in, R. Ritschl, 58; Single Crystals, Recrystallisation Power and Shear Hardening in, Dr. W. G. Burgers, 326

Angiosperms, Origin of, Prof. G. R. Wieland, 360

Animal Growth, Heterogony and the Chemical Ground-Plan of, C. H. Waddington, 134

Anode Spot in a Neon Tube, Dr. T. Takamine, T. Suga and A. Yanagihara, 584

Antimony, Molecular Fluorescence of, Jean Genard, 132 Aphanapteryx, The Dodo and the, G. Renshaw, 728 Aquatic Insect Larvæ, Tracheal and Blood Gills in, Dr. W. H. Thorpe, 549 Arsenic, Nuclear Moment of, M. F. Crawford and A. M.

Crooker, 655 Ascaphus, The 'Tail' of the Male American Toad, Prof. C. G. S. de Villiers, 692

Ascorbic Acid: Prof. A. Szent-Györgyi and Prof. W. N. Haworth, 24; Constitution of, Dr. E. L. Hirst, E. G. V. Percival and F. Smith, 617; Vitamin C and, A. L. Bacharach, 364; (Vitamin C) and Glutathione in Animal Tissues, Estimation and Distribution of, T. W. Birch and Dr. W. J. Dann, 469

Astronomical Radiative Stability, The, Sir Joseph

Larmor, 805; Dr. G. C. Simpson, 875

Atmosphere, Components of the, and Synthetic Gases in Relation to Animal Life, Prof. J. W. Hershey, 238

Atomic Number Zero, The Neutron and Neuton, the New Element of, Prof. W. D. Harkins, 23

Bacterial Growth Curve, The, and the History of Species, Dr. A. S. Corbet, 61 Baltic, Inertia Currents in the, T. Gustafson and B.

Kullenberg, 586 Band Spectra which appear near Visible Triplet Lines

of Mercury, E. Matuyama, 58

Barium: Azide, Acceleration of the Decomposition of Crystals of, by the Emission from Radium Emanation, Prof. W. E. Garner and C. H. Moon, 513; Oxide (BaO), Band Spectrum of, P. C. Mahanti, 402

Barley, Egyptian Neolithic, A. Jackson, 652

p-Benzoquinone Crystal, Orientations of Molecules in the, K. S. Krishnan and S. Banerjee, 653

Beryllium and Helium, Lord Rayleigh, 724

Binary Alloys at Room Temperature, Constitution of, Prof. A. N. Campbell, 438

Biology, Teaching of, W. B. Yapp, 203

Blood Pigments of the Invertebrates, Molecular Weights of the, Prof. The Svedberg and Astrid Hedenius, 325

Boric Acid in Sea Water and its Effect on the Carbon Dioxide Equilibrium, K. Buch, 688

Borrowed Days, The, Sir Herbert Maxwell, Bt., 515 British Solomon Islands, Early History of the, R. J. A. W. Lever, 587

Bromine: Spectra of, Br V, VII and IV, A. S. Rao and Dr. K. R. Rao, 170

B. salmonicida in River Waters, The Presence of a Bacteriophage for, Dr. C. Todd, 360

'Bull-Dog' Calf in African Cattle, J. Carmichael, 878 Cadmium Hydride, Isotope Effect in the Spectrum of, E. Svensson, 28

Cæsium Vapour, Absorption of Light in, Dr. H. J. J.

Braddick and Prof. R. W. Ditchburn, 132 Calcite, Solarisation of, Periodicity in the, S. Iimori, 619 Calcium Isotope with Mass 41 and the Radioactive Half-Period of Potassium, Prof. J. Kendall, W. T. Smith and T. Tait, 688

Cancer-Producing Compounds, Sex Hormones and, Dr. J. W. Cook and Prof. E. C. Dodds, 205

Carbon: Fusion of, A. Egerton and M. Milford, 169; Monoxide in Air, Detection of Traces of, Prof. H. Hartridge, 654; H. R. Ambler and T. C. Sutton, 766

Carbonic Anhydrase and the State of Carbon Dioxide in Blood, late Dr. N. U. Meldrum and Dr. F. J. W. Roughton, 874

Carbonyls of Lithium, Rubidium and Cæsium, Dr. T. G. Pearson, 166

Carotene Derivative giving with Antimony Trichloride an Absorption Band at 610-630 m\u03c0, A. M. van Eekelen and A. Emmerie, 275

Cartilage, Glycogen in, Dr. H. B. Fell and Dr. R. Robison,

Cassiopeiæ, y, Spectrum of, Dr. W. J. S. Lockyer, 134 Cathode Ray: Photography of Random Electrical Transients, F. W. Chapman, 620; Tube, Frequency Range of the, A Method of Extending the, Prof. H. Hartridge, 95

Cellulose Products, Lignin Content of, Dr. A. G. Norman

and S. H. Jenkins, 729

Chain Reactions, Reaction Cells in, A. R. Ubbelohde, 328 Chick Embryo, Induction by Coagulated Organisers in the, C. H. Waddington, 275

Chlorine, Hydrogen and, Photochemical Reaction of, Prof. H. B. Baker, 27; Prof. A. J. Allmand, 656 Chlorophylls, Structural Formulæ of the, Prof. J. B.

Conant and Emma M. Dietz, 131

Christening, Caution in, Prof. H. E. Armstrong, 330 Chromic Salts, Line Absorption of, in Relation to Coordination, Dr. C. H. Johnson and A. Mead, 399

Chrysene: and Certain Derivatives, Synthesis of, G. R. Ramage and Prof. R. Robinson, 205; Substituted βγ-Diphenyladipic Acids and Derivatives of, Dr. A. I. Vogel, 402

Citrus, Mycorrhiza in the Genus, Dr. M. C. Rayner, 399 Coffee, Root Development in, Distribution of Nitrates in the Soil and, V. A. Beckley and F. McNaughtan,

Colloidal: Propellants, Rate of burning of, Dr. A. D. Crow and W. E. Grimshaw, 60; F. R. W. Hunt and G. H. Hinds, 206; Solutions, Surface Tension of, and the Action of Light on Soap Solutions, Dr. P. Lecomte du Noüy, 689

Convention and Fact, Dr. N. R. Campbell, 237 Copepods, Sexual Reproduction in, A. G. Lowndes, 240 Cosmic: Radiation, Origin of, H. Alfvén, 620; Rays,

Energy of, Prof. E. Regener, 130

Crustacea, Respiratory Movements in, Control of, Prof. H. Munro Fox and M. L. Johnson, 514

Crystal: Lattice Constants, Determination of, by Electron Diffraction, Prof. G. I. Finch and A. G. Quarrell, 842; Structure and Orientation in Thin Films, Prof. G. I. Finch and A. G. Quarrell, 877

Crystalline Style, Oxidase of the, C. Berkeley, 94 Cuprous Oxide: A New Type of Photoelectric Effect in, in a Magnetic Field, Dr. I. Kikoin and Dr. M. Noskov, 725; Photo-Cell in Red Light, Reversal of the Current from a, Dr. H. H. Poole and Dr. W. R. G. Atkins, 133 Curare, The Neuro-Muscular Junction and, Tudor Jones,

Davainea proglottina, Life-History of the Fowl Tapeworm, F. J. Brown, 276 Debye's Dispersion of Nitrobenzene, Prof. J. Weigle and

R. Luthi, 327

Dehydro-Ascorbic Acid, Constitution of, Prof. P. Karrer, H. Salomon and K. Schöpp, 800

Diphenyl Series, Crystal Structure of, Lucy W. Pickett,

Dodo, The, and the Aphanapteryx, G. Renshaw, 728 Dyes, Anti-Stokes Fluorescence in, Efficiency of, Dr. A. Jabloński, 839

Earthquakes: in the Holy Land: a Correction, Prof. Bailey Willis, 550; P Wave in, Corrections to the times of the, Dr. H. Jeffreys and K. E. Bullen, 97

Egyptian Neolithic Barley, A. Jackson, 652

Einstein", "The Case Against, Col. A. Lynch, 622 Elastic Electron Scattering in Gases, Dr. S. Werner, 726 Electrical Energy by Frog-Skin, Output of, W. L. Francis, 805

Electrolytic Dissociation, Rôle of the Solvent in, Dr. A. R. Martin, 584; Dr. J. A. V. Butler and Miss L. C. Connell, 800

Electrometer Triode in the X-Ray Ionisation Spectrometer, Dr. W. A. Wooster, 545; Dr. B. W. Robinson,

Electron: Diffraction: by Films of Grease, Prof. G. P. Thomson and C. A. Murison, 237; by Vapours, Dr. H. de Laszlo, 803; Polarisation?, Prof. R. Whiddington, 908; Positive, A Possible Property of the, Dr. W. Elsasser, 764
Electronic Oscillations, Production of, with a Two-Electrode Valve, J. S. McPetrie, 691

Electrons in Amorphous and in Crystalline Antimony,

Diffraction of, J. A. Prins, 760 Elements: Artificial Transmutation of, Chemical Detection of, Prof. F. A. Paneth and P. L. Günther, 653; Light, Disintegration of, by Fast Protons, Dr. J. D. Cockcroft and Dr. E. T. S. Walton, 23

Ergosterol and Cholesterol, Hydroxyl Group in, Prof. I. M. Heilbron and J. C. E. Simpson, 438

'Events', 'Time' and, Ideas of, W. W. L., 727 Explosive Chain: Reaction between Hydrogen and Oxygen, Upper Pressure Limit in the, C. N. Hinshelwood and G. H. Grant, 361; Reactions, Upper Limit in, H. W. Melville and H. L. Roxburgh, 690

Fact, Convention and, Dr. N. R. Campbell, 237 Ferns, Sporangia containing Spermatozoids in, Alice E.

Gairdner, 621

Ferromagnetics, Atomic Moments of, Dr. E. C. Stoner, 433 Fibres from the Coat of a Blackface Lamb, Dr. S. G. Barker, 799

Fluorescence, Negative Polarisation in, K. S. Krishnan and S. M. Mitra, 204

Fluorite from Obira, Japan, Thermoluminescence Spectrum of, Bands in the, Ei-ichi Iwase, 909

Formaldehyde in Dew, Presence of, Prof. N. R. Dhar and Atma Ram, 800

Fossil Bones, Preservation of, L. McL. Mann, 366; W. J. McCallien, 694

Fourier Analysis and Vowel Curves, R. H. Nisbet, 401 Frog, Hermaphrodite, A, A. M. Gardiner, 330

Frogs, Spawning in, Factors Controlling Date of, R. M. Savage, 587

Fuel-Knock, A Theory of, Dr. S. Steele, 724

G., The Constant of Gravitation, Prof. V. V. Narlikar, 134 Gallium Isotopes 69 and 71, Nuclear Moments of the, J. S. Campbell, 204

Gammarus chevreuxi, Sexton, Further Mutations in the Amphipod, Mrs. E. W. Sexton and Miss A. R. Clark,

201

Gases, Mixtures of, Conductivity of, Dr. S. P. McCallum and L. Klatzow, 841

Glucose Metabolism, Glutathione, Iodoacetic Acid and, Dr. J. H. Quastel, 206

Glucuronic Acid, A New Unsaturated Derivative of, J. Pryde and R. T. Williams, 57 Glutathione: and Tissue Glyoxalase, Iodoacetic Acid, Dr. F. Dickens, 130; Iodoacetic Acid and Glucose Metabolism, Dr. J. H. Quastel, 206

Glycogen in Cartilage, Dr. H. B. Fell and Dr. R. Robison,

Glycuronic Acid, Methylnornarcotine, and Vitamin C, W. J. Dann, 24

Glyoxalase, Tissue, Iodoacetic Acid, Glutathione and,

Dr. F. Dickens, 130

Gold, Organic Compounds of, Prof. C. S. Gibson, 130 Gonad-Stimulating Hormones in Hypophysectomised Animals, Prof. J. B. Collip, Dr. H. Selye and Prof. D. L. Thomson, 56

Graticules, Photographic, Sir Herbert Jackson, 766 Heart, Abnormal Movability of the, S. Osawa, 241

Helium, Beryllium and, Lord Rayleigh, 724

Helminthosporium avenæ, Sporulation of, in Artificial Culture, W. A. R. Dillon Weston, 435

'Hexuronic Acid' (Ascorbic Acid) as the Antiscorbutic Factor, Prof. A. Szent-Györgyi and Prof. W. N. Haworth, 24; T. W. Birch, Dr. L. J. Harris, S. N. Ray, 273; W. J. Dann, 274

Hibernian Halos, Radioactivity of Samarium and the Formation of, Dr. J. H. J. Poole, 654

High-Frequency Electric Discharge in Gases, J. C. Wilson, 546

Hydrocarbon Combustion in an Engine, A. Egerton and

F. Ll. Smith, 725
Hydrogen and Chlorine, Photochemical Reaction of, Prof. H. B. Baker, 27; Prof. A. J. Allmand, 656

Hydroxyl Group: in Ergosterol and Cholesterol, Prof. I. M. Heilbron and J. C. E. Simpson, 438; The, and Soap Film Structure, W. J. Green, 873

Hygroscopy, Capacitance, and Some of its Applications, Dr. R. K. Schofield, 96; Dr. W. L. Balls, 329

Immiscible Liquid Layers, Four, Systems of, E. Lester Smith, 167

Impulse Generator, A New, for Three Million Volts, T. E. Allibone, F. S. Edwards and D. B. McKenzie, 129

Insects: and Micro-Climates, H. S. Leeson and K. Mellanby, 363; Dr. H. H. Darby, 837; Inhabiting the Soil Surface, Effects of Rainfall-Evaporation Ratio on, Dr. J. Davidson, 837

Intelligence in Man, Inheritance of, A. F. Dufton, 763;
Dr. C. C. Hurst, 764
Internal Ballistics, The Combustion Problem of, Dr. A. D.

Crow and W. E. Grimshaw, 805

Invar, Dr. Ch.-Ed. Guillaume, 658

Iodine: Arc Spectrum of, Prof. W. E. Curtis, 398; -Oxalate Reaction, Kinetics of the, Prof. N. R. Dhar, A. K. Bhattacharya and B. L. Mukerji, 840; Vapour at High Temperatures, Absorption Bands of, E. Skorko, 366

Iodoacetic Acid: and Glucose Metabolism Glutathione, Dr. J. H. Quastel, 206; Glutathione and Tissue Glyoxalase, Dr. F. Dickens, 130

Ionisation Produced by Radioactive Sources, Amplification of the, J. A. C. Teegan, 277

Ionosphere: Characteristics of the, J. P. Schafer and W. M. Goodall, 804; Fine-Structure of the, Prof. E. V. Appleton, 872; J. A. Ratcliffe and E. L. C. White, 873

Ions, Monovalent, Limiting Mobilities of Some, and the Dissociation Constant of Acetic Acid at 25°, Dr. A. I. Vogel and G. H. Jeffery, 27

Kattegat, Boundary Tides in the, Dr. H. Pettersson and B. Kullenberg, 586

Kennelly-Heaviside Layer, Polarisation of Echoes from the, T. L. Eckersley, 512

Latitude, Variation Effect in, Correlatable with the Moon, Prof. H. T. Stetson, 437

Lattice Distortion, Selective, in Wires under Torsion, W. A. Wood, 842

Lepidosiren, Pelvic Filaments of, J. T. Cunningham and

D. M. Reid; G. E. H. Foxon, 913 Liesegang Rings, Dr. E. S. Hedges, 169

Lightning Flash, A Destructive, Dr. C. V. Boys, 765

Light: Measurement of, for Biological Purposes, Dr. W. H. Pearsall and P. Ullyott, 694; -Spots, Faint Transient, Photography of, Dr. L. F. Richardson, 401 Liquids, Supersaturation of, with Gases, T. N. Richardson

and Dr. K. C. Bailey, 762 Local Lists of Animals, Prof. A. E. Boycott, 94

Lolium: Nitrogen Fixation in the Genus, R. Brown, 169; Seedlings, Fluorescence of, in Ultra-Violet Light, P. A. Linehan and Prof. S. P. Mercer, 202

Loudness, Scales of, B. G. Churcher and A. J. King, 760 Lunar Periodicity in Reproduction, Prof. T. A. Stephenson, 622

Magnetic Disturbances, Recent, Rev. J. P. Rowland, 764 Mars, The Minor Details of the Planet, Dr. E. M. Antoniadi, 802

Mayas, Sacred Sandstone of the, Prof. T. D. A. Cockerell, 656

Mendelian Factors in Quantitative Inheritance, Number of, Dr. R. A. Fisher, 400 Methionine in Wool, J. Barritt, 689

Methylnornarcotine, Glycuronic Acid, and Vitamin C, W. J. Dann, 24

Mice under Continued Treatment with Œstrin, Some Effects Observed in, H. Burrows, Prof. E. C. Dodds, and N. M. Kennaway, 801

Mitochondria, Fixation of, J. H. Davie, 59

Molecules of Intermediate Complexity, Kinetics of the Decomposition of, C. N. Hinshelwood and C. J. M. Fletcher, 24

Molybdenum, Distribution of, Dr. W. A. Roach, 202 Mussel, Common, Strange Spatfall of the, on the Common Cockle, Prof. J. H. Orton, 513

Mutations, Spontaneous, Origin of, M. Navashin, 436 Neoteny and Pædogenesis, Meaning of, G. E. H. Foxon, 93 Neuro-Muscular Junction, The, and Curare, Tudor Jones,

Neuton, The Neutron and, the New Element of Atomic Number Zero, Prof. W. D. Harkins, 23
Neutron: and Neuton, The, the New Element of Atomic Number Zero, Prof. W. D. Harkins, 23; Proton and Positron, Dr. N. Thon, 878
Neutrons, Dr. D. Meksyn, 366
Neutrons, Dr. D. Meksyn, 366

Nerve: Action of Quaternary Ammonium Salts on, S. L. Cowan, 658; Mitogenetic Radiation of, Dr. A. Gurwitsch, 912

Nitrates in the Soil, Distribution of, and Root Development in Coffee, V. A. Beckley and F. McNaughtan, 878 Nitrobenzene, Debye's Dispersion of, Prof. J. Weigle and

R. Luthi, 327

Nitrogen Dioxide: Infra-Red Absorption Spectrum of, C. R. Bailey and A. B. D. Cassie, 239; Molecule, Form and Vibrational Frequencies of the, C. R. Bailey and A. B. D. Cassie, 910; R. Schaffert, 911

Nitrous Oxide in the Glow Discharge, Dissociation of,

E. A. Stewardson, 364

Skaržvýski, 766

NO: Extension of the Visible Absorption System of, to Longer Wave-Lengths, J. Curry and Dr. G. Hertzberg, 842; Molecule, Form and Vibrational Frequencies of the, Dr. L. Harris, W. S. Benedict and G. W. King, 621

Nematodes, Parasitic, Cultivation of, G. Lapage, 583 Nuclear: α-Particles, Fundamental State of, Dr. G. Gamow, 618; Energy Levels, Dr. G. Gamow, 433; Potential Barriers: and Nuclear Structure, Heights of, E. C. Pollard, 398; Heights of, E. C. Pollard, 97

Nucleic Acids and Uracil, Irradiation of, F. F. Heyroth and J. R. Loofbourow, 92 Numerical Coincidence, A, H. P. Hollis, 550

Oestrogenic Substance from Plant Material, An, Dr. B.

Oestrus-Exciting Compound, A Synthetic, Dr. J. W. Cook, Prof. E. C. Dodds and C. L. Hewett, 56

Oldoway Human Skeleton, The, Dr. L. S. B. Leakey, Prof. H. Reck, Prof. P. G. H. Boswell, A. T. Hopwood and Dr. J. D. Solomon, 397

Olefinic Compounds, Catalytic Hydrogenation of, Dr. E. H. Farmer and R. A. E. Galley, 60

Orthopodomyia pulchripalpis, Rondani (Diptera, Culicidæ), A New British Record of, J. F. Marshall and J. Staley,

Ovulation without 'Heat' in the Ewe, Occurrence of, R. Grant, 802

Oxide Catalysts, Positive Ion Emission from, Dr. C. F. Powell and Luang Brata, 168

Oysters (O. edulis), Breeding of, at Port Erin, Prof. J. H.

Orton, Miss M. W. Parke and W. C. Smith, 26 Ozone, Photosensitised Decomposition of, by Chlorine, Dr. R. G. W. Norrish and G. H. J. Neville, 544 P. vulgata, Environment of the Common Limpet, Some

Limiting Factors in the, Prof. J. H. Orton, 693 Pædogenesis, Neoteny and, Meaning of, G. E. H. Foxon,

Paramagnetic Susceptibility, Influence of Light on, P. W. Selwood, 761

Pelagic Organisms, Unusual Occurrence of, Prof. W. J. Dakin, 239; G. P. Farran, 240

Penetrating Radiation, Spectrum and Latitude Variation of, Dr. E. J. Williams, 511

Phenomenal Regression to the Real Object, Prof. W. Peddie; Dr. R. H. Thouless, 544

Phosphorus, Oxide of, Ultra-Violet Bands of, Prof. P. N. Ghosh and A. K. Sen Gupta, 841

12-Phosphotungstic Acid, Structure of the Molecule of, J. F. Keggin, 908

Phytophthora megasperma causing Pink Rot of the Potato, H. Cairns and A. E. Muskett, 277

Piltdown Skull, The Second, Sir Arthur Smith Woodward,

Pituitary Gland, Action of Proteolytic Enzymes on the Oxytocic Principle of the, Dr. J. M. Gulland and T. F. Macrae, 470

Plant Virus, Protective Inoculation against a, Dr. R. N. Salaman, 468

Plasticine Edible? Is, D. I. Clements, N. H. Howes and G. P. Wells, 330

Positive Electron, New Evidence for the, Dr. J. Chadwick, P. M. S. Blackett and G. Occhialini, 473

Positron, Neutron, Proton, and, Dr. N. Thon, 878 Potassium, Radioactive Half-Period of, Calcium Isotope

with Mass 41 and the, Prof. J. Kendall, W. T. Smith and T. Tait, 688

Potatoes, Virus A of, Insect Transmission of, J. B. Loughnane, 838

Priestley: as a Practical Chemist, Prof. T. S. Patterson 690; Dr. A. N. Meldrum, 801; Joseph, The "Leeds

Portrait" of, W. C. Walker, 876
Protein, Synthesis of, by Green Plants, L. Loose and Dr. W. H. Pearson, 362

Proton, Neutron, Positron, Dr. N. Thon, 878

Puccinia Helianthi Schw., Union of Pycniospores and Haploid Hyphæ in, J. H. Craigie, 25

Pycnogonid, A Dodecapodous, Dr. W. T. Calman, 242 Pyrrhotite, Iron Lattice of, Vacant Positions in the, Prof. G. Hägg, 167

Radio-Waves? Interaction between, B. D. H. Tellegen, 840

Rare Earths, Pure Salts of the, Fluorescence of, Prof. R. Tomaschek and O. Deutschbein, 473

'Raw' Weather, Prof. W. A. Osborne, 515 Rectifier Photo-Cells, Reversal of Current in, J. Guild, 327; Dr. H. H. Poole and Dr. W. R. G. Atkins, 547

Reproduction, Lunar Periodicity in, Prof. T. A. Stephenson, 622

Resonance Ag I Lines, Fine-Structure of the, W. E. Williams and A. Middleton, 692

Rickets, Experimental, as a Phosphorus Deficiency Disease, Prof. H. D. Kay and Dr. B. L. Guyatt, 468 Ringworm Fungi, Development of Fuseaux, Aleuriospores

and Spirals on detached Hairs infected by, Dr. A. M. Davidson and Dr. P. H. Gregory, 836

Rubber Research, Dr. S. S. Pickles, 273 Sabellids, Blood Circulation in, Reversible Stoppage of

the, Prof. H. Munro Fox, 26

Samarium: Radioactivity of, and the Formation of Hibernium Halos, Dr. J. H. J. Poole, 654: Range of Radiation from, Prof. G. Hevesy and Dr. M. Pahl, 434 Schneider Mediumship, Further Light on the, Lord Charles Hope, 549; H. Price, 658

Scientific Centralisation in the British Empire, Prof. C. Stanton Hicks, 397

Seismographs, Hydraulic, Prof. S. K. Banerji and K. N. Sohoni, 547

Sex: Artificial Control of, in the Progeny of Mammals, Prof. N. K. Koltzoff and V. N. Schröder, 329; Hormones and Cancer-Producing Compounds, Dr. J. W. Cook and Prof. E. C. Dodds, 205

Sheep, Nutritional Condition of, and Susceptibility to Stomach Worm, A. H. H. Fraser and Dr. D. Robertson,

Silver, Arc Spectrum of, Structure of the Lines of the, D. A. Jackson, 691

Snails and Changes in Sea-Level, Prof. T. D. A. Cockerell, 277

Soap Film Structure, The Hydroxyl Group and, W. J. Green, 873

Sodium Benzoate, Chlorination of, Dr. J. C. Smith, 28 Soot Films and Oil, Interaction between, J. H. Coste, 691; Dr. S. C. Blacktin, 873

Spearman's General Factor, Ambiguity in Sign of, Prof.

H. T. H. Piaggio, 170 Species, History of, The Bacterial Growth Curve and the, Dr. A. S. Corbet, 61

Spectral Lines in Indium and Gallium, Relative Intensity of, Miss R. Payne-Scott, 365

Sponges, Light-Producing Powers of, Prof. E. Trojan. 728

'Stainless-Invar', A New Alloy, Prof. Kôtarô Honda, 587 Strepinema Stage in Reduction, Prof. H. H. Dixon, 437

Sulphur: Hexafluoride, K. G. Denbigh and Prof. R. Whytlaw-Gray, 763; -Hydrogen Reaction, New Experimental Evidence in the, E. E. Aynsley, Dr. T. G. Pearson and Dr. P. L. Robinson, 471

Sunspots on the Solar Disc, Zones of Apparent Inhibition

of, Prof. J. A. Carroll, 548

Supersonic Vibrations set up in a Zinc Bar undergoing Transverse Vibrations, Prof. K. Prosad and S. Sharan,

Syllis ramosa, McIntosh, Distribution of the Polychæte Worm, Dr. C. Crossland, 242

Symmetrical Region, Variation Problems for a, Prof. H. Bateman; Prof. E. T. Whittaker, 472

Tektites, Origin of, F. Chapman; Dr. L. J. Spencer, 876 Tellurium Fluorescence, Magnetic Quenching of, R. Smoluchowski, 914

Thunderstorms, Summer, S. Morris Bower, 473

'Time': and 'Events', Ideas of, W. W. L., 727; Determination, Dr. J. de Graaff Hunter, 515; Measurements, Number 60 in, Prof. H. Chatley, 914

Timothy Grass, A Fly Pest of, L. A. L. King and Agnes

A. Meikle, 837

Trees, Twisted, H. G. Champion, 133

Triatomic Molecules, Structure of, A. B. D. Cassie, 438 Ultra-Sonic Radiation, Chemical and Biological Effects of, Prof. A. Szent-Györgyi, 278

Units, Fundamental, Dimensions of, Prof. H. Stansfield,

59; Prof. F. M. Denton, 585 Uracil, Nucleic Acids and, Irradiation of, F. F. Heyroth and J. R. Loofbourow, 92

Vibrating Metal Bars, Viscous Damping of, Prof. K. Sezawa, 803

Virus Diseases and Intracellular Inclusions in Plants,

Dr. F. M. L. Sheffield, 325 Viscosity Measurements of Liquids by the Oscillating Disc Method, Prof. C. E. Fawsitt, 97

Visual Experience, A Peculiar, Dr. F. A. Bather, 62 Vitamin A: Absorption Spectrum of, at Low Temperatures, Dr. F. P. Bowden, S. D. D. Morris, and Dr. C. P. Snow, 582; Highly Active, Characteristics of, Dr. F. H. Carr and W. Jewell, 92; in Cod Liver Oil, An Inhibitor of the Antimony Trichloride Test for, A. Emmerie, 364; B₁, Crystal Structure of, and of Adenine Hydrochloride, J. D. Bernal and D. Crowfoot, 911; B4 and Adenine, R. D. Heard, H. W. Kinnersley, J. R. O'Brien, Prof. R. A. Peters and V. Reader, 617; C Activity, Indophenol Reducing Capacity of Lemon Juice and its Fractions in Relation to, Dr. S. S. Zilva, 363; and Ascorbic Acid, A. L. Bacharach, 364; Constitution of, Dr. F. Micheel and K. Kraft, 274; E. G. Cox and Dr. E. L. Hirst, 402; in the Adrenal Gland, G. Bourne, 874; Methylnornarcotine, Glycuronic Acid, and, W. J. Dann, 24; Triphenylmethyl Derivative of, L. Vargha, 363

Wave Equations and the Conservation of Energy, A.

Lees, 402

Weather, 'Raw', Dr. G. M. B. Dobson; Sir Leonard Hill, 28; 241; Prof. S. Russ, 131; H. E. Beckett, 132

Whale Shark in the Waters Around Ceylon, The, Dr. E. W. Gudger, 165; Dr. J. Pearson, 729

Wheat-Germ Oil: Absorption Spectrum of the Unsaponifiable Matter from, Dr. R. A. Morton and J. R. Edisbury, 618; Vitamin E Fraction of, Absorption Spectrum of the, Dr. P. Bowden and T. Moore, 512

White Pine Blister Rust, Fusion of Pycniospores with Filamentous Hyphæ in the Pycnium in the, R. K.

Pierson, 728

Wireless Echoes, Recording, at the Transmitting Station, Prof. S. K. Mitra and H. Rakshit; R. A. Watson Watt and L. Bainbridge-Bell, 657

Wool: Fibres, Origin of Curls and Twists in, Dr. J. E. Nichols, 201; Methionine in, J. Barritt, 689

X-Radiation, Properties of, Prof. C. G. Barkla, 166 $\gamma\textsc{-Excitation}$, Mechanism of, by $\beta\textsc{-Disintegration}$, Dr. G. Gamow, 57

Corylophidæ (Coleoptera), New Species of, C. Deane, 287 Cosmic: Dust, Increase in Mass of a Planet by the Effect of, F. Zagar (1), 775; Radiation in South Africa, Intensity of, S. M. Naudé and J. E. C. Coventry, 411; Origin of H. Alfvén, 619; (Symons Memorial Lecture), P. M. S. Blackett, 429; Ray Phenomena, Wilson Photographs of, Blackett and Occhialini, 589; Rays: Prof. A. H. Compton, 807; Energy of, Prof. E. Regener, 130; Latitude Variation of, Prof. A. H. Compton, 769; Nature of, Prof. A. H. Compton, 713; Ultra-Radiation, Corpuscular Nature of, Earth-Magnetic Effect and the, Prof. J. Clay, 136

County Maps of England, Early (Review), 530

Crustacea, Control of Respiratory Movements in, Prof. H. Munro Fox and M. L. Johnson, 514

Crystal: Lattice Constants, Determination of, by Electron Diffraction, Prof. G. I. Finch and A. G. Quarrell, 842; Structure and Orientation in Thin Films, Prof. G. I. Finch and A. G. Quarrell, 877

Crystalline Style, Oxidase of the, C. Berkeley, 94

Crystals: of the Living Body, Sir William Bragg, 125; Orientation of Certain, by Hydrargillite, L. Royer, 596; Paramagnetic, at Low Temperatures, Principal Magnetic Susceptibilities of Some, Dr. L. C. Jackson, 338; The Form and Properties of, an Introduction to the Study of Minerals and the Use of the Petrological Microscope, Miss A. B. Dale (Review), 44; Thermal Expansion in, and Hauy's Law, A. Cavinato

Cubic Ferric Oxide, Dissociation of, A. Girard and G.

Chaudron, 447

Cuckoo, Swallow and, Arrive, 629

Cuckoo's Call, The, 773

Culture, Science and (*Review*), 707 Cultures, Single-Cell, Transferring, to Solid Nutrient Media, L. O. Koblmuller and R. Vierthaler, 287

Cuprite, Mono-Crystals of, The Photoelectric Effect in the, R. Deaglio, 887 Cuprous Oxide: in a Magnetic Field, A New Type of Photoelectric Effect in, Dr. I. Kikoin and Dr. M. Noskov, 725; Photo-Cell in Red Light, Reversal of the Current from a, Dr. H. H. Poole and Dr. W. R. G. Atkins, 133

Curare, the Neuro-Muscular Junction and, Tudor Jones, 693

Curlew, Long-billed, Restriction in Range of the, J. W. Sugden, 439

Cyanide-Cyanamide Equilibrium, The, A. Perret and R. Perrot, 375

Cyclades Islands, A New Snake from the, F. Werner, 287 Cyclone Season in the South Indian Ocean, R. A. Watson and N. R. McCurdy, 517

Cyclops, A New Species of, Found at Tenby, D. J. Scourfield, 135

Cynocephalus, Bone of Adult, and of Human Bone, Differential Microscopic Characters of the, Muller and Desmarez, 739

Cytology: Experimental, Third International Congress of, 797; Recent Advances in, Dr. C. D. Darlington (Review), 5

Daffodils, Narcissi and their Hybrids, H. W. Pugsley, 517 Dahomey, Rainfall in, Dr. R. Pignol, 208

Dairying, National Institute for Research in, Annual Report for year ending July 31, 1932, 870

Dairy Products, Marketing, Co-operation in, R. W. Bartlett. 2 Vols. (Review), 188

Daphnia magna Straus, Antagonistic Action of Ultraviolet Rays on the Phototaxy of, R. J. Wojtusiak, 703 Darwin Where He Did? Stands (Review), 709

Darwin's Barometer, 464

Davainea proglottina, Life-History of the Fowl Tapeworm, F. J. Brown, 276

Davy, Von, und Döbereiner bis Deacon: ein halbes Jahrhundert Grenzflächenkatalyse, Dr. A. Mittasch und E. Theis (Review), 150

Deaf and Dumb in England and Wales, 580

Death Valley a National Monument, 542

'Débacle', The, 373

Debye's Dispersion of Nitrobenzene, Prof. J. Weigle and R. Luthi, 327; Theory (formula of Gronwall, La Mer and Sandved), Application of, to solutions of Copper Sulphate, Mlle. M. Quintin, 630

Deep Drilling, New Record for, 721

Definite Integrals with Bessel's Functions, K. Mayr,

Dehydro-Ascorbic Acid, Constitution of, Prof. P. Karrer, H. Salomon and K. Schöpp, 800

Delinquency: Constitution Types in, Practical Applications and Bio-physiological Foundations of Kretschmer's Types, Dr. W. A. Willemse (*Review*), 188

Delta Orionis, The Variable Star, Dr. J. Plassmann, 661

Democracy: After, Addresses and Papers on the Present World Situation, H. G. Wells (*Review*), 183 Dental Caries, Cause and Nature of, J. J. Enright, H. E.

Friesell, and M. O. Trescher, 918

Depopulation: and Disease, Nutritional Aspects of, Dr. W. M. Strong, 519; The Medical Sciences in Relation to, Dr. R. W. Cilento, 519

De Sitter's World (Review), 487

Development Commissioner, Sir William Cecil Dampier appointed a, 581

Devonian Phosphates of Tennessee, Constitution of the, L. Cayeux, 666

Dew Points and Boiling Points of Mixtures of Volatile Liquids, A Boiling Point Apparatus for the Determination of the, L. Gay and J. Soulié, 630 Diabase Rocks at the You Yangs, near Geelong, A.

Coulson, 71

Diabetes, The Practical Treatment of, Dr. T. I. Bennett (Review), 9

Diamond, Genesis of the, A. F. Williams. 2 Vols. (Review), 255

Diamonds, Dr. A. E. H. Tutton (Review), 255

Diaporthe, Nits, The Genus, L. E. Wehmeyer, 880 Diatomic Molecules, Band-Spectra of, Dr. W. Jevons; Dr. R. C. Johnson, 102

Diazomethane, Photochemical Decomposition of, Kirkbride and Norrish, 404

Dictyoconoides, Lockhartia, and Rotalia, The Genera, Col.

L. M. Davies, 279 Dielectric Phenomena. 3: Breakdown of Solid Dielectrics, S. Whitehead. Edited, with a preface, by E. B. Wedmore (Review), 384

Differential Equations from the Algebraic Standpoint,

Prof. J. F. Ritt (Review), 456 Differenziali e propagazione ondoso, Caratteristiche dei sistemi, Prof. T. Levi-Civita. Lezioni raccolte dal

Dr. G. Lampariello (Review), 45 Diffraction Gratings used with Grazing Incidence, O. U. Vonwiller, 668

'Diffusion' and Disease, Dr. C. Nicolle, 879

Diflavones, Synthesis of, J. Algar, Isabella M. McCarthy, and Eveline M. Dick, 666

Dimetaphosphates, Preparation of the, P. Pascal and Mme. Réchid, 666

Diphenyl Series, Crystal Structure of the, Lucy W. Pickett,

Diphtheria, Past and Present: its Ætiology, Distribution, Transmission and Prevention, Dr. J. G. Forbes (Review), 150

Direction Finding of Birds, An Experiment in, 701

Discontinuous Functions, Effective Integration of, L. Laboccetta (2), 71

Discovery, The Great Age of. Edited by Prof. A. P. Newton (Review), 260

Disease: Classical Descriptions of, with Biographical Sketches of the Authors, Prof. R. H. Major (Review), 895; The Inborn Factors in, an Essay, Sir Archibald E. Garrod (Review), 314

Diselenomesoxanilides and Oxyselenanilides, A. Baroni, 107 Dissociation Formula, A Simple Derivation of the, D. H. Menzel, 739

Dodecolopoda mawsoni, A New Species representing a New Genus of Pycnogonida, Dr. W. T. Calman and Isabella Gordon, 774

Dodo: George Edwards's Picture of the, 615; The, and

the Aphanapteryx, G. Renshaw, 728
Donnan Equilibria: The, and their application to Chemical, Physiological and Technical Processes, Dr. T. R. Bolam (Review), 859

Dove Marine Laboratory, Report for year ending June 30,

542

Downtonian and Dittonian of Great Britain and Northwest Europe, W. W. King, 738

Dragons, G. D. Hornblower, 806

Drink Problem, The Social and Economic Aspects of the (Review), 187

Drosophila: Genetic Studies on, 627; melanogaster, Genetics of certain Chromosome Anomalies in, A. H. Sturtevant and T. Dobzhansky, 627; obscura, Fallen, Spermatogenesis in, C. Koller and Thelma Townson (1), 447

Dublin University, Sir Charles Arthur Kinahan Ball appointed Regius professor of surgery, and Prof. J. Purser professor of civil engineering, 772

Duddell medal of the Physical Society, Award of the, to Prof. W. Gaede, 195

Durham University, Miss Doris L. Reynolds appointed lecturer in geology, 811

Dyes, Anti-Stokes Fluorescence in, Efficiency of, Dr. A. Jabloński, 839

Early Man in East Africa: 427; 462; Reports of Committees on, 477

Earth: The Common, E. L. G. Watson (Review), 636; -Magnetic Effect and the Corpuscular Nature of

(Cosmic) Ultra-Radiation, Prof. J. Clay, 136 Earthquake: Alaskan, of April 26, 757; Californian, of March 10, 509; Damage and Earthquake Insurance, Dr. J. R. Freeman (Review), 381; in the North of England, 88; in North-west China, 88; in Southeast Africa, 19; of March 2, The, C. E. Brazier and L. Génaux; C. Maurain, 596; Series at Nagusa, Japan, A. Imamura, T. Kodaira and H. Imamura, 136; the Californian, 391; the Kansu, of Dec. 26, 304; The Recent Japanese, Dr. C. Davison, 351;

Wensleydale, Father J. P. Rowland, 128
Earthquakes: Apparent Velocity of the Surface Propagation of, in relation to the Hypocentral Depth, C. Alessandri, 923; Bibliography of, 580; Destructiveness of (*Review*), 381; Felt and Recorded in Tokyo, Seismometrical Reports on the, 359; in the Holy Land: a Correction, Prof. Bailey Willis, 550; of Northern Africa, Mme. A. Hée, 807; P Wave in, Corrections to the Times of the, Dr. H. Jeffreys and

K. E. Bullen, 97

East: Anglia, Prehistoric Society of, Annual Business Meeting, Dr. Cyril Fox elected president for 1933, 520; Anglian Flora, The, Prof. E. J. Salisbury, 336 Eclipse of Hi and Ho, Dr. J. K. Fotheringham, 881

Eclipses, Mutual, and Occultations of Jupiter's Satellites, 173

Ecological Studies in Victoria, R. T. Patton (1), 703

Economic: Biologists, Association of, election of officers, 304; Conference, The World, 866; 889; Recovery, Education and, Dr. J. Macmillan Brown, 538; Sanctions, Inadequacy of, Dr. J. J. van der Leeuw, 867 Economics: in Primitive Communities, Prof. R. Thurn-

wald (Review), 315; International (Review), 451 Ecuador, New Fossil Fresh Water Molluscs from, W. B.

Marshall and E. O. Bowles, 589 Eddy Diffusion, O. F. T. Roberts, 916

Edinburgh University: award of the Cameron prize to Dr. George F. Dick and Dr. Gladys H. Dick, 177; Sir Ian Hamilton installed as Lord Rector, 372; offer of honorary doctorates, 408; Prof. I. de Burgh Daly appointed professor of physiology, 700

Education: and Economic Recovery, Dr. J. Macmillan Brown, 538; for Trades and Industries: a Historical Survey, C. T. Millis (*Review*), 44; The Crisis in, 628

Eels—an Early Hypothesis, 665

Egyptian: Neolithic Barley, A. Jackson, 652; Pottery, Corpus of, 357

Eiffel Tower, Radiation of the Radiotransmitter of the, P. David, 179

Einstein: The Case Against, Col. A. Lynch (Review), 260; re review of, Col. A. Lynch, 622

Einsteinian Correction of Time in a Planetary Motion,

M. Cimino, 107 Elastic: Electron Scattering in Gases, Dr. S. Werner, 726; Energy Theory, Prof. J. A. van den Broek (Review), 347

Elasticity, 'Linked', and its Mathematical Representation,

E. Volterra, 107

Electric: Trolley Omnibuses, C. J. Spencer, 869; Weld-

ing in Ship Construction, 796

Electrical: Age: The, being further Everyday Marvels of Science, V. H. L. Searle (Review), 491; and Rubber Equipment on the Farm, 904; Conductivity and Atmospheric Condensation Nuclei during a voyage to Greenland, C. Maurain and J. Devaux, 35; Energy by Frog-Skin, Output of, W. L. Francis, 805; Engineers, Institution of, Ll. B. Atkinson elected an honorary member of the, 128; awards of the, 723; Equipment of Buildings, Regulations for the, 905; Measurements, Advanced, Dr. W. R. Smythe and Dr. W. C. Michels (*Review*), 322; Resistance, National Standards of, First Comparisons of the, A. Pérard and M. R. Romanowski, 887; Terms: A Dictionary of, including Telegraphy, Telephony and Wireless, S. R. Roget. Second edition (*Review*), 322 Electricity: and Magnetism for Beginners, W. C. Bad-

cock and Dr. E. J. Holmyard (Review), 347; Atmospheric, Dr. B. F. J. Schonland (Review), 785; Gas and other Fuels as Heating Agents, A. H. Barker, 32; Supply and Distribution, Costs of, J. M. Ken-

nedy and Miss D. M. Noakes, 358

Electrification: of an Ice Factory, J. H. Lamb, 833; of Railways in Britain, F. Lydall, 19; of the Swedish State Railways, Öfverholm, 199

Electrodynamic Forces, Nature of, G. Eibenschütz, 923

Electrolytes (Review), 382

trolytic: Copper Deposited in the Presence of Colloids, Strains in, P. Jacquet, 70; Dissociation, Rôle of the Solvent in, Dr. A. R. Martin, 584; Dr. Electrolytic: J. A. V. Butler and Miss L. C. Connell, 800

Electromagnet, A Compact, for General Purposes, Dr. L. F. Bates and B. J. Lloyd Evans, 558

Electrometer Triode in the X-ray Ionisation Spectrometer, Dr. W. A. Wooster, 545; Dr. B. W. Robinson, 546

Electron: Diffraction: by Films of Grease, Prof. G. P. Thomson and C. A. Murison, 237; by Vapours, Dr. H. de Laszlo, 803; Energy of the Beams in, Prof. R. Whiddington and F. C. Poultney, 814; Inelastic Scattering in Gases, C. B. O. Mohr and F. H. Nicoll, 143; Motion of an, in a Crystalline Lattice, G. C. Wick, 71; Polarisation? Prof. R. Whiddington, 908; Positive, A Possible Property of the, Dr. W. Elsasser, 764; Scattering by Atoms, Further Calculations of, Massey and Mohr, 368; Polarisation Phenomena in, E. Rupp, 209

Electronic Oscillations, Production of, with a Two-Electrode Valve, J. S. McPetrie, 691

Electrons: in Amorphous and in Crystalline Antimony, Diffraction of, J. A. Prins, 760; in Gases, Large Angle Scattering of, C. B. O. Mohr and F. H. Nicoll, 143; in Helium: at Zero Angle, Inelastic Scattering of, Prof. R. Whiddington and J. E. Taylor, 814; Scattering of, Prof. R. Whiddington and T. Emmerson, 814

Correct Arrangement for, Mlle. Electrophoresis, A Choucroun, 630

Electrostatic Precipitation, Direct-Current Generators for, 162

Elektrischen Ladung der Erde, Die Aufrechterhaltung der, Prof. E. Schweidler (Review), 785

Elektrolyte, Prof. H. Falkenhagen (Review), 382

Elements: Artificial Transmutation of the (Boyle lecture), Lord Rutherford, 832; Chemical Detection of, Prof. F. A. Paneth and P. L. Günther, 652; Light, Disintegration of, by Fast Protons, Dr. J. D. Cockcroft and Dr. E. T. S. Walton, 23; Small Quantities of, Spectroscopic Detection of, W. Späth, 30; Transmutation of the, Recent Researches on the, Lord Rutherford, 388

Elvers, Arrival of, 409

Empire: Broadcasting, 16; Forestry Handbook, new edition, 870; Journal of Experimental Agriculture, 304; No. 1, 759; Timbers for Structural Design. British Columbia Douglas Fir, 244

England: The Map of, or About England with an Ordnance Map, Col. Sir Charles Close (Review),

English Language: Use of the, Dr. C. Norwood, 197;

741

Entomological Society of London, The History of the, 1833–1933, Dr. S. A. Neave, assisted by F. J. Griffin (Review), 678

Enzyme which dehydrogenates Stearic Acid, Presence in the Bile of an, G. Quagliariello, 411

Enzymes: A Discovery and its Consequences, Dr. E. F. Armstrong, 535

Eremophila Mitchelli, Essential Oil from the Wood of, A. E. Bradfield, A. R. Penfold, and J. L. Simonsen,

Ergosterol and Cholesterol, Hydroxyl Group in, Prof. I. M. Heilbron and J. C. E. Simpson, 438

Eros, Mass of, Dr. K. Lundmark, 31

Erysiphe graminis Tritici, Host Specialisation of, E. B.

Mains, 739
Eskimo: Culture Sequence, J. A. Ford, 243; in Greenland, Spread of, Capt. E. Mikkelsen, 367; Vertebræ, Anomalous, Dr. T. D. Stewart, 125

Essential News, 614

Ethane, Slow Combustion of, at High Pressures, D. M. Newitt and A. M. Bloch, 214

Ethics and Archæology, 561 Ethnogenics, Capt. G. Pitt-Rivers, 89

Ethyl Alcohol and Ethyl Ether, Freezing Points of Binary Mixtures of, A. Lalande, 447

α-Ethylene Oxides, Some, R. Lespieau and Mlle. B. Gredy, 447

Eucalypts, Chemical Composition of the Woods of the Ironbark Group of, W. E. Cohen, A. L. Baldock, and A. G. Charles, 333

Eucalyptus, Identification of the Coloured Woods of the Genus, H. E. Dadswell and Maisie Burnell, 280

Eugenics and Marriage Laws, L. M. Crump, 540

Europe: Northern, Spring in, 812; Southern, Dr. Marion I. Newbigin (*Review*), 317 Evelyn's "Fumifugium", Reprint of, 651

Events', 'Time' and, Ideas of, W. W. L., 727 Everest: Mount, Col. H. L. Crosthwait, 10; Flight, Preliminary Tests for, 160; Projected Flight over, 392; Houston Expedition over, 499

Evolution: and Philosophy, Prof. R. R. Gates (Review), 380; Creative, The Mechanism of, Dr. C. C. Hurst (Review), 780; Taxonomy and, Dr. W. H. Longley, 863; The Causes of, Prof. J. B. S. Haldane (Review), 709; The Scientific Basis of, Prof. T. H. Morgan (Review), 380; Up to Date (Review), 780

Examination Methods, Committees on, 141

Exchange of Goods during Economic Depression, 834 Exhibition of 1851, appointments to Senior Studentships, 848

Experimentalphysik, Handbuch der, Herausgegeben von W. Wien und F. Harms. Unter Mitarbeit von H. Lenz. Band 4: Hydro- und Aerodynamik. Rohre, Offene Gerinne, Zähigkeit. Herausgegeben von L. Schiller. Bearbeitet von L. Schiller, F. Eisner, S. Erk (Review), 602

Explosion Flames, Photographic Analysis of, Prof. W. A.

Bone, 300

Explosions, Experimental, Air Waves from, Dr. F. J. W. Whipple, 138

Explosive Chain Reaction: between Hydrogen and Oxygen, Upper Pressure Limit in the, C. N. Hinshelwood and G. H. Grant, 361; Upper Limit in, H. W.

Melville and H. L. Roxburgh, 690
Explosives: Chemical Examination of, A. Marshall (Review), 186; Their History, Manufacture, Properties and Tests, A. Marshall. Vol. 3 (Review), 452; War and Second edition. War and Post-War, Prof. R. C. Gale (Review), 452

Exponential and Hyperbolic Functions, The, and their

Applications, A. H. Bell (Review), 319

Extra-Galactic Nebulæ, The, Dr. H. Knox-Shaw, 247 Extreme Ultra-Violet, Studies in the, and the very soft X-ray Region, Prof. M. Siegbahn (Guthrie lecture), 901 Eyes and Colour Change, Prof. Ll. M. Bertholf, 331

Fabrication, An Intentional, of an Engraving of a Horse's Hoof on Granite, by means of a Stone Tool, M. Badouin, 375

Facial Growth in Children, Corisande Smyth and M.

Young, 29

Fact, Convention and, Dr. N. R. Campbell, 237

Faith, Hope and Charity in Primitive Religion, Dr. R. R. Marett (Review), 9

Faraday Effect of a Camphor Derivative, J. Verhaeghe, 666 Faraday House Journal, 466

Farm, Electrical and Rubber Equipment on the, 904

Fat in Æstivating Animals, 268

Fauna of Hot Springs in North America, C. T. Briers, 403

February Fill-Dyke, 141 Feldspäte, Die, und ihre praktische Bestimmung, Dr. K.

Chudoba (Review), 384

Ferns, Sporangia containing Spermatozoids in, Alice E. Gairdner, 621

Ferromagnetics, Atomic Moments of, Dr. E. C. Stoner, 433 Ferro-Silicons Rich in Silicon, Magnetic Susceptibility of, C. Bedel, 375

Ferrous Metals and Alloys, Cast, Sampling and Analysis of, 845

Fibres from the Coat of a Blackface Lamb, Dr. S. G. Barker, 799

Fifty-five Year Rule, The, 146 Film Institute, A British, 465

Filter for the 3130 Mercury Line, A Method of Preparing a, G. C. Brock, 410

Fire, Safety of Life from, Col. G. Simonds, 269

Fish: and Fishing (Review), 42; Cultivation in the Philippines, 522; Culture, 177; Movements, Directional Control of, Dr. J. Gray, 774; Species (Cyprinides), Three New, from Asia Minor, V. Pietschmann, 704

Fishery Research: Advisory Committee on, of the Development Commissioners, Prof. E. W. MacBride appointed chairman of the, 871; in Newfoundland, 918

Fishes, Muscular Movements of, Dr. J. Gray, 825

Flies Go in Winter? Where do, 249

Flight: 'Instrumental', Stability of, G. A. Crocco, 71; Non-stop, New Record for, Squadron-Leader O. R. Gayford and Flight-Lieut. G. E. Nicholetts, 232

Flint Miners of Blackpatch, The, J. H. Pull (Review), 677

Flood-Lighting and Bird-Life, 55

Flora: of Surrey, late C. E. Salmon. Edited by Dr. W. H. Pearsall (Review), 77; The East Anglian, Prof. E. J. Salisbury, 336

Flow Kinetics: Model of Photolysis, E. Abel and H. Schmid, 251

Flower-Size, Inheritance of, Prof. R. R. Gates, 136

Fluorescence, Negative Polarisation in, K. S. Krishnan and S. M. Mitra, 204

Fluorescent Screens, the Spurious Ring Exhibited by, J. V. Hughes, 558

Fluorite from Obira, Japan, Bands in the Thermoluminescent Spectrum of, Ei-ichi Iwase, 909

Fluorites, Luminescence Investigations with, (2), H. Haberlandt, 851

Flyfisher, The, and the Trout's Point of View, Col. E. W. Harding (Review), 42

Fly Swarms in Dwelling Houses, Composition of, 249

Fog Droplets, Size of, M. Kinoshita and K. Uchiyama, 99 Food: Consumption after Fasting, Capacity for, Famiani, 339; Relation of, to Disease, S. Dixon, 333; Storage at Low Temperature, Sir William Hardy (Hunter Memorial Lecture; Sir William Trueman Wood Lecture), 459; Value, Reconstructive, of the Embryos of Various Cereal and Leguminous Seeds, V. Famiani, 339

Foods, The Structure and Composition of, Dr. A. L. Winton and Dr. Kate Barber Winton.

(Review), 316

Forest: Fires: and Weather, T. Terada and T. Utigasaki, 161; in relation to Soil Fertility, Prof. F. P. Worley, 787; on the Roof of the (Review), 78; Products Research Board, Report for 1931, 847

Forestry: British, Problems of, Prof. R. S. Troup, 900;

Commission, Twelfth Annual Report, 406

Formaldehyde in Dew, Presence of, Prof. N. R. Dhar and Atma Ram, 800

Forthcoming Books of Science, 322

Fossil: Bones: Bakelite Impregnation of, H. W. Nichols and P. C. Orr, 127; Preservation of, L. McL. Mann, 366; W. J. McCallien, 694; Mammals, Some Ancestral (Review), 454

Fossiliferous Deposits in the Homa-Kendu Area, Kenya

Colony, Dr. L. S. B. Leakey, 886 Fourier Analysis and Vowel Curves, R. H. Nisbet, 401 Fowl, Non-Disjunction in the, Prof. F. A. E. Crew, 446 François Joseph Archipelago, Circumnavigation of the,

by the North, J. Schokalsky, 70 Franklin Institute, award of medals to Dr. Orville Wright,

Dr. P. Sabatier and J. de la Cierva, 870, 871
Franz Josef Land, Voyage around, Prof. N. N. Subov, 359
Frazer Lectures, The, 1922–1932, by divers hands.
Edited by W. R. Dawson (Review), 712

French: Chemical Society, Annual Meeting of the, 870; Society of Physical Chemistry, 25th anniversary of

the, 797

Freshwater Biological Association, election of officers, 796 Frog, Hermaphrodite, A. A. M. Gardiner, 330

Frogs: and Fishes, Showers of, 666; Hibernation of, 34; Spawning in, Factors Controlling date of, R. M. Savage, 587; Spawning of, and a remarkable Tadpole, 521

Frost and Shore Animals, 142

Froude, William, Laboratory, Ship Researches at the, 162 Fruits, Hardy, Trials of Varieties of, for Commercial Purposes, 63

Fuel: Gas, Water and Lubricants, The Analysis of, Prof. S. W. Parr. Fourth edition (Review), 860; -Knock, A Theory of, Dr. S. Steele, 724; Research Board, Report for year ending March 31, 1932, 627; Testing: Laboratory Methods in Fuel Technology, G. W. Himus (Review), 259

Fulminating Material, E. Mathias, 482 Fundulus, Xanthophores in, A. E. Warren, 287

Fur-Bearing Animals in Michigan, Dr. N. Dearborn, 127 Furunculosis, 213

Furze or Gorse (Ulex europœus) Flowering, 34

G., The Constant of Gravitation, Prof. V. V. Narlikar, 134 γ-Excitation by β-Disintegration, Mechanism of, Dr. G. Gamow, 57

γ-Rays: Anomalous Absorption of, B. Arakatsu, 696; Internal Conversion of, Hulme; H. M. Taylor and Mott, 99; Ellis and Mott, 517

Gafsa Hills, Tunisia, Pleistocene Dating of, R. Vaufrey, 659 Galactic: Co-ordinates, Tables for, J. Ohlsson, Dimensions, H. Shapley, 739; Rotation, Miss Phyllis Hayford, 173

Galaxy, Absorption of Light in the, Prof. J. Stebbins and Dr. C. M. Huffer, 769

Gallium Isotopes 69 and 71, Nuclear Moments of the, J. S. Campbell, 204

Gammarus chevreuxi, Sexton, Further Mutations in the Amphipod, Mrs. E. W. Sexton and Miss A. R. Clark,

Gannets, Return of the, 373

Garnet in the Dartmoor Granite: its Petrogenetic Significance, A. Brammall and S. Bracewell, 250

Gas, Use of, Development of the, S. Lacey, 478

Gases: Conductivity of Mixtures of, Dr. S. P. McCallum and L. Klatzow, 841; Rare, Discovery of the, Prof. M. W. Travers, 65

Gaseous: Combustion: at High Pressures, Prof. W. A. Bone, and others, 245; Researches on, Prof. W. A. Bone (Bakerian lecture), 494; Hydrogen Iodide, Action of, on some Iodine Derivatives of Hydrocarbons, G. Emschwiller, 738; Ions at Cathodes, Accommodation Coefficient of, K. T. Compton, 631; Oxidation, A. Ritchie and Ludham, 64; Tubes for Lighting, J. N. Addington, 235

Gastropods: Crystalline Style in, Prof. C. M. Yonge, 915;

Style-Sac of, R. V. Seshaiya, 30 Gaza, Old, Sir Flinders Petrie, 868

Geddes's Comet, 441

Gegenschein, The, or Counterglow, Dr. C. T. Elvey, 590 Genetical Studies, Modern (Review), 185

Gentleman, A Very Gallant, Commdr. L. C. Bernacchi (Review), 782

Genus Lolium, Nitrogen Fixation in the, B. Brown, 169 Geographical: Association, Annual Conference of the, Dr. L. D. Stamp, 68; Names, Foreign, Report on, 161

Geography, An Approach to, Dr. H. R. Mill, 68 Geological: Society of London, awards of the, 159; award of the Daniel-Pidgeon fund to J. D. Solomon, 616; election of officers, 432; Survey of Great Britain, Summary of Progress for 1930, 370

Géologie et les mines de la France d'outremer, La (Review),

823

Geologische Erforschung Thüringens in älterer Zeit: Die, ein Beitrag zur Geschichte der Geologie bis zum Jahre 1843, Prof. B. von Freyberg (Review), 531

Geology: and Glaciation of some Islands of the Southern Ocean, etc., Sir Douglas Mawson, 847; and Petrology of the Mount Leinster district, North-east Victoria, E. Broadhurst and J. D. Campbell, 703; and Physical Geography, A French-English Vocabulary in, G. M. Davies (*Review*), 604; Economic, Aerial Survey in Relation to, D. Gill; H. Hemming, 160; in Great Britain, 370; in the Life of a Nation, The Place of, Dr. L. L. Fermor; Dr. L. D. Stamp, 226 Geometry: Coordinate, A Treatise on (Review), 603;

Coordinate, Elements of, J. M. Child (Review), 603

Geophysical Prospecting, 791

Geophysics, The Contribution of Radiotelegraphy to, 642

Georgia and its People (Review), 309 Georgian People: A History of the, from the Beginning down to the Russian Conquest in the Nineteenth Century, W. E. D. Allen (*Review*), 308 Geothermic Gradient in Limagne, G. Grenet, 143

German: Chemical Plant Exhibition ('Achema VII'), 1934, 907; Universities, Resignation of Jewish Professors, 687

Gestalt Theory, The, and the Problem of Configuration, B. Petermann, translated by M. Fortes (Review), 604

Glabratic Acid, G. Koller and G. Pfeiffer, 411

Glasgow, Royal Technical College, Report of, 213

Glasshouses, Disinfection of, with Sulphur, W. H. Read and O. B. Orchard, 844

Glauconite in Limestone, Modes of Existence of, L. Cayeux, 143

α-Glucoheptulite, Identification and Biochemical Oxidation of, Mme. Y. Khouvine and G. Nitzberg, 339

Glucose: and Shedding the Shell of Crustaceans, Mme. Andrée Drilhon, 483; Metabolism, Glutathione, Iodoacetic Acid and, Dr. J. H. Quastel, 206

Glucuronic Acid, A New Unsaturated Derivative of, J. Pryde and R. T. Williams, 57
Glutathione, Iodoacetic Acid and Glucose Metabolism,

Dr. J. H. Quastel, 206

Glyceria aquatica, Hydrocyanic Acid in, P. Guérin, 106 Glycogen: in Cartilage, Dr. H. B. Fell and Dr. R. Robison, 62; Triacetate, W. S. Reich, 106

Glycuronic Acid, Methylnornarcotine, and Vitamin C, W. J. Dann, 24

Gob Fires, Experimental, T. N. Mason; F. V. Tideswell, 521

God, Existence and Nature of, A Study of the Ordinary Arguments for the, Rev. A. C. Bouquet (Review), 677

Gold: in Kenya: 231; and Native Reserves, 37; -bearing strata of the Region of St. Yrieix (Haute-Vienne), Wakker, 483; Organic Compounds of, Prof. C. S. Gibson, 130

Gonad-stimulating Hormones in Hypophysectomised Animals, Prof. J. B. Collip, Dr. H. Selye, and Prof. D. L. Thomson, 56

Gondwanaland, Land Bridges of, Prof. C. Schuchert,

Gordon, Genl., and the Double Coconut, 337

Gorgonacea from the Great Barrier Reef, Prof. S. J. Hickson, 879

Government Chemist, Work of the, 246

Grain Conference, World, 758 Granary, A Modern, 431

Grass: -Land, Manuring of, 373; Silage, C. Boyle and J. J. Ryan, 410

Graticules, Photographic, Sir Herbert Jackson, 766 Gray, Thomas, Memorial Trust, awards under the, 359 Great: Bear Lake Indians, Dr. C. M. Osgood, 623; Crested Grebe in Britain, Spread of the, T. H.

Harrisson and P. A. D. Hollom, 135

Greenkeeping Research, 272

Green Plants in a Confined Atmosphere, Life of, J.

Beauverie and Mlle. S. Monchal. 36 Greenland: East, New Map of, 90; Ice, Thickness of the, Dr. E. Sorge, 807; Total Calorific Radiation in, C. Maurain and J. Devaux, 106

Greenwich, Royal Observatory, Annual Visitation, 882 Grid: in Great Britain, Completion of the, 539; the French, 539

Grouse, Red, Migration of, 579

Growth, Influence of Living-Space upon, Dr. J. Podhradsky, 29

Guatemala, Mexico and, Recent Discoveries in, 101

Guébhard-Séverine, Annales, No. 8, 797

Guggenheim, Daniel, medal, award of the, to Commdr. J. C. Hunsaker, 871 Guiana Forest, A Naturalist in the, Major R. W. G.

Hingston (Review), 78

Gulick: John Thomas, Evolutionist and Missionary, portrayed through documents and discussions, A. Gulick (Review), 532

Gypsy Art in Russia, M. A. Barannikov, 271 Gyrocotyle, Development of, J. S. Ruszkowski, 243 Gyrostats, Self-Erecting, Prof. J. G. Gray, 250

H-rays in Different Gases, Ionisation of, J. Schintlmeister, 251

Halley's Comet: in 1909-11, Dr. A. C. D. Crommelin, 282; N. T. Bobrovnikoff, 282

Halley, Edmond: Dr. A. Ferguson, 153; Correspondence and Papers of, Arranged and edited by E. F. MacPike,

Halogens in Organic Substances, Estimation of the, by the Sodammonium Method, F. Govaert, 179

Hancock's Steam Omnibus, Centenary of, C. E. Lee; C. F. D. Marshall, 647

Harrison Memorial Prize, award of the, to Dr. H. J. Emeléus, 91

Harrison's First Marine Timekeeper, 355

Harvard University: Dr. J. B. Conant elected president of, 700; Physics at, 395

Harveian Oration, 1932, Sir George Newman, 407

Harvey: and Preventive Medicine, 407; and the College of Physicians, The Debt of Preventive Medicine to, Sir George Newman, 407

Haut-Katanga: Petrographical Researches in the, M. Gysin, 411; (2), 815; (3), 887
Hawaii, Kepelino's Traditions of, edited by Martha

Warren Beckwith, 474

Hawaiian Feathercape, H. G. Beasley, 171

Hearing: Evolution and Mechanism of (Review), 147; in Man and Animals, Dr. R. T. Beatty (Review), 147; Tests of, 163

Heart, Abnormal Movability of the, S. Osawa, 241

Heat: in Air, Natural Convection of, when the heating is very small, P. Vernotte, 179; Transmission of. through Fabrics, M. C. Marsh, 558

Heating: Agents, Electricity, Gas and other Fuels as, A. H. Barker, 32; and Ventilating Engineers, Institution of, Journal of the, March, 541

Heathen Baptism in Early Britain, Miss Eleanor Hull, 403 Heaviside Layer, New Observations on the, Ratcliffe and White, 807

Helm or Steering Orders, The New, 20 Helminosporium avenæ, Sporulation of, in Artificial Culture, W. A. R. Dillon Weston, 435

Helium: Beryllium and, Lord Rayleigh, 724; Lique-faction Plant at the Clarendon Laboratory, Oxford, Prof. F. A. Lindemann and T. C. Keeley, 191

Heredity and Hibernation in Insects, 249 Herring Fishery off East Englia, Forecasting the, Dr. W. C. Hodgson, 98

Herrings: and the Origin of Petroleum, 578; Icelandic, Food of, Dr. P. Jespersen, 884

Hertzian Domain, Dispersion in the, J. Weigle, 144 Heteroazeotrope, The Ternary, W. Swietoslawski and E. Wardzinski, 887

Heterochrome Photometry of Incandescent Lamps, Solu-

tion of the Problem of, G. Ribaud, 667 Heterodera schachtii, the Potato Eelworm, J. Carroll, 850 Hexafluorides of Sulphur, Selenium and Tellurium, Structures of the, L. O. Brockway and L. Pauling,

Hexane, Action of Steam on, C. Matignon and M. Séon, 286 'Hexuronic: Acid' (Ascorbic Acid) as the Antiscorbutic Factor, Prof. A. Szent-Györgyi and Prof. W. N. Haworth, 24; and its Chemical Determination, T. W. Birch, Dr. L. J. Harris, and S. N. Ray, 273;

W. J. Dann, 274

Hibernium Halos, Formation of, Radioactivity of Samar-

ium and the, Dr. H. J. H. Poole, 654

High: -Frequency Currents produced by High-Tension Magnetos, J. Jaffray, 559; Discharge in Gases, T. V. Ionesco and Mme. Irène Mihul, 887; J. C. Wilson, 546; Precision Levelling, Influence of the Daily Oscillations of the Vertical on the Results of, E. Prévot, 559; Pressure Boilers, C. H. Davy and C. H. Sparks, 90; Research, Prof. P. W. Bridgman, 280; The Physics of, Prof. P. W. Bridgman (*Review*), 259; Selectivity Tone Corrected Receiving Circuits, F. W. Colebrook, 442; Voltages, Electrostatic Production of, Van de Graaf, Compton, and Van Atta, 475

Highway Research, Developments in, Prof. Steinburg, 334 Hindu Metal Images, Dr. F. H. Gravely and T. N.

Ramachandran, 279

Hispine Beetles, Early Stages of, S. A. Maulik, 171

History: as Science, 525; in Science, 777 Hochelaga, Plan of, W. D. Lighthall, 64 Hochwipfel Strata of the Carinthian Alps, A Fauna from the, K. Metz, 287

Hæmolytic Streptococci: The, their Grouping by Agglutination, late Sir Frederick W. Andrewes and Ethel M. Christie, 593

Honduras, South-Eastern, Archæological Expedition to,

Hong-Kong, Climate of, T. F. Claxton, 104; Col. E. Gold,

Hormones, Gonad-Stimulating, in Hypophysectomised Animals, Prof. J. B. Collip, Dr. H. Selye and Prof. D. L. Thomson, 56

Horticultural Education Association, Year Book of the, 1932, 616

Houston Expedition over Everest, 499

Huddersfield Technical College, Work of the, 557 Human Reflexology: General Principles of, an Introduction to the Objective Study of Personality, Prof. V. M. Bechterev. Translated by Emma and W. Murphy (Review), 675

Hume, David, Dr. B. M. Laing (Review), 321 Humidity Tables, Negretti and Zambra, 589

Humour and Humanism in Baeyer's Laboratory, Prof. J. Read, 294

Hurricanes of the Southern Hemisphere, 34

Huxley and Scientific Education (Huxley Memorial Lecture), Prof. H. E. Armstrong, 684

Hydra, Nerve-Net of, C. H. McConnell, 207 Hydro- and Aerodynamics (Review), 602

Hydrocarbon: Combustion in an Engine, A. Egerton and F. Ll. Smith, 725; Vapours, Polarity of, McAlpine and Smyth, 517

Hydrocarbons Produced by the Action of Indene on Fatty Organomagnesium Derivatives, Velocity of Disengagement of, D. Ivanoff and I. Abdouloff, 483 Hydrocephalus, a Hereditary Character in the House

Mouse, F. H. Clark, 288

Hydrochloric Acid, Influence of Temperature on the Absorption of Aqueous Solutions of, in the Extreme Ultra-Violet. R. Trehin, 179

Hydrodynamics: Sir Horace Lamb. Sixth edition (Review), 313; Evolution of, Dr. H. Jeffreys (Review),

Hydrogen: and Chlorine, Photochemical Reaction of, Prof. H. B. Baker, 27; Prof. A. J. Allmand, 656; and Oxygen: in a Silver Vessel, Combination of, C. N. Hinshelwood, E. A. Moelwyn-Hughes and S. C. Rolfe, 142; Reaction between, C. N. Hinshelwood, E. A. Moelwyn-Hughes and S. C. Rolfe, 625; Electrode, Improvements to the, for the Measurement of Hydrogen Ion Concentration in Solutions, P. Lecomte du Nouy, 179; Isotope, G. N. Lewis, 590; on Mercury, Initial Electrolytic Over Voltage in the Disengagement of, R. Dufour, 482 Hydrogenations, Catalysed, Velocity of, A. Kailan and

O. Stuber, 251
Hydronomics: Report of the Committee on, Division of Physical Sciences, National Research Council

(Review), 112; (Review), 112 Hydroxyl Group: in Ergosterol and Cholesterol, Prof. I. M. Heilbron and J. C. E. Simpson, 438; The, and Soap Film Structure, W. J. Green, 873

Hygrometer, The Wet-and-Dry Bulb, F. J. W. Whipple,

Hygroscopy: Capacitance, Dr. W. L. Balls, 329; Capacitance, and Some of its Applications, Dr. R. K. Schofield, 96 Hyper-Elliptic Surfaces from the Real Point of View,

Classification of, S. Cherubino, 339
Hyperfine Structure and Nuclear Moments, Prof. B. Venkatesachar and L. Sibaiya, 844

'Iberomarusian', E. G. Gobert and H. Vaufrey, 367 Ice Bridge, The: 557; A Correction, 921 Iceland Spar, Pleochroism of, P. Le Roux, 447 Ideal Home Exhibition, Science at the, 500

Ideas, Adventures of, Prof. A. N. Whitehead (Review), 746 Ignition-Coil Discharge Characteristics, Control of, Prof. G. I. Finch and R. W. Sutton, 374

Illuminated Fountains, A. S. E. Ackermann, 396 Immiscible Liquid Layers, Systems of, Four, E. Lester

Impulse Generator, A New, for Three Million Volts, T. E. Allibone, F. S. Edwards and D. B. McKenzie, 129 Incentives in Industry, The Problem of, Dr. G. H. Miles

(Review), 321 Index Veterinarius, No. 1, 906

India: 1931, The Census of, 109; Age of Retirement in, 146; Mechanical Transport in, Col. F. P. Barnes, 578; Mother Goddess in, Kalipada Mitra, 207; South: Prehistoric Gold-Mining in, L. Munn, 730; -West Monsoon in, 812; Wild Life in, Preservation of, 779

Indian: Association for the Cultivation of Science, Appointment of K. S. Krishnan as secretary and Mahendral Sircar research professor in physics, 907; Caste Customs, L. S. S. O'Malley (Review), 638; Federation, 427; Hot Weather, 701; Institute of Science, Bangalore, 607; Progress at the, 834; Microlepidoptera, Life-Histories of, T. B. Fletcher, 474; Philosophy, A History of, Dr. Surendranath Dasgupta. Vol. 1. Second edition. Vol. 2 (Review), 855; Outlines of, M. Hiriyanna (Review), 855; Sage,

An, Prof. H. E. Armstrong (Review), 672; Sleeper Woods, Identification of, K. A. Chowdhury, 30; Thought, The Development of, Dr. T. Greenwood (Review), 855; Tribe, The Changing Culture of an, Dr. Margaret Mead (Review), 711; Woods, Calorific Values of, S. Krishna and S. Ramaswami, 281

Indices of a Stable Population, Prof. R. Firth, 519 Industrial: Administration, Institute of, 395; Aluminas, Velocity of Dissolution of, in Fused Cryolite, N. Parravano and O. D'Agostino, 107; Management, Specialist Posts in, Dr. W. H. Coates, 268; Psychology in Practice, H. J. Welch and Dr. G. H. Miles

(Review), 567; Research, Co-operative, 817

l'Industrie Nationale, Société d'Encouragement pour, award of the medal of the, to P. Chevenard, 867 Industry: The Arts in, 73; The Management Factor in, 52 Infestation of Stored Products by Insects, Prof. J. W.

Munro, 82 Influenza: March of, 22; Spread of, 128

Infusoria, Ciliated, Axial Gradient in the, J. Monod, 339 Insect-Eating in Siam, W. S. Bristowe, 234

Insects: and Micro-Climates: Dr. H. H. Darby, 839; H. S. Leeson and K. Mellanby, 363; Inhabiting the Soil Surface, Dr. J. Davidson, 837

Institut Colonial International: Great Britain and the Conferences of the, Prof. B. Williams, 683; Lord Lugard elected president of the, 683

Integral Radiator (Black Body), of Electrically Heated Carbon, T. N. Panay, 179 Intelligence in Man, Inheritance of, A. F. Dufton, 763;

Dr. C. C. Hurst, 764 Intensities, Relative, Direct Recording of, by means of a Microphotometer, N. Thomson, 558

Interaction between Different Species of Animals, The Quantitative Theory of, V. A. Bailey, 524

Interatomic Distances and Ferromagnetism, E. C. Stoner, 814

Internal Ballistics, The Combustion Problem of, Dr. A. D.

Crow and W. E. Grimshaw, 805

International: Auxiliary Languages, Dr. E. Ritter, 102; Congress: for Applied Mechanics, The Fourth, 907; of Scientific and Applied Photography, 212; Ornithological Congress, The Eighth, 902; Relations: The Background of, Our World Horizons, National and International, Prof. C. Hodges (Review), 45 Invar, Dr. Ch.-Ed. Guillaume, 658

Invertebrates (Review), 76 Invertebrata: The, a Manual for the Use of Students, L. A. Borradaile and F. A. Potts. With chapters by Prof. L. E. S. Eastham and J. T. Saunders (Review), 76

Iodine: Arc Spectrum of, Prof. W. E. Curtis, 398; Atomic Weight of, Guichard, 738; -Oxalate Reaction, Kinetics of the, Prof. N. R. Dhar, A. K. Bhattacharya and B. L. Mukerji, 840; Tincture of, and Asepsis, R. Sorel, 596; Vapour at High Temperatures, Absorption Bands of, E. Skorko, 366

Iodoacetic Acid, Glutathione and: Glucose Metabolism, Dr. J. H. Quastel, 206; and Tissue Glyoxalase, Dr. F.

Dickens, 130

Iodo-Silver-Benzoic Complex, An, and its Application to the Oxidation of Ethylene Compounds into α-Glycols, C. Prévost, 774

Ionisation Produced by Radioactive Sources, Amplifica-tion of the, J. A. C. Teegan, 277

Ionised Gases in the Magnetic Field, T. V. Ionescu and C. Mihul, 106

Ionization in Cable Dielectrics, P. Dunsheath, 404 Ionosphere: Characteristics of the, J. P. Schafer and

W. M. Goodall, 804; Fine-Structure of the, Prof. E. V. Appleton, 872; J. A. Ratcliffe and E. L. C. White, 873; Wireless Investigations of the, An Automatic Recording Method for, C. T. R. Wilson, J. A. Ratcliffe and E. L. C. White, 522 Ions: Gaseous, Prof. A. M. Tyndall, 65; Monovalent,

Limiting Mobilities of Some, and the Dissociation Constant of Acetic Acid at 25°, Dr. A. I. Vogel and G. H. Jeffery, 27; Multiply-Charged Large, J. J. Nolan and J. G. O'Keeffe, 106

Iraq: Northern, Expedition to, 89; Petroleum and Pipe-Line, 864

Ireland, Anthropological Survey of, 160 Iris, Physiology of the, (1), J. Z. Young, 178

Irish: Flora and Fauna, Origin of the: Dr. R. Lloyd Praeger, 279; 579; Plants, Some Noteworthy, Dr. R. Lloyd Praeger, 172; Woods and their Significance, A. C. Forbes, 246

Iron and Steel: Industry, The, Sir William Larke, 335; Institute, K. Headlam-Morley appointed secretary of the, 272; Cast, Analysis of, 845; -working in the Bahr el Gharzal, T. C. Crawhall, 474

Isotopes: of the Radio Elements, 440; Zero-Point Energy and the Separation of, H. Eyring, 739 Israel: from the Beginnings to the Middle of the Eighth

Century, Prof. A. Lods. Translated by Prof. S. H. Hooke (*Review*), 315

Isthmian Links, Prof. B. Willis, 696

January freeze-the-pot-by-the-fire, 34 Japan Sea Floor, Lowering of the, Prof. H. Tsuya, 368

Japanese: Journal of Engineering, Vol. 8, 543; Physiological Standards and Occupational Characteristics of Bodily Functions of the, Tomoyosi Isikawa, 588; Power Station, A Modern, 127

Jehol: City of Emperors, Sir Sven Hedin. Translated

by E. G. Nash (Review), 184

Jews in Germany, 612 Jupiter, Seasonal Changes on, A. S. Williams, 808

Kaiser Wilhelm Gesellschaft, Work of the, 870 Kamet Conquered, F. S. Smythe (Review), 224 Kangaroo Mouse of Western America, 737 Karyology and Systematic Relationship, 807 Katabatic Winds, S. Atmanathan, 696

Kattegat, Boundary Tides in the, H. Pettersson and B. Kullenberg, 586

Kazakstan, Semipalatin District of, Rodents of the, B. A. Kuznetzov, 516

Kea Parrot, The, 463

Kennelly-Heaviside Layer: Prof. E. V. Appleton and R. Naismith, 808; Polarisation of Echoes from the, T. L. Eckersley, 512 Kenya: Gold in, and Native Reserves, 37; 123; 231;

Miners' 'Claims' and Native Lands in, 51

Kerr Cell, The, E. E. Wright, 702

Ketene, Sudden Pyrogenation of, J. A. Muller and Mlle. Eglantine Peytral, 375

Kidston Collection of Fossil Plant Slides (2), Mary G. Calder, 922

Kitchen-Middens at Gordon's Bay, H. A. Shapiro, 411 Knowledge: and Action, 485; and Existence, The Theory of, Dr. W. T. Stace (Review), 455 Kosmos, Prof. W. de Sitter (Review), 487

Kuala Selinsing, Perak, Excavations at, I. H. N. Evans, 135

Kudaru Hills, Nigeria, Younger Intrusive Rocks of the, A. D. N. Bain, 214

K'unlun und Karakorum-Himalaya, Geologische Forschungen im westlichen, Dr. H. de Terra and others (Review), 600

LaAl4, Crystalline Structure of the Compound, A. Rossi,

Lactic Acid, Emetic Derivatives of, Volmar and Betz, 410 Lambing Season in Great Britain, 557

Lamellibranchs, Artificial Ridges of the Valves of (Northern Annam), Mlle. M. Colani, 523

Lancaster Frankland Chemical Society, Inauguration of the, 196; Prof. H. E. Armstrong elected president, 197

Land Utilisation: Maps, 198; Survey, Second Annual Report of the, 396

Landolt's Reaction, Retarding Action of Glass on, A. Juliard, 338

Language: Difficulty and Scientific Progress, 197; in Civilization, The Spirit of, K. Vossler. Translated by O. Oeser (Review), 152; in the Service of Science, 741 Landslide at Toge, Japan, Prof. N. Nasu and Prof. N.

Miyabe, 99

Late-Glacial Re-advance Moraines of the Highland Border West of the River Tay, J. B. Simpson, 522

Latent Image, Evolution of the, C. Jausseran, 410 Latitude, Variation: Effect in, Correlatable with the Moon, Prof. H. T. Stetson, 437; of, Dr. J. C. Dobbie, 769 Lattice Distortion, Selective, in Wires under Torsion,

W. A. Wood, 842 Lead: Effects of the Addition of Tellurium to, W. Singleton and Brindley Jones, 696; from Cyrtolite, Baxter and Alter, 881; Volumetric Determination of, A. Travers and Lu, 523

Least Squares, Method of, Sir Arthur Eddington, 374 Leather, Examination of, for the Presence of Extractable Chromium Compounds, F. E. Humphreys and H. Phillips, 813

Leben und Erkennen: Vorarbeiten zu einer biologischen

Philosophie, Prof. G. Wolff (Review), 529 Leeches, Land, Prof. J. P. Moore, 63 Leeds University: A. V. Williamson elected reader in geography, 372; gifts by C. R. Brotherton and B. Scattergood; renewal of subscription of the West Yorkshire Coal Owners' Association, 665; Report for 1931–32, 700; Dr. W. MacAdam elected professor of clinical medicine; Dr. R. H. Evans appointed lecturer in civil engineering; gift from Mrs. Graham, 772; Pathology and Bacteriology at, 835; Institu-

tion of a diploma course in biology in, 920 Legendre, Centenary of the death of, 18 Leguminous Seeds, Nutrition with, and Reproduction, V. Zagami, 108

Lepidoptera, Nomenclature in (Review), 566

Lepidosiren: Pelvic: Fins of the, G. E. H. Foxon, 732; Filaments of, J. T. Cunningham and D. M. Reid; G. E. H. Foxon, 913; The Pituitary in, and its Development, T. Kerr, 523

Leverhulme Research Fellowships, Appointment of an

Advisory Committee, 795

Leyden Observatory, Report for 1932, 553

Libyan Desert, Volcanic Craters in the, Dr. K. S. Sandford,

Liesegang Rings: Dr. E. S. Hedges, 169; and other Periodic Structures, Dr. E. S. Hedges (Review), 316 Life and Thought: My, an Autobiography, A. Schweitzer. Translated by C. T. Campion (*Review*), 785

Light: and Life, Prof. N. Bohr, 421; 457; for Interferometer Work, Sources of, Rolt and Barrell, 244; in Interstellar Space, Absorption of, J. Dufay and Ssu-Pin Liau, 923; Measurement of, for Biological Purposes, Dr. W. H. Pearsall and P. Ullyott, 694;

-thrposes, Dr. W. H. Fearsan and F. Chyott, 694;
-Sensitive Apparatus, Practical Applications of,
Major C. E. Prince, 430; -Spots, Faint Transient,
Photography of, Dr. L. F. Richardson, 401
Lightning: Flash, A Destructive, Dr. C. V. Boys, 765;
Investigation in South Africa, W. H. F. Tredre, 19; Progressive, A New Stereoscope, Dr. C. V. Boys, 492;

Protection Against, Code for, 509

Lilydale Limestone, Stromatoporoids of the, Elizabeth A. Ripper (1), 106 Limestone Oil Wells, Reopening, with Acid, 541

Limnologie, regionalen, Grundzüge der, Prof. E. Naumann

(Review), 223 Limpets, Age and Growth of, N. Abe, 30

Linear Partial Differential Equations in the Plane, M. Brillouin, 410

Linen Research Association, Report of the, 614

Lines of Varieties, Curvature of, T. Boggio, 71 Linnean Society of London: election of Dr. E. D. Merrill, Dr. E. Baur, Prof. A. Pascher and Prof. F. Silvestri as foreign members, 759; election of officers, 798; Linnean medal presented to Prof. R. Chodat, 793

Liquid Surfaces, Instability of, Dr. L. Rosenhead, 175 Liquids: on Solid Surfaces, Spreading of, Buckley and Snyder, 407; with Gases, Supersaturation of, T. N. Richardson and Dr. K. C. Bailey, 762; Lister Institute, Report of the, 870 Lithium Alloys, A. Baroni (1), 71

Liver and Liver Extracts, Iron and Copper in, H. G. Rees, 702

Liverpool University: bequest to, by Dr. J. M. Hunt, 51; Prof. J. Proudman elected professor of oceanography in, 391; gift to, by W. Horton, 557
Living Machinery, Prof. A. V. Hill (Review), 860
Local Lists of Animals, Prof. A. E. Boycott, 94

Locusts: in South Africa, The Phases of, Prof. J. C. Faure, 423; South African, Phases in, B. P. Uvarov,

Logarithmetica Britannica: being a Standard Table of Logarithms to Twenty Decimal Places, Dr. A. J. Thompson. Part 5: Numbers 50,000 to 60,000 (Review), 785

Lolium Seedlings, Fluorescence of, in Ultra-Violet Light,

P. A. Linehan and Prof. S. P. Mercer, 202 London: Hospital, Collected Researches, 1932, 835; School of Hygiene and Tropical Medicine: Annual Report, 177; Hand List of Periodicals. edition, C. Barnard, 906; Subsidence of, T. E. Longfield, 558; University: Grants to, by City Companies, 104; Prof. L. N. G. Filon elected vicechancellor; title of reader in experimental physiology conferred on H. P. Gilding; F. A. Greene appointed a fellow of King's College, 177; grants from the Leathersellers' Company and the Pilgrim Trust, 248; Prof. J. B. S. Haldane appointed professor of genetics at University College; W. P. Yetts appointed professor of Chinese art and archæology at the Courtauld Institute of Art; Prof. C. L. Burt appointed Heath Clark lecturer for 1933, 336; conferment of the title of reader in experimental physiology on H. P. Gilding; conferment of doctorates, 372; bequest by A. L. Leon; J. Stinton Jones appointed consultant engineer for heating etc. for the new buildings in Bloomsbury, 408; Prof. A. J. S. Pippard appointed professor of civil engineering at Imperial College— City and Guilds College; P. M. S. Blackett appointed professor of physics at Birkbeck College; W. R. Spurrell appointed reader in physiology at Guy's Hospital Medical School; Dr. Vera Anstey appointed Cassel lecturer in commerce at the London School of Economics, 480; conferment of doctorates, 628; grants by City Companies, 700; Report for 1932-33, 736; and the British Museum, 755; C. A. Mace appointed reader in psychology at Bedford College; title of reader conferred on Dr. Katharine H. Coward, Dr. D. C. Jones, and B. Topley; award of Dunn exhibitions to J. C. B. Bone and J. S. Horn, 772; title of professor conferred on Dr. L. S. Stebbing, 811; grants from City Companies, 884; and the Development of Science, 896; The New Buildings for the, T. Ll. Humberstone, 903; award of doctorates, 920

Lonicera, Presence of a Centrosomic Apparatus in Species of the Genus, P. A. Dangeard, 886

Loudness, Scales of, B. G. Churcher and A. J. King,

Low Temperature Research, Prof. Keesom, 768

Lubricants, Hydrogenated Motor, 643

Lubricating Oils, Volatility Range of, Measurement of, R. N. J. Saal and C. G. Verver, 661 Lucerne 'Flea', Ecology of the, D. S. Maclagan, 556

Lunar Periodicity in Reproduction, Prof. T. A. Stephen-

Lund Zone of the Astronomische Gesellschaft Catalogue, Proper Motions in the, W. Gyllenberg, 245

Lyons Atmosphere, Progressive Darkening of the, A. Allix, 179

Machine Age's Starvation Predicted, Prof. R. A. Gortner,

Macrolepidoptera of the World: The, a Systematic Description of the Hitherto Known Macrolepidoptera. Edited by Prof. A. Seitz. Suppt. to Vol. 1: The Palæarctic Butterflies (Review), 566

Madagascar: Social Organisation and Marriage in, R. Linton, 843; Zoologie der, G. Grandidier et G. Petit

(Review), 748

Maglemose in Britain, Age of, H. and M. E. Godwin, 551 Magnesium: Chlorite and Double Chlorites of Copper with Magnesium, Barium and Thallium, G. R. Levi and D. Ghiron, 775; Silicides, Thermal Decomposition of the, G. Gire, 923; Sublimation of, in a Vacuum and Casting in an Atmosphere of Argon, J. Herenguel and G. Chaudron, 179; Zinc and Aluminium Films, Structure of, Prof. G. I. Finch and A. G. Quarrell, 482

Magnetic: and Electric Double Refraction, Anomalous Dispersion in, M. Schwob, 630; Data and Mine Surveying, T. G. Bocking, 768; Disturbances, Recent, Rev. J. P. Rowland, 764; Double Refraction of Some Cerium Salts in Aqueous and Non-Aqueous Solutions, C. Haenny, 887; Measurements at Mogadiscio, M. Bossolasco, 560

Magneto-Rotation in Coloured Glass and Rock Salt, G.

Kürti, 411

Malabar House, The, Dr. M. D. Raghavan, 29

Man: and Medicine: an Introduction to Medical Knowledge, Dr. H. E. Sigerist. Translated by Margaret Galt Boise (*Review*), 894; and Metals: a History of Mining in Relation to the Development of Civilisation, Dr. T. A. Rickard. 2 vols. (Review), 743; Ancient and Modern, Dr. S. Zuckerman, 367; Constitution and Temperament in, K. H. R. Edwards, 832; The Inequality of, and other Essays, Prof. J. B. S. Haldane (*Review*), 529; The Races of, Differentiation and Dispersal of Man, Prof. R. B. Bean (Review), 712

Managerial and Administrative Problems, Study Group

on, 719

Manchester University: Prof. I. M. Heilbron appointed professor of organic chemistry; retirement of Dr. J. Prescott and J. Winterbottom, 772; Dr. M. Polanyi appointed professor of physical chemistry in, 902

Manihiki and Rakahanga, Cook Islands, Ethnology of, Te Rangi Hiroa (Dr. P. H. Buck), 588

Mankind, The Proper Study of, B. A. Howard (Review), 676

Manure, Distribution of, 445

Maps, Reproductions of Early Engraved, 2: English County Maps in the Collection of the Royal Geographical Society. With Introduction and Notes by E. Heawood (Review), 530

Marble, Solution of, in Acids, Velocity of, W. Jacek (3),

887

Marcapata Valley, Eastern Peru, Geology of the, J. A. Douglas, 70

'March Brown' of the Angler, The, 481

March Winds, 285

Archæology of the, Laura Maud Marianas Islands, Thompson, 623

Marine Fishes, Low Temperatures cause Mortality amongst, 373

Market Gardening, Recent Developments in, 395 Marmite, Medicinal and Dietetic Value of, 616

Marquesan Insects, I, 623

Marquis, The, and the Land Agent, Prof. G. N. Watson, 67 Marrubüne, New Method of Preparing, L. J. and F. Mercier, 143

Mars: Canals on, H. B. Brydon, 518; Planet, The Minor Details of, Dr. E. M. Antoniadi, 802; Recent Observations of, Dr. R. L. Waterfield, 845

Mathematical Association, Annual Meeting of the, 67 Matrices and Continued Fractions, Prof. H. W. Turnbull,

Maya and Aztec, Gods of, J. E. Thompson, 171 Mayas, Sacred Sandstone of the, Prof. T. D. A. Cockerell, 656

Mayflies and Man, 701 May Sunshine, 629

Maze, Symbolism of the, W. F. J. Knight, 98

Mechanics and Acoustics, Physical Principles of, Prof. Translated by Winifred M. Deans R. W. Pohl. (Review), 320

Medical: Biography (Review), 40; Research Council and the Rockefeller Foundation, award of travelling fellowships to H. W. Fullerton, Miss M. H. Roscoe, Dr. D. Sheehan and Dr. D. Zuckerman, 907

Medieval Faith and Fable, Rev. Canon J. A. MacCulloch

(Review), 80

Mediterranean Region: The Geography of the, Its Relation to Ancient History, Prof. Ellen Churchill Semple (Review), 317

Memory, Experiments on (Review), 309 Mercury, Visible Triplet Lines of, Band Spectra which Appear Near, E. Matuyama, 58

Mesolithic Age in Britain, 32

Metals in very Thin Plates, Reflecting Power of, P. Rouard, 36

Metamorphism: a Study of the Transformations of Rock Masses, Dr. A. Harker (Review), 310; in Rocks, the Process of, Prof. P. Niggli (Review), 310

Meteor Craters, Prof. F. A. Melton and W. Schriever,

100

Meteoric Craters, Dr. L. J. Spencer, 172

Methylnornarcotine, Glycuronic Acid, and Vitamin C, W. J. Dann, 24

Mexico and Guatemala, Recent Discoveries in, 101 Mechanik und Akustik, Einführung in die, Prof. R. W. Pohl. Zweite Auflage (Review), 568

Mechanical Transport in India, Col. F. P. Barnes, 578 Mechanik, Handbuch der physikalischen und technischen. Herausgegeben von Prof. F. Auerbach und Prof. W. Hort. Band 3: Statik und Dynamik elastischer Körper nebst Anwendungsgebieten, zum Gebrauch für Ingenieure, Physiker und Mathematiker. Band 4, Hälfte 1: Lief. 1. Lief. 2. Band 4, Hälfte 2: Technische Physik der festen Körper, zum Gebrauch für Ingenieure, Physiker und Mathematiker (Review), 748

Medical: Education, Methods and Problems of, Vol. 21, 441; Research Council: Sir Thomas Lewis appointed a member of the, 581; Report of the, for the year

1931–1932, 753; in Great Britain, 753 Medicine: Man and, an Introduction to Medical Knowledge, Dr. H. E. Sigerist. Translated by Margaret

Galt Boise (*Review*), 894 Melanogenesis, Biological Signification of, E. Friedheim,

Melanophore Reactions, Cellular Transmission of Neurohumoral Substances in, G. H. Parker, 776

Melchett Medal of the Institute of Fuel, award of the, to Sir John Cadman, 581

Membrane Equilibria (Review), 859

Mendel medal of Villanova College, award of the, to Prof. H. S. Taylor, 759

Mendelian Factors in Quantitative Inheritance, Number of, Dr. R. A. Fisher, 400

the Procedure for the Mental Deficiency Practice: Ascertainment and Disposal of the Mentally Defective, Dr. F. C. Shrubsall and Dr. A. C. Williams (Review),

Mercury: Hornsby's Observations of, 405; Arc Rectifiers, 54; Fulminate, Thermal Decomposition and Detonation of, Prof. W. E. Garner and H. R. Hailes, 286

Merget's Phenomenon, Production of, by d'Arsonvalisation with Short Waves, H. Bordier, 375

Mesopotamia, Northern, Excavations in, 685 Metallic: Neodymium Free from Iron and Silicon, Preparation of, F. Trombe, 595; Objects in Museums, Preservation of, Dr. A. Scott, 906

Metals: and Civilisation (Review), 743; Cold Working of, Energy absorbed in the, Rosenhain and Stott, 769; in the Service of Human Life and Industry, Sir Harold Carpenter, 733; Institute of, election of officers, 467; Journal of the, Vol. 49. Edited by G. Shaw Scott (Review), 187

Meteor Crater: Age of, Prof. E. Blackwelder, 404; Arizona, D. M. Barringer, 579

Meteoric Irons and Silica-Glass from the Meteorite Craters of Henbury (Central Australia) and Wabar (Arabia), Dr. L. J. Spencer, 250

Meteorite: Scars in Carolina, Probable, F. A. Melton and W. Schriever, 624; The Great Siberian, 614

Meteorological: Data, Correlation of, Sir Gilbert Walker, 284; Records of Southport, 199

Methane: Action of Steam on, C. Matignon and M. Séon, 215; Halogen Derivatives of, Infra-red Absorption Spectra of some, J. Lecomte, 738

Methionine in Wool, J. Barritt, 689

Methylene Blue, Reduction of, by an *Endomyces* at the expense of its Endocellular Hydrogen Donators, F. Chodat and M. Junquera, 815

Metrology by Light Waves at the National Physical

Laboratory, 192
Metropolitan-Vickers Electrical Co., Ltd.: Representatives in Russia, 428; the Research and Testing Departments of the, 126; 649

Mexican and Ecuadorian Copper and Bronze Axes, M. A.

Clement, 279

Mexico's "Houses of the People", 557 Mice under Continued Treatment with Œstrin, Some Effects Observed in, H. Burrows, Prof. E. C. Dodds,

and N. M. Kennaway, 801 Microscope: Evolution of the, R. S. Whipple (Review)

219; The History of the, compiled from original instruments and documents, up to the introduction of the Achromatic Microscope, Dr. R. S. Clay and T. H. Court (Review), 219

Middlesex in British, Roman and Saxon Times, Sir Montague Sharpe. Second edition (*Review*), 677 Migration from and to Great Britain, H. Leak and T.

Priday, 125

Milk: the Colloidal Calcium Phosphate of, G. T. Pyne and J. J. Ryan, 35; the Freezing-Point of, G. W. Monier-Williams, 702; Report of the Reorganisation Committee for (Ministry of Agriculture and Fisheries), 605; Supply, Agriculture and, Prof. H. E. Armstrong, 605

Milne's Theory of the Expansion of the Universe, Dr. G. C. McVittie, 533

Mind-Body Relation, Absurdity of any, Dr. C. S. Myers, 579

Mindouli-Mines Region, Stratigraphical Study of the, H. Lagotala, 144

Mineral Oils, Dewaxing and Acid Refining, Dr. N. O.

Backlund, 31 Mineralogy: A Textbook of, with an extended treatise on Crystallography and Physical Mineralogy, Prof. E. S. Fourth edition, revised and enlarged by

Prof. W. E. Ford (Review), 318 Miner's Nystagmus, Third Report on, 394

Mines: Roof Falls in, Prevention of, 140; Underground Lighting in, R. H. Campin, 465

Mining: and Metallurgy, Institution of, award of the gold medal to Sir John Cadman, 432; Institute of Scotland, C. A. Carlow elected president of the, 650

Minnesota Pleistocene Homo, A. E. Jenks, 739

Mistral, Greatest Strength of the, 338

Mites of Fruit Trees, A. M. Massee, 136 Mitochondria, Fixation of, J. H. Davie, 59

Mitogenetische Strahling, Die, zugleich zweiter Band der "Probleme der Zellteilung", Prof. A. Gurwitsch. Unter Mitwirkung von Lydia Gurwitsch (Review),

Mixtures of Substances of Large Electric Moment, Magnetic Susceptibility of some, J. E. Garssen, 523

Molecules: of Intermediate Complexity, Kinetics of the Decomposition of, C. N. Hinshelwood and C. J. M. Fletcher, 24; Simple, and Elementary Processes, Prof. A. J. Allmand, 173

Mole's Winter Store of Earthworms, 285

Mollusca, The British Marine, R. Winckworth, 334 Molybdenum, Distribution of, Dr. W. A. Roach, 202

Monte Rosso di Verra, Mineral Deposits of, T. Carpanese, 775; (2), 924

Moon: New, An Early Observation of the, 100: Partial Eclipse of the, Sept. 14, 1932, Photometric Study of the, F. Link, 179

Moon's Shadow on the Earth, Photographing the, W. M. Browne, 697

Mornington Earth Tremor of Sept. 3, 1932, W. M. Holmes,

Motor Car Lights on the Road, Dr. Dickinson, 20

Mousterian Age, A Spear-Point of Bone of, Abbé Breuil, Dr. H. Martin, 53

Mole, Captive, Observations on a, L. E. Adams, 466 Mollusca, Spawning in, Spring Outburst of, 445

Morphology, The Old, and the New, Dr. H. Hamshaw Thomas, 47

Mother-Tongue in India, Use of the, 444

Mottramite and Psittacimite, Identity of, with Cupriferous Descloizite (Cuprod-Escloizite), F. A. Bannister, with chemical analyses by M. H. Hey, 250

Mucor, Biometric Researches on the Spores of, W.

Schopfer, 483

Murray, John, Expedition, The, Prof. J. Stanley Gardiner, 640

Muscular Work, Fatigue and Recovery, G. P. Crowden

(Review), 321 Museums: of British Territory and the Mediterranean, 430: of Canada, 84

Musk: -Rat in Britain, The, Prof. J. Ritchie, 385; -Rats at the London Zoo, 755

Mussel, Strange Spatfall of the Common, on the Common Cockle, Prof. J. H. Orton, 513

Mutations, Spontaneous, Origin of, M. Navashin, 436 Mutsu Bay, Tintinnids of, 695

Mycorrhiza on Conifer Roots, 176

Mycorrhizal Theory, Objections to the, J. Costantin, 447

Nationalism and Academic Freedom, 853

National: Physical Laboratory, Collated Researches of the. Vol. 23, 162; New Radio Department of the, R. A. Watson Watt appointed superintendent,

Native Reserves, Gold in Kenya and, 37

Natural History: Intensive, Prof. E. J. Salisbury (Review). 343; The Standard, from Ameeba to Man, G. J. Arrow and others. Edited by W. P. Pycraft (Review),

Naturalist, resignation of T. Sheppard of editorship of the, 304; The, and the Country Side (Review), 636

Naturwissenschaften, Die (birthday of Dr. Arnold Berliner), 127

Natuna Islands, Birds of the, H. C. Oberholser, 207

Naturæ, Vox, E. Terbea (Review), 491 Naval Architects, Institution of: award of premium to Dr. G. Kempf and H. Lerbs, 164; Lord Stonehaven

nominated as president of the, 432 Navy, Health of the, 1931, Vice-Adm. R. St. G. S. Bond,

906 Nebulosity Surrounding a Nova, Illumination of, B. M.

Peck, 808 Nematodes, Parasitic, Cultivation of, G. Lapage, 583

Neodymium-Samarium Fractionation, Radioactivity of a, M. Curie and S. Takvorian, 702

Neolithic Fortress, Late, Homolka, Bohemia, V. J. Fewkes, 207

Neon Discharge Tubes, Behaviour of, in a Flashing Capacity Circuit by Means of a Cathode Ray Oscillograph, J. H. J. Poole, 35

Neoteny and Pædogenesis, Meaning of, G. E. H. Foxon, 93 Nerve: Action of Quaternary Ammonium Salts on, S. L. Cowan, 658; Impulse, Nature of the, Prof. A. V. Hill, 233; 497; 501; Mitogenetic Radiation of, Dr. A. Gurwitsch, 912

Nesting Season, Opening of the, 445

Nettle, Stinging, Eradication of the, G. H. Bates, 208 Neural Energy Constant: The, A Study of the Bases of Consciousness, Dr. J. Bostock (Review), 80

Neuro: Humoral Mechanisms, Prof. R. J. Brocklehurst, 65; -Muscular Junction, the, and Curare, Tudor Jones, 693

Neuton, The Neutron and, The New Element of Atomic Number Zero, Prof. W. D. Harkins, 23 Neutral Aryl Sulphites, Reaction of Phosphorus Penta-

chloride on the, P. Carré and D. Libermann, 667

Neutron: and Neuton, The, The New Element of Atomic Number Zero, Prof. W. D. Harkins, 23; and Proton, Law of Force between, E. C. Pollard, 814; Chemical Nature of the, P. Achalme, 559; Proton, and Positron, Dr. N. Thon, 878; The, Dr. J. Chadwick (Bakerian lecture), 794

Neutrons: Dr. D. Meksyn, 366; Conditions of Emission of, by the Action of α-Particles on the Light Elements, Mme. Irène Curie and F. Joliot, 447; Emission of, by Aluminium under the Action of the α-Particles,

P. Auger and G. M. Herzen, 523

New Britain, 358

Newfoundland: Fishery Research in, 918; Ice off,

409

New South Wales: Association of the Tertiary Alkaline Rocks of, with late Tertiary Tectonic Lines, C. A. Sussmilch, 667; Linnean Society of, Prof. A. N. St. G. H. Burkitt elected president of the, 759; Mineralogy of the Narrabeen Series of, Alma G. Culey, 523

New Year Honours, 17

New Zealand, Trout Fishing in, Prof. E. Percival, 163 Niari: Limestones of, Continental Formation Subjacent

to the, H. Lagotala, 851; Stratigraphic Scale of the,

H. Lagotala, 215

Nickel: and Nickel-Chromium Alloys in the Neighbourhood of their Curie Points, Measurements of the Thermoelectric Powers of, A. W. Foster, 814; Salts, Electrolysis of, Influence of the Magnetic Field on the, H. Forestier, 70

Nicotine, the Toxic Action and Elimination of, N.

Sabatucci (1-2), 483

Niger, Barrages on the, 466

Nigeria, Census of, S. M. Jacob, 516

Nile Flood, 921

Niobium, Determination of, by Orthoxyquinoline, P. Süe, 738

Nitrates in the Soil, Distribution of, and Root Development in Coffee, V. A. Beckley and F. McNaughtan, 878

Nitrobenzene, Debye's Dispersion of, Prof. R. Weigle and

R. Luthi, 327

Nitrogen: Dioxide, Infra-Red Absorption Spectrum of, C. R. Bailey and A. B. D. Cassie, 239; Dioxide Molecule, Form and Vibrational Frequencies of the, C. R. Bailey and A. B. D. Cassie; R. Schaffert, 910; Fixation in the Genus Lolium, R. Brown, 169; Organic, Estimation of, in the presence of Nitrates by Kjeldahl's Method, R. Cambier and L. Leroux, 179; -Uptake of Plants, Prof. A. I. Virtanen, 534

Nitrogenous Manuring of Legumes, H. Burgevin, 660 Nitrous Oxide: and Carbon Dioxide, A Bomb Calorimeter Determination of the Heats of Formation of, R. W. Fenning and F. T. Cotton, 446; in the Glow Discharge, Dissociation of, E. A. Stewardson, 364

NO₂: Extension of the Visible Absorption System of, to longer Wave-lengths, J. Curry and Dr. G. Herzberg, 842; Molecule, Form and Vibrational Frequencies of the, Dr. L. Harris, W. S. Benedict and G. W. King, 621

Nomogram: The, The Theory and Practical Construction of Computation Charts, H. J. Allcock and J. R.

Jones (Review), 785

Non-Linear Equations of the Elliptic Type, G. Giraud,

North: Atlantic Gale, 18; China Amphibia and Reptiles, Handbook of, Dr. Alice M. Boring, C. C. Liu, and S. C. Chou, 304

Northern Lights: The Official Account of the British Arctic Air-Route Expedition, 1930–1931, F. S. Chapman, and others (*Review*), 317

Norton, Thomas, and the "Ordinall of Alchimy", Dr.

E. J. Holmyard, 520

Norwegian: Antarctic Expedition, Capt. H. Riiser-Larsen, 301; Oyster Pools, T. Gaarder and R. Spärck, 806

Nova: A Remarkable Short-Lived, Delporte and Arend, 553; of March 20, Dr. Kukarkin, 590; Phenomenon, Frequency of the, C. Lönnqvist, 64

Nuclear: Potential Barriers, Heights of, E. C. Pollard, 97; α-Particles, Fundamental State of, Dr. G. Gamow, 618; Energy Levels, Dr. G. Gamow, 433; Moments, Predictions of, Prof. B. Venkatesachar and T. S. Subbaraya, 552; Potential Barriers, Heights of, and Nuclear Structure, E. C. Pollard, 398

Nucleic Acids and Uracil, Irradiation of, F. F. Heyroth

and J. R. Loofbourow, 92

Numerical Coincidence, A, H. P. Hollis, 550 Nutrient Salts in the Sea, Depletion of, 445 Nutrition, Recent Investigations in, 357

Oaks in Cultivation in the British Isles, Sir Oscar and E. F. Warburg, 905

Oats, Grain Colour in, Inheritance of, W. Robb, 517 Observer's Handbook for 1933, 369

Obstacles Revolving Round an Axis Parallel to the General Direction of Flow, Vortices Produced by, P. Dupin and M. Teissié, 179

Oceanographical Work of the Carnegie (Review), 114 Oceanography: (Bulletin of the National Research Council. No. 85: Physics of the Earth, 5) (Review),

344; Progress of, 214

Octopus vulgaris, Hæmocyanin of, Svedberg and Erikson, 137 Enothera, Phylogeny in the Genus, Prof. R. R. Gates, 589

Œstrogenic Substance from Plant Material, An, Dr. B. Skarżyński, 766

Estrus-exciting Compound, A Synthetic, Dr. J. W. Cook,

Prof. E. C. Dodds, and C. L. Hewett, 56
Oil: and Natural Gas, Migration of, Prof. V. C. Illing, 475;
and Water, Separation of, Molecular Phenomena at the Surface of, J. J. Trillat and L. Leprince-Ringuet, 850;
Oxidation and the Lubricating Properties of, R. O. King, 476;
Reserves and Production, V. R. Garfias, 868;
Seeds and Vegetable Oils, Survey of, Vol. 2, 542

Oldoway Human Skeleton, The, Dr. L. S. B. Leakey, Prof. H. Reck, Prof. P. G. H. Boswell, A. T. Hop-

wood, and Dr. J. D. Solomon, 397

Old: Trades and New Knowledge, Sir William Bragg (Review), 860; Wine and New Bottles, Dr. A. Ferguson (Review), 417

Olefinic Compounds, Catalytic Hydrogenation of, Dr. E. H. Farmer and R. A. E. Galley, 60

Olive, A Non-Parasitic Alteration of the, R. Gigante, 851 Omaha Secret Societies, Dr. R. F. Fortune, 331

Opiliones-Laniatores, The Arachnid Group, late Dr. W. Sørensen, revised by Dr. C. With and Dr. K. L. Henriksen, 543

Ophiocytium Nageli in Victoria, Two Species of, O. terrestris n.s. and O. arbuscula Rabenhorst, Jean Heyward, 107

Opium Poppy, Atavism in a Strain of the, L. Blaringhem, 178

Optic Tectum, A Phylogenetic Consideration of the, G. K. Huber and Elizabeth C. Crosby, 739

Optical: Rotatory Power, Prof. T. M. Lowry and H. Hudson (4), 374; Rotatory Power, Fundamental Laws of, Prof. W. Kuhn, 771

Ordovician Rocks of the Trefriw District, North Wales,

D. A. Bryn Davies, 595

Organic: Reagents for Metals, Hopkin and Williams, Ltd., 396; Oxides, Dissociable, C. Dufraisse and J. A. Monier, Jr., 887; Syntheses: an Annual Publication of Satisfactory Methods for the Preparation of Organic Chemicals. F. C. Whitmore, editor-in-chief. Vol. 12 (Review), 604

Orthogonal Polynomials with Two Variables, M. Ghermanesco, 107

Orthophosphoric Acid, Action of Solutions of, on Ordinary

Cellulose, G. Champetier, 703
Orthopodomyia pulchripalpis, Rondani (Diptera, Culicidæ),
A New British Record of, J. F. Marshall and J. Staley,
435

Osiris and the Atom, J. G. Crowther (Review), 320 Osler and other Papers, Prof. W. S. Thayer (Review), 40 Outline Series, The. 11 vols. (Review), 8 Ovaries, Transplantation of, Preserved Outside the Organism, A. Lipschütz, 143 Ovulation Without 'Heat' in the Ewe, Occurrence of,

R. Grant, 802

Oxford University: Dr. G. D. Hale Carpenter appointed Hope professor of zoology, 159; Junior Scientific Club, Jubilee of the, 757; Old Ashmolean, 250th Anniversary of the, 684; Report of the Lewis Evans Collection, 884

Oxide Catalysts, Positive Ion Emission from, Dr. C. F.

Powell and Luang Brata, 168

Oxides, Dissociable Organic, C. Dufraisse and R. Buret,

Oximes and Semicarbazones, Colour and Structure of, Mme. Ramart-Lucas and Mme. M. Grumez, 70 Oxyflavonic Compounds, Microchemical Characters of the,

A. Guilliermond and R. Gautheret, 410

Oxygen: Commercial, Manufacture of, G. Claude, 70; Hydrogen and, Reaction between, C. N. Hinshelwood, E. A. Moelwyn-Hughes and S. C. Rolfe, 625; Liquid. Magnetic Double Refraction of, P. Lainé, 850; Molecule, Magnetic Behaviour of the, R. Einaudi, 71

Oxyquinoline, Azo Derivatives of, Utilisation of Some, as Reagents in Qualitative Analysis, G. Gutzeit and

R. Monnier, 144

Oysters (O. edulis), Breeding of, at Port Erin, Prof. J. H. Orton, Miss M. W. Parke and W. C. Smith, 26

Ozone: Liquid, Magnetic Properties of, P. Lainé, 702; Photosensitised Decomposition of, by Chlorine, Dr. R. G. W. Norrish and G. H. J. Neville, 544

Pacific: Influence of the, on the Circulation in the South-West Atlantic Ocean, A. J. Clowes, 189; Islands, Reptiles and Amphibians of the, C. E. and May Danheim Burt, 767; Ocean, Soviet Expedition to the, 391; Science Congress, Fifth, 757; Scientific Problems of the, Lord Rutherford, 831

Pædogenesis, Neoteny and, Meaning of, G. E. H. Foxon, 93

Paint and Varnish Industries, Research in the, 794

Painted Lady Migrates, The, 849

Palæontographical Society, election of officers, 649 Palæozoic: Fossils from Victoria, F. Chapman, 703; Planktonic Faunas of North America, R. Ruedemann,

Palestine: Ancient Man in, Miss Dorothy Garrod, 19; M. T. Dawe appointed director of agriculture and forests, 543; Mesolithic Culture in, 767

Pamir, A Photogrammetric Survey in the, P. Lake (Review), 744

Panda, The, at the Zoological Gardens, 428

Papal Observatory in Castel Gandolfo, 124 Papua, Depopulation in, Sir Hubert Murray, and others, 519

Paraffins, Normal, Crystal Structure of the, Dr. A. Müller, 100

Paramagnetic Susceptibility, Influence of Light on, P. W. Selwood, 761

Parc National Suisse, Les Insectes Forestiers du, Dr. A. Barbey, 234

Paris Academy of Sciences, Prize awards of the, 174 Parsons: Charles, his Life and Work, R. Appleyard (Review), 891

Particle Size and X-Ray Spectroscopy, Fonda, 332
Partridge Disease, 177; and its Causes. Edited by Major
M. Portal and Dr. W. E. Collinge (Review), 224;

Numbers, Fluctuations of, 178; Shooting, 142 Passivity Phenomena, Theory of, W. J. Müller (18), 888

Pasteur, Louis, The Genius of, P. Compton (Review), 40 Pastures, Nutritive Value of, E. J. Sheehy, 30

Patents: British, and Designs Statutes as Amended and Consolidated to 1932. With an Introduction and Index by H. J. W. Bliss (Review), 79; for Chemists, The Law of, Dr. J. Rossman (Review), 115; Trade Marks and Designs: their Commercial Aspect and Development, C. W. Thomas (Review), 79

Patwin and Maidu Cult Origins, E. M. Loeb, 659

Pearl Oysters, Breeding of, coincident with Full Moon,

Pedigree Schedules: the Study and Preparation of Family Records (Review), 712

Peirce, Charles Sanders, Collected Papers of. Edited by C. Hartshorne and P. Weiss. 2 vols. (*Review*), 639 Pelagic Organisms, Unusual Occurrence of, Prof. W. J. Dakin, 239; G. P. Farran, 240

Penetrating Radiation, Photography of the Tracks of,

P. M. S. Blackett and G. Occhialini, 286

Penguin Embryos, C. W. Parsons, 695

Pepys, Samuel: Christ's Hospital and, 267; Tercentenary: of, T. E. James, 228; and the Royal Society, 299; E. Chappell, 833; The Priestley and, Commemorations, 443
Periodic: Functions, Almost, A. S. Besicovitch (Review),

384; Light Beam, Action of a, on Metallic Sheets,

Q. Majorana, 447

Permanganate, Action of Hydrogen Peroxide on, P. Dubois, 923

Persepolis, Excavations of, 233 Persia, "Wind of 120 Days", 773

Perturbed Series, Hyperfine Structure of, E. Gwynne . Jones, 813

Peterhead Sealers and Whalers, Dr. R. W. Gray, 904 Petrography and Petrology: a Textbook, Prof. F. F. Grout (Review), 317

Pétrole, Le, de Mésopotamie et son Pipe-Line, V. Forbin,

Pétroles naturels et artificiels, J.-J. Chartrou (Review), 455 Petroleum: Congress, World, 758; Herrings and the Origin of, 578; Iraq, and Pipe-Line, 864; Production in Russia, P. C. Beckstrom, 541; Technology: and Chemical Industry, 145; Aircraft in Relation to, 160

Petrols, Vapours of, Accidental Ignition of, by Electric Sparks, A. Kling and A. Soulier, 106

Pflanzenanalyse, Handbuch der, Herausgegeben von G. Klein. Band 2: Spezielle Analyse. Teil 1: Anorganische Stoffe; Organische Stoffe, I (Review), 8 Phæocystis and Herrings, 737

Pharmacognosy, A Textbook of, J. W. Cooper and T. C. Denston (*Review*), 419

Pharmacopœia: The British, 1932 (Review), 6; The Extra, of Martindale and Westcott. Revised by Dr. W. H. Martindale. Twentieth edition. In 2 vols. Vol. 1 (Review), 6

Pharmacy, Principles of, H. B. Mackie (Review), 895 Phase, Variations of, by Reflection on very Thin Metallic Films, P. Rouard, 410

Phaseolus multiflorus, Petaloid Formation of the Calyx in, E. Tschermak, 560

Phenols and Naphthols, Action of Gaseous Cyanogen on, (2), G. Machek, 851; (3), 852

Phenomenal Regression to the Real Object, Dr. R. H. Thouless, 261

Philips Sodium Lamp, 440

Philosophy: Biology and (Review), 529; of a Scientific Man, The, P. R. Heyl (Review), 491

Philyrea latifolia, The Heterosides of, Mlle. A. Kramer,

Phlogiston Theory, The History of the, Dr. J. H. White (Review), 531

Phosphorescent Zinc Sulphide, Connexion between the Two General Methods for the Preparation of, R. Coustal, 887

Phosphoric Anhydride, Modifications of, K. Boratynski

and A. Nowakowski, 595

Phosphorus: Liquid, Dielectric Constant of, S. Dobinski, 447; Pentachloride, Mechanism of the Reaction of, on Neutral Alkyl Sulphites, P. Carré and D. Libermann, 143; Ultra-Violet Bands of Oxide of, Prof. P. N. Ghosh and A. K. Sen Gupta, 841

12-Phosphotungstic Acid, Molecule of, Structure of the,

J. F. Keggin, 908

Photocells: and their Application, Dr. V. K. Zworykin and Dr. E. D. Wilson. Second edition (*Review*), 860; of the Dry Rectifier Type for the Measurement of Daylight, Suitability of, Drs. H. H. Poole and W. R. G. Atkins, 850

Photochemical Kinetics, R. Wegscheider, 287

Photoelectric: Experiment, A New, Q. Majorana, 71; 107; Phenomenon, A New, Exhibited by Metallic Sheets, Q. Majorana, 559; Polarimeter, Improve-

ment of the, G. Bruhat and A. Guinier, 630
Photographic: Densities, Exact Measurement of, P. Fleury and G. A. Boutry, 738; Sensibility, Influence of Pressure on the, to Various Monochromatic Radiations, Ny Tsi-Ze and Chien Ling-Chao, 286; Sensitisers for the Infra-Red, Miss N. I. Fisher and Miss F. M. Hamer, 475

Photography, Scientific and Applied, International Con-

gress of, 212

Photometry, Everyday, with Photoelectric Cells, Dr. J. W. T. Walsh, 660

Photovoltaic Elements, Differentiation of the Electronic Effects and the Photoelectric Effects in, R. Audubert, 482

Physic, Modern (Review), 6

Physical: Atomic Weights, Dr. F. W. Aston, 172; Bibliographies, 615; Society: election of officers, 510; presentation of the Duddell medal to Prof. W. Gaede, 901; Society's Exhibition, Lectures at the, 66; World in Cinematography, Image of the, Torahiko Terada, 358

Physics: Elementary, G. Stead. Fourth edition (Review), 748; in the Boot and Shoe Industry, H. Bradley, 756; Institute of: award of prizes to E. Lancaster-Jones, Dr. J. L. Miller and J. E. L. Robinson, 467; election of officers, 797; Intermediate, Dr. C. J. Smith (*Review*), 532; Numerical Examples in, Dr. W. N. Bond (Review), 320; Practical, W. R. Bower and Prof. J. Satterly. Third edition (Review), 491; Theoretical, A Concise (Review), 221; The Method of, Prof. A. Einstein (Herbert Spencer Lecture), 867

Physik: ein Lehrbuch für Studierende an den Universitäten und technischen Hochschulen, Prof. W. H. Westphal. Dritte Auflage (Review), 895; Kosmischen, Ergebnisse der. Herausgegeben von V. Conrad und L. Weickmann. Band 1 (Review), 785; theoretischen, Lehrbuch der, Prof. G. Joos (Review), 221

Physikalisch-chemisches Taschenbuch, Herausgegeben von C. Drucker und E. Proskauer. 2 Band (Review),

676

Phytophthora megasperma causing Pink Rot of the Potato,

H. Cairns and A. E. Muskett, 277

Phytosterol Group, Investigations in the, (2), C. Antoniani, 483

Phytosteryl Acetate Test as a Routine Method for Examining Butter Fats, H. Hawley, 814

Piedmontite in Quartz-Muscovite-Schist from the Shotover Valley, Western Otago, New Zealand, F. J. Turner, 630

Pilot Balloon Observations at Mauritius, R. A. Watson

and N. R. McCurdy, 626 Piltdown Skull, A Second, Sir Arthur Smith Woodward, 242

Pinastric Acid, Constitution of, G. Koller and G. Pfeiffer,

Pitchblende at Great Bear Lake, Canada, H. S. Spence, 208

Pit-Head Baths, J. H. Mitchell, 207

Pituitary Gland, Action of Proteolytic Enzymes on the Oxytocic Principle of the, Dr. J. M. Gulland and T. F. Macrae, 470

Place Names, Spelling of, 161

Plane Algebraic Curves, Prof. H. Hilton. Second edition (Review), 383

Planet: New Minor, 808; The New, 1933 HH, 881 Planetary System, Origin of the, Dr. R. Gunn, 405 Planets: Minor, 441; Spectra of the, Prof. W. M. Slipher (George Darwin Lecture), 734

Planktonic Diatoms, Spring Increase of, 285

Plant: Genetics, Recent Advances in, Dr. F. W. Sansome and J. Philp (Review), 185; Histology, Methods in, Prof. C. J. Chamberlain. Fifth edition (Review), 44; Virus: Protective Inoculation against a, Dr. R. N. Salaman, 468; Research, Dr. Kenneth N. Smith, 915

Plants: Green, Distribution of Nitrates and 'Organisation' of Nitrogen in the Leaves of, E. Parisi and G. De Vito, 107; The Growth Hormone of, K. V. Thimann and J. Bonner, (2), 631; Physiology of, Functions of Radiation in the, F. S. Brackett and E. S. Johnstone, 331; Sexuality in, Discovery of, Dr. C. Zirkle, 392

Plastic: Materials, Exhibition of, 510; Sediments, Relative Impermeability of the, towards Rain Water, Spring Water and various Alkaline Solutions, P.

Urbain, 739

Plasticine Edible ?, Is, D. I. Clements, N. H. Howes and G. P. Wells, 330

Plasticity: a Mechanics of the Plastic State of Matter, Dr. A. Nádai, assisted by A. M. Wahl (Review), 383 Platinum: Compounds, Decomposition of Some Complex, at Progressively Increasing Temperatures, P. Vallet,

143; Thin Spluttered Films of, Crystalline State of, Prof. G. P. Thomson, N. Stuart and C. A. Murison, 522

Pliny's Chemical Knowledge, Dr. E. J. Holmyard (Review),

Polarisationsebene des Lichtes, Drehung der, W. Kuhn und K. Freudenberg (Review), 677

Polarografo: Il, sua teoria e applicazioni, Dr. G. Semerano (Review), 260

Polar Year, International Commission for the, 1932-33,

Police, Technical Training of the, 754

Polish Guide to Zoology, 905

Pollination in Orchards, G. F. Wilson, 659

Polonium Acetylacetonate, Existence of a, Servigne, 375

Polynesia, Ceremonial Exchange in, Dr. I. Hogbin, 439

Population: Density and Egg-Laying in Flies, Prof. R. Pearl; Dr. Alpatov, 176; Problems in the Pacific, 519

Port: Erin Biological Station, Research at the, 21; Sanitary Regulations, Revised, 616

Positive Electron, New Evidence for the, Dr. J. Chadwick, P. M. S. Blackett and G. Occhialini, 473 Positron, Neutron, Proton and, Dr. N. Thon, 878

Potassium, Radioactive Half-Period of: Calcium Isotope with Mass 41 and the, Prof. J. Kendall, W. W. Smith and T. Tait, 688

Potato, Degenerescence in the, Variations in the Virulence of, J. Costantin, 850

Potatoes: Early, 886; in the Fens, 594; Insect Transmission of $\tilde{\text{Virus}}$ A of, J. B. Loughnane, 838; Normal and Leaf-Roll Variations in the Total Nitrogen Content of, G. Cockerham, 375

Poultry Manure, Production of Kiln-dried, R. Sayce and F. Hanley, 198

Power: Alcohol, 341; Station Efficiency, W. S. Burge, 518

Prehistoric Society of East Anglia: Annual Business Meeting, Dr. Cyril Fox elected president for 1933, 520; Proceedings of the, vol. 7, pt. 1, 465

Pressures, Variable, Methods of Measuring and Recording Rapidly, A. Labarthe and M. Demontvignier, 179

Prickly Pear in Australia, Eradication of, 613 Priestley: and Pepys Commemorations, The, 443; as a

Practical Chemist, Prof. T. S. Patterson, 690; Dr. A. N. Meldrum, 801; as a Scientific Man and Theorist, Sir Philip Hartog, 555; Bicentenary Celebrations at Warrington, 498; Celebration at the Chemical Society, 555; The "Leeds Portrait" of, W. C. Walker, 876; The Scientific Work of, Prof. J. R. Partington, 348

Priestley's: Associations with London, H. G. Wayling, 350; Science to Chemistry, Sir Harold Hartley, 555; Work on Gases and Nitrogen Peroxide, Prof. A. N.

Meldrum, 555

Primes and Factors, Emma Gifford (Review), 785 Professions, The, Prof. A. M. Carr-Saunders and P. A. Wilson (Review), 863

Progress and Scientific Method, 1

Prosperity, The Means to, J. M. Keynes (Review), 451

Protein: Foods in Growth, Comparative Value of Various, E. F. Terroine and Mlle. Simone Valla, 375; Synthesis of, by Green Plants, L. Loose and Dr. W. H. Pearsall, 362

Proton, Neutron, and Positron, Dr. N. Thon, 878

Protons: Liberated by Neutrons, Photographic Detection of, Marietta Blau and Hertha Wambacher (2), 287; Produced in the Artificial Disintegration of the Nitrogen Nucleus, E. C. Pollard, 482

Protoselenosulphochloride, A. Baroni, 483 Prout's Hypothesis, Papers by Dr. W. Prout (1815–16), J. S. Stas (1860) and C. Marignac (1860) (Review), 152

Proverbes et maximes Peuls et Toucouleurs traduits, expliqués et annotés, H. Gaden (Review), 347

Psychoanalysis, The Medical Value of, Dr. F. Alexander (Review), 532

Psychologique, L'Année, Publiée par Prof. H. Piéron. Année 32 (1931). Vols. 1 and 2 (Review), 604

Psychology: Genetic, (1), J. L. Gray and Pearl Moshinsky, 922; in Autobiography, A History of, Vol. 2. B. Bourdon, and others (Review), 321 Psychotherapy, Analytical, An Introduction to, Dr. T. A.

Ross (Review), 639

Public Analysts, Society of, election of officers, 359 Puccinia Helianthi Schw., Union of Pycniospores and Haploid Hyphæ in, J. H. Craigie, 25 Pumping Machinery, Science Museum Handbook of, G. F. Westcott, 542

Pycnogonid, A Dodecapodous, Dr. W. T. Calman, 242 Pyridine-Iodine Molecular Association, A, M. Chatelet,

Pyrométrie optique, Traité de, Prof. G. Ribaud (Review),

Pyrophosphoric Acid, Absorptive Power of Soil for, A. Fabris, 775

Pyrrhotite, Iron Lattice of, Vacant Positions in the, Prof. G. Hägg, 168

Pyrrolic Aldehydes, Action of Alkaline Hypoiodites on the, P. Pratesi, 924

Quail, Changed Standing of the, 337

Quantem-mechanik, Die Gruppentheoretische Methode in der, Prof. B. L. van der Waerden (Review), 531 Quantification, Théorie de la, dans la nouvelle mécanique,

Prof. L. de Broglie (Review), 639

Quartz: as a Standard for Accurate Lattice-Spacing Measurements, A. J. Bradley and A. H. Jay, 813; Crystal, Oscillating, as an Accurate Clock, Scheibe and Adelsberger, 281

Quinine Salts, Rotatory Power of, in Aqueous Solution, C. Lapp, 703

Queen: of Bermuda, Electrical Equipment of the, 796; of the Sciences, The, Prof. E. T. Bell (Review), 319 Quekett Microscopical Club, election of officers, 272

Rabbit: Blood Group Inheritance in the, W. E. Castle and C. E. Keeler, 775; Tests for Linkage between the Blood-Group Genes and other known Genes of the, W. E. Castle and C. E. Keeler, 775

Racial Distributions in Palæolithic Europe, M. C. Burkitt, 540

Radiation, Penetrating: Absorption of, Prof. E. Regener, 880; Spectrum and Latitude Variation of, Dr. E. J. Williams, 511

Radioactive: Minerals, Primary, Analysis of, A. Karl, 143; Substances, New, A. Debierne, 630

Radio: Communications by very Short Electric Waves, Marchese Marconi, 292; Direction-Finding, C. E. Horton and C. Crampton; J. F. Coales, 846; -Electric Waves, Short, Apparent Velocity of, N. Stoyko and R. Jouast, 887; Research Board, Report of the, for the year 1931, 156; Telephone Links, 731; Union, International Scientific, Prof. A. E. Kennelly elected president of the, 55; Valve, The 'Catkin', 735; -Waves, Interaction between, B. D. A. Tellegen, 840

Radiometer Action and the Pressure of Radiation, Mary Bell and S. E. Green, 374

Radium in Canada, Assays of, H. S. Spencer, 55

Railway Electrification Experience, Col. C. Leigh, 649 Railways in Britain, Electrification of, F. Lydall, 19

Rainfall: at Capodimonte during 1833-1931, E. Guerrieri, 107; over the British Isles, 1820 to 1929, J. Glasspoole, 737

Raman: Effect (24), H. Kopper and A. Pongratz, 251;

Spectra and Chemistry, 263
Rangoon, Geology and Underground Water of, P. Leicester, 916

Rare Earth: Elements, Radioactivity of some, Libby and Latimer, 368; Group, Use of Non-Aqueous Solvents in the Study of the, B. S. Hopkins and L. L. Quill, 739; Earths, Fluorescence of Pure Salts of the, Prof. R. Tomaschek and O. Deutschbein, 473

Rat: Crested, at the London Zoo, 794; The Crustacean Chromatophore Activator and the Gonads of the,

B. Kropp, 631

Ray Society, election of officers, 466

Rayon Industry, The (Review), 454
Rays Entering the Eye Pupil at Different Points,
Luminous Efficiency of, W. S. Stiles and B. H. Crawford, 250

Reading University: appointment of Dr. R. H. Stoughton as professor of horticulture, 248; Prof. R. Rae appointed professor of agriculture, 772

Recrystallisation and Coloration, Prof. K. Przibram, (3),

Rectifier Photo-Cells, Reversal of Current in, J. Guild, 327; Dr. H. H. Poole and Dr. W. R. G. Atkins, 547 'Red Coal', Prince Ginori Conti, 679

Red Deer Shed their Antlers, 629

Reflexology (*Review*), 675 Regression, Phenomenal, to the Real Object, Prof. W. Peddie; Dr. R. H. Thouless, 544

Regulus, Occultation of, on April 6, 553

Reinmuth Planet, 1932 HA, Dr. G. Stracke, 369 Religious Experience: its Nature, Types and Validity, Rev. A. C. Bouquet (Review), 677

Relativity: Addresses and Discussions on, 722; and the Structure of the Universe (Review), 637; of Time, Kennedy and Thorndike, 136; Special Theory of, Momentum and Energy in the, Prof. A. C. Banerji,

Remembering: a Study in Experimental and Social Psychology, Prof. F. C. Bartlett (*Review*), 309

Renéville (French Congo), Borings in the Marly Limestones of, H. Lagotala, 215

Reproduction, Lunar Periodicity in, Prof. T. A. Stephenson, 622

Research: Financing, 289; in Central Asia, 705; The Spirit of, Dr. T. Brailsford Robertson. Edited by

Jane W. Robertson (Review), 111

Resonance: Ag I Lines, Fine Structure of the, W. E. Williams and A. Middleton, 692; Radiation: Frequency Distribution of, R. W. Ditchburn, 106; Possibility of Separating Isotopes by the Use of, R. W. Ditchburn, 106; through a Gas Transmission of, R. W. Ditchburn, 738

Rhenium: Dr. E. S. Kronman (Review), 224; Metallic, The Constant Paramagnetism of, N. Perakis and L. Capatos, 559; Spectra of, H. Schober, (3), 411; (4), 851

Rhesus Monkey, Sexual Cycle in the, Dr. C. G. Hartmann, 626

Rhodesia, Southern, Meteorology in, 54

Ribaucour's Transformation and Spherical Representation (3), H. Hamburger, 339

Rice Grass, The Economic Possibilities of, J. Bryce,

Richborough, Roman, Dr. J. P. Bushe-Fox, 539

Rickets, Experimental, as a Phosphorus Deficiency Disease, Prof. H. D. Kay and Dr. B. L. Guyatt,

Ringworm Fungi, Development of Fuseaux, Aleuriospores, and Spirals on detached Hairs infected by, Dr. A. M. Davidson and Dr. P. H. Gregory, 836

Road: Island Refuges, Guard Posts for, 394; Research Board, Major F. C. Cook appointed chairman of the, 686; Station, Harmondsworth transferred to the Department of Scientific and Industrial Research, 686

Rock: -Salt, Blue, Recrystallisation and, Prof. K. Przibram, 287; Specimens of, for Measuring their Electrical Resistance, Methods for putting in Circuit,

Mme. F. Bayard-Duclaux, 887

Rocks: Certain Igneous, Magnetic Properties of, A. F. Hallimond and E. F. Herroun, 338; Thermal Conductivities of, Determination of the, H. A. Nancarrow, 702

Rodents, Jumping, A. B. Howell, 623 Roebuck, Dr. John, Work of, 196

Roof Falls in Mines, Prevention of, 140

Root: Nodule Bacteria and Leguminous Plants, E. B. Fred, I. L. Baldwin and Elizabeth McCoy, 99; Studies, III, W. S. Rogers, 695

Roscoe, Sir Henry Enfield, Centenary of the birth of, 18 Ross Institute for Tropical Diseases: Activities of the,

300; Annual Report for 1932, 835

Ross: Ronald, Discoverer and Creator, R. L. Mégroz (Review), 40

Rotatory Power, Curious Case of Change of Sign of the, and of Mutarotation, Pariselle, 702

Royal: Academy: Lessons in Visualisation from the, Dr. Vaughan Cornish, 644; Portraits at the, 646; Winter Exhibition, 52; Aeronautical Society: award of the gold medal to Sir Richard Glazebrook, 359; award of silver medals to D. L. H. Williams and A.H.R. Fedden, 510; foundation of two medals; award of the British silver medal to Capt. C. F. Uwins, Squadron-Leader O.R. Gayford and Flight-Lieut. G.E. Nicholetts, 685; Asiatic Society of Bombay, Sir John H. Marshall awarded the gold medal of the, 683; Astronomical Society: award of the gold medal to Dr. V. M. Slipher, 91; election of officers, 236; Cornwall Polytechnic Society, 99th Annual Report, 833; Geographical Society, awards of the, 543; Danish Academy, Dr. C. Tate Regan elected a foreign member of the, 650; Dublin Society, Prof. P. A. Murphy awarded the Boyle medal of the, 683; Entomological Society of London, The, 678; Institution: Annual Meeting; presentation of portrait of Sir William Bragg, 647; gift from the Rockefeller Foundation, 686; Meteorological Society, election of officers, 200; Microscopical Society, election of officers, 128; Observatory, Greenwich, Annual Visitation, 882; Society: Mond Laboratory, Cambridge University, Opening of the, 210; Recommendations for election to the, 353; Prof. V. F. K. Bjerknes, Prof. H. W. Cushing, Prof. P. Debye, Prof. F. A. F. C. Went elected foreign members of the, 798; of Edinburgh: election of fellows, 354; award of the Makdougall-Brisbane prize to Dr. A. C. Aitken, and the Gunning Victoria Jubilee prize to Sir James Walker, 359; presentation of the James Scott prize to Prof. A. Sommerfeld, 687; of South Africa:

election of officers, 22; election of officers, 687 Rubber: Industry Bill, 650; New Source of, 21; Research, 181; Dr. S. S. Pickles, 273; Bill, 576

Russia: Botany in, 554; Petroleum Production in, R. C. Beckstrom, 541

Russian Arctic Expeditions, 797

Ryegrasses, Fluorescence in, Linehan and Mercer, 474; E. W. Hullett and J. W. Calder, 475

RZ Cassiopeiæ with Eclipses, The Double System, J. Ellsworth, 850

Sabellids, Blood Circulation in, Reversible Stoppage of the, Prof. H. Munro Fox, 26

Safeguarding of Industries Act, Articles Chargeable with Duty under Part I of the, 651

Safety in Mines Research Board, appointments to the, 91 Sage Grouse, Display of, 773

Saharan Rock-Engravings, Dr. T. Monod, 551

St. Andrews University: offer by F. O. Salisbury of a replica of a portrait of Dr. E. Harkness, 372; proposed conferment of doctorates, 480; retirement of Prof. J. E. A. Steggall, 628; J. Anderson appointed professor of surgery, 848

St. Hilary's Day, 69

St. Valentine's Day, 214

Salamander Embryos, Segmentation of Spinal Nerves in,
S. R. Detwiler, 739
Salmon: Fishing, E. Taverner, and others (*Review*), 42;

Salmon: Fishing, E. Taverner, and others (Review), 42; Hatching and Salmon Migrations, Dr. W. L. Calderwood (Review), 42; Season; Disease, 213

Salt, Inorganic, An Optically Active, Mann, 808

Saltpetre, Chile, Origin of the, Dr. C. T. Kautter, 556
Samarium: Radioactivity of, and the Formation of Hibernium Halos, Dr. J. H. J. Poole, 654; Range of Radiation from, Prof. G. Hevesy and Dr. M. Pahl, 434

Sandfell, Laccolith and 'Dome of Elevation', L. Hawkes and Hilda K. Hawkes, 214

San Diego: Know Your, M. J. Rogers, 903; Museum, 903

Santa Monica Mountains, California, Glacial Epochs of the, Prof. W. M. Davis, 288

Schistocerca peregrina, Desert Anhydrobiosis and its Influence on the Animal Cycle of, E. Roubaud, 774

Schistosomiasis in the Philippines, M. A. Tubangui, 99
Schneider: Mediumship, Further Light on the: (Review),
489; Lord Charles Hope, 549; H. Price, 658;
Rudi, An Account of Some Further Experiments
with, a Minute-by-Minute Record of 27 Séances,
H. Price (Review), 489

Schools and the Social Upheaval, 408

Science: A Symposium of, Prof. F. S. Marvin (Review), 674; Anatomy of Modern, an Introduction to the Scientific Philosophy of To-day, B. Bavink. Translated, with additional notes and Bibliography, by H. S. Hatfield (Review), 707; and Culture (Review), 707; and Service in Universities Overseas, 377; and the Textile Industry, Dr. K. Lee, 162; bequests to, Sir Dugald Clerk, 235; Forum, No. 1, 649; General, F. Fairbrother and E. Nightingale. Part 1 (Review), 420; ?, General, What is, 141; History in, 777; in the Changing World, T. Holland, and others. Edited by Mary Adams (Review), 674; in the Schools, 217; Irresponsibility of, Prof. H. Levy, 162; ? Is Man Ethically Fit for the Bounties of, Prof. D. D. Kanga, 233; Language in the Service of, 741; Masters' Association, Annual Meeting of the, F. C. Tompkins, 65; Museum, Photoelectric Exhibition at the, 429; Tabular Data of (Review), 892

Sciences, Meet the, W. M. Malisoff (Review), 639
Scientific: and Industrial Research, Department of,
Report for Year 1931–32, 570; 669; and Learned
Societies of Great Britain and Ireland, The Official
Year-Book of the, 1931–1932 (Review), 45; and
Technical Books, Recent: January 28, iii; February
25, iii; March 25, v; April 29, v; May 27, v;
June 24, iii; Centenaries in 1933, Eng.-Capt. E. C.
Smith, 14; Centralisation in the British Empire,
Prof. C. Stanton Hicks, 397; Method, Ethical Conditions of, Bishop of Durham (Fison Memorial
Lecture), 902; Progress and, 1; Professionalism,
861; Riddles, Sir J. Arthur Thomson (Review), 315

Scientists, A Rebuke to, Prof. A. Findlay, 1
Scotia Sea, Soundings in the, H. F. P. Herdman, 440
Scottish Marine: Biological Station, Millport, Report of the, 904; Fauna, A. C. Stephen, 446; Tectonics,

Help from America in Reading, Prof. E. B. Bailey, 522

Scott's Own Expedition (Review), 782

'Sea: -Blue Bird of March', The, 445; -Lions in California, Breeding Season of, 885; Movements of the, Use of the Cattaneo Wave Pump for Utilising the, J. Richard, 179

Seal, Harp: Breeding Habits of the, 558; Food of the, and its Significance, 558

Sedimentation, Researches on, 394

Seed Germination, Temperature Relations of, T. I. Edwards, 404

Seeds: Germinating, of Pisum sativum and Triticum vulgare, Influence of certain Chemical Agents on the Carbon Dioxide Output of, under Optimum Germination Conditions, J. Kisser and R. Furtauer, 251; Influence of Treatment of, with Chemical Stimulants on the Cell Growth of the Rootlets, J. Kisser and J. Schubert, 251

Seismographs, Hydraulic, Prof. S. K. Banerji and K. N.

Sohoni, 547 Seismology, Theoretical, Introduction to, Rev. J. B. Macelwane and Rev. F. W. Sohon. Part 2: Seismometry, Rev. F. W. Sohon (Review), 824

Selectivity and Radio Communication, 442 Semi-Conductors in a Magnetic Field, J. W. Harding, 731 Seric Proteins Isolated by the Acetone Method and Myxoprotein, Modification of the, A. Bonot, 106

Severn Barrage Scheme, The, 449

Sewage Purification by a New Process, G. H. Gleason

and A. C. Loonam, 697 Sex: Artificial Control of, in the Progeny of Mammals, Prof. N. K. Koltzoff and V. N. Schröder, 329; Hormone, A New Crystallised, A. Girard, G. Sandulesco, A. Fridenson and J. J. Rutgers, 71; Hormones and Cancer-Producing Compounds, Dr. J. W. Cook and Prof. E. C. Dodds, 205

Sexual Cycle in the Rhesus Monkey, Dr. C. G. Hartmann,

Shamal, 885

Shamans and Spiritism on the Mosquito Coast, Honduras, Dr. E. Conzemius, 135

Sheep, Nutritional Condition of, and Susceptibility to Stomach Worm, A. H. H. Fraser and Dr. D. Robert-

son, 94

Sheffield University: Dr. G. A. Clark appointed professor of physiology, W. R. Maddocks lecturer in metallurgy, J. Dick assistant lecturer in mechanical engineering, A. J. Macdougall assistant lecturer in metallurgy, 736; Dr. L. C. D. Hermitte appointed demonstrator in pathology, Dr. E. James junior assistant bacteriologist and demonstrator; resignation of Prof. E. Mellanby and J. W. Kershaw, 884 Shrimp Industry of Leigh-on-Sea, A. L. Wells, 270

Silica: Gel, Heats of Moistening, with various Liquids, R. Berthon, 106; in the Organism and the Siliceous Particles of the Blood, L. Lemarte and E. Kahane, 523; Volatile Transport of, J. W. Greig, H. E. Mervin and E. S. Shepherd, 768

Silkstone Coal Seam of Yorkshire, 64

Silver: Arc Spectrum of, Structure of the Lines of the, D. A. Jackson, 691; Iodomercurate, F. Gallais, 215; the Life-Story of an Atlantic Salmon, R. L. Haig-Brown (Review), 42; Voltameter, New Form of, M. Picard and A. Stampa, 666

Size Limits, Legal, and their Effect upon Fisheries, N. B.

Scofield, 831

Skiddaw Granite and its Residual Products, C. S. Hitchen, 482

Skye, Differentiated Sills in, Dr. F. Walker, 440

Skylark's Song, 105

Sleep, Agglomeration Theory of, W. D. Bancroft and J. E. Rutzler, Jr., 739

Smithsonian Field Expeditions in 1932, 721

Snails and Changes in Sea-Level, Prof. T. D. A. Cockerell,

Snail's Tissue, Culture of, Prof. J. B. Gatenby and E. S. Duthie, 474

Snell Memorial medal of the National Institute of Agricultural Botany, award of the, to Dr. Kenneth M. Smith, 164

Soap Film Structure, The Hydroxyl Group and, W. J.

Green, 873

Social: Anthropology, Dr. P. Radin (Review), 315; Hygiene Congress, Sixth Imperial, 798; Study, Methods of, Sidney and Beatrice Webb (Review),

Sociological Review, Vol. 24, No. 3, Supplement to, 466 Sociology, Institute of, Incorporation of the, 33

Socrates, Before and After, Prof. F. M. Cornford (Review),

Sodium: Benzoate, Chlorination of, Dr. J. C. Smith, 28; Salicylate, Yield of Fluorescence of, P. Dubouloz, 850; Sulphite in Photographic Developing Solutions, G. Gopala Rao and Madhusudanan Pandalai, 100

Soil: Chemical Composition of the, and that of its Colloidal Contents, Relationships between the, O. Bottoni, 560; Electrical Properties of, for Alternating Currents: with Particular Reference to Radio-Frequencies, Dr. R. L. Smith-Rose, 142; Fertility, Forest Fires in Relation to, Prof. F. P. Worley, 787; Microbiology, Principles of, Prof. S. A. Waksman. Second edition (*Review*), 316; Scientific Study of the, An Introduction to the, Prof. N. H. Comber. Second edition (*Review*), 316; Tilling the, 141 Solanine Poisoning, S. G. Willimott, 813

Solar: Activity, a Remarkable Epoch of, H. Mémery; E. Esclangon, 559; Eclipse, Effects of, on Audio Frequency Atmospherics, E. T. Burton and E. M. Boardman, 81; Halo, A Complex, R. C. T. Evans and Dr. C. Evans, 613; Prominences at Meudon, Direct Observation of, B. Lyot, 70; 332; Radiation in Relation to its Warming Effect on the Human Body, H. M. Vernon, 737; Spectrum from 4040 to 4390 A., Microphotometry of the, Dr. R. v. d. R. Woolley, 917; System, Origin of the, A. C. Gifford, 518

Solid Solutions by Precipitation, and Isomorphism between Complexes of Platinum and of Quadrivalent Tellurium, G. Natta and R. Pirani (2), 107

Solids, Decomposition and Detonation of, Prof. W. E. Garner, 65

Soot Films and Oil, Interaction between, J. H. Coste, 691; Dr. S. C. Blacktin, 873 'Sooty Moulds' of Some Australian Plants, Eileen E.

Fisher, 107

Sothic Cycle, A, Mrs. A. S. D. Maunder, 332

Sound: Films: at Chicago University, 593; Synthetic, 125; in Gases in Tubes, Velocity of, Dr. G. W. C. Kaye and G. C. Sharratt, 338; The World of, Sir

William Bragg (Review), 860 South Africa: Geology of, Dr. S. H. Haughton, 624; Royal Society of, election of officers, 22; Vanished Races in, D. S. van der Merwe, 392; African: Culture?, A New, Dr. E. C. N. Van Hoepen, 393; Institute for Medical Research, Report for 1931, 22; Australia, Mining Facilities in, 395; -Eastern Union of Scientific Societies, 38th Annual Congress, 919; Prof. H. L. Hawkins elected president for 1934, 920

Soviet Education, The Broad Highway of, C. A. Harrison,

Spat-fall and Shore Populations, 921

Spatial Distance in a Curved Space-Line, Measurement of, H. S. Ruse, 250

Spearman's General Factor, Ambiguity in Sign of, Prof. H. T. H. Piaggio, 170

Species, History of, The Bacterial Growth Curve and the, Dr. A. S. Corbet, 61

Spectral Lines in Indium and Gallium, Relative Intensity of, Miss R. Payne-Smith, 365

Spectrograph for Study of Fibrous Substances, Adam Hilger, Ltd., 844

Spectroscope, Application of the, to Biology, H. Ramage, 919

Spectroscopic Analysis by X-Rays (Review), 39 Spectroscopie, Handbuch der, Prof. H. Kayser und Prof. H. Konen. Band 8, Lief 1 (Review), 824

Spectrum: Analysis at the Massachusetts Institute of

Technology, 393; Infra-Red Region of the, C. R. Bailey and A. B. D. Cassie (8), 286 Spektralphotometrie, Objective, Drs. L. S. Ornstein, W. J. H. Moll, und H. C. Burger (Review), 824

Spencer's Scientific Correspondence with Sir J. G. Frazer and others. Edited by Dr. R. R. Marett and T. K. Penniman (Review), 187

Sperm Whales in the North Atlantic, 594 Spider Crabs from Japan, Dr. Tsune Sakai, 551 Spiders: of Denmark, The (Review), 149; The Biology of, with especial reference to the Danish Fauna, E. Nielsen. 2 Vols. (Review), 149

Spike Disease in Indian Sandal, 89; 720

Spiral Nebulæ, Motion of the, Dr. Knox Shaw, 697

Sponges: Californian, M. de Laubenfels, 588; from the North-Western Pacific, Dr. Y. Okada, 279; Lightproducing Powers of, Prof. E. Trojan, 728; Locomotion in, M. Burton, 356

Spray Spreaders, Present uses and Future Development

of, H. Martin, 768

Spreng- und Zündstoffe: Chemische Untersuchung der, unter besonderer Berücksichtigung der zu ihrer Herstellung notwendigen Ausgangsstoffe. Unter Verwendung eines von Hermann Kast hinterlassenen Manuskriptes. Herausgegeben von L. Metz (Review),

Spring: Equinox, The, 409; Flush, The, 813; Winds and some Biological Effects, 409

Squirrel, The Grey, A. D. Middleton (*Review*), 45 'Stainless-Invar', A New Alloy, Prof. K. Honda, 587 'Standard Gold' and Silver: the Way Out of the Crisis,

Dr. E. Zucker, 198

Star: Occulted by Jupiter, A, 845; of the Cepheid Type, Mean Absorption Coefficient in a Variable, G. Tiercy, 888; Influence of the Absolute Magnitude of a, on the Width of the lines of Stellar Hydrogen, P. Rossier, 483; A Variable, with Eclipses of Short Period, R.

Tremblot, 815

s: Class Be, Suggested Mechanism of, D. B. McLaughlin, 739; Colour Temperatures of, 281; Composition of the (Halley lecture), Prof. H. N. Stars: Russell, 832; Double, Measures of, at the Union Observatory, Johannesburg, W. H. van den Bos, 476; Ephemerides of Variable, 590; Faint, Spectral Types of, Dr. Humason, 209; Faint, with Common Proper Motion, W. J. Luyten, 625; New Proper Motions of, from Bergedorf Observatory, C. Vick, 625; of the μ-Cephei Type, W. Zessewitsch, 137; of the Cluster M7, Diameter of some, Rossier, 888; of the Type A5, Width of a Photographic Stellar Spectrum for, G. Tiercy and A. Grosrey, 888; Variable, in the Globular Cluster M.53, 441; Radial Velocities of, Dr. P. W. Merrill, 917

State-aided Research in Great Britain, 669

Statistical: Methods for Research Workers, Dr. R. A. Fisher. Fourth edition (*Review*), 383; Tables, Dr.

R. A. Fisher (Review), 893

Steam: Action of, on Heavy Petroleum Oils and on certain Cyclic Hydrocarbons, C. Matignon and M. Séon, 523; -Jet Operated Air Ejectors, F. R. B. Watson, 369; Tables, N. S. Osborne, H. F. Stimson, E. F. Fiock, and D. C. Ginnings, 624

Steel, Medium Carbon, Fatigue Limit of, J. W. Cuth-

bertson, 140

Steels: Alteration of, by Hydrogen, L. Jacqué, 36; Rustless, Measurement of the Degree of Polishing in view of the Determination of the Amount of Corrosion of, J. Cournot and Mlle. Louise Halm, 738

Stellar Spectra, Bright Lines in, 64

Stereochemie: Prof. S. Goldschmidt (Review), 783; eine Zusammenfassung der Ergebnisse, Grundlagen und Probleme, Herausgegeben von Prof. K. Freudenberg. Lief. 1, 2, 3 (Review), 563

Stereochemistry and Physics, Relations between, Prof. P. Debye (Faraday lecture), 498

Sterility in Domesticated Animals, Dr. W. Orr and Dr. F. F. Darling, 200 Stoic, The, 104

Stoke Park (R. G. Burden) Fund for Research in Mental Disorders, 159

Stone, Decay of, Relationship of Micro-organisms to the, S. G. Paine, F. L. Lingood, Freda Schimmer, and T. C. Thrupp, 178

Stork in Western Europe, J. Sokolowski, 403 Strahlenoptik, Dr. M. Herzberger (Review), 747

Stratosphere: New Ascent into, 614; Temperature of the, Dr. B. Rolf, 280 Strawberry 'Yellow Edge' Disease, R. V. Harris, 730

Strepsinema Stage in Reduction, Prof. H. H. Dixon, 437 Sturgeon, Spawning-time of the, 812

Styria: Eastern, Geology of, the Rocks and their Inter-Relationships, R. Schwinner, 560; Flower-Visiting Insects in, 1913, K. Fritsch, 852

Subject Index to Periodicals, 1931 (Review), 711

Sublimation, Inverse, O. Dony, 731

Submarine: Faulting in Kimmeridgian Times, Prof. E. B. Bailey and Dr. J. Weir, 244; Telephone Cables at Carrier Frequencies, Dr. E. W. Smith, 431; Valleys, Prof. F. P. Shepard, 331 Subsidence of London, T. E. Longfield, 558 Sudan: Anglo-Egyptian, Fossiliferous Siliceous Boulders

from the, L. R. Cox, 70; Nilotic, Pagan Tribes of the, Prof. C. G. Seligman and Brenda Z. Seligman (Review), 345; Southern, Tribes of the, E. B. Haddon (Review), 345

Suffolk, East, Rural Education in, 848 Sugar in the Blood, Application of the Iodometric Method to the Estimation of, H. Bierry, B. Gouzon, and Mlle. C. Magnan, 667

Sulphates as Manure, Importance of, G. Bertrand and

L. Silberstein, 215

Sulphur: and Nitrogen, Utilisation of, from Cystine at the level of Endogenous Protein Metabolism, E. F. Terroine, P. Mezincesco, and Mlle, Simone Valla, 483: Hexafluoride, Higher Homologues of, K. G. Denbigh and Prof. R. Whytlaw-Gray, 763; -Hydrogen Reaction, New Experimental Evidence in the, E. E. Aynsley, Dr. T. G. Pearson, and Dr. P. L. Robinson,

Sulphuric Acid, Ionisation of, Prof. H. M. Dawson, 375

Summer Thermocline in Seas and Lakes, 773

Sun, Total Eclipse of the, Aug. 31, 1932, A. de La Baume Pluvinel and D. Barbier, 922

Sun-Fish in British Waters, 665

Sunspot Areas and Numbers for 1932, 661 Sunspots: and Meteorological Phenomena, H. Mémery, 591; Earth's Influence on, P. R. Chidambara Aiyar, 245; Large Group of, 209; on the Solar Disc, Zones of Apparent Inhibition of, Prof. J. A. Carroll, 548

Supersonic: Blower, Some Results obtained with a, L. Santon, 523; Vibrations set up in a Zinc Bar undergoing Transverse Vibrations, Prof. K. Prosad and

S. Sharan, 803

Surface: Catalysis (Review), 150; Tension and its Measurement, Dr. A. Ferguson, 66

Surrey Flora, A New (Review), 77

Surveying from Air Photographs, A Simple Method of, Lieut. J. S. A. Salt, 650

Suspensions Prepared starting with Proteins separated from Serum by the so-called Acetone Method, C. Achard and A. Boutarie, 178

Swallow and Cuckoo Arrive, 629 Sweden, Higher Education in, 480

Sword, They that take the, Dr. E. Wingfield-Stratford (Review), 635

Syllis ramosa, McIntosh Distribution of the Polychæte Worm, Dr. C. Crossland, 242

Symbols, Units and Nomenclature Commission, Conference of the, A. E. Kennelly, 775

Symmetrical Region, Variation Problems for a, Prof. H. Bateman; Prof. E. T. Whittaker, 472

Symmetry, Determination of, A Possible Source of Error in the, from Optical Extinction-Angles, M. H. Hey,

Tables for Statisticians and Biometricians. Edited by Prof. Karl Pearson. Part 1, third edition; Part 2 (Review), 893

Tabular Data of Science (Review), 892

Talbot's Bands, late A. C. G. Beach, 702

Talking Pictures: and Acoustics, C. M. R. Balbi (Review), 188; and Picture Telegraphy (Review), 781

Tapir, Concealing Coloration of the, 867

Tartaric Acid and Chromium, Compounds of, J. P. Mathieu, 106 Tata, Lady, Memorial Fund scholarships, award of, 920

Taxonomy and Evolution, Dr. W. H. Longley, 863

Technical: Education: F. H. Reid, 883; International Bureau of, A. Abbott and E. G. Savage appointed to act as representative and correspondent respectively, 811; Institutions, Association of Teachers in, Annual Conference, 883

Technocracy, 87

Tektites: Origin of, Dr. L. J. Spencer, 117; 596; 876; F. Chapman, 876

Television: Sir Ambrose Fleming, 539; First Principles of, A. Dinsdale (Review), 456

Tell: -Billa, Dr. E. A. Speiser, 806; Duweir, Palestine,

Excavations at, 648; L. J. Starkey, 897 Tellurium Fluorescence, Magnetic Quenching of, R.

Smoluchowski, 914

Temperature: below 0.27° K., Prof. W. J. de Haas, E. C. Wiersma, and Prof. H. A. Kramers, 719; Recorder, A New, Dr. W. Cawood and H. S. Patterson, 332

Tempora Mutantur, 105 Terns, Homing Sense of, 701

Terrestrial Magnetic Field, Diurnal Incidence of Disturbance in the, A. Crichton Mitchell, 446

Tertiary Alcohols, Dehydration of some, by Anhydrous Copper Sulphate, A. Meyer and M. Tuot, 851

Textile: Design: and Colour: Elementary Weaves and Figured Fabrics, W. Watson. Third edition, with appendices on the manufacture of Rayon (Artificial Silk) and Standard Yarns, Weaves and Fabrics (Review), 187; and Fabric Structure, Elementary, J. Read (Review), 187; Industry, Science and the, Dr. K. Lee, 162

Thallous Thiocarbonate, Preparation and Properties of,

Picon, 179

Thames Valley, Lower, Deposits, 756

Theileria dispar, Experimental Suppression of Sexual Reproduction in, E. Sergent, A. Donatien, L. Parrot, and F. Lestoquard, 143 Thermal Balance of a Water Drop or Ice Particle Sus-

pended in the Atmosphere, C. S. Durst, 35

Thermionic Vacuum Tubes and their Applications, Prof.

E. V. Appleton (Review), 9

Thermoelectric: Elements, Construction of, A. Cotton, 559; Phenomena, A New Method of Realisation of, A. Egal, 410

Thicknesses, A Magnetic Apparatus for the Determination of, A. Bricout, 595

Thiobenzophenone, Magnetic Rotatory Dispersion of a Coloured Diamagnetic Compound, A. Cotton and M. Schérer, 215

Thiocyano- and Thio-pyrroles, and Pyrrole Disulphides,

P. Pratesi, 411

Thiosulphuric Acid, Decomposition of, in Dilute Solution at the Boiling Point, E. Carrière and Mlle. Carlini,

Thixotropy of certain Salts, Causes of the, P. Bary, 338 Thompson: Mary Clark, medal of the Washington National Academy of Sciences, award of the, to Dr. F. A. Bather, 304; John, the Work of, 232

Thomsonite, Dehydration of, A. E. Mourant, 630

Thorium Sulphide, Picon, 70

Thought in Progress, The Play of (Review), 746

Thunderclouds, Penetrating Rays from, Schonland and Viljoen, 916

Thunderstorms, Summer, S. Morris Bower, 473

Tide Marks, Quantitative Studies between the, R. Elmhirst, 767

Tidal Shifts in the Earth's Crust, Dr. H. T. Stetson and Dr. A. L. Loomis, 137

Tierwelt der Nord -und Ostsee, Die, Begründet von G. Grimpe und E. Wagler. Herausgegeben von G. Grimpe. Lief. 21. Teil 1.d₂; Teil 2.g (*Review*),

Tilletia Species of Wheat in Australia in 1931, Geographical Distribution of, J. G. Churchward, 631

Timber: During Seasoning, Longitudinal Variation of, M. B. Welch, 667; Home-Grown, Census of Production of, 905; Investigations on, 847 'Time' and 'Events', Ideas of, W. W. L., 727

Time: Determination, Dr. J. de Graaf Hunter, 515; Divisions of, in use in Ancient Mesopotamia, Dr. J. K. Fotheringham, 299; Measurement: Old and New, Hope-Jones, 67; Measurements, Number 60 in: 299; Prof. H. Chatley, 914
Timothy Grass, A Fly Pest of, L. A. L. King and Miss

Agnes A. Meikle, 837

Tissue Structure of Homologous Organs (Prepuce Glands) in Mice and Rats, Differences in the, J. Schaffer, 703 Tobacco Mosaic, Some Effects of the Ordinary, upon the Developmental Anatomy of the Host Plant, J. Grainger and Rachel M. Heafford, 814

Tomato Late-Blight Rot, G. B. Ramsey and Alice A.

Bailey, 440

Tonfilm: Aufnahme und Wiedergabe nach dem Klangfilm-Verfahren. (System Klangfilm-Tobis.) Herausgegeben von Dr. F. Fischer und Dr. H. Lichte (Review), 781

Tonga Archipelago, Fishes from the, H. W. Fowler, 63 Tonphotographie: Einführung in die, photographische Grundlagen der Lichtton-Aufzeichnung. Prof. J. Eggert und Dr. R. Schmidt (Review), 781

Tornadoes in U.S.A., 481

Tortoise, Anatomy of the, late Dr. J. Stuart Thomson, 396 Trail that is Always New, The, W. P. Lowe (Review), 636 Transitory Regimes, E. Crausse and J. Baubiac, 559 Transport Problems of the Empire, W. R. Jeffreys, 126 Trans γ-Oxycrotonic Acid, R. Rambaud, 483

Tree: Growth in the Tropics, Periodicity in, Jean Schweizer, 624; Roots, Dr. E. V. Laing, 176

Trees: of Ireland, H. M. Fitzpatrick, 35; Roots of, Disintegration Action of, M. A. R. Khan, 844; Twisted, H. G. Champion, 133

Tremadoc Shales in the Tortworth Inlier, Gloucestershire, T. Smith, with notes on Fossils by C. J. Stubblefield,

214

Trematodes of Marine Mammals, E. W. Price, 243 Trevithick Centenary Commemorations, 577; 609

Triatomic Molecules, Structure of, A. B. D. Cassie, 438 Trigonometry, Elementary, Dr. J. Prescott and H. V. Lowry (Review), 420

Trinkler Expedition to Central Asia, The, Dr. L. D. Stamp (Review), 600

Trout: Brown, Spring Food of, 337; Fishing in New Zealand, Prof. E. Percival, 163; Sea, The Life of the, especially in Scottish Waters: with Chapters on the Reading and Measuring of Scales, G. H. Nall (Review), 42

Tsetse Fly, Control of the, R. W. Harris, and others, 463

Tumor Tissue, Application of Optical Spectroscopy to Analysis of, Winifred R. Mankin, 668 Tumours, Growth of Transplantable, in Plasma and Serum Culture Media, R. J. Ludford, 250

Tungstates, Metallic, Preparation of the, A. Karl, 923

Turbine, Early Days of the, C. Turnbull, 20

Turnips, Resistance of, to Finger and Toe Disease, 849 Turtle: Leathery, Egg-Laying of, 849; Painted, Egg-Laying and Incubation of, 849

Twins, Intellectual Resemblance of, L. Herrman and Prof. L. Hogben, 446

Uganda: Archæological Discoveries in, E. J. Wayland, 439; Stone Age Culture in, E. J. Wayland and M. C. Burkitt, 730

Ulijanowsk, Plant Geography of, H. Grosset, 696 Ultra: -filtration, Principles of, as applied to Biological Problems, W. J. Elford, 178; -Microscopic Organisms and the troubles which they cause, Sir Henry Dale, 370; -Pressures, Biological Effects of, J. Basset and M. A. Machebœuf, 251; J. Basset, Mme. E. Wollman, M. A. Machebœuf, and M. Bardach, 774; -Shortwave Radio Research, E. C. S. Megaw, 269; -Sonic Radiation, Chemical and Biological Effects of, Prof. A. Szent-Györgyi, 278; -Sound Waves, Measurement of the Absorption of, by Liquids, P. Biquard, 375; -Sounds, Action of, on Photographic Plates, N. Marinesco and J. J. Trillat, 667; -Violet: Light: and Fungi, Alice A. Bailey, 404; at British Health Resorts, 91; Rays, Absorption of, by Methane, etc., W. Kemula and S. Mrazek, 106

Unemployment and the Schools, H. N. Penlington, 578 Units: Fundamental, Dimensions of, Prof. H. Stansfield, 59; Prof. F. M. Denton, 585

Universe: Expanding, Sir Arthur Eddington (Review), 637; Expansion of the, Milne's Theory of the, Dr. G. C. McVittie, 533; Through Wonderlands of the, R. K. Golikere (Review), 823

Universities: of the Empire, Year Book of the, 1933. Edited by Sir H. Frank Heath (Review), 710; Overseas, Science and Service in, 377; Review, April, 885

University: College, Cardiff: Bequest by Prof. C. M. Thompson, 372; Metallurgy Building at, 391; Degrees, Spurious, 686; Graduate, Training the, Sir Charles Grant Robertson, 268; Statistics of Great Britain, 770; Teachers, Association of, and Displaced Teachers in German Universities, 920

Unsaturated Fatty Acids, Monolayers of, Rate of Oxida-

tion of, A. H. Hughes and Prof. E. K. Rideal, 446 Upper: Atmospheric Ionisation, Weekly Measurements of, Prof. E. V. Appleton and R. Naismith, 522; Devonian Rocks of the Chudleigh area, South Devon, L. G. Annis, 482

Ur, Ritual at, C. L. Woolley, 356

Uracil, Nucleic Acids and, Irradiation of, F. F. Heyroth and J. R. Loofbourow, 92

Uranium in Spring Waters and Deposits, Detection of, F. Hernegger, 704

Uric Acid, A New Plant Principle, R. Fosse, P. de Graeve, and P. E. Thomas, 179

U.S.A.: American Museum of Natural History, election of F. T. Davison as president and Dr. H. F. Osborn as honorary president, 236; Bureau of Standards, Annual Report of the, 394; Dr. L. J. Briggs appointed director of the, 580; doctorates conferred in, 480; Effect of the Economic Depression on Schools in, 480; Tornadoes in, 481; Forest Products Research in, 199; Legal Medicine in, 722; Local Archæological Observation in the, 868; National Academy of Sciences, award of the Alexander Agassiz medal to Dr. A. Defant, the Public Welfare medal to Dr. W. H. Park, the John J. Carty medal and award to the late Dr. John Joseph Carty, the Henry Draper medal to Dr. V. M. Slipher, and the Mary Clark Thompson medal to Dr. F. A. Bather, 754; elections to membership, 754; Naval Observatory, Publications of the, 917; Secondary Schools in, 104; Service for the Blind in, 906

U.S.S.R., Academy of Sciences, Sir Robert Hadfield, Bt. elected an honorary member of the, 272

Vacuum Technique, Laboratory, Dr. S. H. Piper, 65

Vai Script, A. Klingenheim, 915

Valoniav entricosa, Penetration of m-Bromo-phenol Indophenol and of Guaiacol Indophenol into, Matilda Moldenhauer Brooks, 631

Vanadium in certain Tunicates, Presence of, J. Cantacuzène and A. Tchekirian, 35

Vancouver Island, Relationships of the Mammals of, E. R. Hall, 879

Variety V_n , Concurrent Directions in a, C. Agostinelli, 851 Vegetation, Influence of Man on, Prof. E. J. Salisbury, 919 Vererbungswissenschaft, Handbuch der. Herausgegeben von E. Baur und M. Hartmann. Lief. 15 (Bd. 3, L): Entstehung der Kulturpflanzen. Von E. Schiemann (Review), 604

Vertebrate Skeletons, Making Whole Mounts of, M. Rahimullah and Prof. B. K. Das, 171

Verulamium 1932, Dr. and Mrs. A. E. Mortimer Wheeler, 300

Veterinary: Publications, 615; Research in the Union of South Africa, 905

Vibrating Metal Bars, Viscous Damping of, Prof. K. Sezawa, 803

Vine, Comparative Composition in the, of Homologous Leaves, H. Legatu and L. Maume, 815

Virus: Diseases, 370; and Intracellular Inclusions in Plants, Dr. F. M. L. Sheffield, 325; of Exanthematic Fevers, Passage of the, by the Digestive Canal in the Rat, C. Nicolle, J. Laigret, and P. Giroud, 375

Viscosity: Measurements of Liquids by the Oscillating Disc Method, Prof. C. E. Fawsitt, 97; of a Molten Metal, Measurement of the, by means of an Oscil-

lating Disc, V. H. Stott, 850

Vision and Colour Vision, Dr. R. A. Houstoun (Review), 532 Visual: Acuity, Measurement of, R. J. Lythgoe and Dorothy E. Corkill, 98; Experience, A Peculiar, Dr. F. A. Bather, 62; Sensations, Measurement of, Dr. N. R. Campbell, 850

Vital Capacity and Occupational Characteristics, 588 Vitamin A: Absorption Spectrum of, at Low Temperatures, Dr. F. P. Bowden, S. D. D. Morris, and Dr. C. P. Snow, 582; Highly Active, Characteristics of, Dr. F. H. Carr and W. Jewell, 92; in Cod Liver Oil, An Inhibitor of the Antimony Trichloride Test for, A. Emmerie, 364; B₁, Crystal Structure of, and of Adenine Hydrochloride, J. D. Bernal and D. Crowfoot, 911; B_4 and Adenine, R. D. Heard, H. W. Kinnersley, J. R. O'Brien, Prof. R. A. Peters, and V. Reader, 617; C Activity, Indophenol Reducing Capacity of Lemon Juice and its Fractions in relation to, Dr. S. S. Zilva, 363; and Ascorbic Acid, A. L. Bacharach, 364; Constitution of, Dr. F. Micheel and K. Kraft, 274; E. G. Cox and Dr. E. L. Hirst, 402; Identification of, Prof. A. Szent-Györgyi, 225; in the Adrenal Gland, G. Bourne, 874; Methylnornarcotine, Glycuronic Acid, and, W. J. Dann, 24; Triphenylmethyl Derivative of, L. Vargha, 363; Content of Australian, New Zealand and English Butters, M. E. F.

Crawford, E. O. V. Perry, and Dr. S. S. Zilva, 770 Vitamines, Les, Mme. Lucie Randoin et H. Simonnet

(Review), 258

Vitamins: A Survey of Present Knowledge (Review), 857; (Review), 258; Bibliographical Survey of, 1650-1930, with a Section on Patents, M. H. Wodlinger (Review), 568; Recent Research on the, 118; Research on, A. L. Bacharach (Review), 857

Vivisection of Criminals, 902

Vocational Guidance and the Health of the Industrial

Worker, A. Macrae, 243 Volcanic: and Tertiary Marine Series at Curlewis, near Geelong, the Older, A. Coulson, 71; Craters in the Libyan Desert, Dr. K. S. Sandford, 46; Fumaroles, G. Ponte, 923

Volcanoes, Central African, Dr. J. Parkinson (Review), 821

Voltages, Very High, Dangers of, 833 Vosges, Schistogreywacke Complex of the, Stratigraphic Subdivision of the, G. Dubois, 339

Wales, University of, award of doctorates, 665

Wanderers Wetterbuch: Einführung in das Verständnis der Wettervorgänge, Dr. O. Myrbach (Review), 786

Wapiti Deer in Wyoming, Census of, 516

War, Civilisation and (Review), 635 Water: Flow of, between Moving Boundaries, Cornish, 731; -Fowl and Game-Birds in Captivity: some notes on Habits and Management, A. F. Moody (Review), 316; -Level, Falling, in the Chalk under London, Dewey, and others, 882; Power Development, Canadian, in 1932, Dr. Brysson Cunningham, 788; Thermal Conductivity of, L. H. Martin and K. C. Lang, 813

Watt, James, Memorial Institute, Birmingham, 757

Wave: Equations and the Conservation of Energy. A. Lees, 402; -Lengths of Light and the Fundamental Standards of Length, A New Apparatus for Determining the Relationship between, J. E. Sears and H. Barrell, 192; Mechanics: Elementary Theory, Prof. J. Frenkel (Review), 860

ther: Prophet: Be your Own, a Book for the Holidays and after, E. S. Player (*Review*), 712; Weather: 'Raw', Dr. G. M. B. Dobson; Sir Leonard Hill, 28; Prof. S. Russ, 131; H. E. Beckett, 132; Sir Leonard

Hill, 241; Prof. W. A. Osborne, 515

World, Tabulation of, Weather, E. W. Bliss (5), 35

Webster's Collegiate Dictionary. Fourth edition (Review),

Weddell Sea, Ice in the, J. M. Wordie and Dr. S. Kemp, 916 Weed-Killers, Effect of, on the Soil, W. E. Bowser and J. D. Newton, 880

Weeds, 737

West Regional Broadcasting Station in Great Britain, 664 'Western Indian' Art, 843

Weston Standard Cell, The Temperature Coefficient of the, P. Vigoureux and S. Watts, 374

Whale Shark in the Waters around Ceylon, The, Dr. E. W. Gudger, 165; Dr. J. Pearson, 729

Wheat: Embryos, Nutritive Value of, V. Famiani, 107; -Germ Oil, Absorption Spectrum of the Unsaponi fiable matter from, Dr. R. A. Morton and J. A. Edisbury, 618; of the Vitamin E fraction of, Dr. P.

Bowden and T. Moore, 512; Spring, 284
White Pine Blister Rust, Fusion of Pycniospores with
Filamentous Hyphæ in the Pycnium of the, R. K.

Pierson, 728

Whooping-Cough, Mortality from, Dr. B. Hill, 269 Wicken Fen, The Natural History of. Edited by Prof. J. Stanley Gardiner. Part 6 (Review), 343

Wild Life in India, Preservation of, 779

Wind: Direction in the British Isles since 1341, Variations of, Dr. C. E. P. Brooks and Theresa M. Hunt, 814; Gradients, Steep, in Inversions, Breakdown of, C. S. Durst, 35; Over Different Surfaces, Variations in the Structure of, C. S. Durst, 814

Windle, Sir Bertram: Bertram Coghill Alan Windle, F.R.S.: a Memoir, Dr. Monica Taylor (Review), 307; Prof. W. Wright (Review), 307

Echoes, Recording, at the Transmitting Station, Prof. S. K. Mitra and H. Rakshit; R. A. Watson Watt and L. Bainbridge-Bell, 657; Waves, The Travel of, Sir Frank Smith, 642

Witchcraft in Africa, F. H. Melland, 195

Wood: Daily Shrinkage and Swelling of, M. B. Welch, 667: Moisture Content of, M. B. Welch, 524

Woods Hole Region, Mass., The Copepods of the, Dr. C. B. Wilson, 698

Wool: Fibres, Curls and Twists in, Origin of, Dr. J. E. Nichols, 201; Industries Research Association, Report of the, for 1932–33, 719; Methionine in, J. Barritt, 689

Woolly Hair in a Nordic Pedigree, Dr. O. L. Mohr, 695 Woo's Formula, An Extension of, C. Cannata, 339

World: A Vanished, Dr. A. Ferguson (Review), 75; Crisis and the Gold Standard, C. Lallemand, 70; Economic Conference, The, 866; 889; Order, New, The Cost of a, R. Brightman (*Review*), 183; Power Conference, Forthcoming, 22

Worms, Collecting and Preserving, 90

Xenohelix in the Maryland Miocene, L. Dryden, 775 Xenon, Ultra-Violet Absorption Bands of, Prof. J. C. McLennan and R. Turnbull, 214

X-Radiation, Properties of, Prof. C. G. Barkla, 166 X-Ray: Spectra, Quadrupole Lines in, E. Sègre (2), 411; Tube Current and Voltage, Measurement of, Dr.

G. W. C. Kaye and G. E. Bell, 552 X-Rays: Diffraction of, by Liquid Sulphur, A. H.

Blatchford, 813; Spectroscopic Analysis by (Review),

Yangtse-Kiang Flood of 1931, L. Brandt, 280

Yeast: Growth, Influence of Sodium and Potassium Ions in, Prof. I. Novi, 63; Permeability of, to Methylene Blue, Combined Influence of pH and Glucose on the, M. Junquera, 411

Yugoslavia, American Archæologists in, Dr. V. J. Fewkes,

Yuman: Ethnology, E. W. Gifford, 915; Music, Miss Densmore, 98

Zea Mays, A Correlation of Ring-shaped Chromosomes with Variegation in, Barbara McClintock, 631

Zentral-Asien und Karakorum-Himalaya, Geographische Forschungen im westlichen, Dr. E. Trinkler (Review), 600

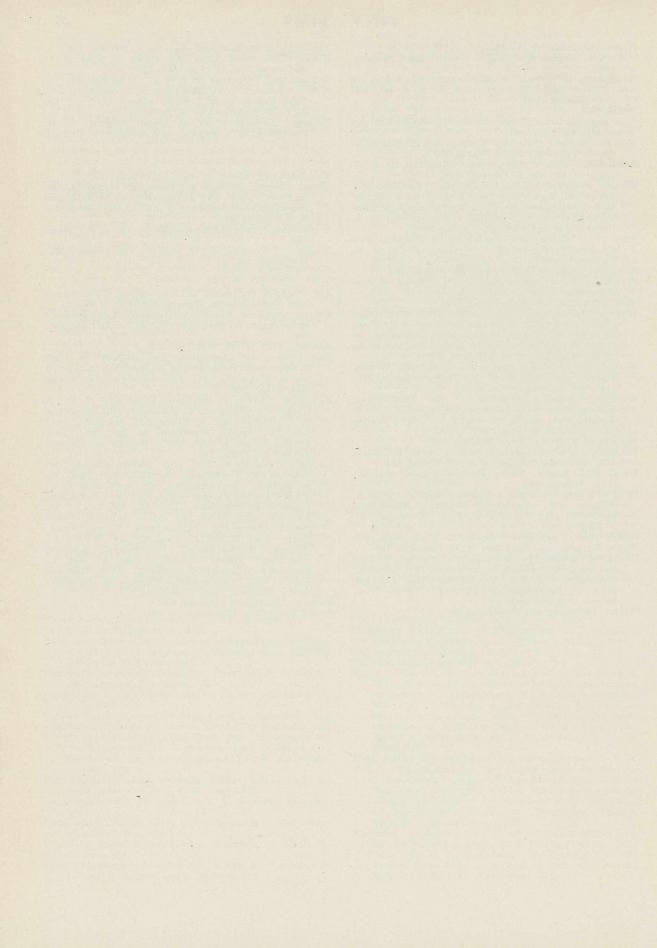
Zeolites, The, M. H. Hey (5), 630

Zinc Oxide: and Anhydrous Zinc Chloride at High Temperature, Behaviour of Mixtures of, A. Ferrari and G. Trampetti, 339; Fluorescence of, Beutel and Kutzelnigg, 917

Zoological: Advice to the State, Importance of, Dr. J. Gray, 88; Expedition to Morocco (1930), Results of a, F. Werner and R. Ebner, (5), Scorpions, 287; Gardens: Indian Rhinoceros at the, 683; New Gorilla House at the, 646; Society of London, Report for 1932, 755

Zoologie: Handbuch der, eine Naturgeschichte der Stämme des Tierreiches, Gegründet von Prof. W. Kükenthal. Herausgegeben von Dr. T. Krumbach, Bd. 2: Vermes Amera, Vermes Polymera, Echiurida, Sipunculida, Priapulida. Lief. 11; Bd. 2, Lief. 12, Lief. 13; Bd. 2, Lief. 14; Bd. 3, Halfte 2: Chelicerata, Pantopoda, Onychophora, Vermes Oligomera. Lief. 1, Teil 3; Lief. 2, Lief. 3; Bd. 6; Acrania (Cephalopoda), Cyclostoma, Ichthya, Amphibia. Hälfte 2, Lief. 1, Lief. 2; Bd. 7; Sauropsida, Allgemeines; Reptilia; Aves. Hälfte 1, Lief. 1; Bd. 7: Sauropsida, Allgemeines; Reptilia; Aves. Hälfte 2, Lief. 6 (Review), 222

Zoology, Descriptive (Review), 222 Zurich University, Centenary of, 684



Supplements should be collated and bound with the numbers with which they were issued.



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"To the solid ground
Of Nature trusts the mind that builds for aye."—WORDSWORTH.

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Official Publications Received

Vol. 131

CONTENTS	PAGE
Progress and Scientific Method	. 1
Properties of the Atom. By Prof. G. Hevesy	. 4
Chromosome Mechanics. By Prof. A. H. Sturtevant	. 5
Modern Physic. By T. A. H	. 6
Short Reviews	. 8
Mount Everest. By Col. H. L. Crosthwait, C.I.E.	. 10
Scientific Centenaries in 1933. By EngCapt. Edgar C. Smith. O.B.E., R.N.	. 14
Empire Broadcasting. By R. L. SR.	. 16
News and Views	. 17
Letters to the Editor:	
Disintegration of Light Elements by Fast Protons.—Dr. J. D. Cockcroft and Dr. E. T. S. Walton	. 23
The Neutron and Neuton, the New Element of Atomic Number Zero.—Prof. William D. Harkins	er . 23
Kinetics of the Decomposition of Molecules of Intermedia Complexity.—C. N. Hinshelwood, F.R.S., and C. J. M. Fletch	te er 24
'Hexuronic Acid' (Ascorbic Acid) as the Antiscorbutic Factor Prof. A. Szent-Györgyi and Prof. W. N. Haworth, F.R.S.	
Methylnornarcotine, Glycuronic Acid, and Vitamin C.—W. Dann	J. 24
Union of Pycniospores and Haploid Hyphæ in <i>Puccinia Helianti</i> Schw.— J. H. Craigie	ii . 25
Breeding of Oysters (O. edulis) at Port Erin.—J. H. Orton, Mis M. W. Parke and W. C. Smith	. 26
Reversible Stoppage of the Blood Circulation in Sabellids.—Pr H. Munro Fox	of. . 26
Photochemical Reaction of Hydrogen and Chlorine.— Prof. H. Baker, C.B.E., F.R.S.	B. 27
Limiting Mobilities of some Monovalent Ions and the Dissociation Constant of Acetic Acid at 25°.—Dr. A. I. Vogel and G. E Jeffery	n L. . 27
'Raw' Weather.—Dr. G. M. B. Dobson, F.R.S.; Sir Leonar Hill, F.R.S.	d . 28
Isotope Effect in the Spectrum of Cadmium Hydride.—Er	ie . 28
Chlorination of Sodium Benzoate.—Dr. J. C. Smith	. 28
Research Items	. 29
Astronomical Topics	. 31
Electricity, Gas and Other Fuels as Heating Agents	. 32
Mesolithic Age in Britain	. 32
University and Educational Intelligence	. 33
Calendar of Nature Topics	. 34
Societies and Academies	35
Tarthanda Variation	0.0

Progress and Scientific Method

AMONG the misfortunes of this present age are the tendencies of the Press to seize on the sensational in the business of science, and of the public to expect such sensational pronouncements. This, equally with the absence of an adequate scientific element in our general educational system, of which the first is indeed a consequence, is mainly responsible for the general failure to understand the scientific spirit and the place of science in our civilisation.

Unquestionably, extravagant claims have been made alike regarding the contribution of science to general welfare and the part which men of science have to play in the State and in the general life of the community, and such claims have done great disservice to the cause of science. The evidence, however, does not indicate that in the main such claims have emanated from scientific workers themselves. They have more frequently been due to the journalistic quest for the sensational, to the quotation of impressive passages in the utterances of a man of science apart from the qualifying context. Of this danger the treatment recently accorded to a scholarly address by Prof. A. Findlay to a joint meeting of the Manchester chemical societies is a pertinent illustra-Under the headlines of "A Rebuke to Scientists" it is hard to recognise the thoughtprovoking address of Prof. Findlay with its wide sweep and philosophic temper, recalling at times the presidential address of General Smuts to the British Association in 1931. Prof. Findlay stressed the idealistic value of science rather than its material results and suggested that the real claim of science to a wider recognition and cultivation in the community is based on such values. The spirit of science, as expressed in the quest for truth, is an essential element complementary to beauty and goodness in the perfecting of humanity, and its inculcation should form an essential element in our system of education.

This plea for science as an essential element of culture and not as part of a technical training was, however, ignored by those who fastened on his refutation of the claims of scientific workers to a privileged position in national administration. Unfortunately, Prof. Findlay himself made one or two incautious remarks which have tended to offset his admirable presentation of a point of view which scientific workers themselves have often been prone to overlook, owing to the widespread neglect of the teaching of the philosophy of science or the scientific outlook in the universities. Few scientific workers with any experience of industry would, for example, care to endorse the statement that, so far as industry is concerned, the battle for scientific control is already won. The position of the Cotton Industry Research Association in the very city in which the lecture was delivered should have been sufficient reminder of how very far we are yet from a general acceptance by industry of the importance of research as an essential element in industrial efficiency and advance. Similarly, so far as national administration is concerned, without pointing to a host of matters in which the neglect of technical factors and scientific opinion has been conspicuous, it is sufficient to recall that the report of the Bridgeman Committee on the Post Office issued recently was the first public inquiry committee to give full recognition to the scientific expert in the Civil Service scheme of things. Only in this report was it recognised officially that the scientific expert is entitled to participate on level terms with the administrative expert and the financial expert in the formulation of policy.

If the principle has been conceded, the application here and elsewhere has yet to be worked out, and those who most welcome Prof. Findlay's insistence on the cultural value of science will most regret his untoward suggestions regarding the capability of men of science for administrative work or the extent to which they have been accepted in such positions. While Prof. Findlay's protest against the mechanisation of industry and his warning against the dangers of materialism

and an excessive emphasis on mechanical efficiency will be warmly welcomed, the influence of science on industry has scarcely been either so materialistic or so mechanising as he implies. On the contrary, the application of scientific methods of management characteristic of large-scale industry pays an attention to the human factor which was formerly rarely encountered. The results which have attended the investigations of the National Institute of Industrial Psychology or of the Industrial Health Research Board already attest the humanising effect which applied science can exert on industrial conditions. Moreover, the increasing attention which is being directed to the theory of management has already emphasised the importance to efficient administration of encouraging the development of individual responsibility, the expression of personality and the organisation of the team spirit.

It is indeed here that Prof. Findlay's address is most open to criticism. So far from human nature being intractable to the scientific method, the evidence suggests that one of the most urgent needs of to-day is an adequate attack on the social sciences and a determined attempt to place sociology, economics, psychology on a really scientific basis. Much patient research in anthropology and human biology will be required before this can be achieved, and it is highly probable that the most important contributions of science to human welfare during the next few decades will come from such fields as these, which enable us to understand and interpret more accurately the human factor and our social, economic and industrial problems.

Apart altogether from these considerations, Prof. Findlay's dislike for extravagance has led him to overlook the essential challenge presented by such spokesmen as Sir Alfred Ewing and Prof. Miles Walker. The failure of our present administrative methods to which our economic and international difficulties bear witness, and of which the bankruptcy of the Disarmament Conference is a recent example, had led many others besides men of science to wonder whether methods in which action is determined less by political prejudice than by dispassionate analysis of the ascertained facts may not offer better hope of success. The disposition to accept the leadership of science is prompted by a growing realisation of the extent to which scientific and technical factors enter into all our problems in this scientific age and that, if such factors are

only one aspect of our problems, they present an aspect which demands accurate assessment and intelligent action if disaster is to be avoided.

This point of view does not suggest that numerous other considerations are not involved in an administrative decision. It does, however, contend that the scientific worker who possesses administrative ability as well as the capacity to assess the technical facts is at least as likely to arrive at a sound decision, as those without such scientific knowledge and training, and that the openmindedness and willingness to face change associated with the scientific outlook is an important and hopeful element at the present time. The situation is so serious that accepted doctrines in finance, economics, politics are being questioned and must stand their trial as every scientific theory has to do in the advance of science.

Scientific management stands essentially for the substitution of exact scientific knowledge for opinions or rule of thumb methods, and not for the disregard of essential elements in a problem, whether those elements are or are not what is usually described as technical.

Even in industry, such exact scientific know-ledge involves to an increasing extent a re-examination of the whole of our economic and indeed of our political machinery and ideas—a re-examination conducted much in the spirit of an engineer faced with a problem in factory or office management and concerned solely with the issue of improving those arrangements so as to secure a higher standard of living for the community. The world believes that the advancement of objective and exact knowledge is inevitable and ultimately beneficial and that the problems of distribution and finance, for example, so long treated in haphazard and fatalistic fashion, will yield to exact knowledge.

It can accordingly scarcely be questioned that in the modern State the scientific expert will occupy a position of increasing responsibility, and the security of our civilisation largely depends upon the efficiency and expedition with which methods are worked out for permitting his effective participation in public as in industrial affairs. Possibly the solution may be found along lines visualised by General Smuts, and the nations learn to look to the organised system of the expert report and accept as a matter of course, just as judicial decisions are accepted, the authoritative and impartial lead which it gives.

An important factor, however, is bound to be

the general outlook of the scientific worker himself. His specialist knowledge of one particular field must be joined to a sense of values, a wide outlook and a public spirit which induce him to place his services to a greater extent at the disposal of society. In the leisure State there should be no room for the excessive specialisation and narrowness of outlook deplored by Prof. Findlay, and it should be possible so to re-organise our training that the cultural value of scientific thought and method is imparted to the student, and the tendency of scientific workers to regard only their special field from a scientific point of view counteracted. Only as scientific workers are prepared to regard the whole domain of life from a scientific point of view and to support as a body those who at particular points are seeking to extend the application of scientific method and the scientific spirit, substituting facts for prejudice, opinion or guesswork in determining action, can we hope for decisive advance. Both the organisation and the functioning of the State must become more scientific, impartial, businesslike and less purely political, and the scientific expert is unlikely to receive his fitting place in the new order without organised support from his fellows as well as keen and intelligent criticism.

Prof. Findlay has done well to remind us that science is more than a body of results and truths. Great as are the benefits which mankind has derived from scientific discoveries, from the new knowledge which science has made available, there are greater services to be rendered. Science has yet to teach man how to use that knowledge, to impart generally that attitude, outlook or method of acquiring knowledge which is essential if mankind, individually or collectively, is to adapt himself to the new conditions. Science alone, indeed, will not save mankind from disaster, invaluable as can be its contribution to the fundamental thinking required. But the balanced wisdom and judgment so imperatively demanded to-day can only be attained as science is regarded not as a storehouse of facts to be consulted from time to time but as one of the great human endowments, to be ranked with art and religion, and the guide and expression of man's fearless quest for truth. This outlook can only receive adequate recognition as more and more in their own personality scientific workers exemplify the unifying power of the spirit of science and its contribution to the ordering of every sphere of conduct on a basis of truth and not prejudice.

Properties of the Atom

The Interpretation of the Atom. By Prof. Frederick Soddy. Pp. xviii+355+20 plates. (London: John Murray, 1932.) 21s. net.

PEW discoveries, if any, have influenced so rapidly and deeply physics, chemistry and many kindred sciences as that of the spontaneous disintegration of radioactive elements. The development following this discovery attracted also the lay mind, which was struck by the idea of evolution of inorganic matter and by the farreaching conclusions in the field of geology and cosmology, to which the application of the science of radioactivity led. The layman's interest was enhanced by the spectacular sides like that of electroscopic discharge at the approach of the experimenter carrying radium in his pocket or suitcase, by the radium clock, and, by no means least, by the possibility of the medical application of radium rays against both tedious and fatal diseases.

The general reader interested in the great advances in our knowledge of the nature of matter and its manifold applications has every reason to be highly indebted to Prof. F. Soddy for his "Interpretation of Radium", first published in 1909. It is a most instructive task to compare the above mentioned book with "The Interpretation of the Atom", prepared by Prof. Soddy to take the place of the first mentioned volume, which has long been out of print. Even the reader familiar with atomic physics and chemistry will be struck when reading this account of the unparalleled progress made by these sciences in the last three decades. Part I deals fully and systematically with the remarkable series of spontaneous disintegration of the atom, while in Part II the general progress of atomic chemistry is discussed. A chapter on cognate geological and astronomical aspects and the cosmic radiation concludes the volume.

Prof. Soddy is known as a thinker and writer of unusual originality, who does not hesitate to oppose himself, even to ideas accepted by most of his contemporaries—a characteristic quality which, coupled with genius and lucidity, is undoubtedly responsible for his greatest achievement, the introduction of the conception of isotopes. The reader will therefore not be astonished to find Prof. Soddy in pronounced opposition to the mathematical presentation of physical theories in the age of great successes of

quantum mechanics, and to learn that he is not among those who can "bow down and worship the square root of minus one". "The Interpretation of the Atom" is primarily intended for the general reader, but it will undoubtedly prove to be a helpful guide to scientific students of all grades; while the man of science interested in the history of this subject will also find matter of interest to him.

The development of scientific thought and progress often follows roundabout paths, which even the onlooker has difficulty in following; not a few scientific workers are inclined, therefore, to the view that the official history of the development of scientific thought not seldom differs materially from the actual course of events. Two examples may be cited, Bohr's fundamental conception of the atomic structure, and the development of the notion of atomic numbers, both outstanding events in natural philosophy, and the development of which are usually described in a way which differs materially from the actual facts. We mostly read that Bohr's conception was due to the effort to explain the appearance of spectral series. It would go too far to pretend that Bohr (who not long before passed his examination with first class honours) would not know of the existence of spectral series, but these were certainly not in the forefront of his thoughts when he first embarked on the philosophy of the atom. It was only at a later date, that in discussing experimental results obtained by one of his Copenhagen colleagues, he noticed the possibility of applying his new conception to explain the appearance of spectral series.

The actual development was as follows. Bohr was deeply struck by Rutherford's conception of the nuclear atom, realising simultaneously, however, the impossibility of reconciling this conception with the requirements of the classical theory of radiation. The most direct step would have been to discard Rutherford's conception; Bohr chose another path and introduced Planck's quantum idea in place of the classical conception, when interpreting the stability of the nuclear atom. By doing so he was led at once to the fundamental distinction between the nuclear and electronic properties of the atom, to the identity of the atomic number and nuclear charge and also to the conclusion that the emission of the disintegration \(\beta\)-particles has its origin in the nucleus. The latter led him to the conclusion that, for example, when an atom of thorium loses consecutively one a- and two \beta-particles, the thorium atom reappears, and he arrived thus at the displacement laws, as a special case of a much wider generalisation. Bohr discussed this result widely without publishing it. These laws were, as is well known, published later by different men of science as an empirical summary of the experimental results of Fleck and others. That the sequence of the elements in their natural system, that is, the atomic number, is identical with the number of the nuclear charge, was published by Bohr at an early date when basing his first calculations on the assumption that the hydrogen nucleus has one, the helium nucleus two, elementary charges, and so on.

The above narrative of the introduction of the conception of the atomic number would, however, not be complete without mentioning that this conception emerged, simultaneously with Bohr's deductive line of thought, in the Manchester physics laboratory on inductive lines. Geiger and Marsden checked completely Rutherford's theory of large scattering of α-particles and found the nuclear charge of gold to be about 100. result was not at all unexpected, as early theories of α-ray scattering and stopping had suggested already that there should be half as many electrons in the atom as its weight. When Moseley started his work which became the foundation of a new and important branch of physics, the conception of atomic number was thus clearly developed. We owe to Moseley's genius a simple method of determining the atomic number of each element and of fixing exactly the number of elements between hydrogen and uranium. From the fact that the nuclear charge of gold was found to be 100, and also from the chemical evidence, it followed already that the number of unknown elements could only be a very restricted one. The actual number was, however, fixed by Moseley.

The introduction of the conception of the atomic number impressed contemporary scientific workers so deeply that its rival conception, that of atomic mass, seemed for a while to be deposed from its long reign. Through Aston's work and the development of later years, contemporary interest is again focused on atomic mass and thus equal rights are secured for these two fundamental properties of the atom.

Every reader interested in the problems and the progress of natural science will find Prof. Soddy's book most fascinating reading. G. Hevesy.

Chromosome Mechanics

(1) Recent Advances in Cytology. By Dr. C. D. Darlington. Pp. xviii + 559 + 8 plates. (London: J. and A. Churchill, 1932.) 18s.

(2) Chromosomes and Plant-breeding. By Dr. C. D. Darlington. Pp. xiv + 112. (London: Macmillan and Co., Ltd., 1932.) 7s. 6d. net.

THE discovery that the chromosomes are the bearers of the hereditary units (genes) has led to a renewed interest in their study by cytological methods. There has developed a vast literature, much of it based on correlated studies of cytology and experimental genetics. Dr. Darlington has worked through this literature, and has attempted in the larger of the two works under notice not only to describe the more important recent results but also to provide a unified general scheme of chromosome behaviour.

This general scheme is based on several fundamental hypotheses. The first of these is that like chromosomes (or parts of chromosomes) are regularly attracted in pairs, and that the attractive force is satisfied when two like parts come together, so that more than two are not attracted. This leads directly to an interpretation of the difference between mitosis and meiosis. At mitosis each chromosome is already split into two daughter chromatids at prophase. These lie side by side and thus satisfy the attractive force. At meiosis the division is delayed, and attraction therefore causes homologous chromosomes to conjugate. Division now follows, and there result four chromatids instead of two chromosomes; these are no longer attracted except in pairs, and at each level there results a separation into two groups of two chromatids each. Observation shows, however, that this separation does not occur between the same pairs of strands at each level, so that there are chiasmata, or crossed strands, at certain points. Darlington argues that it is these chiasmata that maintain the unity of the quadruple structure until metaphase, and thus insure regular segregation. In the absence of chiasmata the chromosomes merely fall apart into two separate pairs, and these segregate at random at the first division.

According to Darlington, each chiasma results from a crossing over. The initial separation of two chromatids from two others is reductional at every level, that is, the two chromatids derived by division of a single chromosome always remain associated. If two that are not sisters have

undergone mutual exchange at a given level, there results a chiasma at that level.

These are Darlington's main theses; meiosis is mitosis with delayed division of the chromosomes, chiasmata result from crossing over, and metaphase pairing is due to chiasmata that prevent separation of divided homologues. He interprets the whole range of chromosome cytology and genetics on the basis of these views. There is a logical and detailed marshalling of an almost overwhelming body of evidence, much of it based on Darlington's own novel and extensive investigations. This evidence relates not only to careful analysis of the normal behaviour of a very wide variety of animals and plants, but also more especially to unusual cases—polyploids, extra-chromosome types, hybrids, and mutant differences in chromosome behaviour.

The reviewer finds himself strongly inclined in favour of two of the main hypotheses-those concerning the nature of meiosis and the effect of the chiasmata in holding homologues together until metaphase. The first of these is contrary to the observations of some cytologists, who have reported that the chromosomes divide before the meiotic conjugation; but Darlington's view has an attractive logical simplicity that makes it difficult to think of the phenomena in any other terms. Acceptance of these hypotheses involves also acceptance of the more fundamental hypothesis of attraction by twos and not by larger numbers. But it does not necessitate acceptance of Darlington's interpretation of the origin of chiasmata. This hypothesis leads to certain contradictions with the genetic data, which to the reviewer seem fatal to the hypothesis. The occurrence of regular pairing in the male Drosophila, in which there is no crossing over, makes it clear that chiasmata are not necessarily associated with crossing over—a conclusion avoided by Darlington only by the expedient of very special and improbable accessory hypotheses. It follows that chiasmata are not due to crossing over, and one may turn to the alternative view, referred to by Darlington as the "classical hypothesis". On this interpretation the initial separation is not always reductional, that is, at certain levels sister chromatids separate. If the separation is in different planes at successive levels, a chiasma will result between these levels. If the initial separations are permanent, crossing over will result (after, not before, the chiasma is formed), but it is possible that, at times, the original separation plane may be only temporary, the chiasmata simply unravelling without leading to crossing over. On such a basis it appears to be possible to give a self-consistent account of the genetic and cytological phenomena.

Evolutionary discussions occupy much space in Darlington's book. His view is that chromosomes are especially favourable for the construction of evolutionary schemes, because their materials control the gross structure of the organism but are not themselves controlled by the rest of the organism, and for several other reasons. discussion of the evolution of chromosomes and chromosome systems is interesting and at times stimulating; but to one who has studied the period in the history of zoology when the making of hypothetical phylogenies was popular, there is a familiar sound to the argument. many who will hope that we are not to see a revival of the fashion of constructing elaborate hypothetical histories that must for ever remain hypothetical.

This is a difficult book. It is closely argued, and so full of facts and ideas as to require slow and careful reading. It will be indispensable for the student of cytology or genetics, but the more general reader will find the smaller book easier to assimilate. This, as its title indicates, is written for the plant breeder; but anyone interested in the new results obtained by the cytological study of polyploid and hybrid plants will find here an admirable and readable summary.

A. H. STURTEVANT.

Modern Physic

- (1) The British Pharmacopæia, 1932. Published under the direction of the General Council of Medical Education and Registration of the United Kingdom. Pp. 1 + 713. (London: Published for the General Medical Council by Constable and Co., Ltd., 1932.) 18s. 6d. net.
- (2) The Extra Pharmacopæia of Martindale and Westcott. Revised by Dr. W. Harrison Martindale. Twentieth edition. In 2 vols. Vol. 1. Pp. xlviii + 1216. (London: H. K. Lewis and Co., Ltd., 1932.) 27s. 6d. net.
- (1) THE duty of preparing "The British Pharmacopœia" was laid upon the General Medical Council by the Medical Act of 1858 and the Medical Council Act of 1862. Since then five pharmacopæias have been issued at irregular intervals varying from three to eighteen years. The present issue is the sixth and in future

revision will take place every ten years. It is explained in the preface that the "Pharmacopœia" includes only the more important "standard articles, which are in use throughout the Empire" and with characteristic British elasticity in governance, provision is made for the issue of supplements or addenda by Governments of overseas parts of the Empire, who may desire to sanction the use of drugs with a local reputation. This is a happy solution of a long-standing difficulty.

The Commission upon which the actual work of preparing the "Pharmacopœia" devolved, was assisted in its labours by a group of sub-committees on clinical medicine, pharmacy, pharmacology, pharmacognosy and pharmaceutical chemistry, whilst the decision for or against biological standards for ergot, cod liver oil, digitalis and strophanthus, arsenobenzene derivatives and similar products, chemical control of which is still doubtful or impossible, necessitated the appointment of eight subsidiary committees, one for each subject of potential biological standardisation.

The revision of a pharmacopæia involves three branches of work, the elimination of drugs which have become redundant or obsolete, the selection of such recent contributions to materia medica as have found a sure footing in clinical medicine, and the definition of each item in the final list. In the new "Pharmacopœia" the omissions are far more numerous than the additions. They are chiefly galenical preparations, some of which are replaced by more modern examples of the pharmacist's art, some crude vegetable drugs, a number of compounded medicines and a few synthetic, pharmaceutical chemicals. No serious objection can be raised to any of these omissions, especially as the articles concerned will still figure in the British Pharmaceutical Codex, the revision of which is now in progress. Indeed this process of elimination might have been carried further. Eucalyptus oil (cineole, 70 or more per cent), cajuput oil (cineole, 50 to 60 per cent) and eucalyptol (cineole, 97.5 per cent) are presumably all included for the sake of their cineole content, this being the component indicated by the prescribed assay, and if so, two of them are redundant.

The additions include several well-known alkaloids, quinidine, ephedrine, ergotoxine and emetine, which have found new and important medicinal applications since 1914, a number of synthetic drugs, including a choice of local anæsthetics, and

hypnotics and the indispensable antisyphilitics, neoarsphenamine and sulpharsphenamine. most notable addition is the group of biological products including insulin, liver extract, Schick control, Schick test toxin, diphtheria prophylactic, antityphoid-paratyphoid vaccine, gas-gangrene antitoxin, and antidysentery serum. Apart from this biological innovation, it cannot be said that any striking change has been made in the list of officially recognised drugs. Some notable candidates for recognition have probably been omitted because they are patented products, others because their present reputation may be based on pioneering enthusiasm and they have still to run the gauntlet of extended clinical trial, which will eventually bring them to their proper places in practical therapeutics.

The list of official drugs, which emerges by the operation of these two kinds of selection, is essentially a matter for comment by clinicians, but if a chemist may venture two opinions upon it they are, that the list is what a list of officially-recognised drugs should be—safe, sound, and not too conservative—and that it is satisfactory because due weight has been given to the pharmacological and chemical evidence for and against each drug.

It is in the third branch of its labours, the definition of the selected products, that the Commission has most clearly improved upon preceding pharmacopæias. Much more attention is given to tests for identity and for degree of purity, to the analysis of chemicals and to the assay of potent galenical and biological preparations. In the 1914 Pharmacopæia the appendices amounted to over eighty pages: in the new issue nearly twice that space is occupied, due mainly to the inclusion of more descriptions of biological, chemical and physical methods for the determination of the quality of drugs, using that term in a broad sense. These appendices now form a useful and interesting manual of methods of assay, which may even find a place for the Pharmacopæia in laboratories not directly concerned with medicinal products.

Daily use of the sixth" British Pharmacopœia" will inevitably reveal minor weaknesses, but taken as a whole it is a satisfactory and in many ways a model contribution to the world's collection of national pharmacopœias. The nation has every reason to be grateful for the labours of the Pharmacopœia Commission, for the time and knowledge which the committees of experts have

expended and for the accumulated results of technical experience which manufacturers have given just as freely. The least it can do in return is to provide for that continuity of investigation which will lighten the work of preparing the Pharmacopæias of the future.

(2) When a technical work has reached its twentieth edition in the course of less than half a century, it may be assumed that the author knows much better than any merely ephemeral reviewer, what his particular public requires and in what form to give it to them. This is the case of the "Extra Pharmacopæia". The book is fittingly noticed here because it provides the

medical practitioner with the information he needs about the latest developments in therapeutics and thus facilitates that extensive clinical experimentation with new materia medica, the results of which alone can finally determine their real value. It thus serves the cause of medical progress and at the same time admirably fulfils its main purpose as a handy and singularly complete reference book for the multifarious needs of the pharmacist and the medical man. In these respects the twentieth edition is as good and as up to date as its predecessors were in their day, and those who know the "Extra Pharmacopœia" need no further commendation. T. A. H.

Short Reviews

The Outline Series. The Physical Nature of the Universe. By J. W. N. Sullivan. Pp. 143. Theories and Forms of Political Organisation. By G. D. H. Cole. Pp. 160. Modern Theories and Forms of Industrial Organisation. By G. D. H. Cole. Pp. 159. An Introduction to the Study of Sex. By Prof. F. A. E. Crew. Pp. 160. An Introduction to Psychology. By Prof. F. Aveling. Pp. 176. An Introduction to Finance. By Prof. T. E. Gregory. Pp. 144. An Introduction to Economics. By Maurice Dobb. Pp. 143. The Arts of Painting and Sculpture. By Roger Fry. Pp. 160. An Introduction to Psycho-Analysis. By Prof. J. C. Flügel. Pp. 159. The Theory and Practice of Architecture. By Prof. C. H. Reilly. Pp. 144. Principles of Literary Criticism. By Prof. Lascelles Abercrombie. Pp. 160. (London: Victor Gollancz, Ltd., 1932.) 1s. 6d. each.

ONE of the more hopeful signs, indicating the approach of an age in which knowledge and power are more closely and harmoniously related, is the reception afforded to the various summaries or outlines of modern knowledge which have appeared in recent years. The publication in separate parts of eleven of the essays by acknowledged authorities which formerly appeared as a composite book under the title "An Outline of Modern Knowledge" is evidence of this demand and the publishers are to be congratulated on the step they have taken in place of issuing a new edition of the composite work.

The publication as individual essays is free from several objections which may be brought against the combined treatise. The essays can be seen as the introductory monographs which they really are and the impression of a reference book is destroyed. The lack of proportion in the original book is less conspicuous in the separated monographs and could indeed easily be remedied by the publication of supplementary volumes to fill the more serious gaps, so far as they are not found in the reprinted monographs themselves.

Handbuch der Pflanzenanalyse. Herausgegeben von G. Klein. Band 2: Spezielle Analyse. Teil 1: Anorganische Stoffe; Organische Stoffe, I. Pp. xi + 973. (Wien: Julius Springer, 1932.) 99 gold marks.

A knowledge of the constituents of plants is becoming both of increasing importance and difficulty, as their number continues to grow, and it is, therefore, of value to have a work of reference which, on one hand, lists the various products according to some logical scheme, whilst on the other, it describes the methods for their identification and analysis. Working in conjunction with the index of nearly ninety pages, it is possible rapidly to have the requisite information about any desired compound.

The book is necessarily the result of collaboration, no less than twenty-three different authorities being responsible for the several sections. The first quarter of it deals with the inorganic constituents of plants including the nitrogen compounds, also the analysis of plant ash and gas: it is done with great attention to detail but in a suitably concentrated form. Therefollows the organic section divided up into groups of allied substances as is customary: it is this part which will be found of immense value to all active workers in the domain

of plant chemistry.

The matter is highly condensed but contains just the information which is usually needed, and the text is not overburdened with countless references to the original literature, which the seeker must find in certain other works indicated, should he require them. Taking the section headed "Phenols" for example: the systematic occurrence in indicated plants is given for 61 phenols, 10 quinones, 11 anthraquinones. The general properties and colour reactions of phenols are described with a note as to the properties and derivatives of each individual phenol, the whole giving the complete state of knowledge in this particular field, which it would otherwise take a great deal of reading to survey.

Climate: a Handbook for Business Men, Students and Travellers. By Dr. C. E. P. Brooks. Second edition, revised. Pp. 199. (London: Ernest Benn, Ltd., 1932.) 10s. 6d. net.

An early second edition testifies to the general utility of Dr. Brooks's handbook. All parts of the globe are dealt with, and a useful feature at the end of each regional division is a table furnishing for a number of selected places data of temperature, humidity, rain, snow and thunder. There are very few illustrations, and, perhaps, it is a recommendation that the book is not encumbered by a lot of commonplace maps and diagrams.

The treatment is essentially statistical, but Dr. Brooks knows his subject too well to allow his outlook to be blinded by the data which he This is an important matter because climatology as a science has suffered badly from the mishandling or misinterpretation of the very

data on which it largely depends.

The book is, however, not so free from loose statements as is to be expected in a second edition. For example, a statement, to the effect that it is the relative humidity which gives the 'feel' of the air, seems to us to be faulty. of the atmosphere is an intricate complex of temperature, wind, radiation, moisture and probably, also, obscure influences like electricity; but, so far as humidity is concerned, it is surely the absolute humidity or vapour pressure as controlling the rate of evaporation from the body which is the important factor, not the relative humidity except at saturation point with fog. Thus on an enervating summer day of high vapour pressure the relative humidity is often much lower than on a bracing winter day of low vapour L. C. W. B. pressure.

Faith, Hope and Charity in Primitive Religion. By R. R. Marett. Pp. vii+181. (Oxford: Clarendon Press; London: Oxford University Press, 1932.) 10s. net.

Dr. Marett's Gifford lectures for 1931-32, an expanded form of lectures delivered under the auspices of the Lowell Institute of Boston in the previous year, have as their theme the evaluation of the religious experiences of peoples of the lower culture, or as Dr. Marett prefers to call them, savages. It must not be thought, however, that Dr. Marett would regard anthropology as one of the normative sciences and that he would attempt to apply an ethical scale to primitive ideas of His evaluation is biological in the behaviour. sense that its aim is to test survival value. The various activities of the savage are passed in review one by one and analysed with the view of the isolation of their emotional content—savage religion being a matter of the emotions rather than of intellect or of action. Dr. Marett then proceeds to show that the religious emotions which colour the whole range of savage activity on the whole make for the virtues or qualities which he designates "Faith, Hope and Charity" and regards

as the effective element in the contribution of religion to the advancement and survival of man. A brief summary does less than justice to the acuity of Dr. Marett's vision—it may be suspected that at times he finds his material a thought intractable. Nor is it possible to do more than refer to the insight shown in the many valuable suggestions on controversial points which he throws off, almost casually, in the course of his argument.

Thermionic Vacuum Tubes and their Applications. By Prof. E. V. Appleton. (Methuen's Monographs on Physical Subjects.) Pp. vii+117. (London: Methuen and Co., Ltd., 1932.) 3s. net.

THE student with a good knowledge of general physics will find this volume very helpful in studying radio-frequency phenomena. Modern thermionic tubes have so many important applications in physics and electrical communication that a knowledge of their action and how they are constructed is essential to almost every research physicist and electrical engineer. The author writes carefully and clearly, so the ordinary reader easily grasps the laws which govern the emission of electricity from hot bodies and how to apply Richardson's formulæ to make calculations. He then explains the internal action of the two electrode tube (diode) and states some of its applications. Finally he discusses the three electrode tube (triode) and describes its applications as an amplifier, a rectifier and an oscillation generator. The book can be recommended to the experienced amateur in radio communication as a scientific introduction to the subject as well as to the physicist and mathematician who intend to read the relevant parts of advanced treatises well before following up some line of research.

The Practical Treatment of Diabetes. By Dr. T. Izod Bennett. Pp. ix+107. (London: Constable and Co., Ltd., 1931.) 6s. net.

This book is a brief and severely practical account of the modern method of treating the diabetic. References to theory are few and are limited to such details as are absolutely necessary to explain the therapeutic steps. A short chapter is devoted to diagnosis, the second and third to general principles of treatment and dieting, the fourth to the use of insulin, and the remainder of the book to complications and special problems and considerations. As described by Dr. Izod Bennett, the restoration of a diabetic to such condition that he is sugar-free on sufficient diet appears an easy matter, and the general practitioner who has not had to deal with many cases will learn with surprise that this object can be attained almost invariably within three weeks. How it can be done is clearly set out in some fifty pages of this book, which, as a practical guide, lacks no essential features. Specimen diets during treatment, and food tables showing the great variety finally available, are included.

Mount Everest

By Col. H. L. CROSTHWAIT, C.I.E.

MOUNT EVEREST, everyone knows, is the highest mountain in the world. It was discovered, and its height determined, during the operations of the Great Trigonometrical Survey of India in the course of carrying out the geodetic triangulation of that country in the years 1849-50. The figure adopted, namely, 29,002 ft. above mean sea level, was derived from the mean of a large number of vertical angles observed to the peak from six different stations situated in the plains of India south of Nepal. These stations were at distances varying from 108 to 118 miles. It was not until some months afterwards, when the necessary computations had been completed, that the great height of Everest was first realised. The actual discovery was made in the computing office at Dehra Dun.

The determination of height above mean sea level by the method of vertical angles, observed from one station only, involves the assumption of a coefficient of refraction—always a doubtful quantity, especially where great differences of height are concerned as in the present case—and also a knowledge of the amount of separation of the geoid and the sphereoid. Suffice it to say, however, that the subsequent extension of observations to the mountain so as to include stations situated within the hills, and the adoption of a more probable coefficient of refraction, made the summit some 29,149 ft., but the estimated separation of the geoid and sphereoid, due to deformation caused by local attraction, would reduce this amount, the net result being to assign to the peak a probable height of about 29,050 ft. above mean sea level. There are also certain other corrections which can be estimated theoretically but these are not taken into account in ordinary trigonometrical levelling.

By height above mean sea level is implied the distance of the summit above the geoid immediately beneath it. Accepting this definition, it is not possible to determine accurately the height when access cannot be had to the summit for purposes of observation. Taking into account that there are still doubtful quantities involved, it has been considered best to allow the time honoured figure of 29,002 ft. to stand, anyhow until sufficient data are available to enable a more exact solution of the problem.

It may be asked why so little was known about the highest mountain in the world beyond its position and height, until comparatively recently, considering it was discovered eighty-two years ago. This was, without doubt, due to its great distance from civilisation and above all to the strict political isolation maintained by the two countries, Nepal and Tibet, on the border of which it is situated. The shortest way, some 110 miles, to reach the mountain from India

would be through Nepal, but even if the Nepalese Government were willing to permit the passage of its country, the route would be through trackless leach-infested jungles impossible for pack transport. Added to this, the snow line is about 2,000 ft. lower on the south side than on the north, for it is subject to the full force of the monsoon and is probably more deeply eroded and, in consequence, more inaccessible than from the Tibet side. For these reasons successive expeditions have taken the longer route, about 350 miles from Darjeeling via the Chumbi valley, Kampa Dzong and Sheka Dzong, made possible since the Tibetan objection to traversing its territory has been overcome.

This route possessed the advantage of passing through country where pack transport was available, and could be used up to the base camp. Further, it lies at an average elevation of about 14,500 ft., so the long march has a certain beneficial effect in acclimatising the party before

the base of operations is reached.

Though the project to climb Mount Everest had been long in the minds of several mountaineers, the first definite move in the matter appears to have been due to Major Rawling and Captain Noel, about the year 1912-13. Both these officers had had experience of Tibetan travel. Then followed the years of the War in which Rawling was killed. The project was revived in 1920, when permission was obtained from the Tibetan Government to approach the mountain through its country. A Committee, consisting of three members each of the Royal Geographical Society and the Alpine Club, was formed to collect funds, select personnel, purchase instruments and equipment and generally to manage the business of an expedition to Everest. The chairman of this Committee was Sir Francis Younghusband.

There have now been three expeditions including the one in 1921, which was of the nature of a reconnaissance. The other two were serious attempts to reach the summit—they were undertaken in the years 1922 and 1924. As already announced, a fourth expedition will leave England

early this year.

The expedition of 1921, led by Col. C. K. Howard-Bury, consisted of Dr. A. M. Kellas, C. Raeburn, G. H. L. Mallory, G. H. Bullock, mountaineers, and A. F. R. Wollaston, doctor and naturalist. Since the local topography of the mountain and its neighbourhood was quite unknown, Majors Morshead and Wheeler, of the Survey of India, with three Indian surveyors, and Dr. Heron, of the Indian Geological Survey, joined the expedition in India. Unfortunately, that veteran mountaineer, Dr. Kellas, died at Kampa Dzong on the way out.

The party started from Darjeeling on May 18 and after the long detour through Tibet, already mentioned, reached Tingri, a place about 45 miles north-north-west of Everest, a month later. On June 23, Mallory and Bullock began the exploration of the western and northern sections of

Everest. Approach was soon found to be, if not impossible at least very difficult, from the west and north-west. They then turned their attention to the great Rongbuk Glacier which drains the northern slopes of the mountain. It was seen to consist of at least two important branches—one trending away to the west, and the main glacier coming down from the northern face of Everest itself.

itself. As the result of these explorations the following facts were established, that the key of the mountain appeared to be an important saddle, the Chang La (North Col), 22,990 ft., situated at the head of the main Rongbuk Glacier, about 13 miles from the summit of Everest, and that it could only be reached with the greatest difficulty from the main glacier. It then became necessary to seek another way of approach to the Chang La. They decided to try what could be done from an easterly direction. They therefore marched round to a new base which, in the meantime, Howard-Bury had established in the Kharta Valley, at the head of which lies the Hlakpa La (22,200 ft.). This point they eventually reached in the face of very adverse weather. It was found to lead to a hitherto unsuspected eastern branch of the Rongbuk glacier, draining into the main valley through a narrow gorge, which had been missed during the first reconnaissance. Mallory, Bullock and Wheeler crossed the East Rongbuk glacier and succeeded in reaching the summit of the Chang La itself on September 24. In doing this they had the most difficult climb, especially

for laden men, that had so far been encountered. "Beyond the Col (Chang La) an easy succession of rock and snow slopes could be seen leading to the north-east shoulder (27,390) of Everest."

The weather and the exhaustion of the party prevented any further advance. This closed the season of 1921. The results may be briefly stated. A practical route had been discovered, via the Chang La, which had been reached from the

Kharta valley, but there was an easier way by the East Rongbuk glacier which avoided the ascent of the Hlakpa La and consequent descent of about 1,200 ft. in order to reach the foot of the Chang La. The best time for high climbing appeared to be the months of May and early June.



Fig. 1.—Everest and the surrounding country. By courtesy of the Royal Geographical Society

In the course of this reconnaissance, 600 square miles in the immediate neighbourhood of Mt. Everest were mapped photographically on the 1-inch scale, which proved of great use in subsequent expeditions.

A way had now been found to make a real attempt to reach the summit. The expedition of 1922 comprised Brig. Gen. the Hon. C. G. Bruce, as leader, Col. E. L. Strutt, as second-in-command,

Mallory, T. H. Somervell, Dr. T. H. Wakefield, Prof.G.I.Finch, Major (now Brigadier) E.F. Norton, mountaineers, Dr. T. G. Longstaff, doctor and naturalist, J. B. L. Noel, photographic and cinema expert. In India the party was joined by the late Lieut.-Col. H. T. Morshead, C. G. Crawford,

Capts. Morris and G. Bruce.

One of the questions which the Mount Everest Committee had to consider was the use of oxygen for climbing the last stages of the mountain, which appeared to offer no great difficulty apart from lack of oxygen inseparable from great altitude. It is a matter on which climbers and physiologists are not agreed. Some are in favour of relying on the acclimatisation of those men selected for the final assault. This means remaining for some time at high altitudes under what, at best, must be very trying conditions, where deterioration may set in from the lack of well cooked food and exposure to great cold. disadvantages of oxygen were the considerable weight of the apparatus, about 30 lb., which the climber himself would have to carry, and the number of porters required to take up the supply of spare cylinders in which the gas is compressed to 120 atmospheres. Then there was the danger of the apparatus going wrong at the critical moment. After taking into consideration all the arguments for and against, "the Committee finally

unanimously agreed" to its use.

Profiting by the experience of the previous year, the expedition left Darjeeling on March 26, 1922, following the same route as the year before to Sheka Dzong where they turned south, reaching the spot selected for the Base Camp at the snout of the Rongbuk glacier, 16,560 ft., at the end of April. By May 20 camps were established up to Camp IV as shown on the accompanying sketch map. Camp III at a height of 21,000 ft. at the foot of Chang La on rock, Camp IV on the Chang La, on snow, just under 23,000 ft. What they had now "to aim at was getting two tiny tents carried up the North Face to some niche near the North-East Ridge at a height of 27,000 ft.", that is somewhere above the position of Camp VI on the map, and 2,000 ft. from the top. From such a point it was hoped the final assault on the summit could be successfully made. "So the crux of the situation was the capacity of porters to carry two tents and sleeping bags, provisions and light cooking apparatus for this 27,000 ft. camp" sufficient for four climbers. This arduous work was to be carried out by a specially selected corps of forty Nepalese porters.

The first assault was made on the morning of May 20 from Camp IV, where four climbers—Mallory, Somervell, Norton and Morshead—had slept the night before, but they were only able to reach Camp V, 25,000 ft., and had to spend the night there. The next day was fine and they succeeded in reaching a height of 26,985 ft. just below the north-east shoulder of Everest (marked 27,390 ft. on the map). They were "within one mile of the summit and there appeared

to be no serious obstacle in this last remaining lap—only it was more than could be accomplished in one day from a camp at 25,000 ft." They were compelled to turn back on account of the slow progress made and the lateness of the hour. It was between 10 and 11 p.m. before the party reached the shelter of Camp IV on the Chang La. All four were more or less frost bitten during the descent. The worst case was Morshead, who afterwards had to have the top joint of three of his fingers amputated.

Finch, G. Bruce, and the Gurkha Lance-Corporal Tejbir, then made another attempt, this time using oxygen. They were accompanied from Camp IV by twelve porters carrying spare oxygen cylinders and camp gear. The climbers themselves carried the oxygen apparatus weighing about They hoped to pitch a camp at 26,000 ft., but they only succeeded in reaching 25,500 ft. when they were forced by the weather to camp and send the porters back to Chang La. thirty-six hours the storm lasted, and there was great danger of the whole camp being blown away. On the third day, still using the remaining oxygen, they reached a height of 27,235 ft., about half a mile from the top, when they were compelled to return.

One more attempt was made early in June by Mallory, Somervell and Crawford, but an avalanche, on the steep ascent to the Chang La, carried away seven of the porters, who were killed instantaneously. The monsoon had now arrived and all further attempts for the season were abandoned.

The results of the 1922 expedition were—heights of 27,235 ft. and 26,985 ft. were reached with and without oxygen; it was ascertained that there are no serious physical obstacles between the highest point attained and the summit. Camps were made at 25,000 ft. and 25,500 ft., but it was found that it was not possible from there to reach the top and return in one day. Oxygen was used for the first time, though its efficacy does not appear to have been definitely proved or disproved.

We now come to the third expedition, which was undertaken in 1924. This year Gen. Bruce was again the leader, but owing to illness he was obliged, shortly after the start, to hand over to Col. Norton who had been appointed second-incommand. The other members were Mallory, Somervell, G. Bruce, Odell, Beetham, Hazard, Irvine, Shebbear, Major R. W. G. Hingston, and Noel who was in charge of the photographic equipment which was to contribute so much to the finances of the expedition.

Darjeeling was left on March 25, the old Base Camp at the foot of the Rongbuk glacier being reached on April 29. The three camps in the glacier valley had then to be established and stocked, and the route to Camp IV on the Chang La made practicable for laden porters. This done, Camp V, 25,500 ft., would have to be similarly established and stocked and finally

Camp VI, at 26,500 ft., and possibly another camp (not marked on the map) at about 27,200 ft. "All this had to be done before the actual attempt could be made."

Camps I and II were established and stocked with Tibetan labour, while all above this were worked by the specially enlisted Nepalese porter corps, fifty-two strong. Early in these operations the occurrence of violent winds and snow impeded the work, obliterating the way which had been prepared a short time before. On the night of May 6-7, the thermometer fell to $21\frac{1}{2}^{\circ}$ F. below zero. There were many sick porters at Camp III who had to be sent down to Camp II. In these circumstances it was not easy to keep up the morale of the porter corps. This first attempt got no further than Camp III. It was beaten by the weather and in the end the whole expedition had to retreat to the Base Camp; even this was only carried out with considerable difficulty. "So ended Round One with the mountain."

After a complete reorganisation of the porter corps a fresh start was made on May 17. By May 19 the expedition was in full occupation up to Camp III. The next operation was to tackle the route to Camp IV, which has already been mentioned as the most difficult obstacle. It was here that the seven porters lost their lives in 1922, so the greatest care had to be taken to guard against a similar accident, by finding a way clear of avalanches. This was found by Norton, Mallory and Odell. It proved to be a most exhausting operation, accompanied by several narrow escapes, especially for Mallory, who nearly lost his life by a fall into a crevasse. The establishment of Camp IV was undertaken by Somervell, Hazard and Irvine, using the way marked out by Norton and Mallory. Ropes were fixed at the worst places and a rope ladder was used for negotiating the chimney. Then came the rescue by Norton, Mallory and Somervell of four porters stranded at Camp IV. Incidents like this are most trying and exhausting for members of the climbing party just before they have to make the supreme effort to reach the summit, yet such occurrences cannot be avoided.

The next effort was made by Norton and Somervell with the help of three gallant porters. They pitched a tent at Camp VI, 26,800 ft., and sent the porters back to the Chang La. night was spent at this height and they slept fairly well. The next morning, June 4, they started at 6.45 for the summit which was about a mile off and 2,200 ft. above them. The going was good, the day perfect. At 27,500 ft. they both began to feel symptoms of distress. Norton felt very cold and had trouble with his eyes; Somervell also was in trouble with his throat and had to stop often and cough. "Now seven, eight or ten complete respirations were necessary for each single step forward." They struggled on, however, to about 28,000 ft., when Somervell could go no further. Norton went on to 28,126 ft., when his progress became very slow. It was then 1 P.M.

when he, too, was forced to give up. The cause of their failure was most likely due to want of fitness following the strenuous time they, with other climbers, went through in rescuing the marooned porters, and other exertions which should be avoided if the final climb is to be successful.

The last and final attempt was now to be made. Two climbers without oxygen had failed. It was the turn of others to try with oxygen: Mallory, the experienced mountaineer whose third expedition it was to Everest, and Irvine, the young undergraduate and Oxford blue. On June 7 they set out for Camp VI while Odell was in support at Camp V. On arrival Mallory returned his four porters to Odell and he and Irvine spent the night at Camp VI. In a note which he sent back with the porters to Odell he said, "the weather was perfect but the oxygen apparatus was a nasty

load for climbing."

Nothing is known of what happened after this except that Odell saw them at 12.50 P.M. the next day much nearer their camp than he had expected they would be had they started early in the morning. Shortly afterwards the mist shut out the view and Odell saw them no more. That evening Odell returned to Camp IV, where Hazard was. The following day, as the climbers did not return, Odell, with two porters, reascended the mountain intending to visit Camp V and VI, but did not get further than Camp V, where he spent the night. The following morning the weather had changed, a strong and bitter wind was blowing. As he could not persuade the porters to go any higher he sent them back to Camp IV, going on himself alone to Camp VI. This time he took oxygen, though he does not seem to have derived much benefit from it. He eventually reached Camp VI but found no sign of Mallory and Irvine. After searching in vain further up the mountain for a couple of hours, he too was obliged to return. In a storm of gale force he reached Camp IV; it was evidently a forerunner of the approaching monsoon. Here Hazard was awaiting him. Next day they evacuated the Chang La camp. Thus ended the last assault on Mount Everest.

A great adventure deserves a great narrative. The story of Mount Everest has been told in three large volumes splendidly illustrated by the leaders and members of the three expeditions, in which every aspect of the subject has been dealt with in great detail. But this is not all, for these narratives have been admirably summarised and woven into a single story entitled "The Epic of Mount Everest", by Sir Francis Younghusband.

What are the chances of success in 1933? A great deal, in fact almost everything, depends on the weather, and in weather we include wind. The very best laid plans may be wrecked at the last moment by one of those sudden changes which occur in the high Himalaya without warning. Wireless might be used to receive weather reports from the Indian Meteorological Department, but whether they could give indications of the local changes is more than doubtful.

The climbers must be absolutely fit and have had an opportunity of acclimatisation, undisturbed by any incident calling for great exertion on their part, such as having to rescue porters from dangerous positions. Yet such incidents are liable to occur.

The margin between success and failure must always be small—that is what makes it such a great adventure. Many favourable conditions must conspire to take the climbers to the summit;

without these, success is doubtful.

We have no information as to the plans of the new expedition but we fancy the same camps will be occupied up to, and including, Camp IV on the Chang La (22,990 ft.) with, perhaps, some form of portable hut at Camp III as has been suggested. Higher up an attempt may probably be made to establish three camps in order to climb the remaining 6,000 ft.

As regards oxygen, since 1924 a much lighter form of apparatus has been constructed, weighing only about 12 lb.* While a high degree of acclimatisation will be aimed at by the climbers selected for the final assault, no doubt the latest

* See NATURE, 128, 1037, Dec. 19, 1931.

form of light oxygen equipment will be provided, so that it can be used if required for the last portions of the ascent. Aptitude for acclimatisation appears to vary with the individual. With some persons, deterioration, due to remaining at high altitudes, may set in before acclimatisation has been fully attained; this has always to be reckoned with. We think, given reasonable luck, Everest will be climbed. If fortune smiles the summit will be reached.

The new expedition will be led by Mr. Hugh Ruttledge, formerly of the Indian Civil Service, who has had a great deal of Himalayan experience. The other members of the expedition are Mr. F. S. Smythe, leader of the successful Kamet expedition, with three of his companions—Capt. Burney, Dr. Raymond Green, and Mr. E. E. Shipton; Mr. Noel Odell, Mr. C. G. Crawford, Mr. Shebbear, members of former Everest expeditions; Messrs. Wyn Harris, J. Longland, G. Wood-Johnson, and Capt. Hugh Boustead, and possibly Dr. W. McLean as second doctor; Mr. L. R. Wager, of the Arctic Air Route Expedition, 1930-31, and Mr. T. Brocklebank.

Scientific Centenaries in 1933

By Eng.-Capt. Edgar C. Smith, o.B.E., R.N.

THE year which has just closed saw the commemoration of the centenaries of many famous men, among whom were Scott, Goethe, Leeuwenhoek, Locke and Warren Hastings. Of especial interest to men of science was the commemoration of the tercentenary of Wren, while the celebration at Cologne of the centenary of Otto attracted considerable attention in the

engineering world.

The closing month of the year also saw the commemoration of four other men distinguished as engineers or inventors. On December 7, the American Society of Mechanical Engineers at its annual dinner paid tribute to the memory of John Edson Sweet (1832–1916), a founder of the Society and once professor of engineering at Cornell, and to the memory of Alexander Lyman Holley (1832–1882), a singularly gifted man, whose monument in Washington Square, New York, describes him as "Foremost among those whose genius and energy established in America and improved throughout the world the manufacture of Bessemer steel". Eight days later, on December 15, a ceremony took place at the base of the Eiffel Tower to commemorate the centenary of the birth of its constructor, Gustave Eiffel (1832-1923), while on December 23 various tributes were paid to the memory of Sir Richard Arkwright (1732-1792) who reduced to practice the principle of roller drawing in spinning machines, and was the founder of the factory system of cotton manufacture as we know it to-day. Lecky, the historian, speaking of the wealth which accrued to Great Britain through the cotton mill and the steam engine, and the power this wealth gave the country for carrying on the great French wars, declared that Arkwright and Watt deserved statues beside those of Wellington and Nelson, but so far no monument has ever been erected

by his countrymen to Arkwright.

Earlier in the year, at Camborne, Prince George unveiled a statue of the Cornish engineer, Richard Trevithick, the centenary of whose death, falling on April 22 this year, will be commemorated by the engineering world on a scale in keeping with Trevithick's place as a pioneer. Trevithick died at Dartford, Kent, a poor man and for half a century was almost forgotten. The publication of his life by his son in 1872 did much to rescue his achievements from oblivion, and in 1888 a memorial window was erected to him in Westminster Abbey.

The forthcoming celebration is being supported by the leading engineering societies of Great Britain, and the arrangements are in the hands of a committee of which Sir Murdoch Macdonald is chairman. There will be memorial services in Dartford Parish Church and Westminster Abbey, an eminent engineer will deliver an address on Trevithick's work as an inventor, and the committee hopes to be able to erect tablets to Trevithick at his birthplace, Illogan, Cornwall, and also at Pen-y-daren, South Wales, and near Euston Road, London, to record his early attempts to introduce steam locomotion on roads and railways. Trevithick's reputation has never stood higher than it does to-day. He may indeed be proclaimed the father of the high-pressure steam engine. Many other engineering centenaries also occur this year. A far less well-known inventor who died three months before Trevithick was Frederick König (1774–1833), whose four patents of 1810–14 led to the construction of the flat-bed printing press in which the paper was pressed against the type by a revolving cylinder. A red letter day in the history of printing was November 28, 1814, when the *Times* was first produced on a König machine driven by a steam engine.

Two centuries ago, on February 11, 1733, John Perry died at Spalding at the age of sixty-three years. For many years he was comptroller of maritime works under Peter the Great. He constructed harbour works in Great Britain and as the inscription on his tomb in Spalding Parish Church records, he was "Employed by ye Parliament to stop Dagenham Breach which he Effected and thereby Preserved the Navigation of the River of Thames and Rescued many Private Familys

from Ruin".

Engineers born a hundred years ago include Sir Richard Tangye (1833–1906), the best known of the brothers who built up the great Cornwall Works at Birmingham; Thomas Edwards Vickers (1833–1915), another engineer and captain of industry of world-wide fame; James Robson (1833–1913), a pioneer of the gas engine; Henry Wilde (1833–1919), one of the leading inventors of the dynamo and Rudolph Haack (1833–1909) who at Stettin built the first ironclad constructed in Germany, and helped to lay the foundation of the

German shipbuilding industry.

reconnaître sa fille"

Turning to other fields of human endeavour, it is not necessary to stress the interest which is attached to the bicentenary of the birth of Joseph Priestley (1733-1804) who was born at Birstal, Leeds, and died at Northumberland, Penn-The statues of Priestley at Leeds, sylvania. Birmingham and Oxford, testify to the esteem in which his memory has been held, and it may be recalled that it was at the celebration on August 1, 1874, of the centenary of his discovery of oxygen, on the piazza of his dwelling-house in Northumberland, that the American Chemical Society was founded, one of many instances of the value of such commemorations. Priestley was one of the few Englishmen elected a foreign associate of the Institute of France, the secretary

Among Priestley's intimate friends was the geologist and chemist, Richard Kirwan (1733–1812), who, like Priestley, received the Copley medal of the Royal Society. Kirwan was in correspondence with many of the leading savants of Europe, his London house was the meeting-place of the learned and after he returned to his native country, he became president of the Royal Irish Academy.

of which, Cuvier, in his éloge, referred to him as

"le père de chimie modèrne qui ne voulait pas

The year 1733 also saw the birth of Jean Charles Borda (1733–1799), the eminent French mathematician and astronomer; of Dr. Thomas

Hornsby (1733–1810), the successor of Bradley as Savilian professor of astronomy at Oxford and the first Radcliffe Observer, and of the famous German traveller, Carsten Niebuhr (1733–1815), the pupil and friend of the astronomer, Tobias Mayer.

Somewhat later than these eighteenth century worthies came Joseph Nicephore Niepce (1765–1833), the French chemist, whose statue at Chalons recalls his achievement of obtaining sun prints on metal plates; Thomas Allan (1777–1833), the Edinburgh mineralogist; Adrien Marie Legendre (1752–1833), the French mathematician, writer of many works and the contemporary of Lagrange and Laplace. Legendre's "Elements of Geometry" was translated into English by Thomas

Carlyle.

To the names of these three men who died in 1833 may be added that of Dr. William Babington (1756–1833), whose statue in St. Paul's Cathedral is regarded as a fine example of portraiture in marble. Babington was physician for many years to Guy's Hospital, and it was at his house that the meetings took place which led to the foundation of the Geological Society. Active to the last, when seventy-six years of age, he presided over the Priestley centenary festival on March 26, 1833, but died of influenza three days later. He is buried in St. Mary Aldermanbury in the City.

Coming nearer our own times, it is but natural that with increase of opportunity, the lists of men who make notable contributions to discovery and progress should grow longer. "Life," said Emerson, "is girt round with a zodiac of sciences. the contributions of men who have perished to add their point of light to our sky. These road makers on every hand enrich us." To this everincreasing group of men belong many born in 1833, among them Sir Henry Roscoe (1833-1915); the German mathematician, Rudolph Clebsch (1833-1872); and the meteorologist, Robert Henry Scott (1833-1916), the successor of FitzRoy at the Meteorological Office. On April 15 was born Maurice Loewy (1833-1907), the constructor of the equatorial coudé and the successor of Tisserand as director of the Paris Observatory; on May 5, Ferdinand, Baron von Richthofen (1833-1905), the geologist and geographer; on March 16, Hilary Bauerman (1833–1909), the metallurgist and benefactor of the Iron and Steel Institute and the Royal School of Mines; and on June 29, Peter Waage (1833-1900), the distinguished Norwegian chemist, student of Bunsen, successor of Strecker and collaborator with Guldberg.

Then, too, on October 15 and October 21, 1833, respectively, occurred the births of Frederick Guthrie (1833–1886), through whose efforts the Physical Society of London was founded, and of Alfred Bernhard Nobel (1833–1896), the Swedish engineer, chemist and inventor of explosives, by whose will of November 27, 1895, the bulk of his great fortune was used for founding the famous Nobel prizes for physics, chemistry, medicine,

literature and peace.

Empire Broadcasting

IN the field of radio communication, the last few weeks of the year 1932 were conspicuous for the inauguration of a regular broadcasting service between Great Britain and the various portions of the British Empire. The developments which took place in this connexion during 1932 are somewhat striking, and they are recorded in some detail in the B.B.C. Year Book for 1933. From the previous edition of this Year Book, it appeared that little progress was likely in connexion with the Empire broadcasting scheme drawn up by the British Broadcasting Corporation owing to lack of support by the Colonial and Imperial Conferences of 1930, and the responsible authorities in the overseas countries concerned. In November 1931, however, the B.B.C. announced its intention of developing the Empire broadcasting scheme on its own initiative, and the rapid progress made during the ensuing twelve months is illustrated by the fact that the regular service from the new station erected at Daventry for the purpose began on December 19.

As an aid in the development of this new broadcasting service, the B.B.C. had the benefit of about five years' experience with an experimental short wave station erected at the Marconi Company's works at Chelmsford. Owing to the fact that the distances from Great Britain to the various Dominions and Colonies are in the range 2,000-13,000 miles, it is necessary to make use of the short wave-length band of approximately 12-60 metres for radio communication. Chelmsford station operated on a wave-length of 25 metres, and enabled a large amount of technical data to be obtained in connexion with short wave broadcasting. A little consideration of the matter will show, however, that one transmitting station working on a single wave-length cannot provide a satisfactory Empire broadcasting service. In the first place, account must be taken of the wide differences in local time in the various countries concerned, and it is obvious that the broadcast reception must be possible in general during the leisure hours of the inhabitants of those countries. Secondly, as a result of experience, not only in broadcasting but also with ordinary telephonic and telegraphic radio communication on short waves, it is known that the choice of the best wave-length varies with the distance to be covered, and whether the portion of the earth over which the communication takes place is in sunlight or darkness. Following on the latter condition, it is frequently found that diurnal and seasonal variations in the ionosphere, without which longdistance short wave communication would apparently not be possible, make it necessary to have a choice of at least two wave-lengths in order to maintain regular communication between two fixed points. Lastly, a directive antenna system at either or both ends of the communication link, makes such a vast improvement in general efficiency that the arrangement is always employed

where possible in commercial short wave communication.

Based upon the considerations outlined above, the British Empire has been divided up into five zones and the new station at Daventry is designed to supply a direct broadcasting service to listeners in each zone, a two-hour programme being given daily between 6 P.M. and midnight local time, with the hours 8-10 P.M. as a focus wherever possible. For each zone a separate antenna array has been erected at Daventry, each array being so oriented as to give a not too highly concentrated beam of radiation in the desired direction. In the case of the first array, for Australasia, the transmission is limited to a single wave-length of 25.5 metres, but the radiator and reflector units of the array are reversible so that the waves can be transmitted around the earth in either direction. In the case of the other four zones, two or three wave-lengths are available as required. approximate constitution of these zones, with the wave-lengths available, and the periods in G.M.T. at which the transmissions take place in Great Britain are given in the following table:

Zone	. Countries.	Wave-lengths (metres).	Approximate programme period (G.M.T.).
	Australasia, including New Zealand, Borneo, New Guinea and the Pacific		
	Islands	25.5	9.30-11.30 A.M.
2	India, with Burma and the Federated Malay States	17, 25 and 32	2.30- 4.30 р.м.
3	South Africa, with East Africa, Palestine, Sudan	G MA SAIL	an endant in
4	and Somaliland West Africa, Nigeria and Gold Coast, with Tristan	14 and 32	6.0 - 8.0 P.M.
	da Cunha and Falkland Islands	32 and 48	8.0 -10.0 P.M.
5	Canada, Newfoundland and West Indies	19, 32 and 48	1.0 - 3.0 A.M.

In addition to the directional antenna arrays, six omnidirectional aerials have been provided and these will be used for transmitting any special programmes which it may be desired to receive at

any hour in any part of the world.

These arrays and aerials are supplied by feeders or transmission lines from two transmitters housed in the new station building at Daventry. Each of these transmitters is of modern design and construction, and comprises a crystal-controlled master oscillator followed by the necessary frequency multiplying, modulating and energy amplifying stages: these terminate in a final amplifier which consists of four 15 kw. valves connected in push-pull arrangement, the output circuits of which are coupled to the aerial feeder.

The programme supplied through this station will comprise items largely of an original or topical nature, having entertainment or news interest of a type not normally obtainable in the countries where reception takes place. Owing to the difficulties of arranging repeat performances, particularly at the more inconvenient times shown in the above table, the B.B.C. will make extensive

use of modern electrical methods of programme recording for use in the transmitters. example of the use of the station, reference may be made to the occasion of the broadcasting of the King's Christmas message on December 25 last, when the vision of Sir Ambrose Fleming and other pioneers of radio communication, of a single human voice addressing listeners over the whole surface of the earth, was realised. In addition to being transmitted by all the B.B.C. stations serving Great Britain, it was sent through the two Empire transmitters at Daventry. One of these was connected to the Indian zone array, for which zone the transmission took place at the normal The other short wave transmitter was operated on a wave-length of about 20 metres,

and supplied an omnidirectional aerial, so as to broadcast so far as possible in all directions. In addition, the programme was recorded by Blattner-phone and used for re-transmission in the various zone programmes later in the day.

In addition to the direct reception from Daventry on listeners' private receiving sets, it is likely that the broadcasting authorities in the Dominions will arrange for the relaying of some programmes through their local stations operating on medium wave-lengths. From the point of view of the B.B.C., the whole Empire service will be experimental for a period of about six months, during which arrangements will be made to collect reports of reception from selected listeners in all parts of the Empire.

R. L. S.-R.

News and Views

New Year Honours

THE New Year Honours List includes the following names of scientific workers and others associated with scientific work: Baron: Sir Thomas Horder, Bt., senior physician to St. Bartholomew's Hospital. Knight of the Thistle: The Right Honourable Sir Herbert Maxwell, Bt., chairman of the Royal Commission on Scottish Historical Monuments, president of the Society of Antiquaries of Scotland, 1900-13. K.C.B.: Sir Frederick Leith-Ross, chief economic adviser to His Majesty's Government. K.C.I.E.: Major-General J. W. D. Megaw, Director-General of the Indian Medical Service. K.C.V.O.: Mr. F. J. Willans, surgeon apothecary to H.M. Household at Sandringham. Knights: Mr. C. A. Cochrane, chairman of the Council of Armstrong College, University of Durham. Mr. H. H. Dalrymple-Hay, consulting engineer, for his inventions and services in connexion with the construction of tube railways. Prof. F. T. G. Hobday, Principal and Dean of the Royal Veterinary College. Mr. A. J. C. Huddleston, lately economic adviser to the Sudan Government. Mr. J. L. McKelvey, for services to surgery in the Commonwealth of Australia. Mr. E. R. D. Maclagan, Director and Secretary, Victoria and Albert Museum. Mr. W. Perry, president of the Royal Agricultural Society, New Zealand. Mr. R. S. Rait, Principal and Vice-Chancellor of the University of Glasgow since C.H.: Rev. John Scott Lidgett, president of the Methodist Church, Vice-Chancellor of the University of London in 1930-31 and 1931-32. C.M.G.: Dr. J. J. C. Bradfield, Government engineer, Sydney Harbour Bridge, State of New South Wales. Prof. D. B. Copland, professor of commerce, University of Melbourne. Ennis, constructing engineer for Messrs. Dorman, Long and Co., Sydney Harbour Bridge, New South Wales. Mr. W. B. Johnson, director of Medical and Sanitary Service, Nigeria. Mr. W. J. U. Woolcock, formerly general manager of the Association of British Chemical Manufacturers, chairman of the Committee of Non-official Advisers associated with the industrial advisers of the United Kingdom delegation at the Ottawa Conference. C.I.E.: Major-Gen. W. C. H. Forster, Surgeon-General with the Government of Bombay. Lieut.-Col. R. B. Seymour Sewell, director of the Zoological Survey of India. Mr. C. G. Trevor, chief conservator of forests, Punjab and North-West Frontier Province. Col. J. N. Walker, director of the Medical Department and Sanitary Commissioner, his Exalted Highness the Nizam's Government, Hyderabad, Deccan. C.B.E.: Mrs. Alice Baker, one of the founders of the Thomas Baker, Alice Baker, and Eleanor Shaw Medical Research Institute, Melbourne, Commonwealth of Australia. Dr. F. H. A. Marshall, reader in agricultural physiology in the University of Cambridge. Dr. S. W. Smith, chief assayer, Royal Mint, and president of the Institute of Mining and Metallurgy. Dr. H. A. Tempany, director of agriculture, Straits Settlements and Federated Malay States. Prof. H. E. Whitfeld, Vice-Chancellor of the University, State of Western Australia. O.B.E.: Dr. J. T. Bradley, chief medical officer, Seychelles. Mr. A. V. Elsden, War Department chemist, Royal Arsenal, Woolwich. Mr. A. McCallum, senior inspector for agricultural education, Department of Agriculture for Scotland. Mitharam Pribhdas Mathrani, executive engineer, Left Works Division, Lloyd Barrage Circle, Sukkur, Bombay. Dr. S. A. Neave, assistant director, Imperial Institute of Entomology. Mr. J. Smith, director of animal health and acting secretary for agriculture, Northern Rhodesia. Mr. H. W. O. Taylor, executive engineer, Right Works Division, Lloyd Barrage Circle, Sukkur, Bombay. Mr. R. S. Taylor, principal medical officer, Somaliland Protectorate. M.B.E.: Mr. J. Coelho, assistant Crown surveyor and assistant engineer, Public Works Department, Gibraltar. Mr. W. M. Schutte, agricultural engineer to the Government of Bombay. Mr. H. N. Williams, assistant engineer, Irrigation Department, Iraq.

Calendar of Nature Topics

For the past nine years calendars have been published in our columns recording week by week notes of historic interest relating to people and

institutions, discoveries and inventions, customs, festivals, geographical exploration and other matters appropriate to a scientific "Book of Days". We began in 1924 with "Early Science at the Royal Society", and during last year appeared a "Calendar of Geographical Exploration", which was contributed by Miss R. M. Fleming and most admirably fulfilled its purpose. With this issue begins an annual cycle of a different kind relating to natural history in a wide sense, as was understood, for example, by Gilbert White in the notes and letters which make up his famous "Natural History of Selborne". Prof. James Ritchie, Regius professor of natural history in the University of Aberdeen, will be chiefly responsible for the weekly notes in this "Calendar of Nature Topics", and Dr. C. E. P. Brooks will deal with meteorological events of topical interest. In addition, we hope to receive occasional notes on such subjects as agriculture, botany, marine biology, fisheries and similar branches of pure and applied natural history from other contributors.

IT is not intended that this year's Calendar shall be of the usual type, recording aspects of Nature or country life week by week in Great Britain, but that it shall take a much wider outlook. The main idea will be to bring together, in chronological sequence throughout the year, observations and conclusions representing ascertained knowledge to-day on the subjects of the notes; and the range of the natural occurrences or phenomena may be that of the whole world. It is unlikely, therefore, that there will be any lack of suitable material; nevertheless, suggestions of topics of interest for inclusion in this new Calendar, or short notes which might be used, would be helpful and should be sent to Prof. Ritchie at the University of Aberdeen. It need scarcely be said that any such communications should be sent well in advance of the dates to which they

Centenary of Legendre, 1752-1833

THE centenary occurs on January 10 of the death of the eminent French mathematician, Adrien Marie Legendre, whose labours over a period of sixty years were contemporary with those of Lagrange and Laplace, with whom he formed part of "that constellation of mathematical talent of which Paris was for more than two generations the main centre". Legendre was eighty years of age when he died, having been born at Toulouse on September 18, 1752. He was educated at the Collège Mazarin and at the age of twenty-five became a professor at the military school in Paris. He published his first important memoir, on attractions, in 1783, and in that year he was elected a member of the Paris Academy of Sciences. Four years later, with Cassini and Mechain, he was appointed to conduct the geodetical operations for connecting the Observatories of Paris and Greenwich. Through this he visited London, and was made a foreign member of the Royal Society. Unlike many of his contemporaries, he passed through the Revolution unscathed and by his writings and his work on commissions continued to add to his reputation. In 1795, he became a member of the staff of the famous Ecole Normale. His chief works were his "Géométrie" (1794), which was translated into English by Thomas Carlyle, his "Théorie des Nombres" (1798), "Calcul Intégral" (1811–1826) and "Fonctions Elliptiques" (1825–26). A few weeks before his death he added to the last of these another volume, which contained some of the researches of the younger mathematicians, Abel and Jacobi, the value of whose work Legendre quickly recognised. Among the best-known pupils of Legendre were Cauchy and Arago. The death of Legendre took place at his house at Auteuil.

Sir Henry Roscoe, 1833-1915

Among those whose efforts aroused Great Britain to a realisation of the value of scientific education few did more than Sir Henry Enfield Roscoe, whose birth took place in London on January 7, 1833, a century ago. The son of a judge and a grandson of William Roscoe the historian, he got his second christian name from a great-grandfather, Enfield, a colleague of Priestley's at Warrington. He was sent first to Liverpool High School and afterwards to University College, London, where he came under the influence of Graham and Williamson. Later, he spent some time under Bunsen, working in the historic old laboratory at Heidelberg where "beneath the stone floor at our feet slept the dead monks, and on their tombstones we threw our waste precipitates". Returning from Germany, Roscoe at the age of twenty-four years was appointed to succeed Frankland at Owens College, Manchester, a position he held with conspicuous success for thirty years. He was one of the foremost in engendering a spirit of research and many of his students afterwards rose to high rank. His collaboration with Dittmar, Harden and Schorlemmer, the first professor of organic chemistry in Great Britain, led to the publication of many notable works some of which are still sought after. One of his achievements as an experimentalist was the isolation for the first time of vanadium. He was elected a fellow of the Royal Society in 1863 and awarded a Royal medal in 1874; he served as president of the Society of Chemical Industry in 1881, and as president of the Chemical Society in 1882. He was elected member of parliament for South Manchester in 1885; in 1887, the year in which he retired from Owens College, he was president of the British Association. He was a member of various Royal commissions, and from 1896 until 1902 was Vice-Chancellor of the University of London. His eightieth birthday was marked by the presentation of his bust to the Chemical Society. He died on December 18, 1915, at Woodcote Lodge, West Horsley, Surrey, and was buried four days later in Brookwood Cemetery.

North Atlantic Gale

THE last day of the old year and the first few days of the new have proved remarkably tempestuous on the North Atlantic. During the near approach to Ireland on December 31 of an exceptionally intense

cyclonic depression, a destructive gale occurred in Ireland. At Valentia Observatory the pressure tube anemograph registered a gust of 96 miles an hour, which is the highest gust recorded there for at least sixteen years. Barometric readings in Iceland were unusually low during the three first days of 1933. On January 3, pressure at sea-level was less than 928 millibars (27.4 in.) near the centre of a depression lying off the south-west coast of Iceland, but it is not possible to say by how much it fell below that value. That depression was certainly among the deepest of which we have any record since daily synoptic weather charts of the North Atlantic were first begun. On those prepared and published by the Danish and German Admiralties, there is only one depression which looks to have had such a low reading, that of February 24, 1903, which had been preceded five days earlier by another only slightly less intense.

Earthquake in South-East Africa

The strong earthquake that occurred shortly after 8.30 a.m. on December 31 in south-east Africa possesses some interest as it visited a region in which destructive shocks are almost or quite unknown. Its epicentre seems to have been in Zululand, for damage to buildings was caused at Eshowe and other places. The shock was felt all over Zululand and Natal, in various parts of the Transvaal and the Free State, and from Lorenzo Marquez on the north to the Transkei on the south. Its disturbed area must therefore have contained about a quarter of a million square miles, which is more than that shaken by the great Japanese earthquake of 1923.

Ancient Man in Palestine

IT is reported that Miss Dorothy Garrod, director of the Joint Expedition of the School of Archæology in Palestine and the American School of Prehistoric Research, has discovered further remains of Palestine man. A massive and powerful lower jaw has been found in the cave of the Oven at the foot of Mount Carmel. In an announcement of the discovery by Dr. Grant MacCurdy, of Yale University, director of the American School of Prehistoric Research, it is stated that the character of the newly discovered jaw fully conforms to the view, based upon the evidence previously discovered in the caves of Mount Carmel, that Palestine man, while presenting Neanderthaloid characters, is of a distinct type. It is also announced that a cap or veil made of dentalia shells has been discovered in an adjacent cave. It will be remembered that in the course of the excavations of 1931, Miss Garrod found a cap of dentalia shells still attached to a skull from a mesolithic series in one of the Mugharet el-Wad caves.

Cave Paintings in the Pyrenees

FURTHER details of an interesting discovery of the prehistoric painting of a horse in a cave in the Pyrenees, briefly announced some weeks ago, are now supplied by Science Service, Washington D.C. The cave, to be known as La Grotte de la Bastide, is situated near the village of La Bastide, Hautes Pyrénées, and was discovered by M. Norbert Casterat, pupil of Count Bégouen. At the entrance of the cave were intact Magdalenian hearths, and on the walls were a number of engravings and polychrome paintings, including human figures, and as the central object, a polychrome painting of a horse. The horse is described as 'superb' and is an artistic production comparable with the famous horse of the Altamira cave at Santander. The figure is more than six feet long, and is executed in red with black muzzle. The mane is erect; the eve, ear and nostril being delicately engraved. High lights are indicated on shoulders, stomach and flanks. Judging from this description, the painting would appear in every way to be an exceptionally fine example of cave art.

Lightning Investigation

MR. W. H. F. TREDRE, honorary technical secretary of the Educational Section of the South African Institute of Electrical Engineers, Kelvin House, 100 Fox Street, Johannesburg, has favoured us with some interesting particulars relating to the organisation which has been established for the study of lightning in South Africa. The movement was initiated by Mr. T. P. Pask in a paper read before the South African Institute of Electrical Engineers in April 1930; as a result a committee was formed under the chairmanship of Mr. Pask. The present organisation consists of a main committee and three subsidiary committees dealing with each of the subjects research, statistics and education. regard to the research, the chairman is Dr. B. F. J. Schonland, of Capetown—the scope of the work it is proposed to undertake includes the collection of photographs of lightning strokes and their effects taken by means of revolving lenses of the type suggested by Prof. Boys, of klydonograms and cathode ray oscillograms of lightning waves, point discharge work, etc. The activities of the statistical section may be illustrated by the fact that there are 3,500 observers throughout the Union who are collaborating by making notes on the history of storms. These observers are working under the supervision of Mr. G. W. Cox, acting chief meteorologist of the Union. The educational section will disseminate information on protective measures through the medium of the Press, schools, pamphlets, etc. During the 1933 session it has been arranged for certain of the investigators to read papers on the subjects of their work before the South African Institute of Electrical Engineers. It is anticipated that interesting results will be obtained at the end of the present lightning season.

Electrification of Railways in Britain

In a paper by F. Lydall to the Institute of Transport read on December 12, the electrification of railways is considered under two headings, 'suburban passenger' and the 'general' electrification usually referred to as main line electrification. The main characteristics of the former type of traction are rapid acceleration and increased terminal capacity due to the elimination of locomotives and the ability

of the trains to run equally well in either direction. The practically universal adoption of the multiple unit system for suburban traffic, where several coaches throughout the train are provided with motors, proves that the flexibility this gives to the make-up of the train is of great value in practice. It is usual to divide the trains into units, each unit consisting of one motor coach and several trailer coaches. In the new express service from London to Brighton, the trains are made up of six or twelve coaches, two or four of which are motor coaches, each being equipped with four 225 horse power motors. Over a new portion of an American railway where the stops are 1.55 miles apart, the average speed including stops is 31 miles per hour; on another portion of the line where the stops are 2.9 miles apart the average speed is 40 miles per hour. The increase over the speed of steam trains in Great Britain is about 50 per cent. This speed could be easily increased; it is merely a question of cost. Mr. Lydall considers that on main lines it would be found advisable to work not less than one third of the passenger train mileage by multiple unit By electrification the average speed of passenger trains in Britain could be raised by 25 per cent. The combination of greater comfort, higher speed, and more frequent service would attract many more passengers, and the latter two would also enable the railways to recover much of their goods traffic which at present goes by road transport.

Motor Car Lights on the Road

WHEN motor cars pass each other at night time, there is often a blinding glare in the drivers' eyes. We learn from Science Service that, in the United States, the Bureau of Standards has been conducting an extended research on head-lights to discover how glare can be avoided. Dr. Dickinson of the Bureau of Standards concludes that the most important difficulty in obtaining safe head-lighting is the great disparity in brightness between beams from different lamps. One head-light beam is frequently ten times as intense as another. The driver with the dim lights experiences an almost complete lack of visibility when his car plunges into the bright light of the approaching car. Dr. Dickinson suggests that if the lights were kept so that no head-lamp was more than two or three times brighter than another, most of the glare problem would be solved. Most drivers rely on what they can see of the curb rather than what they see of the oncoming car. Hence the light is increased for a hundred feet in front of the car and the beam is widespread horizontally and slightly depressed. Few motorists realise that it is more dangerous to pass a car that is standing still than one that is running fairly fast. A driver in judging whether the road is clear relies on what he has seen during the past few seconds by the light of the oncoming car. But the road immediately at the back of a car at rest is not illuminated in this way and so danger may lurk there unseen. Exposed lights along the roads sometimes increase the risks of night driving. They often make objects almost invisible which could easily be seen by the head-lights alone.

Early Days of the Turbine

In his inaugural address as chairman of the North-Eastern Centre of the Institution of Electrical Engineers, Mr. C. Turnbull gave interesting reminiscences of some of the initial difficulties Sir Charles Parsons met in perfecting the steam turbine. His experiments with early forms of the turbine were in entirely new and unexplored regions of engineering. Everything had to be found out. Steel discs were run under stresses that no one could calculate and no one knew whether they were safe or not. The early high-speed turbines ran at 4,800 revolutions a minute. But thanks to the wonderful care always taken at Parsons's works, accidents were very rare. When driving dynamos at high speeds, the armature reaction caused great difficulty. Several solutions were adopted for turbo-alternators. In one way the brushes were moved automatically with the load by steam pressure and the variation of the strength of the field was counteracted by special windings. A further difficulty was that owing to the springing of the shaft, the connexions between the armature and the commutator used to give trouble. was overcome by the use of flexible connectors, a device first proposed by Parsons. It has proved of the greatest value. Details are given of the famous Turbinia and the heartbreaking experiences with the destroyers Viper and Cobra. Mr. Turnbull tells how Parsons refused to give up and how his perseverance ultimately led to success. Another great invention that came from Parsons's works was the invention of means for balancing high-speed machinery. The early days of the steam turbine were hard days and the labour expended seemed to lead to nothing. At one time it was doubtful if the steam turbine would ever become practical. The story of Parsons's life should prove very encouraging to young and old inventors.

The New Helm or Steering Orders

THROUGH the work of the International Safety at Sea and Load Line Convention, and the passing of the Merchant Shipping Act, 1932, and in accordance with the subsequent instructions of the Board of Trade, on January 1 the 'direct' system of helm orders came into use on all British vessels. centuries the order to "Starboard the helm" or "Port the helm" has caused the ship's head to go in the opposite direction, the practice having come down through the centuries when tillers were in use. Under the new regulations the order "Starboard" will be given, when it is intended that the wheel, the rudder blade and the head of the ship should go to starboard, and the order "Port" will be given when it is intended that the wheel, the rudder blade and the ship's head should go to port. Though it is expected that little difficulty will be experienced in changing over from the 'indirect' system to the 'direct' system, for a time the orders will be given in the words "Wheel to Starboard" and "Wheel to Port", thus enabling the helmsman to adapt himself gradually to the new system. Like most innovations of the kind, the change in established practice has not been introduced without considerable criticism and opposition, but it is probable that in a very short time it will be regarded as an eminently sensible and desirable alteration.

Research at the Port Erin Biological Station

A SEPARATE report on the work of the Port Erin Biological Station has been discontinued and a survey of the research done is now published in the "Report for 1931 (No. 40) on the Lancashire Sea-Fisheries Laboratory at the University of Liverpool and the Annual Report of the Marine Biological Station (No. 45) at Port Erin, Isle of Man", 1932, edited by the late Prof. James Johnstone and Dr. R. J. Daniel. In this report a large amount of original work is also published. This covers a wide field and deals with the abdominal musculature of Crustacea, herring investigations, hydrography and the fauna and flora of the Isle of Man. The Port Erin Station has recently (1931-32) had a new laboratory added, fully equipped for practical teaching and with about thirty work places. New engines and pumps have been installed. The plaice hatchery and lobster culture are still continued. Besides liberating lobsterlings hatched and reared in the laboratory, a number were placed singly in glass rearing jars and fed on crab, boiled and ground to a fine meal. Nearly half of the number survived and cast their shells four or five times, making eight or nine casts between the period of hatching and the end of the first year. The newly hatched young are fed on fresh plankton; after the fourth moult (lobsterling) they are fed on crushed crab, small pieces of fish being added after the fifth month. These interesting experiments are to be continued on a larger scale next year. Dr. R. J. Daniel has completed his comparative work on the abdominal muscles of the Malacostraca and has drawn up an elaborate phylogenetic table to express the relationships that exist between the main ventral system of musculature.

A New Source of Rubber

In order to avoid the necessity of importing rubber from tropical countries, the Soviet Government has organised investigations of native plants likely to contain this valuable product in their latex. Amongst various plants studied, several species of Chondrilla (Compositæ) occurring mainly in the southern sandy regions, proved to be very promising and their cultivation is being carried out on an extensive scale. Green parts of the plant are cut and rubber prepared from the latex. The quantities obtained must be rather small, since only up to 2 per cent of the green mass represents rubber. Recently, however, it was found that certain insects feeding on roots of Chondrilla can be utilised for extracting rubber from the latex (Veltischev and Luppova, Priroda, No. 10, 1932). One of these is a caterpillar of a pyralid moth, Bradyrrhoa gilveolella Tr., which feeds on the roots and constructs round its body a tube formed of condensed latex and sand grains. Up to thirty and more such tubes can be found on the roots of a single plant, and the tubes contain 9-17 per cent

of rubber. Another useful insect is the larva of a buprestid beetle, *Sphenoptera foveola* Gebl., which also feeds on roots of *Chondrilla* and causes a large outflow of latex solidifying round the root. These swellings contain up to 4 per cent of rubber. Neither insect produces any serious effect on the infested plant, and healthy plants can be infested artificially in order to increase their productivity. Experiments are being made to test the practical and economic side of this method of obtaining rubber.

Cultivation of Green Crops

Bulletin No. 53 of the Ministry of Agriculture and Fisheries ("Cabbages and Related Green Crops". London: H.M. Stationery Office, Price 9d.) has recently been published. It traces the botanical origin of the various green crops, and deals in detail with the soils, rotation, manuring, harvesting and marketing most suitable for cabbages, savoys, Brussels sprouts, cauliflowers and several miscellaneous green crops. A very interesting section deals with the saving of seed in the counties the climate of which is most suitable for this purpose. Pests and diseases are not mentioned in the present pamphlet, since they have been described very fully in other publications of the Ministry.

Astrolabes and Their History

ASTROLABES are not as generally available for study in the museums of the world as their scientific importance and artistic qualities would merit, but all who may desire to become better acquainted with this instrument in its various forms are now given the opportunity. Subscribers are invited for a comprehensive work, entitled "The Astrolabes of the World", based upon the series of instruments in the Lewis Evans Collection in the Old Ashmolean Museum at Oxford, in the Science Museum at South Kensington, and in several other public and private collections in Europe and America. The early Greek treatise on the astrolabe, by Philopon, and the Syriac treatise by Sabokt-both dating from the seventh century—will appear in English for the first time. Illustrations are given of Chaucer's astrolabe, now clearly identified by the character of the rete as depicted in MSS. and many instruments contemporary with Columbus and Drake, are figured. The subject is of fundamental importance to all students of the history of astronomy, geography and surveying, and indeed to the history of science generally, for it may truly be said that the astrolabe kept alight the torch of the scientific method of observation and of computation of results, in many countries, and through many dark ages, when larger instruments and well equipped observatories did not exist. It is hoped that the principal reference libraries may obtain copies of this monumental work, which it is proposed to issue in two quarto volumes, containing more than 600 pages and 155 plates, of which 12 are in colletype, and 216 text figures. The price to subscribers is ten guineas. Subscription forms may be obtained from Dr. R. T. Gunther, Curator of the Lewis Evans Collection, in the Old Ashmolean, Broad Street, Oxford.

World Power Conference

The preparatory work of the organising committee at Stockholm of the next World Power Conference, which will take place in 1933 in Scandinavia, is proceeding steadily. The first plenary World Power Conference was held in London in 1924, the next in Berlin in 1930. There have also been sectional meetings with special programmes, for example, at Basel in 1926 and London in 1928. The Scandinavian Conference will be such a special meeting, dealing with the energy problems of large industry and transport. Participation and collaboration of fifteen countries outside Scandinavia is assured and more than 170 reports are announced. Some forty reports to be published at the meeting deal with problems of energy supply in large-scale industry, such as combined power and heat supply, the rôle of large-scale industry in national power schemes, etc. Many of the technical papers deal with the problems of long distance gas transmission, while other papers are devoted to more special power problems concerning the iron and steel industry, pulp and paper, and cement, sugar, textile and other steam heat consuming industries. Energy questions of transport provide the subjects for 62 reports; railway and marine transport, the peculiarities of city and suburban traffic are to be discussed with due emphasis on the new aspects which have been introduced by electric traction and Diesel engines.

Medical Research in South Africa

THE annual report of the South African Institute for Medical Research, by the director, Sir Spencer Lister, summarises the work of the Institute during 1931. Methods of dust estimation and studies on underground humidity in the air of the Witwatersrand mines have been pursued. The pneumonia attacking the native labourers in the copper mines of Northern Rhodesia has been investigated, and has been found to be caused by the Streptococcus pyogenes and other organisms, but rarely by the pneumococcus. This change of type of the organisms causing pneumonia has also been found in the Kimberley diamond mines. Much experimental work has been done on cancer, and Dr. des Ligneris has found that a microorganism, Salmonella gallinarum of fowl typhoid, if cultivated in the tumour filtrate of the Rous fowl sarcoma for several generations, appears to have the power on inoculation of prolonging the life of fowls with the Rous sarcoma by about fifteen per cent.

The March of Influenza

According to a report of Science Service, Washington, D.C., a considerable outbreak of influenza is in progress in several States of the Union, particularly in the south and west, the United States Public Health Service reporting 14,291 cases during the week ended December 3. In Great Britain outbreaks are reported in Southampton and Birmingham, and during the week ended December 17 in 117 great

towns, including London, of England and Wales, 85 deaths from influenza were certified. In the same week Glasgow reported 28 deaths from this disease.

Announcements

The following have been elected officers of the Royal Society of South Africa for 1933: President: Dr. W. A. Rogers; Treasurer: Dr. L. Crawford; General Secretary: Dr. B. F. J. Schonland.

The Royal Society Mond Laboratory at Cambridge will be opened by the Chancellor of the University, the Right Hon. Stanley Baldwin, on February 3. It will be recalled that the hydrogen liquefaction plant of the Laboratory was described in NATURE for February 13, 1932, p. 224.

WE regret to learn of the death on December 25 of Prof. Paolo Enriques, professor of zoology in the University of Padua and president of the last International Congress of Zoology; he was just starting for Naples to pursue his researches in the Zoological Station. Also of Prof. James Johnstone, professor of oceanography in the University of Liverpool, formerly director of the Marine Biological Station, Port Erin, aged sixty-three years.

CATALOGUE No. 552 of Messrs. Francis Edwards Ltd., High Street, Marylebone, W.1, includes a miscellaneous collection of books, autographs and manuscripts, falling under such headings as Africa, America, Economics, Geography, History, First Editions, Scripture, Shakespeare and the like. There are also special series such as the publications of the Hakluyt Society and books from special presses such as the Kelmscott and Doves Presses. Among the books listed under Africa to be noted especially are Angas "The Kaffirs" with the coloured lithograph plates (£21) and a copy of the 1727 folio edition in Dutch of Peter Kolben's famous and invaluable account of the Cape of Good Hope (£15). The most notable item offered, however, is George Lily's map of Britain, 1546, of which the only other known, and slightly inferior, copy is in the British Museum, and was one of the six treasures selected from the British Museum for the International Geographical Congress in 1928. Apart from Ptolemy's atlas, this is the first printed map of Great Britain (£105).

APPLICATIONS are invited for the following appointments, on or before the dates mentioned:—A secretary of the Iron and Steel Institute and editor of the Proceedings—The Chairman, Appointments Committee, Iron and Steel Institute, 28, Victoria Street, S.W.1 (Jan. 21). A full-time tutor for the practical course in social science at the University of Cape Town—The Secretary, Office of the High Commissioner for the Union of South Africa, 73, Strand, W.C.2 (Jan. 31). A professor of textile industries at the University of Leeds—The Registrar (Feb. 28). A principal of the School of Metalliferous Mining (Cornwall)—The Secretary to the Governors, Camborne (March 31).

Letters to the Editor

[The Editor does not hold himself responsible for opinions expressed by his correspondents. Neither can he undertake to return, nor to correspond with the writers of, rejected manuscripts intended for this or any other part of NATURE. No notice is taken of anonymous communications.]

Disintegration of Light Elements by Fast Protons

SINCE the publication of our paper¹ on the disintegration of elements by fast protons, we have examined some of the light elements more carefully, using much thinner mica windows than we had previously employed on the high voltage tube. With the present arrangement, we can count particles which have passed through only 6 mm. air equivalent of absorber on their way from the target to the ionisation chamber.

In the case of lithium, we have found, in addition to the α -particle group of 8.4 cm. range, another group of particles of much shorter range. The number of these is about equal to that of the long range particles and their maximum range is about 2 cm. The ionisation produced by them indicates that they are α -particles. It will be of interest to examine whether any γ -rays are emitted corresponding to the difference of the energies of the α -particles in the two groups, but on account of the smallness of the effect to be expected, a sensitive method will be necessary.

In the case of boron, the number of particles observed increases rapidly as the total absorption between the target and the ionisation chamber is reduced. The maximum range of these particles is about 3 cm. and in our earlier experiments we determined the number of particles only after passing through the equivalent of $2 \cdot 9$ cm. of air, so that we were very nearly at the end of the range. Decreasing the absorber to 6 mm. of air gives an enormous increase in the number of particles. In this way about twenty-five times as many particles have been obtained from boron as from lithium under the same conditions. We estimate that there is roughly one particle emitted per two million incident protons at 500 kilovolts. The ionisation produced by the particles suggests that they are α -particles, and the energy of the main group would support the assumption that a proton enters the B11 nucleus and the resulting nucleus breaks up into three a-particles. There also seem to be present a small number of particles with ranges up to about 5 cm.

> J. D. COCKCROFT. E. T. S. WALTON.

Cavendish Laboratory, Cambridge. Dec. 22.

1 Proc. Roy. Soc., A, 137, 229; 1932.

The Neutron and Neuton, the New Element of Atomic Number Zero

SINCE neutrons were first recognised by Chadwick in the rays from the beryllium nucleus, it may be of interest to note that in 1915 the hydrogen-helium theory¹ considered this nucleus to consist of two doubly charged helium nuclei and a condensed or nuclear hydrogen atom, now called a neutron. In a paper written early in 1919, the formula of the beryllium nucleus was given as $\alpha_2(\eta\beta)$ which in more

recent symbols is $\alpha_2(pe)$, where the parentheses were used to emphasise the idea that the proton p and the electron e are united to form a neutral group or neutron.

The suggestion that neutrons exist as separate atoms was made independently at practically the same time by Lord Rutherford (June 3, 1920) and

myself (April 12, 1920).

The basis for my assumption of the existence of neutrons, was that it would be difficult for α -particles, on account of their double positive charge, to pass through the region of repulsion (now called the potential barrier) around nuclei of high positive charge in order to unite with the nucleus, but that electrically neutral particles "could easily pass into and through this region". "Such atoms might have masses 1, 2, 3 and 4, and possibly other values, and they would contain no non-nuclear electrons, so they would have no chemical, and almost none of the ordinary physical properties, aside from mass." 2

While the question of stability is not discussed in this paper, it may be stated that at that time I did not consider any neutron of higher mass than 2 to exist more than momentarily except in a moderately heavy nucleus, but that the first quadruple neutron present in other nuclei is that in argon 40, the nuclear formula of which may be written $\alpha_0^{++}(pe)_4^{\circ}$ or

 $\alpha_9 + + (\alpha e_2)^{\circ}$.

Since neutrons of unit mass (according to Chadwick about 1.006), and possibly those of mass two also, probably exist throughout space, and are concentrated by gravitation in the planets, and still more in the stars, they may be considered to constitute collectively an element. Since the atomic number of an element is determined by the magnitude of the charge on the nucleus of an atom of the element, the atomic number which corresponds to a neutron is zero; that is, the neutron is a nucleus with a zero charge. As a name for this new element, neutronium, neutronon, or neuteron have been suggested to me, but the name 'neuton' is more simple and preserves in it the suggestion of neutrality and also the final 'on' of the chemically indifferent elements.

To what extent the atoms of neuton partake in the partition of heat energy between molecules in general, is an interesting problem, since the collisions of the neutrons would be with nuclei and with electrons, rather than with atoms and molecules.

It is not improbable that the general formula for any nucleus, $(p_2e)_z(pe)_i$, proposed independently in 1921 by Masson and by me, may be shown to have a theoretical significance. If n represents a neutron, then this formula may be written $(pn)_zn_i$ in which z is the atomic, and i the isotopic number. For the neutron z is 0 and i is 1, for the proton z is 1, and i is -1, for oxygen 16, z is 8 and i is 0, and for the principal isotope of argon z is 18 and i is 4. It is evident that if all of the electrons are present as neutrons, then z+i gives the number of neutrons, z the number of extra protons, and 2z+i the total number of protons. The numbers z and i are the most important in the classification of atomic species, and the value of i defines specifically the particular isotope, either known or unknown, of any element, either radioactive or non-radioactive.

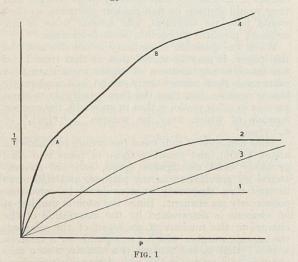
WILLIAM D. HARKINS.

University of Chicago. Nov. 8.

Harkins and Wilson, J. Amer. Chem. Soc., 37, 1396; 1915.
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Kinetics of the Decomposition of Molecules of Intermediate Complexity

In a large class of gaseous chemical reactions, the mechanism is generally supposed to be the following. Molecules are brought into an activated state by collision. Most of the activated molecules are deactivated again, but during their existence there is a definite probability of their chemical transformation. As is well known, the relation between reaction rate and pressure under these conditions is such that the reciprocal of τ , the half life of the reacting substance, gives a curve of type 1 or 2 (see Fig. 1) when plotted against the initial pressure1. (The precise shape of the curve before it has become horizontal depends upon how the above mentioned probability varies with the total energy of the molecule2.)



In a molecule of moderate, but not too great, complexity, it seems possible that there may be several different modes of activation, corresponding to particular divisions of part of the energy among a limited number of vibrational (or rotational) states. To a first approximation, each of these modes may be associated with a separate probability of transformation, because, in the absence of collisions, internal redistribution may be difficult. Thus the total rate of reaction will be roughly the sum of several virtually independent reactions, each rate varying according to a curve of the type 1, 2, or 3. Since each of these reaches its limiting rate at a different pressure, the total rate may vary with pressure according to a segmented curve of type 4, with fairly pronounced changes of direction at A, B, and so on.

The simplest types of molecule may give curves which do not bend, at least up to high pressures, moderately simple molecules may give curves with a limited number of segments of decreasing slope, while complex molecules with many degrees of freedom give curves in which the segments merge into a single line without any noticeable changes of direction at particular stages.

The equation of the segmented curve in real examples will be very complicated, since the assumption of virtually independent reactions is to some extent an idealisation. Further, the variation of all the transformation probabilities with total energy may also complicate matters.

The object of this note, however, is simply (1) to

suggest the desirability of exploring the hypothesis that the most general form of curve for molecules of 'intermediate complexity' is of the segmented type, and (2) to state that experiments on the decomposition of nitrous oxide at fairly low pressures (already published3) and further experiments on the decomposition of acetaldehyde (in progress) yield indications of the real existence of these different types of activated state.

> C. N. HINSHELWOOD. C. J. M. FLETCHER.

Balliol and Trinity College Laboratory, Oxford. Dec. 8.

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 Rice and Ramsperger, Kassel.
 Musgrave and Hinshelwood, Proc. Roy. Soc., A, 135, 23; 1932.

'Hexuronic Acid' (Ascorbic Acid) as the Antiscorbutic Factor

In view of the facts that (1) hexuronic acid is the name of a class of substances rather than that of one individual compound, and that (2) the material described as hexuronic acid isolated from adrenal cortex and now from Paprika contains a molecule of water less than is required for a hexuronic acid, we wish to ascribe the name ascorbic acid to the crystalline substance C₆H₈O₆ which has been the subject of earlier communications from our laboratories.

A. SZENT-GYÖRGYI. W. N. HAWORTH.

Universities of Szeged and Birmingham. Dec. 19.

Methylnornarcotine, Glycuronic Acid, and Vitamin C

THE view that vitamin C is identical with methylnornarcotine has been advanced by Rygh^{1,2}, who claims that narcotine occurs in unripe fruit and disappears with corresponding formation of vitamin C during ripening. According to Rygh, guinea-pigs receiving a scurvy-producing basal diet together with a suitable daily supplement of synthetic methylnornarcotine lose weight and die but without any of the typical symptoms of scurvy. Other workers^{3,4,5,6} have been unable to confirm this observation or to find narcotine in unripe fruit, and Dalmer and Moll⁵ have shown that methylnornarcotine as prepared by Rygh is not a single chemical substance.

In a later communication Rygh⁷ has changed his ground and taken up the position that vitamin C is a complex, of which one component must be methylnornarcotine while the other may be any uronic acid. Thus according to Rygh, the recently demonstrated anti-scorbutic activity of hexuronic acid is due to the contamination of the hexuronic acid by methylnornarcotine, giving a mixture of the two components necessary for the make-up of vitamin C. This hypothesis was drawn from the observation that guinea-pigs were not only protected from scurvy but also grew normally if in addition to the basal diet they received daily 10 γ of "30 per cent methylnornarcotine" and ½ mgm, of glycuronic acid, although the glycuronic acid alone neither prevented scurvy nor supported growth.

I have attempted without success to confirm these

observations, using a specimen of glycuronolactone kindly given by Dr. E. L. Hirst, and two specimens of "30 per cent methylnornarcotine", one prepared by Rygh and the other prepared in this laboratory. Young guinea-pigs were fed upon a scurvy-producing basal diet, and given a daily supplement of methylnornarcotine, or of glycuronolactone, or of both together, in the amounts specified by Rygh.7 the sixteenth day the animals were losing weight rapidly, so they were killed and examined for (a) hæmorrhage at the knee-joint, (b) rib-beading, and (c) suprarenal staining with dilute silver nitrate solution after Moore and Ray.8 All those animals which had been receiving any of the supplements showed signs of marked or severe scurvy indistinguishable from the condition of the negative controls, while the positive control animals (which had received 10 gm. of cabbage daily) were free from scurvy. The lower jaws of all the guinea-pigs were removed for histological examination of the teeth, and full details of the experimental observations will be published elsewhere when this examination has been completed.

W. J. DANN.

The Dunn Nutritional Laboratory, University of Cambridge and Medical Research Council, Milton Road, Cambridge. Dec. 7.

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Union of Pycniospores and Haploid Hyphæ in Puccinia Helianthi Schw.

THE discovery that, in a heterothallic rust like Puccinia graminis Pers., or P. helianthi Schw., the transfer of pycniospore-containing nectar from a monosporidial (haploid) pustule of one sex to a similar pustule of the opposite sex induces the development of æcia in the pustule receiving the nectar has stimulated interest in the process by

which the diploidisation is effected.

In the examination of free-hand sections of monosporidial pustules of P. graminis and P. helianthi, I have observed that usually two types of hyphæ protrude through the ostiole of a pycnium: (1) the stiff tapering slightly-curved paraphyses, which have been frequently figured; and (2) flexuous hyphæ which show considerable variation in length, diameter, regularity of outline, and a few other features. In some of the pycnia—probably the older ones—of a pustule, these flexuous hyphæ may show rather profuse development; in others, only a few of them, sometimes none, are discernible. They may be shorter, but usually they are as long or longer than the paraphyses, not infrequently two or three times as long. They may branch, but they rarely show septations. Some may be swollen at the tip. Occasionally a short spur or peg, of less diameter than a branch, juts out at a side or tip.

Several pycniospores in union with such hyphæ have been observed in sections of haploid pustules of P. helianthi in which the nectar had been previously intermixed, so that presumably both (+) and (-)pycniospores were present on the surface of each pustule and in close proximity to the protruding

Fig. 1 shows such a union. The pycniospore is empty; and the hypha has lost most of its cytoplasm. Presumably the nucleus of the pycniospore has passed through the connecting tube into the hypha and proceeded down it, to associate itself in conjugate relationship with some nucleus or the mycelium.

With regard to the short spurs or pegs on these hyphæ, it is assumed that a hypha of one sex, in response to the presence in its immediate vicinity of a pycniospore of the opposite sex, sends out a short tube to establish contact with that pycniospore.

A nucleus in the act of passing from a pycniospore into a hypha has not been seen, but empty pycniospores found connected by short tubes to these hyphæ furnish strong circumstantial evidence that nuclei migrate from pycniospores to these hyphæ by way of fusion tubes.

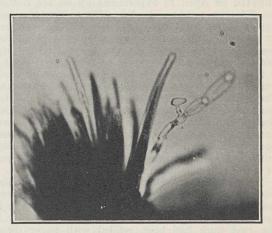


FIG. 1.—The union of a pycniospore and a flexuous hypha in a haploid pustule of *P. helianthi*. The pustule was fixed in formalin acetic alcohol 13½ hours after the intermixing of nectar was done. × approx. 1,500. Photograph by A. M. Brown.

Hyphæ which emerge through stomata and between epidermal cells were observed by Andrus² in haploid pustules of Uromyces appendiculatus and U. vignæ, and by Allen,³ in haploid pustules of Puccinia triticina. Andrus regards these hyphæ as the tips of functioning trichogynes. Allen designates them "receptive hyphæ". However, neither of these investigators found direct evidence of fusions between 'spermatia' (pycniospores) and trichogynes or receptive hyphæ, although both assume from their studies that direct fusions do occur.

There is little doubt that, in haploid pustules of P. helianthi (or of P. graminis), the function of the protruding hyphæ is to establish contact between mycelia of one sex and pycniospores of the opposite sex and thus to serve as an avenue by which the nuclei of the pycniospores reach the internal mycelia of the pustules. The type of union so far observed simulates that of oidium and hypha in the Hymenomycetes.

J. H. CRAIGIE.

Dominion Rust Research Laboratory, Winnipeg, Manitoba. Nov. 17.

Craigie, J. H., NATURE, 120, 765, Nov. 26, 1927.
 Andrus, C. Frederic, J. Agr. Res., 42, 559-587; 19
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Breeding of Oysters (O. edulis) at Port Erin

THE success obtained by the Department of the Ministry of Agriculture and Fisheries in breeding oysters (O. edulis) in the experimental tanks at Conway and Lympne¹ suggested that a similar use might be made of the large outdoor tanks or ponds at the Port Erin Marine Biological Station. normal summers the sea temperature in Port Erin Bay rarely rises above 59° F. seawards of the beaches, owing to the steepness of the beaches and the midsea position of the Isle of Man; the maximum is reached at about the end of August². The stagnant water in the ponds, however, follows air temperature much more closely than the sea, and this year gave surface readings during July and the first three weeks of August ranging mostly between 62° and 65° F.; a range favourable for oyster breeding.

In February, a hundred Whitstable oysters were imported into the Port Erin tank system and distributed on April 14 experimentally on the bottom in different outdoor tanks. These oysters were examined on July 20. High mortality had occurred in the east tank where about a hundred plaice had been stored for breeding. On the bottom of this tank it was found that there had become deposited a black sludge bearing a luxuriant growth of green weed. Oysters which occurred in places where this sludge had accumulated were suffocated. Somewhat similar conditions obtained in parts of the west tank, though the sludge was less dense and the loss was not so heavy as in the east tank. Only one individual, however, among twenty-five had died in a small clean outside store tank through which a small circulation of water had been maintained.

The oysters from this latter tank (24 in number) were opened on July 20, when two individuals were found black sick, that is, carrying larvæ ready to be voided from the parent into the water. On July 13 the east tank had been refilled with water direct from the sea after the pond had been scrubbed out. In this pond most of the larvæ taken from the two black sick oysters were distributed on July 20. To this tank had also been added about a hundred plaice stored for breeding experiments in the following

year.

On November 24 the tank was emptied and examined for oyster spat. A sprinkling only was found. Sixteen were found attached to the north wall of the tank in underhung places along with balanids, mostly remote from direct sunlight. Three occurred in the full glare of daylight on the bottom and near the middle of the tank; one settled on a varnished board and one on a stone, both in deep water near the outlet pipe; two occurred on mussel valves also on the bottom of the tank. The bottom and lower parts of the walls of the tank were covered with a growth of long delicate green weeds, 98 per cent of which were Cladophora fracta, Kütz, var. flavescens Batt. (C. flavescens Harv.), along with a little Urospora isogona Batt. and Percursaria percursa Rosenv.; there was very little accumulation of black mud. It is probable that a good fall of spat might have been obtained had settlers or a supply of cultch been placed in the pond. The spat varied in size from 6 mm. long by 6 mm. deep to 20 mm. long by 18 mm. deep, and fell into two size groups, suggesting that two periods of settling occurred, possibly a settlement from each brood of larvæ.

Ordinary sea-water of a salinity about 33.5 per thousand2 pumped into the tank direct from the sea was used in this experiment; the salinity probably varied little during the first month of the experiment. as this was a period of drought and a little fresh sea-water was pumped into the tank from the sea daily. The plaice in the tank were fed on boiled mussels, the remains of which would contribute with the waste products from the plaice themselves towards manuring the water for the growth of those micro-organisms on which the oyster larvæ and spat thrived. A heavy growth of green weed is evidence of considerable production of motile reproductive bodies from these plants. These reproductive bodies may very well have formed a portion of the food of the larvæ and/or spat at some period. Scott's investigations³ on the plankton of these ponds in 1924-26 indicate that in July and August the phytoplankton (diatoms and dinoflagellates only) may normally be relatively scarce, or even absent: no plankton observations were made in 1932. A few Gobius ruthensparri and about four specimens of various species of small flatfish in the pond may have been inimical to a larger spatfall by feeding on the larvæ, as Dodgson and Sherwood¹ have found. 106 living plaice were recovered from the experimental tank in good condition on November 25.

The experiment is interesting in proving that under such meteorological and other conditions as prevailed in the summer of 1932, (a) oysters will spawn in the tanks at Port Erin; (b) the tanks are suitable for the production of young oysters; and (c) when the method of cultivating oysters in tanks has been improved so as to become a sound economic proposition, it may be possible in favourable summers to produce oyster spat in great numbers at Port Erin. The experiment also encourages the reasonable hope that such tanks as may be built in the future for mussel purification in the north of England may become as useful as those at Conway for oyster

breeding.

We are indebted to members of the staff for taking valuable routine temperature records.

J. H. ORTON. M. W. PARKE. W. C. SMITH.

Zoology Department, University, Liverpool. Dec. 8.

R. Dodgson and H. P. Sherwood, Oyster Breeding Experiments;
 in Reports on Sea Fisheries, Ministry Agric. and Fish., England and Wales, 1919 et seq.
 J. R. Bruce, J. Mar. Biol. Assoc., Plymouth, 15, 542; 1928.
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Reversible Stoppage of the Blood Circulation in Sabellids

Sabellids are marine polychæte worms, the blood system of which is peculiar, both anatomically and physiologically. Numerous capillaries are present, in the body wall, in the branchial crown, and projecting freely into the body cavity, all of them ending blindly. Most of the blood vessels in the body contract rhythmically. By peristalsis blood is forced along the continuous vessels, and at intervals of 10-20 seconds blood is expelled from the blind capillaries by regularly rhythmic centripetal contractions of their walls, to flow back again into these capillaries almost immediately. There may thus be said to be a true Galenic circulation in the capillaries. The blood contains the respiratory pigment chlorocruorin.

If a sabellid worm is put, under water, into a narrow glass tube, the rhythmic contractions of the blood vessels in its crown can be seen through the glass. Under these conditions the contractions continue for half an hour and then cease. But as soon as even the tip of the crown projects again beyond the end of the tube, the contractions recommence. Thus when worms in Nature have retired into their tubes for over half an hour their blood no longer circulates. Yet Spirographis can remain uninjured for eight hours inside its tube when the front end of the latter is artificially closed. After about eight hours the animal re-expands its crown through a new anterior lateral aperture which it forms by local digestion of the mucoid tube.

When a sabellid is put into sea water saturated with carbon dioxide, the pulsations of its blood vessels cease almost immediately. After the animal has been put back into pure sea water the pulsations recommence. Neither water lacking dissolved oxygen, nor water acidified with hydrochloric acid to the same pH as water saturated with carbon dioxide, stops the pulsations. The threshold pH for stoppage in water containing carbon dioxide is in the region of 6.0, varying slightly with species.

in the region of 6·0, varying slightly with species. It has been concluded from a variety of experimental evidence that the natural inhibition of blood circulation when worms are in their tubes is due to an accumulation of excreted carbonic acid between the filaments of the closed crown, and that this carbonic acid does not act on the blood vessels through a nervous reflex, but it prevents the normal exit of carbonic acid produced in metabolism, which accumulates in the body and has a direct inhibitory action on the blood vessel walls

action on the blood vessel walls.

A full account of this work will be published shortly.

H. Munro Fox.

Zoology Department, University of Birmingham. Dec. 6.

Photochemical Reaction of Hydrogen and Chlorine

I am much surprised to read, in Nature of December 17, that so fine an experimenter as Prof. A. J. Allmand has failed to inhibit the action of light on a mixture of hydrogen and chlorine, by drying. It is, with one exception, the easiest reaction with which to demonstrate the influence of traces of water on chemical action. So much so, that for many years it was one of my stock experiments in courses of elementary lectures.

Hydrogen and chlorine were generated by the electrolysis of hydrochloric acid, washed with water and conducted through a tube of phosphorus pentoxide for rough drying. The mixture then passed through a dozen thin bulbs blown in series on a tube of soft glass, in some of which had been placed phosphorus pentoxide. After about six hours, the wash water was saturated with chlorine, and half an hour later the bulbs were sealed off at their capillary joinings. The last bulb was exposed to the light of burning magnesium, and if it exploded after five seconds, the batch was assumed to be perfect. In lecture a pair of bulbs, one containing phosphorus pentoxide and the other not, were exposed to the same intensity of light and in no case did the dry gas ever explode, while the moist ones did so without exception. One of the dried

bulbs was exposed to bright sunlight for three days, and analysis of the gas afterwards showed that one third of the mixture was still in the uncombined state. Mellor and Russell¹, in repeating this experiment, found that two thirds of the gas had resisted the action of light.

In describing these experiments to the Chemical Society, I did the experiment successfully; in fact, on the principle that 'seeing is believing' I have, almost without exception, demonstrated my results experimentally at the meetings at which the papers were read. Prof. Bodenstein and Prof. Allmand owe, I think, the failure of their experiments to the complexity of their apparatus. It is almost impossible, on account of the slowness of diffusion, to dry an apparatus which has any dead ends. Mere 'baking out' at 200° is insufficient, as I have shown over and over again, to get rid of deep-seated moisture in glass. Long continued contact of a gas with pure phosphorus pentoxide is the only way in which success can be attained.

At the risk of seeming egotistical, I would ask all who contemplate working in this field to read my paper² on "Manipulation in Intensive Drying". If they do so, they would, I feel sure, save themselves much fruitless effort.

H. B. BAKER.

Imperial College of Science, London, S.W.7. Dec. 22.

¹ J. Chem. Soc.; 1902. ² J. Chem. Soc.; 1929.

Limiting Mobilities of some Monovalent Ions and the Dissociation Constant of Acetic Acid at 25°

MacInnes, Shedlovsky, and Longsworth in reply¹ to our recent letter with the above heading² have asserted, without experimental evidence, that our conductivity technique was subject to errors and that the discrepancies between the results are to be attributed to this cause. We would, however, point out that by using the cell constant determined with 0·01 N potassium chloride by Parker and Parker's method³ we have been able to reproduce Kohlrausch and Maltby's results⁴ for KCl, NaCl, and KIO₃ at 18° over the range 0·0001 – 0·01 N to within 0·1 per cent,⁵ and this, it appears to us, gives definite support to the view that our experimental results are trustworthy to at least 0·1 per cent.

Two limiting cation transport numbers for KCl are given in our paper⁵, namely, 0·497 the old classical figure of Kohlrausch and Maltby⁴, on which basis all limiting mobilities have hitherto been calculated, and 0.490 the most probable figure deduced from the results of MacInnes and Dole6. It was our original intention to redetermine this constant but owing to the transfer of the senior author to Woolwich, this project was abandoned. We now employ the figure 0.490, which appears to be the most probable value deduced from the determinations of MacInnes and Dole and of Longsworth7. The American authors state in their letter that their value for the limiting conductivity of the chloride ion, based on conductivity and also transport number measurements of KCl, NaCl, LiCl, and HCl at 25° , is $76\cdot 32$ "within a few hundredths of a per cent"; the value deduced previously by us was 76.45 $(n_k \text{ for } \text{KCl} = 0.490).$

The use of the empirical formula of Shedlovsky⁸

for extrapolation to infinite dilution seems unnecessary: the ordinary square root formula holds over a considerable concentration range when applied to Shedlovsky's results9 for KCl, NaCl, and HCl at 25°. Further, the Shedlovsky equation is of limited applicability (it breaks down for KNO3, chlorates and iodates) and the values for Λ₀ deduced from it for NaCl and KCl at 18° have actually been exceeded at experimental concentrations by Kohlrausch and Maltby¹⁰ and by Wieland¹¹ respectively.

A. I. VOGEL.

The Woolwich Polytechnic, London, S.E.18.

G. H. JEFFERY.

University College, Southampton. Nov. 30.

NATURE, 130, 774, Nov. 19, 1932.
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 J. Amer. Chem. Soc., 46, 312; 1923.
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911.

5 J. Chem. Soc., 1715; 1931.

6 J. Amer. Chem. Soc., 53, 1357; 1931.

7 Ibid., 54, 2741; 1932.

8 Ibid., 54, 1405; 1932.

9 Ibid., 54, 1429; 1932.

10 Ges. Abh., 2, 886; 1911.

11 J. Amer. Chem. Soc., 40, 146; 1918.

'Raw' Weather

WITH the return of winter the following physicophysiological question once more calls for solution:-Very hot but dry weather may be tolerable, while moderately hot but very damp weather may be almost unbearable. The explanation of this in terms of cooling by evaporation is known to everyone. Again, while very cold dry conditions may be pleasant and invigorating, damp and rather cold weather what is generally described as raw weather—is very unpleasant even to normal people, while it is peculiarly bad for those suffering from rheumatism and the like. So far as I know no explanation of this last effect has been generally accepted; indeed, inquiries of many physicists and not a few physiologists have failed to produce any suggestions regarding even the general lines along which a solution might be found. Any complete explanation must also account for the fact that damp air with a moderate temperature is far from unpleasant, as exemplified by the mild southwesterly conditions often occurring in the British Isles.

G. M. B. Dobson.

Robinwood, Boar's Hill, Oxford.

HEAT expands and softens the tissues, bringing more blood and lymph into the parts: cold tightens them up. It is, I suggest, this change which leads to pain in rheumatic people. A sudden thaw is most trying.

Cold moist air has a much higher conductivity than cold dry air, and acting on the skin produces the raw feeling and excites the nerve endings which

reflexly cause the tightening-up effect.

The action of certain infra-red rays in producing reflexly congestion of the nasal air way or the opposite affords a striking example of how skin stimulation produces reflex effects on deeper organs.

LEONARD HILL.

Isotope Effect in the Spectrum of Cadmium Hydride

In an earlier paper¹, I dealt with the structure of different band systems in the spectrum of cadmium hydride. In several respects, however, my analysis suffered from incomplete resolution of the spectrum, especially regarding the isotope splitting of the band lines. Recently I repeated the investigation, using a large concave grating in the third order (0.6 A./mm.). Much work was devoted to the analysis of an extensive system ${}^{2}\Sigma^{*} \rightarrow {}^{2}\Sigma$, covering a wide range of vibrational levels (v'=0 to 13, v''=0 to 6) and thus forming a favourable case for an examination of the isotope effect. The band lines are here split into components belonging to Cd (114, 112, 110, 116), already known from the work of Aston.2 Besides this, faint components appear belonging to isotopes Cd (118, 108), which have not been reported before. These components were found in every rotational line-group of a great number of bands, falling exactly in the position calculated (± 0.02 cm. $^{-1}$). I was not able to detect the components of the odd isotopes Cd (111, 113), due to their positions between the intense lines of Cd (110, 112, 114). Only in favourable cases could a diffuse blackening be distinguished between the components of Cd (110, 112), which may be related to Cd (111).

There are some peculiarities regarding the intensity distribution among the isotope components which may be of interest to mention. In bands belonging to the transitions $v' \to 0$, the lines of Cd (114) are somewhat more intense than those of Cd (112), while there is a considerable intensity difference between the lines of Cd (110) and Cd (116), in agreement with Aston's values. In transitions $0 \rightarrow v''$, the intensity distribution is changed in favour of the heavier isotopes. Due to this, I was able to trace Cd (118) only in the latter case, where it appeared stronger than Cd (108), while in transitions $v' \to 0$ only Cd (108) could be observed. These relations can be accounted for partly3 by influences of the Boltzmann

factor $e^{-h\nu/kT}$

Detailed account of the analysis of this spectrum will be given shortly.

ERIK SVENSSON.

Laboratory of Physics, University of Stockholm. Nov. 24.

E. Svensson, Z. Phys., 59, 333; 1930.
 F. W. Aston, Phil. Mag., 49, 1191; 1925.
 G. Stenvinkel, NATURE, 126, 649; 1930.

Chlorination of Sodium Benzoate

THE chlorination of benzoic acid in alkaline solution by means of sodium hypochlorite is often quoted as an example of the effect of ionisation on orientation. Lossen's experiments1 have been repeated, and it is now found that the product of chlorination contains 48 per cent ortho-, 32 per cent meta- and 20 per cent para-chlorobenzoic acid. This is an unusual type of orientation.

J. C. SMITH.

The Dyson Perrins Laboratory, University, Oxford. Dec. 14.

¹ Friedländer, 7, 115; 1903.

Research Items

Burial of a Bari Rain-maker. The burial of a rainchief in the Bari country, Mongalla Province, Sudan, is described by Mr. A. C. Barton in Sudan Notes and Records, vol. 15, pt. 1. The rain-maker's last illness took place during a drought, and, as was ascertained by the performance of a special rite, had been caused by a more powerful rain-maker, who attributed to him the failure of his own efforts to bring rain. The grave was of a special type reserved for rain-makers, chiefs and influential freemen. It lay from east to west, and from the west, descent to the bottom of the excavation was by two steps cut in the earth. The interment chamber proper was cut laterally and frontally into the north and east sides of the excavation. The funeral ceremonies began with secret rites, to which only the close kin were admitted, in the house, which was closed immediately after death. These rites included the shaving and anointing of the corpse, the hair being placed in a calabash to be disposed of later in the bush. At the graveside, women who had not been present inside the house again anointed the body over its clothes, and the sons after anointing the back of the corpse down to the waist, again over its clothes, with ground and burnt semen returned to the house, observing a grass taboo, in walking, where they remained until earth had been thrown into the grave. A small son of a serf had been chosen as a serf of the dead, and when the corpse had been laid to rest in the chamber as on a bed, this boy remained by the corpse until it opened in final decay, when he came out of the grave and proclaimed the chief really dead. Large slabs of stone, sacking, etc., blocked the entrance of the cavity, so that no earth could fall on the chief, and above the grave were placed carved grave stakes, of which one had two prongs, a 'male' and a 'female', while the smaller is the 'sentinel'. These were brought from the father's grave and will remain on the chief's grave until required for his son's burial. A mourning feast and dance follow the burial.

The Malabar House. In commenting on a ballad of Kerala in the Indian Antiquary for November, Dr. M. D. Raghavan appends some notes on the Malabar house, which is highly conventionalised in plan and clearly has retained its main features unchanged for a very long period of time. The most characteristic dwelling-house of Kerala is called nâlu-pura, literally 'four-houses', being built round the four sides of a courtyard which is open to the sky, each room being named relatively to its situation with regard to the courtyard, that is, southern room, northern room or the three western rooms. Every house, however small, is regarded conventionally as a nâlu-pura, and hence a house of the smaller type, though facing east, is called padinnarra pura, that being the western block consisting of three rooms with a verandah in front. Usually behind the central room of the western block there is a small room called the 'lean-to'; and beside the inner verandah is an outer verandah with a long ridged roof on a beam supported by high pillars. The central of the three western rooms is the principal room of the house, containing the valuables and sacred to all household ceremonies. The main entrance is through a portico which serves the purpose of a drawing-room. Each house stands in a compound of its own, which is thickly planted with trees, and is enclosed by a massive bank of earth. A broad smooth walk, well rammed and plastered with cowdung, leads to the courtyard, of which the surface has been treated in the same way. The courtyard is used for drying paddy, pepper, etc., in the sun. It serves as a threshing floor, as the recreation-ground of younger members of the family, as an exercise ground in the use of arms, and as a place for ceremonial and social functions.

Facial Growth in Children. "Facial Growth in Children", by Corisande Smyth and Matthew Young (Medical Research Council. Special Report Series, No. 171), gives the results of a study of twenty characters of the face in 1,400 London children. The primary object of the investigation was to establish standards of normality. Measurements were made of some 1,200 selected children aged 8-14 years in the London County Council schools, 100 boys aged 9-10 years taken at random from a group of boys of the same age, and 100 children aged 2-5 years attending a welfare clinic. No measurements were made of children aged 5-8 years. This study confirms the findings of John Hunter, Tomes and Bolk that the dental arch does not increase in length after complete eruption of the milk teeth, but Smyth and Young find there is a definite increase in breadth of the dental arches after four years of age. As regards inter-relationships of the facial measurements, the most interesting result is the high association between the zygomatic breadth and breadth of the dental arches, an association denied by Korkhaus and other observers. There appears to be some tendency for a narrow face to be associated with a high palate but not necessarily with a narrow dental arch. Although all the results are based on selected children, that is, on those in whom normal occlusion has occurred. Smyth and Young consider they may be taken as fairly representative of London school children because of the close agreement found in the results for two groups of boys aged 9-10 years, one group selected because of normal occlusion, the other taken at random from the London schools.

Influence of Living-Space upon Growth. In the course of two years, experiments, fifteen in number, have been carried out by Dr. Jan Podhradsky with the object of deciding whether the size and shape of the living-space influenced in any way the growth and development of tadpoles of Rana fusca (Bull. de l'Instit. Nat. Agron. Brno, CSR., Sign. C. 20, 1931). It was found that size of living-space had a bearing on growth and development so that they followed their normal course only within certain limits of space, above and below which growth was depressed. In a small living-space the adverse factor appeared to be mutual disturbance amongst the tadpoles; in a large living-space the isolation of individuals seemed to be largely responsible. Shape of livingspace also had a measurable influence: narrow and high vessels depressed growth because they enforced and exaggerated vertical movements on the part of the tadpoles and caused greater reciprocal disturbance, as well as allowing only a low absorption of oxygen at the surface. These effects were intensified with the growth of the tadpoles. It was found also that abnormally low water depressed growth and development, and this relation was thought to be due to the unusual accumulation of excretory products and the fouling of food.

Age and Growth of Limpets. N. Abe (Science Reports, Tôhuku Imp. Univ., vol. 7, No. 3, 1932) has shown that colonies of Acmæa dorsuosa formed in spring and summer break up in autumn and winter but the individuals do not migrate more than 5-6 metres and hence are subject to the same environmental conditions every year. He has measured the individuals of such colonies and also other examples in which the shell clearly shows the annual rings. In limpets which live in wet places the growth-rate is greater, the thickening of the shell is slower and the height of the shell is relatively less than in those from a drier locality. The ratio of breadth to length of the shell is practically constant in individuals more than three years old. The weight of the shell is greater than that of the body (soft parts) in the proportion of 1:0.8. The frequency distribution of age in the colonies is asymmetrical; individuals of four years of age are most numerous. Specimens older than twelve years are few and it would appear that the maximum age of this limpet is about seventeen years.

Style-Sac of Gastropods. R. V. Seshaiya (Rec. Indian Mus., vol. 34, pt. 2, 1932) notes that a crystalline style and style-sac have been recorded in ten families of prosobranch gastropods, and he now records them in .two more families-Cerithiidæ and Assimineidæ. He refers to Randles' description (1902) of the posterior chamber of the stomach of Turritella in which the presence of a single fleshy fold and a crescentic groove is recorded, the groove being considered to be a vestigial cæcum and a very primitive feature. The author's examination of the stomachs of Turritella and of several other style-bearing gastropods does not support this view. The groove and the adjoining ridges or folds are functional structures and serve to accommodate the gastric shield and to direct the gastric contents towards the style.

Nutritive Value of Pastures. The fact that pastures which appear very similar may differ widely in nutritive value (stock-carrying and fattening capacity) has long remained without a satisfactory explanation, and an investigation of the problem has been made by E. J. Sheehy (Sci. Proc. Roy. Dub. Soc., vol. 20). So far as the chemical composition of the dry matter of the organic portion of the herbage is concerned, no consistent difference was observed between good, mediocre and poor pastures in the same neighbourhood, and although a slight inferiority in the phosphate content of the inorganic portion did occur in the poorer herbage, the disparity was not large enough to account for its reduced feeding value. Further, the chemical composition of the dry matter of fiorin grass and plantain grown on soils carrying pastures of very different value, remained unaffected. The digestibility of such different types of pasture plants as perennial rye grass, Yorkshire fog and plantain were also found to be very much alike, so that differences in nutritive value could not be explained on this score either. The factor which proved to indicate the value of the pasture, however, was the percentage of dry matter, and in consequence the density of the sward is an important feature.

The latter is largely determined by the botanical personnel, the narrow-leaved grasses yielding a higher dry weight per unit area than broad leaved plants such as weeds. Clovers occupy an intermediate position. Fattening capacity is, therefore, definitely related to the dry matter content and density of the herbage, but stock-carrying capacity is also affected by the rate of growth, as upon this property the provision of abundant feed depends.

Identification of Indian Sleeper Woods. In Forest Bulletin No. 77 (Economy Series, of the Forest Research Institute, Dehra Dun, 1932) Mr. K. A. Chowdhury, wood technologist, has produced a most useful little guide to enable forest officers, railway passing officers and others to identify on the spot some of the more common Indian sleeper woods. This bulletin is the first of its kind. Capt. Trotter in a preface says that it is hoped to publish similar bulletins from time to time dealing with timbers of individual provinces, that is to deal with timbers according to localities rather than uses. In the present case that method of treatment would not have been so useful, as the map appended to the bulletin well illustrates. This map "shows the various species of sleeper woods that grow in the areas in which different railway groups are concerned". These groups are five in number, namely, Northern, Central and Terai, Eastern, Southern The map serves two purposes—the and Burma. first and most obvious, the species which can be obtained from a given locality, and the second and the more striking, it can help in identification, as a species may easily be eliminated which does not occur in a particular locality. A few simple details on wood structure and so forth lead up to the key for the identification of more than fifty species of important Indian sleeper woods. This key is drawn up in the simplest fashion and appears to be easily workable, but as the author states, "quick and accurate identification of timbers can only be achieved after much practice". A sharp knife and a hand lens magnifying 10-12 times are all that are required to assist the key. Incidentally, the bulletin serves once again to show the importance to India as a whole of the research work being carried out at Dehra Dun in connexion with the enormously valuable Indian forest estate.

Spectroscopic Detection of Small Quantities of Elements. The Wiener Berichte IIb 141 contains a paper by W. Späth on the detection of very small quantities of elements by the spectroscope. Droplets of solutions were evaporated on silver or copper electrodes and the spectra were excited by condensed spark or by a break-contact arc, according to whether spark or arc lines are the more persistent in a particular case. The preparation of very pure silver electrodes was a long and troublesome matter, and the methods finally adopted are described in considerable detail. 10⁻¹⁰ gm. of cadmium was the smallest quantity which could be detected, using the arc line 2288 A. and the spark line 2265 A. In experiments with other metals, 10^{-10} gm. manganese, 10^{-7} gm. arsenic, 10^{-7} gm. tellurium, 10^{-9} gm. lithium and 10-11 gm. strontium could be detected. limit is apparently set by the presence of a continuous background in the spectra which masks very faint lines, and it is suggested that a spectrograph with higher dispersion combined with high light gathering power would enable yet smaller masses to be detected.

Low Temperature Carbonisation of Coal. In accordance with the policy adopted by the Government some years ago, the Department of Scientific and Industrial Research has examined the Turner plant for the low temperature carbonisation of coal installed at the Comac Oil Co. Ltd., Coalburn, Lanarkshire. The report on the test issued by the Department (H.M. Stationery Office, 9d. net) shows that the retort is of the continuous vertical type, internally heated by superheated steam. A peculiarity is the use of a fluctuating pressure claimed to facilitate transfer of heat from the steam to the coke. From 1 ton of coal were obtained 13.4 cwt. of coke, found to be a good domestic fuel, 21.3 gallons of tar and spirit, 2.170 cub. ft. of gas of total heating value 18.7 therms. and 334 gallons of liquor of no value, although as an effluent it must be regarded as a distinct liability. The alternations of pressure are regarded as a necessary feature of this process, but tests made with the plant adjusted to give steady and fluctuating steam pressures to the retort gave substantially the same results.

Dewaxing and Acid Refining Mineral Oils. At a Congress of Polish Petroleum Technologists two years ago, the De Laval S-N method of dewaxing was for the first time made public. The data then were based principally on tests with Polish oils from Schodnica and Urycz. Tests were being carried out on a small plant, but since that time a larger centri-

fugal separator has been employed successfully, and it has been possible to treat pipe-still distillates in addition to those produced by other methods. An account of this process was given by Dr. Nils Olof Backlund on December 13 at the Institution of Petroleum Technologists. Among interesting points made was the substitution of trichlorethylene as a more suitable solvent for separating the wax from the oil than the time-honoured benzene. Rate and degree of cooling of the oil-solvent mixture are of the utmost importance to the process. The advantages of the De Laval S-N trichlorethylene method compared with the benzene method include a smaller quantity of solvent used, a shorter period of cooling and the possibility of working at higher temperatures. Particulars were also given regarding acid treatment of petroleum products, and it was concluded that this process had not reached an entirely satisfactory stage. Fundamentally, acid treatment still remains 'discontinuous process', and suffers from the disadvantages thereof when compared with efficiency of distillation, dewaxing and cracking in continuous plants. The author pointed out directions in which modernisation of acid-refining is desirable and gave an account of the De Laval S-N acid sludge separator. which represents a definite advance on anything so far designed. The main importance of this process, however, would seem to be in the field of lubricating oil refining, where it is destined to effect considerable economy in production costs.

Astronomical Topics

Astronomical Notes for January. Mercury can be observed as a morning star early in the month; Venus is also a morning star, but far from the earth, and approaching superior conjunction; its disc is almost fully illuminated. Mars is approaching opposition, and is visible for most of the night; this is an aphelion opposition; the diameter on March 3 will be just under 14". Jupiter also reaches opposition in March, and will be near Mars for some months. Saturn is in conjunction with the sun in January; Uranus is still observable in the evening. Neptune is well placed for observation in the middle of Leo.

A star of mag. 5.6 is occulted by the moon on Jan. 9, disappearing at 8.28 P.M. κ Geminorum (mag. 3.6) disappears at 10.50 P.M. on Jan. 11, reappearing at 11.24, angle 221°; the moon is full on

that day.

The following are the positions of Comet Dodwell-Forbes at the beginning of Jan. 14 and 24, according to the Whipple-Cunningham orbit:

Jan. 14 R.A. 0h38m34s S.Decl. 4° 52′; Jan. 24 R.A.

1h15m4s N.Decl. 4° 43'.

Comets Faye and Geddes may also be seen with moderate telescopes; there are ephemerides in the B.A.A. Handbook for 1933.

There are minima of Algol at convenient times for observation on Jan. 8 at 7.54 P.M. and on Jan. 28 at 9.42 P.M.; a full table is given in B.A.A. Handbook.

Mass of Eros. Soon after the announcement by W. H. van den Bos and W. S. Finsen that Eros appeared like a figure-of-eight in the 26½ in. Johannesburg refractor in January 1931, Prof. W. H. Pickering derived the mass of Eros on certain assumptions as to its figure. Dr. Knut Lundmark has made a further investigation based on the same material (Lund. Obs. Circ., No. 7). The diameter of Eros was

taken as 23.4 km., and it was supposed to consist of two spheres in contact, each having a radius of 5.85 km.; the reciprocal of the mass in terms of the earth is 259,900,000. This gives a density only a quarter of that of the earth. If the distance of centres is increased to 18.6 km., the density becomes equal to that of the earth.

Comets of A.D. 868 and 1366. It was established by Dr. Hind that the second of these comets is in all probability identical with Tempel's Comet, 1866 I, associated with the Leonid meteors; Hind thought that the comet of 868 might also be identical with it. The Japanese Astronomical Herald for October, 1932, contains a re-investigation of the orbits of these two comets, by Dr. S. Kanda, based on the original observations; for the first comet he used observations made in Japan, Europe, and Korea; for the second he does not appear to have found any further observations than those used by Dr. Hind, and the new orbit is quite close to that of Hind.

T 868 March 4 1366 Oct. 18.54 U.T. 277° 164·8° ω 305 Ω 218.5 i 65 149.8 0.42 0.9749q1.00 e0.9059Period 33.35 years (assumed) Equinox 868.0 1366.0

It may be concluded that the first comet is not identical with Tempel, but that the second probably is identical. This identity had been assumed by the Computing Section of the British Astronomical Association in investigating the perturbations of the comet from 1366 until 1932. Search ephemerides are given in the B.A.A. Handbooks for 1932 and 1933.

Electricity, Gas and Other Fuels as Heating Agents

N a paper read by Mr. A. H. Barker to the Institution of Electrical Engineers on December 1, an explanation was given of the conditions under which electrical power can be used competitively for the heating of buildings. It is well known that, reckoned on the heat unit basis alone, electricity is the most expensive of all the sources of heat in common use, and that comparatively, gas is at present not very far behind it. Since the heat in a gas company's therm equals 29.4 electric units, it follows that if gas cost 6d. per therm and electricity 0.5d. per unit, the cost of the crude heat per therm delivered by the gas company would be 6d. and by the electric company 14.7d. In a few years' time we hope that more companies will be supplying at these cheap rates. If the thermal electric storage were employed the price of the electric heat would be reduced by about forty per cent. With oil at 80s. a ton and coke at 40s. a ton, the costs would be 2.25d. and 1.64d. a therm respectively. Looking merely at the costs of the 'crude' heat, the solid fuels are much cheaper.

The most fundamental difference between gas and electric heat is that with the former it is purified fuel which is transported to the spot where the heat is required, while with the latter it is energy only. There are three points arising from this. The street main is much larger and more difficult to accommodate with gas than with electricity but the housepipes for gas are smaller and cheaper. In order to convert the gas energy into heat, it has to undergo the process of combustion with the disadvantages of high temperature and waste products of combustion. The use of gas causes less actual destruction of fuel than electricity and the heat from it is therefore cheaper. Gas is in fact crude fuel which has had all the ashes and smoke purified out of it at the sacrifice of the manufacturing costs and about twenty per cent of the energy of the crude fuel. Electricity carries the purification a stage further. In its manufacture, everything, including all the labour, is purified out of the fuel but there is a loss of about twenty per cent of the energy in the mains. There are obviously cases, where so far as its usefulness is concerned, this further stage of purification is a pure waste of money, just as it is a waste of money to soften water needed for sanitary fittings.

In Mr. Barker's opinion, gas and electric supply have each their own sphere of usefulness, in which one is either economically or functionally superior to the other. There is only a smell area of supply in

which reasonable competition is legitimate. Both industries ought to be amalgamated, in their own and in the public interest. It is very difficult to assess the money value of their relative advantages and disadvantages in each particular case.

Apart from its cost, electrical energy is almost an ideal means for room warming. By its agency, pure heat can be delivered through light and flexible wires in any quantity, at any temperature and in any desired form to any particular spot. Turning on and off involves nothing but a mechanical motion of a switch and this can be readily done automatically by a simple and trustworthy form of thermostat. The use of any other fuel gives, along with the heat, products of combustion of a more or less deleterious nature. It is only in very special cases that heat derived from the combustion of fuels can be employed without the use of chimneys.

Gas possesses the advantage that it can be exactly regulated to requirements. It can be conveniently stored so that a sudden overload need not affect the supply. Breakdown is less probable than with the more complicated electrical plant. In some cases when combustion is effected completely, the products are innocuous and so can be allowed to mix with the air of the building and so secure—like the electrical fire—an efficiency of one hundred per cent. The drawbacks are that gas needs to be ignited and supplied with air before the heat can be developed. It has an objectionable smell and is dangerous if it escapes or is incompletely burnt.

The advantages of oil are that it is fluid and so can be pumped with little smell or trouble into a tank through a pipe-line. It is much easier to ignite, to regulate and to extinguish than coke, though more difficult than gas. It is very clean in operation, and when properly burnt highly efficient. Owing to the high degree of the combustibility of oil, there is a good deal of potential danger attached to it. It is apt also to give off smoke and odorous fumes. The advantages and disadvantages of coke and other solid fuels are well known. For example, they will allow any sort of combustible material to be destroyed in the furnace. It is probably the most economical form of heating but it involves greater labour than any other fuel. It produces dust, is dirty to deliver and clumsy to handle.

Mr. Barker deals exclusively with the cost of the fuel and the labour involved in handling it. When the heating required is intermittent, both electrical and gas heating have advantages over the other fuels.

Mesolithic Age in Britain*

IN 1926 the Royal Anthropological Institute held an exhibition illustrating the microlithic industries of Britain, to which all who were then known to be interested and engaged in forming collections of implements of this phase of the Stone Age were asked to contribute. The mesolithic period had been somewhat neglected by British archæologists; and it is probable that it came as a surprise, even in archæological circles, to find how considerable was the amount of material which it had been possible

*Based on certain papers read before Section H (Anthropology) of the British Association at York, September 6, 1932. to get together and the increase in the interest taken during the early years following the War in these remarkable products of man's skill and ingenuity.

In the period which has elapsed since that exhibition was held, further progress has been made in the study of the mesolithic age, and the time was fully ripe for a detailed discussion of the position of microlithic industries in relation to preceding and succeeding cultures. The opportunity for such discussion was afforded by a series of papers in the programme of Section H (Anthropology) when the

British Association met at York last summer. The conditions were exceptionally propitious, as not only were there within reach of York sites of sandy heath typical of the geographical environment preferred by Tardenoisian man, upon which, moreover, numerous finds of implements had been made, but also the fact that Mr. J. G. D. Clark was to open the session with a paper on the mesolithic age in Britain ensured that the otherwise preponderating attention to be given to the north of England in this series of communications would be placed in proper perspective.

The advances which have been made in the study of the typology of mesolithic implements now make it possible to work out the geographical distribution of characteristic forms with a relative certainty, if not with an absolute precision, while a number of pregnant suggestions emerges as to their derivation and development. Thus Mr. Clark holds that, while the Upper Palæolithic industries of Britain already showed microlithic tendencies, the Tardenois culture is to be regarded as intrusive in both its first and second phases, and late Tardenois, though in all probability a local development in its British manifestation—the 'trapezoid' implement is said to be peculiar to Britain—also has indications of continental influence. Thus in the Pennines the 'broad blade' industry, the non-geometric Early Tardenois industry, is certainly of continental derivation, while the Middle Tardenois, which extended as far as the Isle of Man, points to Belgium. At the same time, the view thus taken of late Aurignacian industries must be kept in mind in considering sites which show a sequence of cultures, such as those described by Mr. A. L. Armstrong.

The study of types leads Mr. Clark to divide Britain into two provinces, of which Province A is characterised by the absence of the tranchet axe or pick, while the microliths are present both in the early non-geometric forms and in the later geometric forms, some of them of extremely small size. In Province B, the south-east of England, on the other hand, the tranchet axe is found, but the place of the geometric forms of Tardenois industry is taken by the still surviving non-geometric

forms.

In the working out of the sequence and relation of the phases of Tardenois cultures, the investigations of Mr. F. Buckley in the Pennine chain are especially significant, although the whole chain has not been covered and his conclusions are drawn only from certain selected sites, such as Standedge Ridge, Yorks, where the whole chain narrows down to a single ridge, along which mesolithic man must have passed in avoiding the valleys. mesolithic sites under the peat have been excavated. These sites contain the relics of two distinct peoples or races of Tardenois folk. Of these one is known as the folk of the 'broad blade' industry, the other as the people of the 'narrow blade' industry. The implements of the former are predominantly of the non-geometric type and include many pointed blades, while among those of the latter are numerous small geometric tools. This people used open-air encampments and wandered freely over the hills; but the 'broad blade' folk travelled along the watershed ridge and erected huts or wigwams on their camping sites. The hearths have yielded wood remains, giving some data as to tree distribution in mesolithic times.

An important contribution to the discussion of

mesolithic problems, comprehensive in its view, was made by Mr. A. L. Armstrong's account of his investigation of the Tardenois and pre-Tardenois cultures of north Lincolnshire in the light of evidence afforded by a number of sites in that area. Here a series of stratified sections and occupied sites exhibits a continuous sequence of industries, embracing the Upper Palæolithic and the whole of the Mesolithic periods. Mr. Armstrong has recently discovered Aurignacian flint implements in glacial gravels at Hardwick Hill, east of the Trent. These are heavily rolled, owing, it is suggested, to the wave action of an estuary or glacial lake, and point to the existence here of palæolithic man before the last glaciation—possibly a band of hunters who had penetrated the swamps and taken up their residence on the dry uplands of the Cliff Range and, probably, also on the Wolds. They appear to have inhabited this region through the last phases of the glacial epoch and to have remained until the appearance of mesolithic peoples, the Azilian and Tardenois. The latter eventually dominated Lincolnshire, according to the indications of a number of stratified sites.

Of these sites the earliest is a late-Developed Aurignacian (Creswellian) station, discovered by Mrs. E. H. Rudkin, and excavated in February last, on the western escarpment of the Lincolnshire cliff above Willoughton. At Sheffield's Hill, near Scunthorpe, a similar occupation site, but of later date, gives evidence of the final phase of the developed Aurignacian, upon which early Tardenoisian was imposed. At Risby Warren, Scunthorpe, where systematic researches have been carried out for eleven years, there is stratified evidence of occupation levels ranging from developed Aurignacian, which is quite free from Tardenois influence, to the earliest neolithic, full neolithic and bronze ages. The Tardenois culture is represented by several horizons and can be classified broadly as early and late. This site, Mr. Armstrong claims, in virtue of its abundance of stratified material and its numerous occupation zones, representing the whole of the Tardenois period, is to be regarded as the type station of Tardenois culture in England.

University and Educational Intelligence

Wales.—The University Court at its meeting on December 15 decided to award the degree of D.Sc. honoris causa to Prof. Francis Ernest Lloyd, professor of botany in McGill University, and Prof. Robert Robinson, Waynflete professor of organic chemistry in the University of Oxford.

The Institute of Sociology, Le Play House, 65 Belgrave Road, London, S.W.1, has now been incorporated as a company limited by guarantee. This is the final step in a scheme for permanently establishing and endowing the Institute, which was initiated in 1920 through the generosity of the late Mr. and Mrs. Branford. Under an order of the Chancery Court, the whole of Mr. Victor Branford's estate, subject to the temporary reservation of a portion for the benefit of relatives, passes to the Institute. The present officers of the Institute are Dr. R. R. Marett (president), Rector of Exeter College, Oxford; Mr. C. H. Rigg (honorary treasurer), and Mr. A. J. Waldegrave (chairman of council).

Calendar of Nature Topics

January freeze-the-pot-by-the-fire

January, the coldest month of the year in England, is proverbially associated with snow and ice. In the French Revolutionary Calendar of 1793, the period from December 21 to January 19 was the month 'Nivôse' or 'snowy'. During the twentieth century, however, January has not lived up to its name, but has been much more open, mild and stormy, with few prolonged frosts. At Greenwich the mean January temperature during the decade 1921–30 was 41·3° F., more than 2° F. above the long-period normal and probably 5° or 6° higher than in some of the decades of the late eighteenth and early nineteenth centuries. This persistent period of mild winters has extended over the whole of western and central Europe; it is associated with a greatly increased frequency of south-westerly winds and may almost be regarded as a change of climate.

Life in the Southern Atlantic Ocean

"January 1-Wilson's petrels, two kinds of prion petrels, albatrosses and a bosun bird were observed to-day. Sun-fish (Orthagoriscus mola) were also seen; at 2.30 p.m. a boat was lowered to get one. After several ineffective attempts at harpooning we managed to stun one with shot. Davidson after a while drove the harpoon into the gills, and we then made fast to one of its fins and towed it to the ship. Three or four sucker-fish accompanied it to the boat, and on cutting the sun-fish up, one was discovered in the gills. A parasitic copepod (Argulus) was found externally, as well as a polyclad. There was also a parasitic copepod on the gills and a barnacle on the lip. In the intestines numerous tape-worms were found and another leech-like parasite. There was in all probability about 10 lbs. weight of tape-worms in the gut. Its weight by the dynamometer was half a ton. Several others were seen twice as large. Brown saw a globe-fish with a small fish attendant on it. Salpa were seen frequently floating past." From "Zoological Log of S.Y. Scotia", on January 1, 1903, in lat. 39° 01′ S.; long. 53° 40′ W.

The lethargy of the sun-fish has often been referred to. The individual mentioned above, now exhibited in the Royal Scottish Museum, Edinburgh, was regarded by its captors as a tiny specimen compared with others seen the same day "about the size of a small haystack", and of it they said, "Its stupidity was amazing; unable to swim faster than a boat could row, all it had to do to escape was to sink and this they can do quite well, but although struck by a harpoon a dozen times before one held, it made no attempt to escape. . . . When cut up it was easily seen why the first harpoons would not hold, as under the skin was a layer two or three inches thick of a hard cartilaginous material. The dissection was performed mainly with axes. central nervous system was very interesting on account of its minute size relative to the body, the spinal cord being only about half an inch long and barely coming outside the cranial cavity. degeneracy is doubtless correlated with the feeble musculature and swimming powers. Intelligence and mobility have become superfluous, the size and thick hide being sufficient protection against most enemies." From "The Voyage of the *Scotia*", by R. N. Rudmose Brown, R. C. Mossman, and J. H. Harvey Pirie.

Hurricanes of the Southern Hemisphere

January 7.—The greatest frequency of hurricanes or revolving storms in the southern hemisphere occurs about the second week in January, though they may be experienced at any time during the southern summer. Four areas are subject to these storms, the islands of Polynesia, especially the Samoa, Fiji and Tonga groups, the Coral Sea between Queensland and the Fiji Islands, the north-west coast of Australia and the South Indian Ocean between the Chagos Archipelago and Madagascar. The Australian storms are locally known as 'Willy-willies', and in addition to causing loss of shipping, frequently result in disastrous floods. The majority of the hurricanes in the southern hemisphere originate between lat. 5° and 15° S., travel at first towards west-south-west, recurve in about 20° S. and finally pass away to the south-east, decreasing rapidly in intensity.

"Furze or Gorse (Ulex europæus) flowering"

This record made by Gilbert White at Selborne on January 8 is a reminder of the winter activity of gorse which makes it an important item in the economy of the countryside. The closely cropped conical bushes, which in many places dot a pasture like old-fashioned bee-hives, are evidence of the part it plays in the winter food-supply of rabbits and sheep. Before roots were commonly available for the winter feeding of farm stock, gorse partly took their place: "The sowing of whins for feeding of cattle takes mightily about London just now [1725]. this improvement comes from Wales, where it has been practised these hundred years." In Scotland rough whins from waste ground were used in place of the more tender sown crop, and these had to be broken and pulped before being served to the cattle. The apparatus used was a cumbersome whin-mill, the essential part of which consisted of a huge stone roller dragged by an ox or a horse round a paved circuit, upon which the whins were laid. Remains of such whin-mills are still to be seen at occasional farmhouses in Scotland, although whins have long since dropped out of use. In the south of Ireland, however, on hillside farms where hay is scarce, whins are still used for feeding horses during the winter months. The seed is sown on spare ground, and the crop, cut in the following year, is passed through an ordinary chaff-cutter before being served.

The Hibernation of Frogs

In early January 1932, Francis B. Bent (as recorded in the Observer, May 1, 1932) emptied a pond and cleaned it out. The pond was ten to fifteen feet deep and the normal depth of water was four feet, but in the clay ooze at the bottom there were discovered thirty or forty frogs. In the same month H. C. Davies (Field, May 8, 1932) ran the water out of a small spring-fed pond in order to dig the mud out. In so doing he dug out, not an isolated frog, but dozens. "Their appearance was not attractive—discoloured and skinny—but all alive, though showing very little energy."

Although the older naturalists believed in the hibernation of frogs in winter in the mud of ponds, doubt has recently been thrown upon the possibility

of survival of air-breathing, lung-possessed creatures in such circumstances. However, in the height of the mating season, when activity is not at its lowest, one has seen beneath the clear water of a pond mated frogs remain for long periods inactive on the bottom, and in face of such experiences as are related above it seems probable that in cold weather the lowering of activity in these cold-blooded animals reduces metabolism to so low a pitch that the oxygen required can be obtained by transpiration through the skin. The problem still to be solved in regard to cold-blooded hibernators, as P. A. Gorer has pointed out, lies in the physiological changes which enable the tissues of an adapted animal such as the common frog (Rana temporaria) to recover from cooling which is not excessive, while such recovery is impossible in unadapted animals.

Societies and Academies

LONDON

Royal Meteorological Society, Dec. 14. C. S. Durst: "The thermal balance of a water drop or ice particle suspended in the atmosphere." From the examination of the long wave radiation received and given out by a water drop or ice particle, it is shown that such a particle will lose heat if it is above a certain critical temperature and gain heat if it is below, from which it follows that if a particle exists in the stratosphere it will gain heat. It is assumed that the base of the stratosphere is saturated and consideration is given to the conditions under which particles could be formed. If a small air mass were raised in the stratosphere the particles formed in it would be melted in a very short time and the temperature of the air would once more be that of its surroundings, the entropy of the air having been increased in the process.—E. W. Bliss: The tabulation of world weather (5). (Discussion by Sir Gilbert Walker.) (Mem. Roy. Meteor. Soc., 4, No. 36.) In order to form more definite ideas regarding the oscillations named the North Atlantic, the North Pacific, and the Southern, series of figures have been derived to express the variations of each, and from these have been obtained their relations with pressure, temperature, and rainfall over wide regions as well as the relations of the three oscillations with each other and with sunspots. The southern oscillation in the southern winter is extremely persistent, and its departure has a correlation coefficient of 0.84 with that of the following summer.—C. S. DURST: "The breakdown of steep wind gradients in inversions." On certain occasions when inversions have formed, a violent eddying arises, which is shown on an anemometer as an abrupt change in the type of trace. This change over occurs when the wind gradient becomes great. On the ground that the eddies formed in these circumstances are different in character from those formed with an adiabatic temperature gradient, a suggestion is put forward for the mechanism of the diurnal variation of wind.

DUBLIN

Royal Dublin Society, Nov. 22. J. H. J. POOLE: An investigation of the behaviour of neon discharge tubes in a flashing capacity circuit by means of a cathode ray oscillograph. The effects of leakage currents in the oscillograph were eliminated by using the oscillograph

in conjunction with a valve anode resistance amplifier. For small shunting capacities the flashing may be extremely irregular, and quite considerable currents pass through the tube before the flash occurs. The presence of radium lowers the striking potential and, by decreasing the maximum dark current, increases the regularity of flashing. The effect of the shape of the electrodes has also been investigated. For concentric cylindrical electrodes, at the filling pressures used, the difference between the striking and extinction voltages is less when the inner cylinder is negative. H. M. FITZPATRICK: The trees of Ireland, native and introduced. A catalogue of the tree species growing in Ireland giving the dimensions attained by each in different parts of the country with, in the case of exotics, an account of their natural distribution and introduction into cultivation. 150 broad-leaved and 215 coniferous trees are recorded. G. T. PYNE and J. J. RYAN: The colloidal calcium phosphate of milk. Some samples of milk out of a large number tested developed a marked alkalinity to phenolphthalein on addition of oxalate. As the wheys prepared from the same milks did not do so, it appeared that the alkalinity must arise from the interaction of the (potassium) oxalate with the casein calcium phosphate complex of milk, presumably owing to the conversion of insoluble tri-calcium phosphate into the strongly alkaline tri-basic potassium salt. The amounts of tricalcic phosphate required to account for the observed alkalinities approximated to those usually accepted for the entire colloidal phosphate of milk, suggesting that the greater part of this colloid must consist of the tribasic salt. The bulk of the casein calcium phosphate complex was removed by prolonged high speed centrifuging from two very different types of milk (as regards their reaction with oxalate) and submitted to analysis. The results appeared to show that the bulk of the colloid in each case consisted of tricalcium phosphate, and that the variations in the behaviour of different samples to oxalate was thus connected with the relative quantity of the colloid present rather than with variations in its composition.

PARIS

Academy of Sciences: Nov. 14. CH. MAURAIN and J. DEVAUX: Electrical conductivity and atmospheric condensation nuclei during a voyage to Greenland. There is a general resemblance between the electrical conductivity of the air in the polar regions and that on mountains at high altitude, possibly due to the purity of the atmosphere and the dryness. measurements were too few to enable any deductions to be drawn as to the effects of the meteorological conditions. MARIN MOLLIARD: Aseptic tuberisation and morphological characters resulting from the action of saccharine food on the onion, Allium cepa. E. MATHIAS: Death by the return stroke (lightning). J. CANTACUZENE and A. TCHEKIRIAN: The presence of vanadium in certain tunicates. Vanadium has been found in nine species of tunicates: the proportion is higher in young animals than in adults. Potron: The Riemann spaces admitting a group of isometric transformations with n(n+1)/2 parameters. Marcel BRELOT: The study of the point singularities of subharmonic functions. Pierre Humbert: Besselintegral functions. D. Pompeiu: A theorem, analogous with that of Rouché, relative to the zeros of holomorph functions. NICOLAS APRAXINE: A calculating machine worked electrically. B. GALERKIN:

The general solution of the problem of elastic equilibrium of a hollow circular cylinder and of a part of the cylinder. R. DE FLEURY, H. PORTIER and S. Benmakrouha: The comparative and reciprocal influences of individual values, for each alloy and for each state of the alloy, of the modulus of elasticity, the elastic limit and the density on the dimensions of beams and elements of framework under bending. Louis de Broglie: The electromagnetic field of the light wave. L. GOLDSTEIN: The quantum theory of the diffusion of electrons. Andre Marcelin and D. G. Dervichian: The direct measurement of superficial pressure of superficial solutions formed by soluble substances. P. ROUARD: The reflecting power of metals in very thin plates. The reflecting power varies with the thickness of the film, passing through a minimum depending on the wave-length of the light. CH. BEDEL: The temperature coefficient of the electrical resistance of silicon and a thermoelectric phenomenon of unipolar substances. A. Sanfourche and B. Focet: The calcium salt of a complex ferrophosphoric acid. F. Gallais: Potassium iodomercurate. Leon Jacque: The alteration of steels by hydrogen. The modification of the micrographic structure of steel submitted to the action of hydrogen under pressure and at temperatures between 400° and 700° C., appears to be the result of two simultaneous phenomena: the elimination of the carbon of the steel by the hydrogen, and a diffusion of the carbon from the unaltered regions towards the decarbonised regions. The process is a reversed cementation. The change depends on the composition of the steel, chrome-nickel steels, for example, being less altered under the same conditions. M. Lesbre: An imperfect silver-guanidine complex. R. Levaillant: The action of acid chlorides on orthoformic The preparation of symmetrical esters esters. of sulphuric acid. The reaction CCl₃.COCl + $\mathrm{HC}(\mathrm{OC_2H_5})_3 = \mathrm{CCl_3.CO_2C_2H_5} + \mathrm{C_2H_5Cl} + \mathrm{HCO_2C_2H_5}$ gives a 90 per cent yield of the ethyl trichloracetate. The reaction is a general one, and can be extended to the sulphonchlorides. G. DARZENS: A new method of glycidic synthesis of aldehydes. LESPIEAU and WIEMANN: The synthesis of allodulcite. Y. MILON: The antiquity of the depression of the gulf of Morbihan. G. DUBAR and H. TERMIER: The facies of the Lias underlying the Toarcian in the Moroccan middle Atlas. J. GAUZIT: An attempt at the estimation of atmospheric ozone by visual photometry. The method detailed, which has the advantage over other methods in use of taking less time, measures the thickness of the ozone layer in the atmosphere with an uncertainty of less than ten per cent. Ph. Joyet-Lavergne: Oxidising power, chondriome and cytoplasmic sexualisation in the Fungi. J. BEAU-VERIE and MLLE. S. MONCHAL: The life of green plants in a confined atmosphere. Plants contained in glass vessels, with moist earth, and sealed with paraffin wax, can live and grow for several years. Examples are given of plants which have remained in good condition for three or four years thus sealed up. MICHEL GRACANIN: Contribution to the study of the relation between transpiration and the resorption of ions. Marc Simonet: New interspecific hybrids in Iris and their cytological study. Armand Dehorne: New observations on the asexual multiplication of an annelid of the genus Dodecaceria. Etienne Rabaud and L. Verrier: The evacuation of the gases from the air bladder and the working of the pneumatic canal. Paul Wintrebert: The two phases of segmentation and the subdivision theory in amphibians.

Forthcoming Events

Monday, Jan. 9

ROYAL GEOGRAPHICAL SOCIETY, at 8.30.—B. Roberts: "The Cambridge Expedition to Vatna-Jökull, 1932".

Tuesday, Jan. 10

ROYAL ANTHROPOLOGICAL INSTITUTE, at 8.30 .- Dr. N. Gordon Munro: "The Ainu Bear Ceremony" (Film).

Thursday, Jan. 12

King's College, London, at 5.—Prof. R. J. S. McDowall: "The Integration of the Circulation" (succeeding lectures on Jan. 19, 26 and Feb. 2).

Official Publications Received

GREAT BRITAIN AND IRELAND

Great Britain and Ireland

The H.E.A. Year Book: the Annual Publication of the Horticultural Education Association. Vol. 1, 1932. Pp. 92+xl. (Wye: South-Eastern Agricultural College.) 3s. 6d.

County Borough of Southport: Meteorological Department: The Fernley Observatory, Southport: Report, and Results of Observations for the Year 1931. By Joseph Baxendell. Pp. 31. (Southport.)

Brompton Hospital Reports: A Collection of Papers recently published from the Hospital. Vol. 1, 1932. Pp. iv+144. (London.) 2s. 6d.

Proceedings of the Royal Irish Academy. Vol. 41, Section B, No. 8: Recent Views bearing on the Problem of the Irish Flora and Fauna. By Dr. R. Lloyd Praeger. Pp. 125-145. (Dublin: Hodges, Figgis and Co.; London: Williams and Norgate, Ltd.) 1s.

True Temperance Scientific Committee. Monograph No. 8: In Chase of Truth of Alcohol. By Prof. Henry E. Armstrong. Pp. 32. (London: Donington House.) 1s.

The Scientific Proceedings of the Royal Dublin Society. Vol. 20 (N.S.), No. 30: A Suggested Mode of Radiotherapy when Longcontinued Feeble Gamma Radiation may be Desirable. By Dr. J. Joly. P. 469. (Dublin: Hodges, Figgis and Co.; London; Williams and Norgate, Ltd.) 6d.

OTHER COUNTRIES

Other Countries

International Hydrographic Bureau. Report of the Proceedings of the Third International Hydrographic Conference held at Monaco, 12–23 April, 1932. Pp. 435. (Monte Carlo.) 2.50 dollars. Colony of Mauritius. Annual Report on the Royal Alfred Observatory for the Year 1930. Pp. 4. (Mauritius.) Miscellaneous Publications of the Royal Alfred Observatory for the Year 1930. Pp. 4. (Mauritius.) By R. A. Watson and N. R. McCurdy. Pp. 17+3 plates. No. 12: The Cyclone Season 1929–1930 at Mauritius. By R. A. Watson and N. R. McCurdy. Pp. 17+3 plates. No. 12: The Cyclone Season 1929–1930 at Mauritius. By R. A. Watson and N. R. McCurdy. Pp. 3+43 plates. (Mauritius.) Indian Central Cotton Committee: Technological Laboratory. Technological Bulletin, Series A, No. 22: Technological Report on Banilla Cotton, 1930–32. By Dr. Naxir Ahmad. Pp. 1i+17. (Bombay: Times of India Press.) 8 annas.

The Quarterly Journal of the Geological, Mining and Metallurgical Society of India. Vol. 4, No. 1, August. Pp. 27. (Calcutta.) 6 rupees. Paleontologiese Navorsing van die Nasionale Museum. Deel 2, Stuk 5: Voorlopige Besbrywing van Vrystaatse Soogdiere. By Dr. Ir. E. C. N. Van Hoepen. Pp. 63–66. (Bloemfontein.) Publications of the Observatory of the University of Michigan. Vol. 5, No. 2: The Orbit of Comet Peltier-Whipple, Second Paper. By Allan D. Maxwell. Pp. 4. (Ann Arbor, Mich.)

U.S. Department of the Interior: Geological Survey. Professional Paper 166: Physiography and Quaternary Geology of the San Juan Mountains, Colorado. By Wallace W. Atwood and Kirtley F. Mather. Pp. vi+176+34 plates. Professional Paper 167: Lower Triasic Ammonoids of North America. By James Perrin Smith. Pp. v+194+81 plates. 70 cents. Professional Paper 171: Geology and Ore Deposits of the Pioche District, Nevada. By Lewis G. Westgate and Adolph Knopf. Pp. vii+77+8 plates. 85 cents. Professional Paper 173: Geology and Ore Deposits of the Stockton and Fairfield Quadrangles, Utah. By James Gelilluly. Pp. vi+171+32 plates. (Washington, D.C.: Government Pr

CATALOGUES, ETC.

The Protexray Tube. Pp. 32. (London: Cuthbert Andrews.)
F/2 Spectrograph. (Lispec 33.) Pp. 4. Vacuum Thermocouples
for Measuring Weak Alternating Currents. (Brug 31.) Pp. 2. (Delft:
P. J. Kipp en Zonen.)
Calendar for 1933. (London: The Chemical Trade Journal.)
John G. Stein and Co., Ltd. Diary for 1933. Pp. 64+Diary.
(Bonnybridge.)

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